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Inhibition of thioredoxin A1 from *Corynebacterium pseudotuberculosis* by polyanions and flavonoids



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ABSTRACT

In pathogens, the thioredoxin system forms part of the defense against oxidative stress and ensures the formation of the proper disulfide bonds to ensure protein function. In *Corynebacterium pseudotuberculosis*, the role and mechanism of TrxA1 has not been elucidated, but, the significant homology among different Trxs and the conservation of the residues that form their active sites underline the importance of the Trx systems. Proteins involved in redox metabolism and low molecular weight thiols, which might interact with them, become attractive targets to modulate the activity of pathogens.

The activity of the protein was investigated using a turbidimetric assay system. The influence of different pH and low molecular weight thiols were tested. Additionally, this assay was used to investigate the inhibitory potential of ligands from different molecular families, such as, polyanions (suramin and heparin) and flavonoids (hesperetin and hesperidin). All four compounds showed inhibition of the protein activity by approximately 80%. The interactions between these compounds and *Cp*-TrxA1 were investigated using CD spectroscopy, NMR, molecular docking and dynamics. Our results demonstrate that suramin and hesperetin can serve as lead molecules for the development of specific inhibitors for the *C. pseudotuberculosis* TrxA1.

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1. Introduction

Corynebacterium pseudotuberculosis (*C. pseudotuberculosis*) together with the *Mycobacterium*, *Nocardia*, and *Rhodococcus* genus constitutes the heterogeneous CMNR-group of pathogens [1]. *C. pseudotuberculosis* is the causative agent of caseous lymphadenitis (CLA), a disease encountered in sheep, goats, equids (ulcerative lymphangitis), and cattle (cutaneous excoriated granulomas). CLA infection results in significant reductions in wool and milk yields, causes cattle weight loss, leads to carcass condemnation and, can eventually, lead to death thereby resulting in considerable economic loss [2, 3] and isolated cases of infection have also been reported in humans [4]. As a facultative intracellular parasite, this bacterium is capable of survival and growth in macrophages and is thus able to evade detection by the host immune system [5].

Redox metabolism represents an alternative, attractive target to control and/or modulate rapidly growing pathogens such as *C. pseudo-tuberculosis*. Low molecular weight (LMW) thiols play key roles in ensuring and modulating a reducing intracellular environment which is essential for the maintenance of regular metabolic activities [6, 7]. Besides the LMW thiols, many organisms contain thiol-disulfide oxidore-ductases such as thioredoxins that function through their redox-active disulfides as reducing agents on protein disulfide bonds [8, 9].

Thioredoxins (Trx) are low molecular weight oxidoreductases (~12 kDa), structurally characterized by a central four-stranded distorted anti-parallel β -sheet surrounded by three α -helices [10, 11]. *In vitro* experiments indicate that the disulfide bonds in several proteins are efficiently reduced by thioredoxin [9, 12, 13]. Upon the reduction of the disulfide bonds, Trxs are oxidized and are subsequently reduced by thioredoxin reductase (TrxR) by utilizing NAD(P)H [9]. Trxs play crucial roles in the regulation of cellular redox homeostasis in pathogens, for

Abbreviations: CLA, Caseous lymphadenitis; CD, circular dichroism; CoASH, coenzyme-A; DTT, dithiothreitol; ESH, ergothioneine; GSH, glutathione; Hst, hesperetin; Hsp, hesperidin; LMW thiol, low-molecular weight thiol; Trx, thioredoxin; SEC, size exclusion column; MD, molecular dynamics; CpHMD, constant pH-titration molecular dynamics; RMSF, root mean square fluctuation; RMSD, root mean square deviation; MM/GBSA, molecular mechanics/generalized Born surface area; NVT, constant volume and temperature; NPT, constant pressure and temperature; PME, particle-mesh Ewald method; STD-NMR, saturation transfer difference-nuclear magnetic resonance.

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example, *M. tuberculosis* relies on the Trx redox system for survival in conditions of elevated environmental stress [14]. *M. tuberculosis* and *C. pseudotuberculosis* are members of the CMNR-group and the actinomycetes and the Trx systems are essential for the survival of both pathogens.

