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Isolated and mixed effects of diuron and its metabolites on biotransformation enzymes and oxidative stress response of Nile tilapia (*Oreochromis niloticus*)



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ABSTRACT

Diuron is one of the most used herbicide in the world, and its field application has been particularly increased in Brazil due to the expansion of sugarcane crops. Diuron has often been detected in freshwater ecosystems and it can be biodegraded into three main metabolites in the environment, the 3,4-dichloroaniline (DCA), 3,4-dichlorophenylurea (DCPU) and 3,4-dichlorophenyl-N-methylurea (DCPMU). Negative effects under aquatic biota are still not well established for diuron, especially when considering its presence in mixture with its different metabolites. In this study, we evaluated the effects of diuron alone or in combination with its metabolites, DCPMU, DCPU and 3,4-DCA on biochemical stress responses and biotransformation activity of the fish Oreochromis niloticus. Results showed that diuron and its metabolites caused significant but dispersed alterations in oxidative stress markers and biotransformation enzymes, except for ethoxyresorufin-O-deethylase (EROD) activity, that presented a dose-dependent increase after exposure to either diuron or its metabolites. Glutathione S-transferase (GST) activity was significant lower in gills after exposure to diuron metabolites, but not diuron. Diuron, DCPMU and DCA also decreased the multixenobiotic resistance (MXR) activity. Lipid peroxidation levels were increased in gill after exposure to all compounds, indicating that the original compound and diuron metabolites can induce oxidative stress in fish. The integration of all biochemical responses by the Integrated Biomarker Response (IBR) model indicated that all compounds caused significant alterations in O. niloticus, but DCPMU caused the higher alterations in both liver and gill. Our findings imply that diuron and its metabolites may impair the physiological response related to biotransformation and antioxidant activity in fish at field concentrations. Such alterations could interfere with the ability of aquatic animals to adapt to environments contaminated by agriculture.

1. Introduction

Brazil is the world's largest sugarcane producer in the world, covering this crop an area of about 10 million hectares, being the third largest area of cultivation in the country after soy and corn. In the years 2015/2016, sugarcane production reached 700,000,000 t (UNICA, 2016). However, the large-scale production means the overuse of pesticides for crops maintenance, which has generated concerns related to its environmental impacts on different ecosystems. Diuron (3-(3,4-dichlorophenyl)-1,1-dimethylurea) is one of the most used herbicide on

sugarcane crops in Brazil and it is a frequently detected pesticides in freshwater ecosystems around the world (Morin et al., 2009; Schlenk et al., 2012). Diuron is a substituted phenylurea compound with a relatively low $K_{\rm OC}$ (418–560, according to ARSUSDA, 2004) but with long hydrolysis and aqueous photolysis half-lives, which indicates a relatively low tendency to sorb to soils and sediments, making it available to the water fraction. This herbicide has moderate to high persistence in soils, with and average field dissipation half-life of 90 days, although it can be highly variable according to soil and abiotic characteristics. According to Kidd and James (1991), diuron residues may persist for

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more than one year at higher application rates. Due to its low tendency to sorb to soils and its moderate persistence, diuron is therefore prone to off-site movement in surface runoff, and migration to both surface and ground water (Troiano et al., 2001; Field et al., 2003; Giacomazzi and Cochet, 2004), posing risks to aquatic organisms.

According to World Health Organization (Abass et al., 2007), diuron is classified as a slightly hazardous pesticide (class III toxicity), and considered as moderately toxic to aquatic life (ECA, 2017). In soil, diuron can be metabolized by fungi and bacteria originating three main metabolites: 3,4-dichloroaniline (DCA), 3,4-dichlorophenylurea (DCPU) and 3,4-dichlorophenyl-N-methylurea (DCPMU) (Tixier et al., 2002; Abass et al., 2007). Spontaneous hydrolysis can also occurs in aquatic environments, generating 3,4-DCA (Salvestrini et al., 2002), the main product of diuron biodegradation and the most persistent metabolite in the environment (Tixier et al., 2000). 3,4-DCA is mentioned for causing several negative effects on aquatic organisms, such as alterations on morphological (Scheil et al., 2009; Mhadhbi and Beiras, 2012), biochemical (Sánchez-Muros et al., 2013), physiological (Miranda et al., 2008; Scheil et al., 2009; Freitas et al., 2016) and behavioral (Saglio and Trijasse, 1998) parameters. Studies have also shown that diuron metabolites (especially DCPMU and DCPU) have anti-androgenic and estrogenic effects in male and female Nile tilapia (Pereira et al., 2015, 2016; Felício et al., 2016). Although clear evidences of negative effects of diuron and its metabolites on endocrine system of fish, there are no studies regarding the effects of these compounds in biochemical parameters often used as classical pollutant biomarkers, such as the biotransformation enzymes and oxidative stress parameters, which are also important to clarify physiological aspects involved in the defense response of aquatic organisms against environmental pollutants.

