



## Enzymatic modification of grapeseed (*Vitis vinifera* L.) oil aiming to obtain dietary triacylglycerols in a batch reactor

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### ABSTRACT

*Structured lipids* (SL) are chemically or enzymatically modified oils and fats with respect to their original fatty acid composition or position in acylglycerols. These compounds present improved functional or nutraceutical properties. The present work aimed at the enzymatic synthesis of SL, MLM-type dietary triacylglycerols, that is, those with medium chain fatty acids (M) at the *sn*-1 and *sn*-3 positions, and long chain fatty acids (L), in the internal position of the triacylglycerol. Grapeseed oil was selected based on its composition rich in unsaturated fatty acids, principally linoleic acid. This oil was submitted to batch acidolysis with medium chain fatty acids (caprylic or capric) in solvent-free media. Reactions were catalyzed by different immobilized commercial lipases, namely: Lipozyme TL IM<sup>®</sup> (*Thermomyces lanuginosa* lipase), Lipozyme RM IM<sup>®</sup> (*Rhizomucor miehei* lipase) and Novozym 435<sup>®</sup> (*Candida antarctica* lipase B). The incorporation degree (ID) ranged from  $23.62 \pm 1.34$  to  $34.53 \pm 0.05$  mol%, after 24 h reaction at 45 °C, using a molar ratio (MR) fatty acid:oil of 2:1. The best results were obtained using capric acid and Lipozyme RM IM<sup>®</sup> lipase ( $34.53 \pm 0.05$  mol%). In the experimental design, the influence of MR and temperature on ID were evaluated. ID increased with MR and T and was fitted to a saddle-like surface.

### 1. Introduction

Modified oils and fats with improved functional and/or nutritional properties have received special attention, due to the importance of the diet on the promotion and maintenance of health. These novel fats and oils, known as *Structured lipids* (SL), are usually chemically or enzymatically synthesized by modification of triacylglycerols with respect to their original fatty acid composition or fatty acid position in glycerol backbone (Cao et al., 2013; Liu et al., 2015).

Within Structured lipids, the class of dietary triacylglycerols (TAGs) is highlighted, which are synthesized for specific nutritional applications (Cao et al., 2013). This class of lipids has lower caloric value (5–7 kcal/g) than conventional oils and fats (9 kcal/g). This is because they have a lower amount of long chain fatty acids at the *sn*-1 and *sn*-3 positions, which consequently increases their solubility. During digestion, the medium chain fatty acids, released from TAG, are metabolized

as rapidly as glucose and, therefore, they are not stored as fat, with weight control benefit (Ferreira-Dias, Sandoval, Plou, & Valero, 2013; Morales-Medina, Munio, Guadix, & Guadix, 2017). The interest in these TAGs has recently increased due to its main nutritional characteristic. They can be administered orally, through probes or mixed with other foods, in the treatment of patients who present metabolic difficulties related to diseases such as AIDS, cystic fibrosis and cirrhosis. They also are used in severely malnourished patients (Liu et al., 2015).

The enzymatic synthesis of Structured lipids is normally catalyzed by regio-selective *sn*-1,3 lipases, which allows synthesizing compounds that are not possible to obtain chemically because chemical catalysts are non-regioselective, acting at random in TAGs (Li et al., 2017; Rehman, Wang, Bhatti, Bilal, & Asgher, 2017).

The aim of this work was to synthesize dietary triacylglycerols of the MLM-type (medium: long: medium), triacylglycerols containing medium chain fatty acids (M) at the *sn*-1 and *sn*-3 positions and long

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chain fatty acids (L) in position *sn*-2. In this context, grapeseed oil was selected as source of polyunsaturated long-chain fatty acids at position *sn*-2 in TAGs, due to its high nutritional value. It is also derived from waste generated from the production of juice and wine (Da Porto & Natolino, 2017). Grapeseed oil has a high content of unsaturated fatty acids, such as essential linoleic acid (58–78%), which is higher than in other oils such as corn (34–65.6%) and soybean (48–59%) oils (FAO, 1999). In addition, it has a high amount of antioxidant compounds, such as tocopherols (vitamin E) and tocotrienols (Raeisi, Tajik, Aliakbarlu, Mirhosseini, & Hosseini, 2015). The use of grapeseed oil for producing MLM-type TAGs by acidolysis with caprylic or capric acids, in solvent-free media, catalyzed by non-commercial *sn*-1,3 regioselective recombinant *Rhizopus oryzae* lipase and *Carica papaya* lipase self-immobilized in papaya latex was recently reported (Costa et al., 2018).