We investigated the inhibitory potential of representative members from two distinct molecular classes *i.e.*; two polyanions (heparin, suramin) and two flavonoids (hesperetin and hesperidin) against *C. pseudotuberculosis* thioredoxin A1 (*Cp*-TrxA1). Polyanions are macromolecules and macromolecular complexes with multiple negative charges, and among others, are represented by proteoglycans (heparin), DNA, RNA, actin (microfilaments), tubulin (microtubules), polysialic acids, ribosomes, suramin, *etc.* [15].

In contrast, flavonoids belong to a large group of mildly charged phenolic compounds that are widely distributed in plants. Several biological properties have been reported for flavonoids, including antioxidant, anticancer, cancer chemo preventive, and anti-inflammatory properties [16, 17]. Hesperidin (Hsd) is a flavanone glycoside (consisting of the disaccharide rutinose), which is abundant in citrus fruits. Its aglycone form is hesperetin (Hst) and both molecules are hydrophobic [18, 19].

Our study describes the production, purification and characterization of *Cp*-TrxA1. Suramin, heparin, Hst and Hsp inhibited (>80%) *Cp*-TrxA1 and were evaluated using activity assay. The evaluation of the binding modes between polyanions and flavonoids by CD spectroscopy, NMR techniques and molecular dynamics indicated significant differences which are relevant for the development of specific inhibitors against this family of proteins.

2. Material and methods

2.1. Chemical and biochemical reagents

Kanamycin, chloramphenicol, MgSO₄, IPTG, K₂HPO₄, KH₂PO₄, NaCl, glycerol, imidazole, DTT, diamide, glutathione, coenzyme-A, ergothioneine, low molecular weight heparin (MW: 6000 kDa), suramin, hesperetin, EDTA, methanol, D₂O, 4,4-dimethyl-4-silapentane-1-sulfonic acid, DMSO, acetic acid and isopropanol were purchased from Sigma-Aldrich (USA). Human insulin was purchased from Eli Lilly (Brazil). Hesperidin was produced and purified as described in Section 2.7.

2.2. Overexpression and purification of Cp-TrxA1

The E.coli BL21-CodonPlus (DE3)-RIL strain harboring the pD441-SR (DNA 2.0-USA) plasmid was cultured overnight in Luria-Bertani (LB) medium supplemented with sufficient amounts of kanamycin and chloramphenicol at 37 °C. The bacterial cultures were then inoculated into fresh LB medium (with the same antibiotic), till the final A₆₀₀ reached 0.6. At that point, Cp-TrxA1 was supplemented with 50 μ mol L⁻¹ MgSO₄ and induced with 0.35 mmol L⁻¹ IPTG, the incubation continued for 5 h at 30 °C. The cultures were harvested and the cell pellet re-suspended in 20 mmol L^{-1} K₂HPO₄/KH₂PO₄, pH 7.4, 500 mmol L^{-1} NaCl, 5% (v/v) glycerol. The cells were lysed by sonication in four sets of 30 s pulses of 30% amplitude, with 10 s intervals, followed by centrifugation at 8000 \times g, 6 °C for 90 min. The supernatant containing the recombinant Cp-TrxA1 with 6x-His-tag was loaded onto a Ni-NTA column pre-equilibrated with the same buffer. To remove the unbound protein, the column was extensively washed with 20 mmol L^{-1} K_2 HPO₄/KH₂PO₄, pH 7.4, 500 mmol L⁻¹ NaCl, 5% (v/v) glycerol, containing 20 and 60 mmol L^{-1} imidazole. Cp-TrxA1 was eluted with 500 mmol L^{-1} imizole. Eluted fractions were individually pooled, and injected onto a Superdex 75 10/300 GL (GE Healthcare) size exclusion column (SEC), pre-equilibrated with 20 mmol L⁻¹ K₂HPO₄/KH₂PO₄ pH 7.4, 150 mmol L^{-1} NaCl. Sample purities, were assessed by SDS-PAGE (20%) gels after each purification step. Protein concentrations were determined spectrophotometrically by applying the Lambert-Beer law [20].