The evaluation of biochemical alterations in environmental monitoring studies is a common practice to indicate the exposure of aquatic organisms to pollutants and it can be used as an early stage alert for detection of environmental contamination. Biotransformation enzymes are considered relevant biochemical parameters evaluating chemical disturbance on animal health, since they contribute to the detoxification process, minimizing the deleterious effects of the xenobiotics in the organisms. Cytochrome P450 isoforms, glutathione S-transferase (GST) and multixenobiotic resistance (MXR) efflux proteins are involved in the biotransformation and elimination of xenobiotics from cells and they usually act as one of the first cellular response to xenobiotic input (Van der Oost et al., 2003; Luckenbach et al., 2004; Klobučar et al., 2010). Intoxication can also increase oxygen consumption, increasing the rate of the generation of reactive oxygen species (ROS) that can lead to oxidative lesions to biomolecules and oxidative stress. In response to increased ROS production, cells alters their antioxidant defenses, which includes the enzymes superoxide dismutase (SOD), glutathione peroxidase (GPx), catalase (CAT) and glutathione reductase (GR), as well as glucose-6-phosphate dehydrogenase (G6PDH), which provides NADPH for the GR-catalyzed regeneration of glutathione from glutathione disulfide and for cytochrome P450-catalyzed biontransformation reactions (Stegeman et al., 1992; Lopez-Torres et al., 1993). If the generation of ROS surpasses the antioxidant capacity, an oxidative stress condition takes place, leading to the oxidation of key cellular macromolecules such as lipids, proteins, nucleic acids and carbohydrates (Van der Oost et al., 2003). One common biomarker of oxidative lesion to lipids is malondialdehyde (MDA), an aldehyde that is a by-product derived from the decomposition of lipid hydroperoxides formed by the oxidation of polyunsaturated fatty acids (Van der Oost et al., 2003; Almeida et al., 2005, 2007). MDA is a highly reactive molecule (Trivic and Leskuvac, 1994) that can react with other macromolecules (Bartels, 2001), including nucleic acids, generating mutagenic DNA adducts (Van der Oost et al., 2003; Almeida et al., 2005, 2007), and therefore must be eliminated from cells. The enzyme aldehyde dehydrogenase (ALDH) can act metabolizing MDA and other lipid peroxidation-derived aldehydes, assisting in cellular detoxification process (Nakazono et al., 2000; Kirch et al., 2001, 2004).

Considering the importance of the biotransformation enzymes and oxidative stress parameters as indicators of health status of aquatic animals in natural systems, we are interested in investigating how diuron and its metabolites can impair physiological mechanisms of fish at environmental relevant concentrations. As mentioned before, diuron has relatively high persistence in the environment, which is likely to occur in combination with other degradation products, such as its metabolites DCPMU, DCPU and 3,4-DCA. Thus, this study evaluated the effects of diuron and its metabolites alone, or in mixture, on the activities of enzymes related to biotransformation process (ethoxyresorufin-O-deethylase, EROD, benzyloxyresorufin-O-deethylase, BROD, and pentoxyresorufin-O-deethylase, PROD, GST and MXR) and oxidative stress response (activities of SOD, CAT, GPx, GR, G6PDH, ALDH and MDA levels) in gills and livers of Nile tilapia. Biological responses triggered by chemical exposure were assessed based on individual biomarkers and on an integrative index, the integrated biomarker response (IBR, Beliaeff and Burgeot, 2002). We hypothesize that diuron and its metabolites may increase detoxification activity of biotransformation enzymes and Mxr levels as consequence of intoxication process caused by chemical exposure. We also suggest that antioxidant response and lipid peroxidation will be associated to biotransformation performance, since ROS production can be rised by intoxication process, which would stimulate antioxidant enzymes and on the opposite way, increase lipid peroxidation as consequence of damage of cellular components. The information brought by this study is relevant to better clarify how aquatic animals are dealing with pesticides presence in their environments.

2. Material and methods

2.1. Chemical

All chemicals were ordered from Sidma-Aldrich Chemical Co (ST. Louis, Mo), including Diuron, DCPMU, DCPU and DCA (> 98% pure).

2.2. Test organisms

Male Nile tilapias (*Oreochromis niloticus*) were obtained from the Aquiculture Center of the Sao Paulo State University - UNESP. The weight and length of tilapias used in this study were, 77.81 ± 12.98 g and 13.68 ± 0.83 cm (mean \pm standard deviation), respectively. Before exposures, the animals were acclimatized and maintained in the laboratory under ideal conditions of temperature, pH and oxygen (28 °C, pH 7.5–8.0) during 7 days, and fed to satiation with commercial fish food (commercial pellets for tropical fish, 32% crude protein - Guabi-Pira/Brazil). Our experiments had permission from Ethics Committee from Animal Use in research of the Sao Paulo State University (CEUA-IBILCE/UNESP) (71/2013).

2.3. Exposure experiments

After acclimatization, 66 fish were separated in eleven groups, with six fish per group (N=6). The animals were maintained in 17 L individual aquariums (one fish per aquarium) with dechlorinated water and controlled temperature (25 \pm 1 °C), constantly aerated and kept in a 12:12 h light-dark cycle. For isolated exposures, the animals were exposed to each individual compound – diuron, DCPMU, DCU or 3,4-DCA - at two different concentrations of 40 and 200 ng L $^{-1}$, totalizing eight individual exposures. For the mixtures, fish were exposed to a mix containing diuron and its three metabolites (diuron + DCPMU + DCPU + DCA) at concentrations of 10 ng L $^{-1}$ and 50 ng L $^{-1}$ each, and totalizing two combined exposures. The concentrations were chosen based on mean diuron values found in contaminated aquatic environments (up to 160 ng/L) (Köck-Schulmeyera et al., 2013; Masiá et al., 2015), and based on previous studies done by our research group on the effects

Table 1
Concentration of diuron, DCPMU, DCPU and 3,4-DCA measured in water of the treatments.

Treatment / Group	Expected concentration (ng L^{-1})	Measured concentration (ng L^{-1})	pН	Dissolved [NH ₃]
Control	0	< LD	7.9 ± 0.06	0.19 ± 0.07
Diuron 40 ng L ⁻¹	40	57.35 ± 0.48	7.9 ± 0.04	0.22 ± 0.07
Diuron 200 ng L ⁻¹	200	284.01 ± 0.34	7.9 ± 0.02	0.26 ± 0.06
DCPMU 40 ng L ⁻¹	40	16.53 ± 3.37	7.9 ± 0.10	0.31 ± 0.09
DCPMU 200 ng L ⁻¹	200	88.50 ± 1.37	7.8 ± 0.25	0.22 ± 0.11
DCPU 40 ng L ⁻¹	40	23.85 ± 0.21	7.8 ± 0.10	0.31 ± 0.15
DCPU 200 ng L ⁻¹	200	231.04 ± 0.72	7.7 ± 0.13	0.22 ± 0.05
DCA 40 ng L^{-1}	40	47.21 ± 0.47	7.7 ± 0.11	0.25 ± 0.08
DCA 200 ng L ⁻¹	200	131.23 ± 0.20	7.8 ± 0.14	0.35 ± 0.16
Mixture of 10 ng L ⁻¹	10 (diuron)	10.91 ± 0.48	7.8 ± 0.05	0.59 ± 0.28
_	10 (DCPMU)	47.15 ± 3.13		
	10 (DCPU)	5.94 ± 2.76		
	10 (DCA)	23.75 ± 0.25		
Mixture of 50 ng L ⁻¹	50 (diuron)	17.69 ± 0.60	7.8 ± 0.08	0.28 ± 0.13
· ·	50 (DCPMU)	49.58 ± 1.20		
	50 (DCPU)	82.80 ± 0.64		
	50 (DCA)	29.23 ± 1.36		