The objective of this work was to optimize the enzymatic synthesis of dietary triacylglycerols of the MLM-type by batch acidolysis of grapeseed oil with medium chain fatty acids, caprylic (C8: 0) or capric (C10: 0) in solvent-free media, employing commercial immobilized lipases as catalysts.

## 2. Material and methods

### 2.1. Material

#### 2.1.1. Biocatalysts

The commercially immobilized enzymes used as biocatalysts were: Lipozyme TL IM<sup>®</sup> (*Thermomyces lanuginosa* lipase), Lipozyme RM IM<sup>®</sup> (*Rhizomucor miehei* lipase) and Novozym 435<sup>®</sup> (*Candida antarctica* lipase B), kindly provided by Novozymes A/S (Bagsvaerd, DENMARK). Porcine pancreatic lipase was purchased from Sigma-Aldrich.

#### 2.1.2. Raw material

Grapeseed (*Vitis vinifera* L.) oil was obtained by cold pressing and filtration from Distriol (São Paulo, BRAZIL), Capric acid was purchased from Merck (Darmstadt, GERMANY), and caprylic acid from Sigma Aldrich (São Paulo, BRAZIL).

### 2.2. Methods

#### 2.2.1. Characterization of grapeseed oil

The acid value (Ca 5a-40 method) and peroxide value (Method Cd 8b-90) were determined according to the official methods of the American Oil Chemists' Society (AOCS, 2011). Fatty acid composition was determined by gas chromatography, according to the American Oil Chemists' Society Method Ce 2-66 (AOCS, 2011).

#### 2.2.2. Determination hydrolytic and esterification activities of the biocatalysts

The hydrolytic activity of the biocatalysts was assayed by the titration method using an olive oil emulsion (Paula, 2012; modified). The esterification activity was quantified by the consumption of oleic acid according to the method described by Pinto, Freire, and Pinto (2014). One unit of activity (U) was defined as the amount of enzyme that released 1 μmol of fatty acid per minute under the assay conditions.

#### 2.2.3. Acidolysis reaction for the synthesis of dietary triacylglycerols (MLM) from grapeseed oil

Acidolysis reactions between the vegetable oil (grapeseed) and the medium-chain fatty acids (caprylic or capric) were carried out in a jacketed cylindrical glass reactor (internal diameter of 3 cm, height of 6 cm), operating in a batch regime. The reactions occurred at a temperature of 45 °C, with constant agitation of 400 rpm, molar ratio (MR) fatty acid:oil of 2:1 and enzyme load of 5% (relative to the total mass of medium). The MR used in these experiments corresponds to the

stoichiometric value for acidolysis catalyzed by *sn*-1,3 regioselective lipases. Samples were collected at 0 time (oil without modification) and after 24 h reaction, and analyzed with regard to acidity, peroxide value, fatty acid composition and fatty acid composition at the *sn*-2 position.

#### 2.2.4. Modeling acidolysis reaction

After the selection of the best reaction system (oil: acid and lipase), the influence of the molar ratio (MR) oil:capric acid and temperature (T) was evaluated in the acidolysis reaction, aiming to model acidolysis and optimize the reaction conditions that maximize the incorporation of capric acid. A Central Composite Rotatable Design (CCRD), consisting of a factorial matrix 2<sup>2</sup>, (levels (+1) and (−1)), complemented with four tests of the axial points and the repetition of the central point, was used. The experimental conditions tested (temperature: 41–69 °C; MR: 1:1 to 7:1) were selected from similar works on acidolysis reactions aimed at MLM production in solvent-free media, catalyzed by the same immobilized enzymes (Caballero, Soto, Olivares, & Altamirano, 2014; Nunes, Pires-Cabral, & Ferreira-Dias, 2011).

The incorporation degree of capric acid into grapeseed oil was assumed as the response. The experimental results were analyzed using the Statistica program, Statsoft, Tulsa, version 6. The polynomial model adjusted to the experimental results was validated by performing independent tests with different values of temperature and molar ratio within the experimental range considered in the study.

#### 2.2.5. Separation and methylation of triacylglycerols

After 24 h reaction, 1.5 g of the sample were weighed. Then, 30 mL of hexane and 10 mL of the hydroalcoholic solution (30% ethanol) of potassium hydroxide (0.8 M) were added. The mixture was kept under stirring and then allowed to stand for phase separation. The organic (upper) phase was collected, repeating the triacylglycerol extraction with hexane (10 mL) again. The solvent was extracted in a rotavaporator (P = 130 mbar and T = 40 °C) and the TAGs were stored at 8 °C until further analysis (Wang, Xia, Xu, & Xie, 2012). Methylation was performed according to ISO 12966-2 (2011), after modification. Approximately, 50 mg of the purified triacylglycerols were weighed and 2 mL of a sodium hydroxide methanolic solution (0.2 M) was added and the mixture was heated at 80 °C/15min under vigorous stirring. Then, 2 mL of a methanolic solution of sulfuric acid (1 M) was added, repeating the steps of heating and stirring (once time). One mL of a saturated NaCl solution and 2 mL of hexane were added. The solution was vortexed and then kept at rest for phase separation. The upper (organic) phase was removed and the extraction procedure with hexane (2 mL) was repeated. The hexane was removed at 40 °C in the rotavapor and the fatty acids methyl esters were stored at 6 °C for further analysis.