2.3. Circular dichroism spectroscopy (CD)

For CD measurements, 15 repeated scans were performed, of which 5 were used to establish the baseline. The wavelength range used to obtain the far-UV spectra was from 200 nm to 260 nm, in a constant time of 1 s and 100 nm/min continuous scanning mode, using a Jasco J-107 spectropolarimeter (Jasco, Japan). *Cp*-TrxA1 was diluted in 20 mmol L^{-1} K₂HPO₄/KH₂PO₄. 20 mmol L^{-1} K₂HPO₄/KH₂PO₄ at pH 7.4 was used to determine the effect of 1.4 µmol L^{-1} : DTT, diamide, GSH, CoASH, ESH, on the protein secondary structure. The effect of the ligands heparin, suramin, hesperidin (Hsd) and hesperetin (Hsp) were individually tested with 10 µmol L^{-1} of *Cp*-TrxA1. The protein was incubated with a fourfold molar excess of the ligands (40 µmol L^{-1}) at 4 °C for 2 h. Subsequently, the protein was extensively dialyzed against 20 mmol L^{-1} K₂HPO₄/KH₂PO₄ to remove unbound ligand molecules and the results were analyzed using the CDpro software package [21].

2.4. Nuclear magnetic resonance (NMR)

2.4.1. Saturation transfer difference by nuclear magnetic resonance (STD-NMR)

Nuclear Magnetic Resonance (NMR) spectroscopy was used to identify the *Cp*-TrxA1 interaction with potential modulators. All experiments were performed utilizing a Bruker AVANCE III HD spectrometer (Bruker, Germany) operating at 600 MHz for ¹H, equipped with a triple-resonance, pulsed-field, z-gradient cryoprobe. An off-resonance at 20 ppm was chosen, and the on-resonance frequencies were selected according to the ligand signals. The Bruker pulse sequence STDDIFFESGP was used for the STD-NMR spectroscopy experiments. The protein was saturated for 2 s, the recycle delay was set to 3 s and a saturation power of 35 dBW was used. The protein signal was suppressed by utilizing a spin-lock filter of 30 ms.

Each experiment used a solution of 400 μ mol L⁻¹ of each ligand (heparin, suramin, Hsp, and Hst) with 20 μ mol L⁻¹ of *Cp*-TrxA1 diluted in 20 mmol L⁻¹ K₂HPO₄/KH₂PO₄, pH 7.4 and possible interactions were evaluated based on the STD-NMR results.

2.4.2. Chemical shift mapping by nuclear magnetic resonance

For protein-ligand identification and quantification, the protein chemical shifts which are usually sensitive to ligand binding were identified by ¹⁵N-HSQC (Heteronuclear Single Quantum Coherence) experiments [22]. Since the NMR spectra have been assigned, binding sites could be identified by mapping them on the protein molecular surface. Chemical shift mapping using the ¹H NMR protein spectrum enabled the identification of shifts of tryptophan side chain indole groups upon ligand binding.

A Bruker Advance (Bruker, Germany) 600 MHz equipped with a 5 mm triple resonance cryoprobe with pulse-field gradient along the z-axis was used and for the regular excitation sculpting experiment (ZGESGP). Data was acquired for *Cp*-TrxA1 in 20 mmol L⁻¹ K₂HPO₄/ KH₂PO₄, pH 7.4 in 10% D₂O both in the presence and absence of each ligand. A temperature of 20 °C was set and 128 transients were collected, with a slide window of 22 ppm and acquisition time of 1.21 s. Each spectrum was referenced to DSS, presented in 25 µmol L⁻¹ in the solution. For *Cp*-TrxA1-suramin interaction, a 0.9 mmol L⁻¹ protein concentration was used and its spectrum was compared to one using 1.8 mmol L⁻¹ ligand. For the flavonoid hesperitin, due to the hydrophobic nature, a 0.1 mmol L⁻¹ protein sample was used and compared to the spectrum in the presence of 0.2 mmol L⁻¹ of the ligand. Experiments were processed in Bruker TopSpin 3.2, applying a line broadening of 0.3 Hz prior to Fourier transformation.