LD = Limit of detection.

of diuron and metabolites in the levels of sexual hormones and gametogenesis (Felício et al., 2016; Pereira et al., 2015, 2016; Boscolo et al., 2017). Control group was maintained under the same conditions as the treatments; however no chemicals were added into the aquariums. Each compound was diluted (50 µL) in the aquariums at the respective concentration from a stock solution previously weighted and dissolved in acetone to give the desired concentration in a volume of 50 µL of the stock solution. The corresponding volume of solvent was also added in the control group. The water and chemicals were totally changed every 2 days, by replacing the animals to another aquarium with the same treatments. The same procedure of water change was done with the control group. Animals were fed with ration for tropical fish (Guabi-Pira/Brazil) corresponding to 3% of biomass at each water change. After seven days of exposure, all fish were collected, anesthetized by immersion in a benzocaine solution (90 mg L⁻¹) and their livers and gills were removed and stored in -80 °C. Water pH and NH₃ levels (Table 1) were monitored throughout the experiment. In the first day, water samples (10 mL) were collected from each aquarium just before placing the fish into the aquariums, in order to check chemical concentrations.

2.4. Chemical analyses

Concentrations of diuron, DCPMU, DCPU and 3,4-DCA in water from the different experimental aquariums were checked by HPLC in the first day, just before adding the fish into the aquariums. The HPLC system (Shimadzu Corporation, Kyoto, Japan) consisted of one CBM20A communication bus module, two LC20AD-XR pumps, one CTO20AR column oven, and one SPDM20A photodiodearray (PDA) detector. The column used was a Shimadzu Shim-Pack XR-ODS (2.0 imes100.0 mm, 2.2 µm particle size, 8 nm pore size). PDA detector wavelength was adjusted between 200 and 600 nm for all analyses, and the compounds were quantified at 250 nm. The mobile phase consisted of acetonitrile and water (40:60, v/v), and was isocratically pumped at a flow rate of 0.5 mL $\rm L^{-1}$. Column oven temperature was set to 40 °C. The water sample (50 µL of water) was injected into the HPLC through an autosampler (Shimadzu, Nexera XR, SIL-20AC XR) and monitored during 5 min. Retention times for diuron, DCPMU, DCPU and DCA were, respectively 2.9, 2.2, 1.5 and 3.6 min. Chromatogram peaks were identified and quantified using LAB Solution 5.71 software (Shimadzu Corporation). The calculation of concentration of each compound was based on a calibration curve, previously constructed by injecting authentic standards into the HPLC system $(10 - 1000 \text{ ng L}^{-1})$. The limit of detection for all the compounds was 10 ng L^{-1} .

2.5. Biochemical analyses

2.5.1. Enzymatic assay and protein quantification

Liver and gill were homogenized (1:4, weight/volume) in Tris buffer 0.05 M (pH 7.4) containing sucrose 0.005 M, KCl 0.015 M and protease inhibitor (phenylmethanesulfonylfluoride - PMSF) 0.001 M. The homogenized samples were centrifuged at 10,000g for 20 min at 4 °C, and the supernatant fraction was collected and re-centrifuged at 50,000g for additional 60 min at 4 °C. The second supernatant fraction was used for SOD, CAT, GPx, GST, G6PDH and GR assays, while the pellet, suspended in 100 μ L of Tris buffer 0.1 M (pH 7.5), containing EDTA 0.001 M, dithiothreitol (DTT) 0.001 M, KCl 0.1 M and glycerol 20%, was used for EROD, BROD and PROD assays.

SOD activity was evaluated by the inhibition of cytochrome C reduction in the presence of hypoxanthine/xanthine oxidase O2. generator system at 550 nm (McCord and Fridovich, 1969). CAT activity was quantified at 240 nm by H₂O₂ decomposition according Beutler (1975). GPx activity was assayed using the oxidation of NADPH (linked to GSSG reduction by excess glutathione reductase) at 340 nm, and using t-butyl hydroperoxide as substrate, as described by Sies et al. (1979). GST activity was quantified at 340 nm using reduced glutathione (GSH) and 1-chloro-2,4-dinitrobenzene (CDNB) as substrate, according Keen et al. (1976). GR activity was measured using the consumption of NADPH at 340 nm in the presence of oxidized glutathione (GSSG), monitoring the reduction of GSSG to GSH by GR, according Carlberg and Mannervik (1985). G6PDH activity was quantified according the production of NADPH through consumption of NADP at 340 nm, using glucose-6-phosphate as substrate and Mg₂⁻¹ (from MgCl₂) as co-factor, according Glock and McLean (1953).

ALDH activity was measured only in the liver because the activity was very low in the gills. The samples were homogenized in potassium phosphate buffer 30 mM, pH 7.5 (1:5; v: v) and centrifuged at 10,000g for 10 min. After homogenization, the samples were analyzed using the method described by Sydow et al. (2004), by monitoring the formation of NADH at 340 nm for 10 min in Tri-HCl buffer 100 mM, pH 8.5, containing 1 mM NAD+, 1 mM pyrazole and 1 mM acetaldehyde as substrate.

EROD, BROD and PROD activity were measured according to Burke and Mayer (1974), by monitoring the conversion of the substrates 7-ethoxyresorufin, 7-benzyloxyresorufin and 7-pentoxyresorufin, respectively, into the fluorescent product resorufin (λ_{excit} =537 nm, λ_{emis} =583 nm) during 3 min, in the presence of NADPH. Protein amount in samples was quantified by the method of Bradford (1976).