#### 2.2.6. Fatty acid composition of structured lipids and incorporation degree (ID, %)

The fatty acid composition of the structured lipids obtained by acidolysis of grapeseed oil was determined by gas chromatography, according to the American Oil Chemists' Society Method Ce 2-66 method (AOCS, 2011), as reported for fatty acid composition of grapeseed oil (cf. 2.2.1.).

The incorporation degree (ID) was calculated according to equation (1) (Casas-Godoy, Marty, Sandoval, & Ferreira-Dias, 2013).

$$ID(\%) = \frac{MFA}{MT} * 100. \quad (1)$$

Where: MFA is the number of moles of medium chain fatty acids (C8:0 or C10:0) in TAGs and MT is the number of total moles of fatty acids in TAGs.

#### 2.2.7. Fatty acid at the *sn*-2 position in triacylglycerols

The fatty acid composition at position *sn*-2 of the TAGs was analyzed following the porcine pancreatic lipase (PPL) hydrolysis

procedure according to Caballero et al. (2014), modified. One mL of Tris-HCL buffer (pH 8.0), 0.250 mL of a solution of bile salts (0.05%, w/v), 0.1 mL of a calcium chloride solution (22%), the TAGs (100 mg) and the porcine pancreatic lipase (100 mg) were mixed. The reaction occurred at 40 °C for 30 min. The hydrolysis was stopped with the addition of 1 mL hydrochloric acid (6M) and 2 mL of diethyl ether. After, the blend was centrifuged (5000 rpm/10 min) and the upper layer was collected and applied to a silica-gel thin layer chromatography (TLC) plate. The TLC plate was put into a chromatographic chamber containing the mobile phase, consisting of hexane, ethyl ether and acetic acid (70/30/1; v/v/v). After elution, the plate was visualized by spraying with an alcoholic solution of the dye 2,7 dichlorofluorescein (0.2%) under u.v. light at 366 nm. The *sn*-2 monoacylglycerol band was scrapped off, methylated and analyzed by gas chromatography (c.f. 2.2.1.).

### 3. Results and discussion

#### 3.1. Characterization of grape seed oil

##### 3.1.1. Physico-chemical characterization of oil

The grapeseed oil used in this study has an acid value of  $0.225 \pm 0.021$  mg KOH/g of oil and a peroxide value of  $0.905 \pm 0.148$  mEq O<sub>2</sub>/kg of oil. These results showed that the raw material is within the specified standards (FAO, 1999): 0.6 mg KOH/g of oil for the acid value and 10 meq O<sub>2</sub>/kg of oil for the peroxide value.

##### 3.1.2. Fatty acid profile

Table 1 shows the results of the fatty acid composition of the grapeseed oil. The oil has high amounts of unsaturated fatty acids (c.a. 78%). The essential linoleic acid (C18:2) is the major fatty acid (46.2%) in our sample, followed by oleic acid (C18:1; 26.6%). Palmitic (C16:0, 11.1%) and stearic acids (C18:0; 6.72%) are also important saturated fatty acids in grape seed oil. The highly polyunsaturated fatty acids C18:3 and of C20:3, which are rather prone to oxidation, are present in relative low amounts (3.46 and 2.11%, respectively).

Due to the presence of high amounts of these fatty acids, the use of grapeseed oil has been gaining prominence and has been applied in different industrial sectors, such as in the food and cosmetic industries (Glampedaki & Dutschk, 2014; Durante et al., 2017).

The intake of oils rich in these fatty acids brings great benefits to human health, ie the consumption of oils with large amounts of oleic acid is associated with the production of oleoylethanolamide, a compound that promotes catabolism and helps satiety in food intake (Bowen et al., 2017). In addition, the consumption of this fatty acid helps in the prevention of cardiovascular diseases, decrease of LDL (low density proteins), suppression of tumorigenesis and improvement of inflammatory diseases (Dhakal, Jung, Chae, Shannon, & Lee, 2014).