2.5. Turbidimetric assay of insulin disulfide reduction

The rate of insulin disulfide reduction by *Cp*-TrxA1 was monitored spectrophotometrically following evaluation of the turbidity at 640 nm [23–26], using a Biomate UV–Visible Spectrophotometer (Thermo Fisher Scientific, USA). The reaction mixture contained a total volume of 500 μ L, with 0.2 mmol L⁻¹ insulin, 50 mmol L⁻¹ Tris/HCl, 2 mmol L⁻¹ EDTA, and 5 μ mol L⁻¹ protein. The optimum pH for the *Cp*-TrxA1 activity was determined by using Tris/HCl buffers at different pH values (7.0, 7.4, 7.8, 8.0, 8.4, 8.8 and 9.0). The reactions occurred at 25 °C by adding 1 mmol L⁻¹ DTT. In another experiment, instead of DTT, the reductive agents GSH, ESH, CoASH and the oxidant diamide, were used at a final concentration of 1 mmol L⁻¹. As a control, the non-enzymatic reduction of insulin by the corresponding reductive agent was used.

Cp-TrxA1 activity was assayed in the presence of the following ligands: heparin, suramin, hesperidin, and hesperetin, all at final concentrations of 0.14 µmol L⁻¹. The protein was incubated with the ligands overnight at 4 °C and, subsequently, the samples were injected onto a Superdex 75 10/300 GL (GE Healthcare) column pre-equilibrated with 20 mmol L⁻¹ Tris/HCl pH 7.4, 150 mmol L⁻¹ NaCl, to remove nonspecifically bound ligands, and then tested using the reaction mixture described above. The experiments were performed as triplet measurements and repeated twice.

2.6. Molecular dynamics and computational analysis of Cp-TrxA1

2.6.1. Starting structures and molecular docking

The initial model of the *Cp*-TrxA1 protein was obtained by homology modeling with the *M. avium* Trx structure (PDB: 4WXT; 50% homology), in order to obtain a refined and relaxed model, the protein was solvated and ions were added to neutralize the system. Following which, 100 ns Molecular Dynamics (MD) simulations were carried out and clustering analysis were performed to obtain a stable and representative model (Fig. S1A, Supplementary material). Structural superposition of the *Cp*-TrxA1 homology model and the *M. avium* Trx structure showed a RMSD of 0.188 Å (Fig. S1B, Supplementary material).

The model was used as a starting conformation for constant pH MD (CpHMD), for docking calculations and posterior MD simulations with ligands. To prepare the ligands, Gaussian 09 [27] was used at the level of theory HF/6-31G* to optimize and calculate their ESP (Electrostatic Surface Potential). Following which, the antechamber program [28] was used to calculate their RESP (Restrained Electrostatic Potential) charges and parmchk program to obtain the missing GAFF (General Amber Force Field) parameters [29].

Docking calculations were performed using AutoDock Vina 1.1.12 [30]. The AutoDockTools program [31] was used adding polar hydrogens and partial charges to the protein and set the rotational bonds in the ligands. The search space was defined in the gridbox near the binding site. As output results, for each calculation, several poses were ranked according to the scoring function of AutoDock Vina. Thus, the first poses for *Cp*-TrxA1-Hst and *Cp*-TrxA1-suramin complexes were used to initiate MD simulations.

2.6.2. General setup and parameterization of the molecular dynamics simulations

All simulations were carried out using the AMBER16 [32] program package. The FF14SB force field [33] was used to describe the all-atom protein interaction, whereas the GAFF and RESP charges were used to describe the ligands. The H++ web-server [34] was used to set the amino acid protonation state at pH 7.0. The systems were neutralized with Na⁺ ions and placed in a rectangular box of TIP3P water extended 10 Å away from any protein atom. To exclude bad contacts from initial structures, each system was minimized in two steps. First, the energy minimization of the protein-ligand constrained complexes (force constant of 50.0 kcal/mol-Å²) were performed with 2500 steepest descent steps followed by 2500 conjugate gradient steps followed by unconstrained energy minimization rounds (10,000 steps). After minimization, the system was gradually heated from 0 to 293 K for 300 ps under NVT ensemble (constant volume and temperature), while the protein was restrained with force constant of 25 kcal/mol-Å². Subsequently, an equilibration step was performed using NpT ensemble (constant pressure and temperature) for 500 ps. Finally, the production runs, 100 and 150 ns for *Cp*-TrxA1-Hst and *Cp*-TrxA1-suramin, respectively, were performed in NpT ensemble without any restraints. The temperature (293 K) and pressure (1 atm) were controlled by Langevin coupling. The SHAKE constraints were applied to all bonds involving hydrogen atoms to allow a 2-fs dynamics time step. Long-range electrostatic interactions were calculated by the particle-mesh Ewald method (PME) using 10-Å cutoff.