Table 2
Enzymatic activities of ethoxyresorufin-O-deethylase (EROD), benziloxiresorufin-O-desbenzilase (BROD), pentroxiresorufin-O-despentilase (PROD), glutathione S-transferase (GST), in gills and liver, and multixenobiotic resistence (MXR) in gills of Oreochromis niloticus, exposed to 40 and 200 ng L^{-1} of diruon, DCPMU, DCPU and DCA and the mixture of all compounds in 10 and 50 ng L^{-1} of each compound for 7 days.

Tissue	Treatment	Concentration (ng L^{-1})	Biochemical biomarkers					
			EROD ^A	BROD ^A	PROD ^A	$\mathbf{GST}^{\mathrm{B}}$	MXR ^C	
Gill	Control	0	11.7 ± 1.9	2.8 ± 1.44	14.8 ± 4.6	12.3 ± 1.0	2.7 ± 0.18	
	Diuron	40	10.6 ± 2.8	1.1 ± 1.5	15.7 ± 1.5	12.0 ± 1.2	2.6 ± 0.06	
		200	15.9 ± 3.3	$6.3 \pm 1.1^{*}$	13.3 ± 2.6	11.5 ± 1.0	$2.0 \pm 0.14^{*}$	
	DCPMU	40	15.0 ± 1.1	6.2 ± 0.7	22.8 ± 2.4	$9.5 \pm 0.5^*$	$1.7 \pm 0.09^*$	
		200	$20.0 \pm 0.5^{*}$	$7.5 \pm 0.8^*$	14.0 ± 6.7	$9.3 \pm 0.6^{*}$	3.0 ± 0.23	
	DCPU	40	$20.6 \pm 1.4^{*}$	2.5 ± 1.0	19.9 ± 4.8	$10.1 \pm 0.3^*$	3.1 ± 0.20	
		200	13.2 ± 2.5	5.7 ± 2.5	21.6 ± 7.9	10.2 ± 0.7	2.7 ± 0.27	
	DCA	40	16.5 ± 3.3	3.6 ± 1.6	17.4 ± 6.7	$10.0 \pm 0.7^*$	$1.9 \pm 0.14^*$	
		200	14.1 ± 1.7	2.6 ± 0.6	27.8 ± 1.6	$9.0 \pm 0.5^*$	$2.2 \pm 0.10^*$	
	Mixture	10	13.8 ± 1.3	2.7 ± 0.6	18.3 ± 1.6	$8.4 \pm 0.3^*$	$2.3 \pm 0.07^*$	
		50	$5.3 \pm 1.0^{*}$	0.08 ± 0.1	19.4 ± 1.8	$9.8 \pm 0.5^{*}$	2.9 ± 0.06	
Liver	Control	0	8.6 ± 1.9	15.7 ± 12.8	12.5 ± 3.0	43.1 ± 8.3	_	
	Diuron	40	103.5 ± 31.9*	9.2 ± 2.5	17.5 ± 0.9	56.6 ± 15.1	_	
		200	92.9 ± 28.1*	12.5 ± 7.5	14.0 ± 2.7	44.5 ± 17.1	_	
	DCPMU	40	46.7 ± 17.2*	5.7 ± 6.2	10.7 ± 0.9	43.0 ± 15.9	_	
		200	$108.4 \pm 22.8^*$	16.0 ± 17.5	8.9 ± 1.9	44.9 ± 7.7	_	
	DCPU	40	$63.1 \pm 6.8^*$	17.3 ± 12.2	7.8 ± 1.5	46.0 ± 12.2	_	
		200	82.7 ± 24.1*	17.0 ± 10.1	8.8 ± 1.9	51.4 ± 9.3	_	
	DCA	40	47.1 ± 9.6*	13.2 ± 11.0	17.8 ± 1.3	43.5 ± 14.0	_	
		200	148.9 ± 44.1*	25.6 ± 12.4	19.0 ± 5.0	38.2 ± 11.7	_	
	Mixture	10	77.1 ± 16.2*	14.8 ± 10.0	6.5 ± 1.8	32.2 ± 13.1	_	
		50	182.5 ± 40.7*	20.8 ± 13.1	19.9 ± 3.0*	41.3 ± 10.3	_	

⁽⁻⁾ No result in this tissue.

All data are mean ± standard deviation;

2.5.2. Multixenobiotic resistance (MXR) activity

MXR analyses were performed in the gills according Luckenbach et al. (2004), with some alterations. Fragments (1 cm²) were cut out from the gills, washed with pure water (to eliminate blood and mucus), dried and then incubated in pure water with 1 μM of rhodamine B (RB) for 90 min at 20 °C in the dark. After this period, the fragments were washed with pure water, dried in filter paper and then placed in 550 μL of butanol. Samples were then sonicated for 30 s to extract intracellular RB and centrifuged for 10 min at 14,000g. The amount RB in supernatant was determined in a spectrofluorimeter (Perkin-Elmer, Victor), using and emission of 545 nm and excitation of 575 nm. Calculations were based on a RB calibration curve.

2.6. Lipid peroxidation

Levels of lipid peroxidation in liver and gills were estimated by measuring the product formed from the reaction of malondialdehyde (MDA) and thiobarbituric acid (TBA) by HPLC coupled to photodiodearray (PDA) detector (Almeida et al., 2003, 2004). The HPLC system (Shimadzu Corporation, Kyoto, Japan) was the same used for chemical analyses of the water. The column used was a Shimadzu Shim-Pack XR-ODS (3.0 \times 100.0 mm, 2.2 μ m particle size, 8 nm pore size). For this analysis, 100 mg of the sample was homogenized in 0.3 mL of Tris buffer 0.1 M (pH 8.0). Then 40 mg of TBA was dissolved in 10 mL of HCl 0.2 M and 0.3 mL of this solution was added to each sample. This mixture was heated at 90 °C for 40 min. After that 1 mL of n-butanol was added and samples were centrifuged at 1500g for 3 min to extract the MDA-TBA derivative. The supernatant was collected and quantified by HPLC at 532 nm, in terms of a malondialdehyde (MDA) standard calibration curve that had been previously prepared (with tetramethoxypropane - TMP) using the same procedure used of the sample.