On the other hand, the consumption of essential linoleic acid is related to the reduction of carcinogenesis, atherosclerosis, diabetes, as well as promotion and aid in bone formation and also acts as an anti-

**Table 1**  
Fatty acid composition of grapeseed oil.

Fatty acid	(%)
C16:0	11.10
C18:0	6.72
C18:1	26.57
C18:2	46.25
C18:3	3.46
C20:3	2.11
C22:0	1.16
U <sup>a</sup>	2.63

<sup>a</sup> U = unidentified fatty acid.

**Table 2**

Hydrolytic and esterification activities and moisture content of Lipozyme RM IM<sup>®</sup>, Lipozyme TL IM<sup>®</sup> and Novozym 435<sup>®</sup>.

Parameter	Lipozyme RM IM	Lipozyme TL IM	Novozym 435
Hydrolytic Activity (μmol/gmin) <sup>a</sup> (U/g) <sup>a</sup>	568 ± 42.02	5678 ± 262.99	–
Esterification Activity (μmol/gmin) <sup>a</sup>	5543	–	6286
Moisture (% , d.w)	5.11	7.86	6.86

<sup>a</sup> Mean ± standard deviation.

inflammatory agent (Viladomiu, Hontecillas, & Bassaganya-Riera, 2016; Yang et al., 2015).

Some studies report that the amounts of oleic acid in grapeseed oil can range from 10 to 22% and linoleic acid from 60 to 75% (Irandoust, Ebrahimi-Mameghani, & Pirouzpanah, 2013; Mironeasa, Leahu, Codinã, Stroe, & Mironeasa, 2010; Rombaut et al., 2014). According to the results presented in Table 1, the grapeseed oil used in the present study presents oleic and linoleic acid contents within the variation range indicated in the Codex Alimentarius (Codex, 1999). Differences in fatty acid contents reported in literature, are justified by several factors, such as grape variety, edapho-climatic conditions and agronomic practices used (Duba & Fiori, 2015; Fiori et al., 2014).

#### 3.2. Hydrolytic and esterification activities of the biocatalysts

The hydrolytic and esterification activity data of the biocatalysts, as well as their moisture content are presented in Table 2. Novozym 435<sup>®</sup> only presents esterification activity in contrast with Lipozyme TL IM<sup>®</sup>, which only shows hydrolytic activity. However, Lipozyme RM IM presents both lipolytic and esterification activities. However, the esterification activity of this enzyme is almost 10-fold its hydrolytic activity. The moisture content of these biocatalysts are not very different and therefore do not explain their different behavior towards hydrolysis.

#### 3.3. Acidolysis reactions of grapeseed oil

##### 3.3.1. Incorporation degree of capric and caprylic acids

Six reactions of acidolysis of the grapeseed oil were conducted at 45 °C for 24 h, as previously described (c.f. 2.2.3.). The molar ratio fatty acid/oil of 2:1 corresponds to the stoichiometric value for acidolysis of TAGs when *sn*-1,3 regioselective lipases are used. Table 3 shows the fatty acid profiles of the TAGs obtained by acidolysis of the grapeseed oil with the caprylic or capric acids. Comparing the fatty acid composition of the crude oil (Table 1) with that of the modified TAGs (Table 3), a decrease in the percentage of palmitic (C16:0), stearic (C18:0), oleic (C18:1) and linoleic (C18:2) acids is observed. This decrease is related to the incorporation of the medium chain acids (C8:0 or C10:0), which did not exist in the raw material. Concerning the other fatty acids, little or no variation was observed.

From the composition in fatty acids, the incorporation degree (ID) of capric and caprylic acids in grapeseed oil was calculated. The results are shown in Fig. 1.

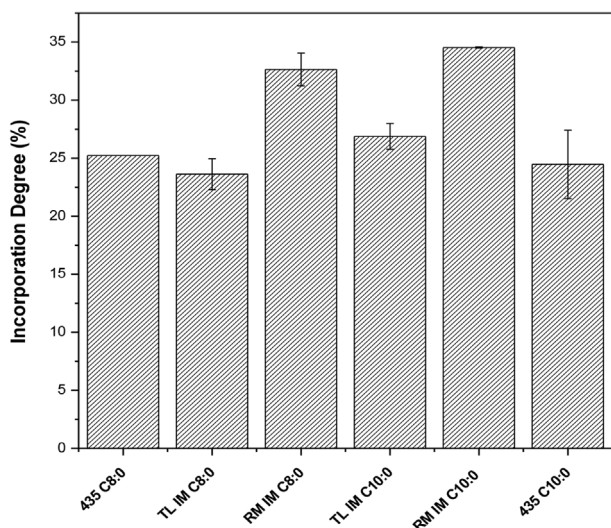
ID values for grapeseed oil ranged from  $23.62 \pm 1.34$  to  $34.53 \pm 0.05$  mol%. For all the biocatalysts tested, higher incorporations were observed for capric acid. The best results were obtained using C10:0 as medium chain fatty acid and Lipozyme RM IM<sup>®</sup> lipase as catalyst. Nunes et al. (2011), using Lipozyme RM IM<sup>®</sup> as a catalyst for the acidolysis of olive oil and capric acid, also in solvent-free medium, obtained an incorporation degree of 27.1 mol%. With Lipozyme TL IM<sup>®</sup> and Novozym 435<sup>®</sup>, the results of the present study are close to those described by Nunes et al. (2011): 28.8 mol% and 30.4 mol% of