2.6.3. Constant pH-titration MD

To understand how pH affect the protein dynamics, we used the CpHMD. First, to prepare the system, the names of the residues Glu and Asp were changed, respectively, to GL4 and AS4 to allow the correct treatment by AMBER. During the simulation, all the GL4, AS4, LYS and CYS residues were titrated, while the rest of residues existed in a fixed protonation state.

The CpHMD simulations were performed following a procedure that resembles a titration experiment in a wet-laboratory, as introduced by Socher and Sticht [35]. The idea is to divide the MD into a large number of small steps and change the pH after each step. In our setup, the MD started at pH 7 and lasted for 1 ns in each step. After that, using the final coordinates and velocity of the previous MD as an input, we increased the solvent pH by 0.05 pH unit and run another round. The pH range was from 7.0 to 9.0, generating a total of 140 short 1-ns MD simulations and a total time of 140 ns. The protonation was checked every 10 steps, according to the MC (Monte Carlo) sampling of the Boltzmann distribution of protonation states. For the protonation check, it strips all the explicit solvent and uses the generalized Born solvent model (igb = 2). After protonation check the MD simulation could follow two directions: if the protonation changes, it ran 100 steps of MD simulation to allow the solvent relaxation around the new titratable site; or, if the protonation did not change, it restored the solvent and continued the MD simulation from the same point.

2.6.4. Structural and dynamical analyses

All the analyses were performed with the CPPTRAJ program [36] of AmberTools17. VMD and Pymol were used for visualization of the systems. RMSD of C α was calculated to determine the equilibration and convergence of the systems. Clustering analysis was done with kmeans method ranging from 2 to 6. To access the quality of clustering we use the DBI values and also silhouette analyses. Protein flexibility was studied by root mean square fluctuation (RMSF) for the C α . The RMSF were calculated residue-by-residue over the equilibrated trajectories.

The interaction energy was calculated using the molecular mechanics/generalized Born surface area (MM/GBSA) approach. MM/GBSA energy was computed between the protein and the ligands in the stable regime comprising the last 10 ns of the MD simulation. For these calculations, all the solvent and ions were stripped.

The secondary structure content of the *Cp*-TrxA1 homology model and the complexes with suramin and Hst after MD simulations were evaluated with PDBsum [37].

2.7. Production and purification of hesperidin (Hsp)

Hesperidin was obtained according to Tasic et al. [38].

3. Results and discussion

3.1. Enzyme production and purification

The recombinant expressed *Cp*-TrxA1 was purified by affinity chromatography using a Ni-NTA column. The recombinant protein was eluted with 500 mmol L^{-1} imidazole; a subsequent purification step was performed using SEC to remove trace protein impurities. After SEC, a single protein band with a molecular mass of around 14 kDa was observed on a 20% SDS-PAGE gel, the result correlated with the predicted molecular mass (~13,5 kDa; with 6x-His-tag) (Fig. S2, Supplementary material).

3.2. Enzyme activity and pH effect on the catalytically process of Cp-TrxA1

The *Cp*-TrxA1 activity was tested by a turbidimetric insulin reduction assay. The catalytic effect of *Cp*-TrxA1 resulted in the rapid precipitation of insulin as observed after 10 min, and reached the saturation level after 38 min. The non-enzymatic reduction of insulin by DTT, used as a control, started after 20 min and a lower precipitation level of insulin was observed (Fig. 1A).

The effects of the pH on the *Cp*-TrxA1 activity were determined for the insulin reaction and, the results demonstrated that the optimum pH at room temperature was 7.4. Interestingly a second pH optimum was observed at pH 8.4. Constant pH-titration molecular dynamics (CpHMD) were used to investigate the influence of the pH on the relevant amino acids for the protein activity, Cys32, Cys35 and, Trp31. As a parameter the solvent accessible surface $[Å^2]$ of these residues were followed during the MD.