2.7. Statistical analyzes

Statistical analyzes were performed using the software Statistica 8. The normality and homogeneity of the data were tested by Shapiro-Wilk and Levene test, respectively. For parametric data, one-way ANOVA was used followed by the Tukey post-hoc test. For nonparametric data, the Kruskal-Wallis test was used, followed by multiple comparisons of mean ranks. Values of P < 0.05 were considered as a reference to assign statistical significance (Zar et al., 1999). To analyze the integrative effect of all biomarkers we used the "Integrated Biomarker Response" test (IBR), described by Beliaeff and Burgeot (2002), with modifications according Sanchez et al. (2013) ("Integrated Biomarker Response version 2"- IBRv2). These calculations are based on reference deviation concept. The IBRv2 test use control results to determine the values of other groups. The IBR number is the sum of the deviation of each parameter. The "star plot" was done with the average of the value of each treatment and contributed to observe how much the treated group differed from the control.

3. Results

3.1. Chemical analyses

The measured concentrations of diuron, DCPMU, DCPU and DCA in each experimental group are depicted in Table 1. The concentrations were near the expected, despite some slight discrepancies, especially for the group exposed to DCPMU at 200 ng $\rm L^{-1}.$ No mortality was observed for $O.\ niloticus$ after seven days of exposure.

3.2. Biotransformation enzymes

EROD activity was increased in gill after exposure to DCPMU 200 ng L^{-1} and DCPU 40 ng L^{-1} , however it was decreased after

A pmol/min/mg protein;

 $^{^{\}rm B}$ mU/mg protein.

^с µМ.

^{*} Statistical difference compared to the control group (p < 0.05).

exposure to the mixture of contaminants at the higher concentration when compared to the control group (p = 0.0076). In liver, EROD activity increased in all treatments when compared to the control group, with an observed increase of the enzymatic activity with the increase of the exposure concentration (p = 0.0047) (Table 2).

BROD activity was only increased in gill after exposure to diuron and DCPMU at 200 ng L^{-1} compared to control group (p = 0.018). No alteration in BROD activity was observed in the liver (p = 0.09). PROD activity was unchanged in gill after exposure in all experimental groups (p = 0.28). In liver, PROD increased after exposure to mixture of contaminants at 50 ng L^{-1} when compared to the control (p = 0.001) (Table 2).

GST activity in tilapia gills was decreased after seven days of exposure to DCPMU 40 and 200 ng L^{-1} , DCPU 40 ng L^{-1} , 3,4- DCA 40 ng L^{-1} and 200 ng L^{-1} , and mixture of contaminants at both concentrations of 10 and 50 ng $L^{-1}(p=0.01)$ (Table 2). With respect to MXR activity, a significant decrease was observed in RB efflux in gill of fish treated with Diuron 200 ng L^{-1} , DCPMU 40 ng L^{-1} , 3,4-DCA 40 and 200 ng L^{-1} , and the mixture of all compounds at 10 ng L^{-1} (p < 0.001) (Table 2).

3.3. Oxidative stress response

Activities of the antioxidant enzymes SOD, CAT, GPx, GR and G6PDH of fish from all experimental groups are show in Table 3. SOD activity was not altered in gills after exposure to any isolated contaminants or mixtures (p = 0.13). In liver, SOD activity was increased after exposure to DCPMU 200 ng L $^{-1}$, DCPU 40 ng L $^{-1}$ and to the mixture of compounds at 10 ng L $^{-1}$ when compared to the control group (p = 0.007). CAT was decreased in gills after exposure to diuron 40 ng L $^{-1}$, DCPMU 40 ng L $^{-1}$, DCPU 200 ng L $^{-1}$, 3,4-DCA 40 ng L $^{-1}$ and the mixture of contaminants at 10 ng L $^{-1}$, (p = 0.005). In liver, CAT activity was increased after exposure to 3,4-DCA 200 ng L $^{-1}$ and

the mixture of contaminants for both 10 and 50 ng L $^{-1}$ (p < 0.001). GPx activity increased in gills after exposure to all compounds, (p = 0.008), however no changes were observed in the liver (p = 0.5). GR activity in gills was increased after exposure to diuron 200 ng L $^{-1}$, DCPMU 40 and 200 ng L $^{-1}$, and decreased after exposure to DCPU 40 and 200 ng L $^{-1}$, 3,4-DCA at 200 nd. L $^{-1}$ and mixture of contaminants at 10 and 50 ng L $^{-1}$ (p < 0.001), while no changes were observed in the liver (p = 0.47). G6PDH activity was decreased in gills after exposure to DCPU 200 ng L $^{-1}$ and 3,4-DCA 40 ng L $^{-1}$ compared to the control group (p = 0.007). On an opposite way, G6PDH was increased in liver after exposure to DCPMU 200 ng L $^{-1}$, 3,4-DCA 40 and 200 ng L $^{-1}$ and mixture of contaminants at 50 ng L $^{-1}$ (p < 0.001) (Table 3). Hepatic ALDH activity in tilapia was increased after exposure to DCPMU 200 ng L $^{-1}$ (p < 0.001) and to 3,4-DCA 40 ng L $^{-1}$ (p = 0.007) (Table 3).

Lipid peroxidation levels were higher in gills of fish exposed to diuron 40 ng L^{-1} , DCPMU $40 \text{ and } 200 \text{ ng L}^{-1}$, DCPU $40 \text{ and } 200 \text{ ng L}^{-1}$, 3,4-DCA $40 \text{ and } 200 \text{ ng L}^{-1}$ and the mixture of contaminants at $10 \text{ and } 50 \text{ ng L}^{-1}$ (p < 0.001). In liver, no alteration was observed after exposure (p = 0.06) (Table 3).