**Table 3**

Fatty acid composition (%) of modified grapeseed TAGs with caprylic (C8:0) or capric acid (C10:0), by acidolysis catalyzed by Novozym 435<sup>®</sup>, Lipozyme RM IM<sup>®</sup> or Lipozyme TL IM<sup>®</sup>.

Fatty acid (%)	Samples					
	C8:0			C10:0		
	Novozym 435	Lipozyme TL IM	Lipozyme RM IM	Lipozyme TL IM	Lipozyme RM IM	Novozym 435
C6:0	0.22 ± 0.00	–	–	–	–	–
C8:0	15.55 ± 0.34	13.90 ± 0.74	20.05 ± 1.07	0.08 ± 0.00	0.01 ± 0.00	–
C10:0	0.05 ± 0.00	0.03 ± 0.00	–	18.49 ± 0.80	24.61 ± 0.03	16.65 ± 0.90
C14:0	–	–	0.02 ± 0.01	0.03 ± 0.00	0.02 ± 0.00	–
C16:0	8.80 ± 0.15	8.97 ± 0.06	5.93 ± 0.04	8.03 ± 0.00	5.55 ± 0.14	7.97 ± 0.13
C16:1	0.05 ± 0.00	0.02 ± 0.00	0.05 ± 0.00	0.04 ± 0.00	0.05 ± 0.00	0.05 ± 0.00
C17:0	–	–	0.04 ± 0.00	0.04 ± 0.00	0.03 ± 0.00	0.04 ± 0.00
C17:1	0.04 ± 0.00	–	0.02 ± 0.00	0.02 ± 0.00	0.02 ± 0.00	–
C18:0	3.16 ± 0.05	3.19 ± 0.06	2.10 ± 0.30	2.90 ± 0.08	1.95 ± 0.00	2.84 ± 0.05
C18:1	21.27 ± 0.37	20.97 ± 0.98	19.30 ± 0.18	19.70 ± 0.25	18.42 ± 0.01	19.79 ± 1.01
C18:2	48.37 ± 0.81	47.30 ± 1.35	46.90 ± 0.92	44.45 ± 0.49	44.16 ± 0.44	46.17 ± 0.89
C18:3	5.14 ± 0.09	5.17 ± 0.06	4.73 ± 0.05	4.83 ± 0.18	4.44 ± 0.00	4.98 ± 0.25
C18:3n6	0.22 ± 0.00	0.021 ± 0.00	0.20 ± 0.00	0.20 ± 0.00	0.19 ± 0.01	0.21 ± 0.00
C20:0	0.23 ± 0.00	0.25 ± 0.01	0.19 ± 0.00	0.23 ± 0.02	0.18 ± 0.00	0.23 ± 0.01
C20:1	0.27 ± 0.00	0.31 ± 0.03	0.31 ± 0.01	0.30 ± 0.04	0.29 ± 0.013	0.33 ± 0.02
C21:0	–	–	0.01 ± 0.00	0.02 ± 0.00	0.01 ± 0.00	–
C20:2	–	0.02 ± 0.00	–	–	–	–
C20:4	0.13 ± 0.02	–	–	–	–	–
C22:0	0.27 ± 0.02	0.32 ± 0.02	0.25 ± 0.00	0.29 ± 0.03	0.23 ± 0.01	0.29 ± 0.01
C24:0	0.13 ± 0.02	–	–	–	–	–

<sup>a</sup>Mean ± standard deviation.



**Fig. 1.** Incorporation Degree of capric and caprylic acids in grape seed oil.

incorporation, respectively. The acidolysis of grapeseed oil with caprylic or capric acid was also performed in solvent-free media, using non-commercial *sn*-1,3 regioselective lipases namely: the recombinant lipase from *Rhizopus oryzae* immobilized in Amberlite™ IRA 96 and *Carica papaya* lipase self-immobilized in papaya latex (Costa et al., 2018). Yields of new TAGs were between 38 and 69 mol%, after 48 h acidolysis at 40 °C, molar ratio fatty acid:oil of 2:1.