At pHs 7.4 and 8.4, the molecular surface is less accessible than at pHs 7.3 and 7.6, for both cysteines. Our results indicate that the libration of the Trp31 indole ring plays a role in modulating the accessibility of both cysteine (32 and 35) and may be involved in guiding the substrate (Fig. S3, Supplementary material).

MD simulations at pH 7.3 and pH 7.6 suggest that the position of the Trp31indole ring leads to formation of an 'open' active site Fig. S3. In this orientation, Trp31 stabilizes the interaction with the substrate [39] and, also forms hydrogen bonds with Cys32 [40]. The pH induced molecular movement of Trp31 can perturb this effect resulting in decreased protein activity as described above.

3.3. Enzyme specificity and identification of potential inhibitors

The experiments to evaluate the influence of redox agents (DTT, GSH, ESH, CoASH and diamide) and ligands (heparin, suramin, Hsd and Hst) on the enzymatic reduction of insulin by *Cp*-TrxA1 were performed at pH 7.4.

DTT, showed the strongest effect on the protein activity, followed by ESH and GSH. The other tested reduction agents (CoASH and diamide) presented reduced effects on the protein (Fig. 2A).

Besides, DTT, ESH produced the strongest effect on insulin reduction by *Cp*-TrxA1, an event that has not been described previously. ESH is a 2-thiol-L-histidine betaine [12, 41] and possesses antioxidative activity [42].

The tripeptide GSH decreased activity of *Cp*-TrxA1 about 50% when compared with DTT. GSH interacts preferably with Grx proteins and is not the preferred substrate of Trx proteins and, the ability of GSH to reduce *Cp*-TrxA1 indicates the variability for substrate preference of *Cp*-TrxA1.

The oxidation of the *Cp*-TrxA1 thiol active site caused by diamide led to inhibition of the protein.

The polyanions heparin, suramin, and the two flavonoids, Hsp and Hst, where tested for their inhibitory potential against the insulin reducing capacity of *Cp*-TrxA1. The molecular structures of these ligands are presented in Fig. S4. These molecules inhibit manifold proteins [43–48]; however, their interactions with Trx proteins have not been investigated, recently, their effects on the *C. pseudotuberculosis* Grx protein were described [49].

To eliminate the effects of unbound ligands on insulin, an additional gel filtration step was used. The insulin reduction activity of *Cp*-TrxA1 was inhibited (>85%) by all four tested ligands (Fig. 2B). The flavanone glycoside Hsp, and its aglycone form (lacking the disaccharide - rutinose) Hst [37], possesses anticancer, cancer preventive, anti-inflammatory, neuroprotective and antioxidant activities [45–47, 50].

Heparin is a linear, polydisperse polysaccharide, consisting of repeating units of $1 \rightarrow 4$ -linked pyranosyluronic acid and 2-amino-2-deoxyglucopyranose (glucosamine) residues [51]. The negatively charged sulfo or carboxyl groups of heparin interact with positively charged basic amino acids in the proteins [45]. A similar binding mode is observed for suramin, a symmetric divalent molecule possessing two naphthalene-trisulfonic acid heads, with negative charges. Suramin is a well-characterized analogue of heparin and both molecules compete for the same binding region in a number of proteins [44, 52, 53].

During the Trx activity the active site thiols need to be reduced and after this process they are oxidized, the formation of an intramolecular disulfide bridge includes changes in the secondary structure of the protein, visible using CD spectroscopy.

Far-UV CD spectra of *Cp*-TrxA1 at pH 7.4 in phosphate buffer were used to investigate the changes of the secondary structure activated by DTT and GSH (Fig. 2C). When incubated with DTT or GSH, the secondary structure composition of *Cp*-TrxA1 changed. When the protein was incubated with diamide, which oxidizes the thiol at the active site, the secondary structure composition is identical to the untreated *Cp*-TrxA1. The secondary structural changes in the reduced and oxidized *Cp*-TrxA1 homology model are presented in Fig. S5, Supplementary material. ESH and CoASH caused slight modifications of the *Cp*-TrxA1 CD signals, which are divergent, to very significant changes induced by GSH and DTT. The results of the insulin reduction assay indicated a strong influence of ESH on the *Cp*-TrxA1 activity, but since the binding mode is unknown, further investigations are required. The



Fig. 1. Insulin reductase activity by Cp-TrxA1 and determination of the pH optimum. A: The Cp-TrxA1 mediated precipitation of reduced insulin monomers was followed turbidimetrically at 640 nm; the non-enzymatic reduction of insulin by DTT served as the control. B: pH profile of Cp-TrxA1 in the pH range 7.0 to 9.0 using the insulin reductase assay.