3.4. Integrated biomarker response

All treatments caused significant increases on IBR scores in gill tissues. The gills of animals exposed to DCPMU 40 and 200 ng L^{-1} presented the greatest increase from the control group (Fig. 1). In the liver, we observed that the metabolite DCPU did not influence IBR index for both concentrations. All other compounds increased IBR values in liver, excepted by diuron at concentration of 200 ng/ L^{-1} and DCPMU at 40 ng/ L^{-1} (Fig. 2). Similarly to that observed in gills, fish exposed to DCPMU 200 ng L^{-1} presented the greatest increase from the control group (Fig. 2). The "star plots" representations for gills (Fig. 3) and liver (Fig. 4) indicate how each individual biomarker contributed

Table 3
Enzymatic activities of superoxide dismutase (SOD), catalase (CAT) and glutathione peroxidase (GPx), glutathione reductase (GR), glutathione-6-phosfato dehydrogenase (G6PDH) and malondialdehyde (MDA) levels in gill and liver, and aldehyde dehydrogenase (ALDH) in liver of Oreochromis niloticus, exposed to 40 and 200 ng L^{-1} of diruon, DCPMU, DCPU and DCA and the mixture of all compounds in 10 and 50 ng L^{-1} of each compound for 7 days.

Tissue	Treatment	Concentration (ng L^{-1})	Biochemical biomarkers						
			SOD ^C	CAT^{C}	GPx ^B	$\mathbf{G}\mathbf{R}^{\mathrm{B}}$	G6PDH ^C	$\mathbf{MDA}^{\mathrm{D}}$	ALDH ^B
Gill	Control	0	4.90 ± 0.54	660 ± 12.7	1.0 ± 0.13	0.58 ± 0.04	0.71 ± 0.03	10.4 ± 1.5	_
	Diuron	40	6.15 ± 0.54	$582 \pm 20.3^{*}$	$2.5 \pm 0.11^*$	0.69 ± 0.03	0.60 ± 0.04	$17.4 \pm 2.4^*$	_
		200	5.10 ± 0.25	662 ± 9.8	$2.6 \pm 0.27^*$	$0.76 \pm 0.06^*$	0.58 ± 0.03	$15.2 \pm 3.1^*$	_
	DCPMU	40	5.88 ± 0.46	561 ± 30.2*	$2.3 \pm 0.39^*$	$0.82 \pm 0.02^*$	0.72 ± 0.08	$22.2 \pm 1.6^{*}$	_
		200	5.33 ± 0.20	605 ± 32.4	$2.4 \pm 0.14^*$	$0.75 \pm 0.01^*$	0.64 ± 0.07	$22.0 \pm 2.2^{*}$	_
	DCPU	40	5.91 ± 0.44	649 ± 24.2	$2.1 \pm 0.09^*$	$0.22 \pm 0.07^*$	0.64 ± 0.02	$16.9 \pm 1.7^{*}$	_
		200	5.02 ± 0.32	584 ± 10.8*	$2.1 \pm 0.08^*$	$0.17 \pm 0.03^*$	$0.57 \pm 0.02^*$	$19.0 \pm 2.9^*$	_
	DCA	40	4.81 ± 0.23	547 ± 5.3*	$2.1 \pm 0.13^{*}$	$0.33 \pm 0.01^*$	$0.49 \pm 0.05^*$	$14.2 \pm 1.8^*$	_
		200	4.92 ± 0.25	632 ± 30.1	$2.2 \pm 0.09^*$	$0.24 \pm 0.05^*$	0.80 ± 0.03	$13.7 \pm 1.9^*$	_
	Mixture	10	4.81 ± 0.25	544 ± 45.8*	$2.1 \pm 0.05^*$	$0.27 \pm 0.09^*$	0.69 ± 0.01	$15.1 \pm 1.1^*$	_
		50	4.96 ± 0.52	621 ± 29.4	$1.9 \pm 0.34^{*}$	$0.06 \pm 0.02^*$	0.62 ± 0.05	$18.4 \pm 2.7^{*}$	_
Liver	Control	0	9.7 ± 2.1	12.1 ± 2.2	17.0 ± 4.0	9.4 ± 3.1	0.47 ± 0.12	19.3 ± 1.8	3.43 ± 1.35
	Diuron	40	9.8 ± 1.7	10.6 ± 2.7	18.3 ± 4.6	9.6 ± 3.9	0.64 ± 0.10	11.3 ± 1.4	1.54 ± 0.03
		200	9.3 ± 1.9	10.6 ± 2.0	14.3 ± 5.4	11.6 ± 4.9	0.69 ± 0.31	13.6 ± 1.9	3.82 ± 0.09
	DCPMU	40	9.1 ± 3.0	11.2 ± 2.5	16.7 ± 7.2	9.3 ± 4.6	0.70 ± 0.28	12.7 ± 2.9	4.75 ± 0.59
		200	$13.0 \pm 0.6^{*}$	10.3 ± 2.8	18.4 ± 4.3	13.1 ± 2.9	$0.85 \pm 0.15^*$	14.4 ± 2.6	10.84 ± 2.0
	DCPU	40	12.2 ± 2.5	10.2 ± 3.1	15.7 ± 3.4	10.0 ± 2.1	0.65 ± 0.15	12.3 ± 2.6	6.05 ± 0.23
		200	11.2 ± 0.5	12.2 ± 3.0	17.9 ± 4.3	11.5 ± 1.4	0.64 ± 0.13	17.1 ± 1.2	5.96 ± 1.36
	DCA	40	11.0 ± 0.9	13.4 ± 1.7	19.6 ± 7.6	9.4 ± 2.0	$0.99 \pm 0.29^*$	11.1 ± 1.4	7.38 ± 1.28
		200	11.5 ± 2.0	$14.1 \pm 0.9^*$	16.4 ± 4.2	9.8 ± 2.5	$1.03 \pm 0.15^{*}$	16.7 ± 2.8	3.71 ± 0.98
	Mixture	10	$14.3 \pm 0.6^*$	$14.5 \pm 3.3^{*}$	19.8 ± 6.6	10.1 ± 2.4	0.63 ± 0.22	16.9 ± 4.4	5.85 ± 0.12
		50	9.5 ± 0.8	$16.4 \pm 2.0^{*}$	16.7 ± 5.1	11.9 ± 2.7	$1.11 \pm 0.17^*$	13.2 ± 0.7	4.60 ± 0.22

All data are mean \pm standard deviation;

⁽⁻⁾ No result in this tissue. EpM/mg tissue.