In our work, Lipozyme RM IM<sup>®</sup> provided a better incorporation degree (34.53 ± 0.05 mol%) because, conversely to the other biocatalysts tested (Novozym 435<sup>®</sup> and Lipozyme RM IM<sup>®</sup>), it has both high hydrolytic and esterification activities (Table 2). In fact, acidolysis reaction has been considered as a two-step mechanism reversible reaction of hydrolysis-esterification (Xu, 2003). However, for some lipases, no relationship has been observed between the acidolysis and the hydrolytic activities. This was observed for the recombinant lipase of *Rhizopus oryzae* immobilized in Eupergit C and for *Carica papaya* lipase

self-immobilized in the latex, when used as catalysts for the acidolysis reaction aimed at the production of low calorie TAGs or Human Milk Fat Substitutes, respectively (Nunes, Pires-Cabral, Guillén, Valero, & Ferreira-Dias, 2012; Tecelão, Rivera, Sandoval, & Ferreira-Dias, 2012).

Moreover, the higher incorporation of capric acid (C10:0) with Lipozyme RM IM<sup>®</sup> may be due to the longer carbon chain of this acid when compared to caprylic acid (C8:0). The specificity of this lipase is directly related to the size of the fatty acid chain, and the enzyme has a preference for fatty acids with longer carbon chain (Rodrigues & Fernandez-Lafuente, 2010a; Rodrigues & Fernandez-Lafuente, 2010b; Caballero et al., 2014). On the contrary, the recombinant lipase of *Rhizopus oryzae* immobilized in Amberlite IRA 96 showed a preference towards caprylic acid while *Carica papaya* lipase self-immobilized in papaya latex showed no preference towards caprylic or capric acid (Costa et al., 2018).

### 3.3.2. Fatty acid at the *sn*-2 position in triacylglycerols

Considering the best result of incorporation degree in the acidolysis reaction of grapeseed oil with C10:0, catalyzed by Lipozyme RM IM<sup>®</sup>, the modified product obtained was evaluated for the fatty acids present at the *sn*-2 position of triacylglycerol.

**Table 4**

Fatty acid profile at the *sn*-2 position of the grapeseed oil and the modified oil, after 24-h acidolysis catalyzed by Lipozyme RM IM<sup>®</sup>.

Fatty acid	Grapeseed oil	Modified grapeseed oil
	%	%
C10:0	0	3.36
C16:0	2.89	2.19
C18:0	0.90	0.89
C18:1	28.15	24.33
C18:2	63.23	63.56
C18:3	4.83	5.03
U <sup>a</sup>	–	0.64
Total	100.00	100.00

<sup>a</sup> U = unidentified.

**Table 5**  
Central Composite Rotatable Design (CCRD) used in modeling acidolysis and optimization of reaction conditions as a function of Temperature (T) and acid/oil molar ratio (MR) (coded and decoded factors).

Experiment	Stage	Coded factors		Decoded factors		Response
		T	RM	T (°C)	MR (acid:oil)	
1	Factorial points	−1	−1	45	2:1	35.91
2		−1	+1	45	6:1	42.79
3		+1	−1	65	2:1	34.40
4		+1	+1	65	6:1	54.36
5	Axial points	−1.41	0	41	4:1	41.70
6		+1.41	0	69	4:1	47.26
7		0	−1.41	55	1:1	21.29
8	Central point	0	+1.41	55	7:1	48.45
9		0	0	55	4:1	38.60
10		0	0	55	4:1	43.72
11		0	0	55	4:1	41.09

Table 4 shows the results obtained for the determination of fatty acid composition at the *sn*-2 position of the original grapeseed oil and the oil after modification by acidolysis with capric acid, catalyzed by Lipozyme RM IM<sup>®</sup>.

It can be seen (Table 4) that the *sn*-2 position of triacylglycerols in grapeseed oil mainly contains long chain unsaturated fatty acids (96.2%). After acidolysis, the compositional fatty acid profile at this position was practically unchanged. In fact, only 3.36% of capric acid was found at position *sn*-2, which may be related to the acyl migration of this fatty acid. This result was expected, since a regioselective *sn*-1,3 lipase was employed. The obtained results show that Lipozyme RM IM<sup>®</sup> is an adequate biocatalyst for the synthesis of TAGs with the MLM configuration by acidolysis of grapeseed oil.

### 3.3.3. Modeling acidolysis reaction

From the CCRD, it was possible to obtain the incorporation degree for 5 different levels for temperature and molar ratio (MR). Table 5 presents the experimental matrix and the incorporation degree of capric acid in grapeseed oil achieved in each experiment after 24-h acidolysis in solvent-free media, catalyzed by Lipozyme RM IM<sup>®</sup>.