Fig. 2. Insulin reductase activity and influence of redox agents and ligands on the insulin reduction by *Cp*-TrxA1. A: Reduction activity of insulin in the presence of DTT, GSH, ESH, CoASH, and diamide. B: Inhibition of *Cp*-TrxA1 activity by different ligands: heparin, suramin, hesperetin, hesperidin (the mean with standard deviation of each measurement is shown in Table S1, Supplementary material). C: Far-UV CD-spectra of *Cp*-TrxA1 incubated with diamide, ESH, DTT, CoASH and GSH. D: Far-UV CD-spectra of *Cp*-TrxA1 treated with heparin, suramin, Hsp or Hst (The CD spectroscopy results were evaluated by utilizing the CDpro software, Table S2, Supplementary material).

inhibitors identified in the insulin reduction assay, were investigated regarding their effect on the secondary structure of the proteins (Fig. 2D). The strongest influence on *Cp*-TrxA1 was observed in the presence of Hst. It has been shown that the interactions of several Trx proteins with inhibitors or ligands result in conformational changes [14, 54, 55]. Suramin and heparin showed similar effects but not so strong like Hst. Interestingly the two tested molecule classes, flavonoids (Hst and Hsp) and polyanions (heparin and suramin) showed each a different effect on the protein secondary structure, which would assume a different mode of interaction. 3.4. Binding studies between Cp-TrxA1 and potential inhibitors using nuclear magnetic resonance (NMR) and MD simulations

STD-NMR experiments of *Cp*-TrxA1 could confirm the interaction of the protein with suramin and Hst. However, experiments involving *Cp*-TrxA1 with heparin and Hsp did not result in observable STD-NMR signals in the difference spectra. For Hsp interaction, it may be due to the affinity cut-off associated with STD-NMR experiments, which requires the ligand to dissociate from the protein receptor to enable successful measurement of the binding event. For heparin, its relatively high



Fig. 3. Illustration of the STD-NMR results for *Cp*-TrxA1 with suramin, molecular docking and MD simulations. A: STD-NMR results for *Cp*-TrxA1 with suramin, NMR spectra, modes of interactions and relative amplification factors (AF) are shown. B: *Cp*-TrxA1 binding position of suramin after molecular docking and MD simulations. C: Effect of suramin on the *Cp*-TrxA1 active site thiols, Trp28 and Trp31. In blue *Cp*-TrxA1 with suramin, in white the protein without the ligand. The arrows mark movements of the Trp residues dependent on the ligand binding.

molecular mass impairs STD from being effective. Protein-ligand complexes with dissociation constants lower than a few micro molars are not directly observed in STD-NMR experiments.

Suramin interacts predominantly with *Cp*-TrxA1 *via* the sulfonic acid groups, as the highest relative STD-AF of 100% was observed for this moiety (Fig. 3A).

Molecular docking of suramin with *Cp*-TrxA1 followed by MD was used to identify the region of interaction between suramin and the protein (Fig. 3B). The interaction energy calculated out of these experiments for the *Cp*-TrxA1 and suramin complex was -32.5 ± 0.3 kcal/mol. Interestingly, the results of molecular docking and MD simulations demonstrated that suramin interacts with the protein in the region which forms the binding surface between the protein and substrate or TrxR (Fig. S6A) [56–58]. Obviously, the interaction of suramin in this protein region interferes with substrate binding and demonstrates a possible mode of inhibition. Furthermore MD results have demonstrated the effect of suramin on Trp31 and the active site cysteine (Fig. 3C). Trp31 interacts thereby directly with suramin as demonstrated by the decomposition energy, which was determined for all residues (Fig. S7A).