B mU/mg protein;

^C U/mg protein;

D nmol/mg tissue;

^{*} Statistical difference compared to the control group (p < 0.05).

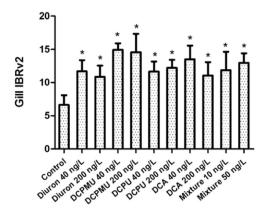


Fig. 1. Integrated biomarker response (IBR) index summarizing the responses of biochemical biomarkers to diuron, DCPMU, DCPU, DCA and the mixtures in gills of fish O. niloticus. Asterisk (*) indicates statistical difference compared to the control group (p < 0.05).

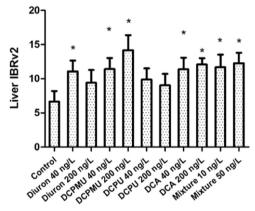


Fig. 2. Integrated biomarker response (IBR) index summarizing the responses of biochemical biomarkers to diuron, DCPMU, DCPU, DCA and the mixtures in liver of fish O. niloticus. Asterisk (*) indicates statistical difference compared to the control group (p < 0.05).

for the IBR value obtained for each experimental group.

4. Discussion

Diuron and its main metabolite, 3,4-DCA, have been shown to exert toxic effects in aquatic organisms (Ensenbach et al., 1996; Guilhermino et al., 1998; Sosak-Swiderska et al., 1998; Girling et al., 2000; Tixier et al., 2001; Osano et al., 2002; Pereira et al., 2015). However, information about the toxicity of other intermediate metabolites, such as DCPMU and DCPU, which participate in the 3,4-DCA formation pathway, are still limited for aquatic and other organisms, although some studies have pointed their higher toxicity in relation to diuron (Tixier et al., 2000). Until the moment, other effects than endocrine disruption caused by diuron and its metabolites on fish physiology have not been demonstrated, especially when they are present in mixture conditions, which makes it difficult to interpret how these animals are adapting to diuron contamination in water bodies. Understanding how aquatic organisms deal with contaminants in their environments is extremely urgent in a context where the toxicity of most agrochemicals used in crops in Brazil is still unknown for many species. In this study, we showed that diuron and its metabolites, in both isolated or mixture conditions impair biotransformation enzymes and antioxidant response of fish, leading the organisms to a condition of oxidative stress, which was evidenced by increased lipid oxidation.

In general, both diuron and metabolites caused significant but disperse alterations in oxidative stress parameters and biotransformation enzymes. In addition, no clear dose-response relationship was observed,

except that EROD activity was increased in the liver of fish after exposure to all treatments. This increase in EROD suggests the involvement of this enzyme in the metabolism of diuron and all metabolites, and could account to indicate it as a suitable biomarker for diuron or diuron metabolite exposure in fish. In accordance to our result, Zhao et al. (2006) demonstrated that diuron is capable to bind the AhR receptor, thus activating CYP1A in mammals. Previously, it was also demonstrated that diuron was able to increase EROD activity in rats, corroborating the involvement of this enzyme in diuron metabolism (Schoket and Vincze, 1990). In contrary, BROD and PROD presented very disperse response in our study, and they were apparently not involved in the metabolism of the studied compounds. In the gills, the significant decrease in GST activity in those fish exposed to diuron metabolites, but not diuron, could indicate a higher negative impact of diuron metabolites compared to the parental compound on fish health, since decreased GST activity turns the fish less efficient on biotransformation processes. Indeed, diuron, DCPMU and DCA caused a decrease in MXR activity, also indicating a negative impact of these compounds in defense mechanisms of the gills against intoxication. MXR is a protein efflux transporter that keeps toxicants out of the cells, protecting the cells from environmental contaminants (Ferreira et al., 2014). The inhibition of MXR as seen in this study can compromise the effectiveness of the defense system, since toxic substances that would normally be excluded, will remain in the cell and exert their toxic effect (Luckenbach et al., 2004; Fisher et al., 2013; Ferreira et al., 2014).

With respect to oxidative stress markers, it was clear that all the compounds triggered oxidative stress in the gills, since all of them caused a marked increase in MDA levels in this tissue. Nevertheless, no such increase was observed in the liver, which indicates the gills as a more susceptible organ to oxidative stress. It is well known that several classes of pesticides are able to induce peroxidation of membrane components in response to a non-compensatory action of antioxidant system against excessive amount of ROS, produced under stress conditions (Regoli and Giuliani, 2014). In this study, we observed that antioxidant enzyme responses were very puzzled for all enzymes, except for GPx activity in gills that increased after exposure to all treatments, also indicating a probable increase in ROS production that lead to increased MDA levels. The remaining antioxidant enzymes presented very disperse responses, without any dose-response relationships. We also observed a varied response between the different studied tissues. For example G6PDH activity in the gill was decreased only after exposure to the higher concentration of DCPU and the lower concentration of 3,4-DCA, while in the liver this enzyme was increased only in those animals exposed to the higher concentration of DCPMU, the mixture of all compounds and to both concentration of 3,4-DCA. SOD activity was increased only in the gills of animals exposed to the higher concentration of DCPMU and to the lower concentration of the mixture. CAT activity also showed random decreases in the gill, and increases in the liver, and GR presented a significant increase in the gills of animals exposed to both concentrations of diuron and DCPMU, while this enzyme was lower in the same organ of animals exposed to the remaining treatments. Finally, ALDH was increased only in the gills of animals exposed to the higher concentration of DCPMU and to the lower concentration of DCA, showing no correlation with MDA levels.