From the obtained results, the main effects and the interaction of the variables temperature and molar ratio on the incorporation degree of capric acid in grapeseed oil were calculated (Table 6).

As can be observed, the molar ratio showed a highly significant positive linear effect ( $p < 0.001$ ) on the ID of capric acid into grapeseed oil. The positive linear effect of temperature, the positive quadratic effects of both T and MR and the linear interaction effect showed  $p$ -values higher than 0.05. However, these effects are important enough to be considered. According to Haaland (1989), it is better to accept factors with effects with  $p$  values higher than 0.05, rather than to ignore an important factor. This could lead to a considerable decrease in both  $R^2$  and  $R^2_{adj}$  of the model fitted to the experimental points.

**Table 6**

Linear (L) and quadratic (Q) main effects of temperature (T) and molar ratio (MR), and the interaction T x MR that influenced the incorporation degree of capric acid in grape seed oil, obtained by acidolysis catalyzed by Lipozyme RM IM<sup>®</sup>.

Factor	Effect	p
T (L)	4.48	0.0752
T (Q)	4.44	0.1214
MR (L)	16.31	0.0005
MR (Q)	−5.17	0.08201
T x MR	6.54	0.0688

The positive linear effects of temperature and MR on ID indicate that the incorporation degree increases with the values of these factors. The quadratic effect is positive for T and negative for MR. This indicates that the ID is described by a convex surface, as a function of temperature, and by a concave surface, as a function of the molar ratio. The interaction T x MR is positive indicating that an increase in both factors promotes an increase in ID. Fig. 2 shows the three-dimensional response surface, and respective projection, as a function of T and MR, fitted to the experimental results. As can be seen, it is a saddle-like surface that does not present any optimum points in the experimental region. However, it can be observed that the higher ID of capric acid are observed for temperatures higher than 60–65 °C and higher molar ratios (higher than 4:1).

Also, in acidolysis optimization, maximum incorporation levels of caprylic acid in sesame oil and in corn oil were expected with a large excess of FFA (MR of 8:1 and 3.9:1 respectively) when Lipozyme RM IM and Lipozyme TL IM were used as catalysts (Kim & Akoh, 2005; Öztürk, Ustun, & Aksoy, 2010). In addition, the temperature and reaction time positively affected caprylic acid incorporation (Kim & Akoh, 2005). Moreover, best results for the incorporation of capric acid in tripalmitin, catalyzed by Lipozyme TL IM were predicted for a molar ratio of 6.5:1 (Bektas, Sevil, Ustun, & Aksoy, 2008).

In fact, higher molar ratios FFA:oil promote acidolysis by shifting the chemical equilibrium towards product formation. However, for some biocatalysts, an inactivation effect due to the presence of large amounts of free fatty acids may also occur (Nunes, Pires-Cabral, Guillén, Valero & Ferreira-Dias, 2012; Tecelão et al., 2012).

At industrial-scale, it is important to notice that the use of large amounts of free fatty acids as substrate will increase operation costs in purification process. The unreacted medium-chain fatty acids, which are mixed with the final product, will be recovered by distillation. Increasing the MR also implies an increase in the cost of substrates.

The response surface for the ID (%) of capric acid in grape seed oil after 24 h acidolysis, catalyzed by Lipozyme RM IM<sup>®</sup>, can be described by the following second-order polynomial model (equation (2)):

$$ID = 105.26 - 2.87 T + 0.022 T^2 + 0.257 RM - 0.646 RM^2 + 0.164 (T \times RM) \quad (2)$$

The values of the coefficient of determination ( $R^2$ ) and the adjusted coefficient of determination ( $R^2_{adj}$ ) are 0.946 and 0.893, respectively. This shows a high fit of the models to the experimental results.

After obtaining the statistical model, an experimental test was performed using the reaction conditions that maximized the ID (RM = 7:1, T = 69 °C). The ID predicted by the model (equation (2)) was 61.70 mol%, and the experimentally obtained was 61.50 mol%. Therefore, this result proves that the model obtained in the present work adequately describes the effect of temperature and molar ratio on the incorporation degree of capric acid in grapeseed oil by acidolysis reaction catalyzed by Lipozyme RM IM<sup>®</sup>.