Complementary NMR experiments indicated that the interaction with suramin changes the micro-environment of both Trp residues in the protein (Trp28 and Trp31). Indicating the binding of suramin at a nearby site, the tryptophan indole group was used to identify binding of suramin. The ligand is observed to promote significant chemical shift change in the area around 10 ppm (Fig. S8). As described earlier, Trp31 is important for protein activity, the interaction of suramin with this residue can affect protein catalysis.

Evaluating the Hst binding epitopes is more challenging since hydrogen signals are superposed in the 1D spectra. However, the highest relative STD-AF of 100% is determined for the hydrogens of the aromatic ring system in the Hst molecule (Fig. 4A).

Molecular docking and MD simulations indicates the interaction of Hst with *Cp*-TrxA1 near the protein active site (Fig. 4B), mainly with α -helix three, thereby the calculated interaction energy was -14.8 ± 0.1 kcal/mol. Compared with the energy calculated for the *Cp*-TrxA1-suramin complex the value is around two times lower. Suramin is a strong negatively charged molecule with a net charge of -6, Hst has a net charge of zero, these differences in the charges and the resulting ionic interactions with positively charged amino acids of the protein can explain the difference in the interaction strength.

Hst interacts with α -helix, which is involved in the Trx protein substrate binding surface [56–58] (Fig. S6B). Moreover, compared with the suramin position, the Hst location is different and appears does not directly restrict access to the active site, but MD results indicated an effect on Trp31 (Fig. 4C), which is indicated additionally by the decomposition energy for the interaction between Trp31 and Hst (Fig. S7B). The effect on *Cp*-TrxA1 Trp28 and 31 could also be observed by additionally NMR experiments, which demonstrated the effect on the micro environment of the Trp residues similar as described before for the suramin interaction (Fig. S8).



Fig. 4. Illustration of the STD-NMR results for *Cp*-TrxA1 with Hst, molecular docking and MD simulations. A: STD-NMR results for *Cp*-TrxA1 with Hst, NMR spectra, modes of interactions and relative amplification factors (AF) are shown. B: *Cp*-TrxA1 binding position of Hst after molecular docking and MD simulations. C: Effect of Hst on the *Cp*-TrxA1 active site thiols, Trp28 and Trp31. In blue *Cp*-TrxA1 with Hst, in white the protein without the ligand. The arrows mark movements of the Trp residues dependent on the ligand binding.

The binding of suramin and Hst with *Cp*-TrxA1 influence molecular movements in the protein structure effecting the secondary structure content. It is demonstrated by MD simulations of the protein complex with the ligands and the CD results as described before. The β -sheets content in the protein core remain unaffected, however, after ligand binding changes in the α -helix content can be observed (Table S2, Supplementary material).

4. Conclusion

Trx proteins play crucial roles in the regulation of cellular redox homeostasis in pathogens, which is essential for their survival in situations of elevated environmental stress. The pathogens are in close contact with many molecules, which possess a strong influence on the cellular redox homeostasis. The high degree of conservation of the Trx active sites, and their similar three-dimensional folds indicate that inhibition of these proteins is likely to play an important role in pathogen viability.

The combination of activity assays, CD spectroscopy, NMR and MD simulations of the *Cp*-TrxA1 Hst and suramin complexes provides interesting insights regarding these interactions. Both molecules inhibit protein activity by >80%, but the localizations and binding modes of these inhibitors are different. The inhibitors belong to different molecular classes; suramin is a polyanion whereas Hst is a flavonoid. Both molecules display significant differences in their electro physical behaviors. Our results demonstrate that polyanions and flavonoids can influence the activity of *Cp*-TrxA1 and depending on the different binding modes as indicated by MD simulations, the Trx inhibitor does not interact directly with the active site cysteins. Restricting access to the substrate binding surface or Trp31, the active site plays a crucial role in determining the protein activity. These results can support the search and development of specific inhibitors for *Cp*-TrxA1 as a means to combat *C. pseudotuberculosis* infections.

Conflict of interest

Authors declare that there exists no conflict of interest.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi. org/10.1016/j.ijbiomac.2018.06.022.

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