Increases in antioxidant enzymes are generally a response to increases in ROS production, while decreases can be a result of an inhibitory effect of the contaminant along the exposure time (Regoli and Giuliani, 2014). The lack of any dose-response or conserved response of antioxidant enzymes to the tested contaminants observed in this study indicates that the different compounds exert distinct mechanisms in the analyzed organs without a clear pattern. Nevertheless, it should be considered that the contaminants interfered in the antioxidant defense enzymes of the fish, which could indicate the instigation of ROS production in the toxicity mechanism of all tested compounds. Although less evident in the liver due to the lack of alterations in MDA levels, this hypothesis can be sustained by the significant increase in GPx activity

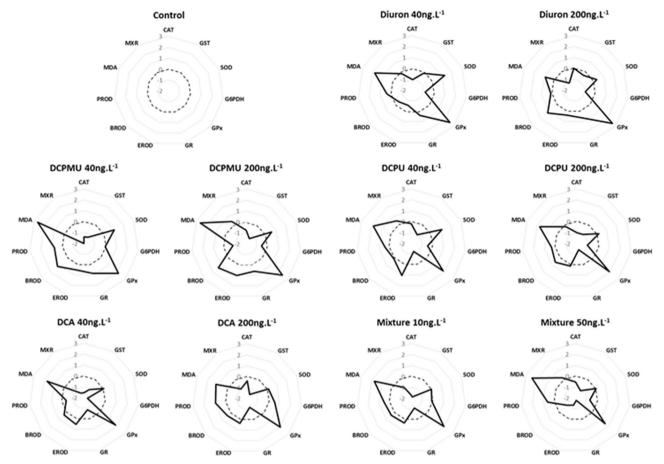


Fig. 3. Star plot of the mean of the standardized differences of the biomarkers in gills of Oreochromis niloticus of each treatment in relation to control group. The dotted circle represents the control group as reference and the black line represents treated groups. If the black line have being outside the dotted circle, the parameter analyzed was increased, if the black line have being inside the dotted circle, the parameter was decreased, compared to the control group.

and MDA levels in the gills.

Similarly to those responses observed on biotransformation enzymes, the analysis of the individual oxidative stress parameters also indicated that diuron metabolites had a more prominent action in the antioxidant defense system of fish than the original compound diuron. These responses were especially evident in animals exposed to the metabolite DCPMU, which impaired most of the enzymes in liver and gills. These results were confirmed by the applied IBR model, which distinguished the effects of DCPMU as the most pronounced in relation to the control group. Previous studies with O. niloticus also showed that DCPMU, but in the same proportion as the other intermediates metabolites, caused prominent anti-androgenic and estrogenic effects compared to diuron (Pereira et al., 2015, 2016). This evidences the relevance of studying intermediate metabolites, and not only the final product or the original compound, in monitoring studies involving pesticides. In addition, exposure of fish to mixture conditions apparently did not cause additive effects or interaction effects on antioxidant response or oxidative stress, even for those exposed to DCPMU. Synergistic effects could account for this lack of response, since contaminant concentrations in the mixture were lower than that used for exposure to the isolated compound.

Taking into account the two sets of biomarkers analyzed, oxidative stress markers and biotransformation enzymes, we noted that EROD in the liver and GPx activity and MDA levels in the gill presented a clear response to the studied contaminants, which would suggest these parameters as general biomarkers for diuron and metabolites monitoring studies. Although the disperse responses of the other biomarkers (BROD, PROD, MXR, SOD, CAT, GR and ALDH), it actually represented an advantage for a better attribution of general effects of the

contaminants on fish health in a integrated approach.

5. Conclusion

Our data demonstrated that diuron and its biodegradation metabolites at environmental relevant concentrations are able to alter different physiological response of fish, evidenced by changes of biochemical parameters of oxidative stress and biotransformation. These findings corroborate with other previous studies that showed that diuron metabolites can be as deleterious as diuron (Saglio and Trijasse, 1998; Miranda et al., 2008; Scheil et al., 2009; Mhadhbi and Beiras, 2012; Sánchez-Muros et al., 2013; Pereira et al., 2015, 2016; Felício et al., 2016). In fact, analysis of individual and integrated parameters pointed that diuron metabolites may promote more evidenced changes on antioxidant response and biotransformation process than the original compound. These results were especially prominent for DCPMU, which was noted as the main compound causing alterations in exposed fish in an integrated biomarker analysis. We also observed that EROD in the liver, and GPx and MDA levels in the gills were especially responsive to the treatments, being considered the parameters with greater responsiveness to the general effects of all contaminants. Our results bring important issues about the using of different biomarkers on monitoring studies approach. It evidences that a variety of tools not only demonstrate the most appropriate biomarkers that should be applied for exposure investigations, but they also contribute to the understanding of which pathways environmental contaminants take to harm the health of aquatic organisms.

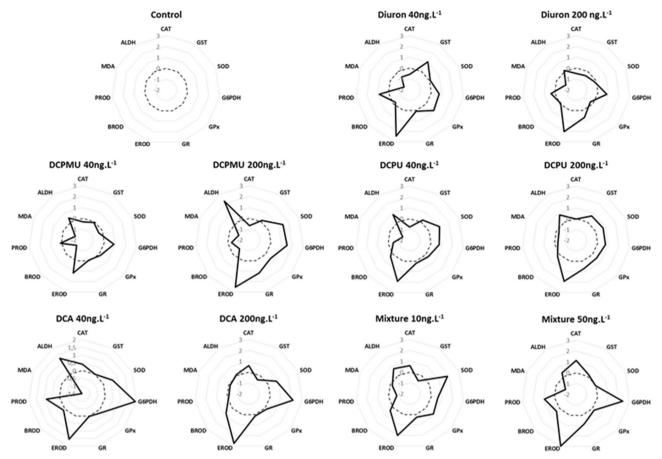


Fig. 4. Star plot of the mean of the standardized differences of the biomarkers in liver of Oreochromis niloticus of each treatment in relation to control group. The dotted circle represents the control group as reference and the black line represents treated groups. If the black line have being outside the dotted circle, the parameter analyzed was increased, if the black line have being inside the dotted circle, the parameter was decreased, compared to the control group.

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Conflict of interest statement

The authors disclose any potential sources of conflict of interest.

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