## 4. Conclusions

From the results obtained, it is possible to conclude that grapeseed oil is an adequate raw material to be used in acidolysis reactions with medium chain acids (C8:0 or C10:0) aimed at producing TAGs of the MLM type rich in linoleic acid, using different commercial biocatalysts (Lipozyme RM IM<sup>®</sup>, Lipozyme TL IM<sup>®</sup> and Novozym 435<sup>®</sup>). The best incorporation degree of the medium chain fatty acids was achieved using Lipozyme RM IM<sup>®</sup> and C10:0 (34.53 ± 0.05 mol%) after 24 h-reaction. In addition, with the use of response surface methodology, it was possible to evaluate the influence of molar ratio and temperature on the incorporation degree. The experimental results could be well fitted to a saddle-like surface, where the ID increased with MR and temperature.

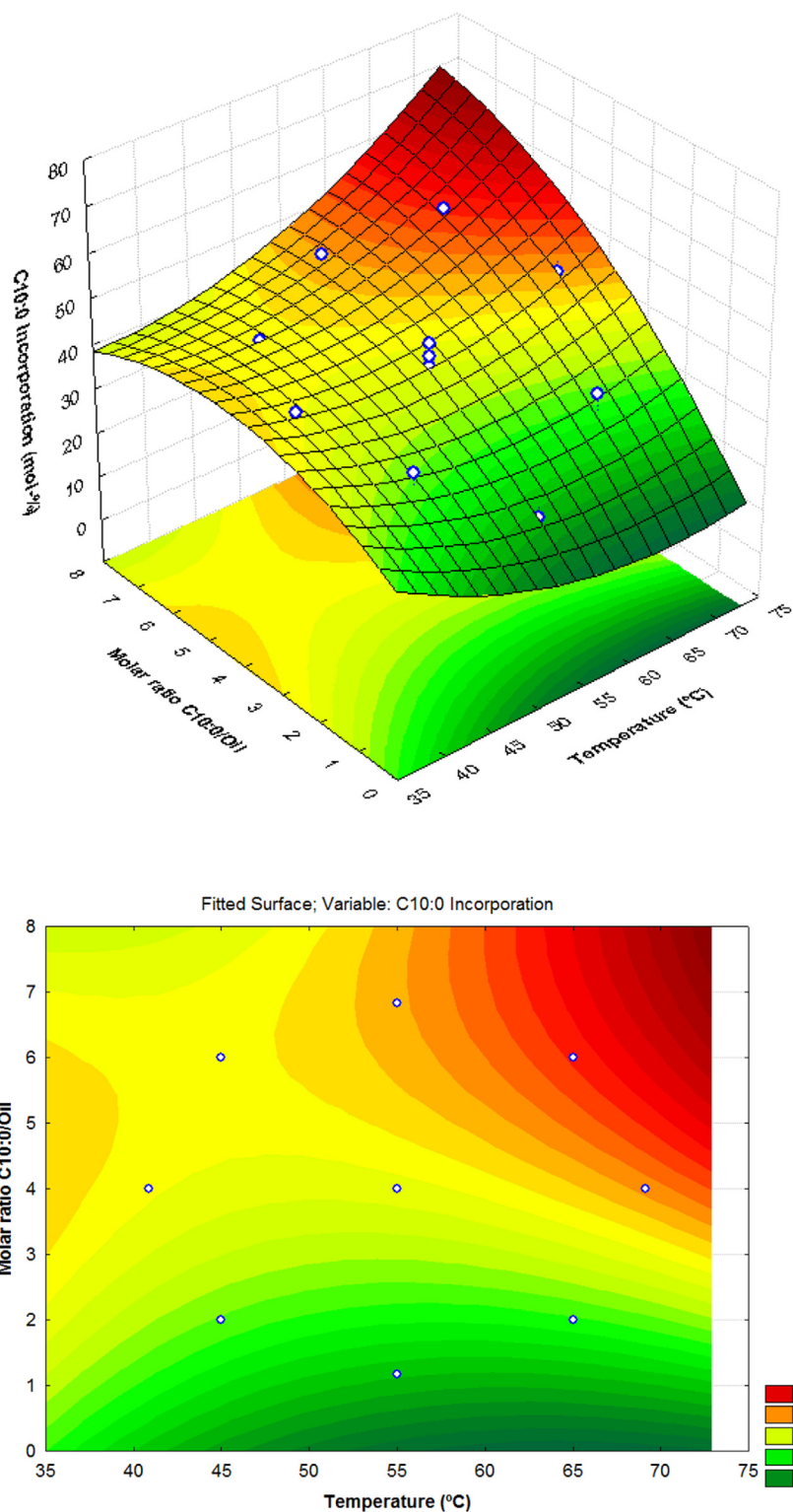


Fig. 2. Response surface (A) and respective projection (B) adjusted for degree of incorporation of capric acid in grapeseed oil, obtained by acidolysis catalyzed by Lipozyme RM IM.

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