



## Influence of pasteurization on antioxidant and *in vitro* anti-proliferative effects of jambolan (*Syzygium cumini* (L.) Skeels) fruit pulp



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### ABSTRACT

Here we analyze jambolan pulp phenolic compounds in order to establish a correlation with antioxidant and *in vitro* anti-proliferative effects, both before and after pasteurization. Total levels of phenolic compounds, flavonoids and anthocyanins were quantified using UV–vis techniques. Major phenolic compounds were identified by standard compound co-injection in HPLC–DAD/UV–vis. Antioxidant activity was measured by radical scavenging ability, as determined by DPPH assay. *In vitro* anti-proliferative activity was determined against nine human tumour cell lines using the methodology described by the Developmental Therapeutics Program at NCI/NIH. Pasteurization led to an increase in the levels of total soluble solids (6.7%), phenolic compounds (7.2%) and flavonoids (16.4%). Anthocyanin content was largely preserved (91%) when compared with pulp without treatment. *S. cumini* preserved 56% of its original antioxidant activity after pasteurization, while thermal treatment revealed cytostatic activity in kidney (786-0) and ovary (OVCAR-3) lineages. Therefore, pasteurization can be applied successfully to *S. cumini* pulp.

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### 1. Introduction

Jambolan (*Syzygium cumini* (L.) Skeels), also known as jambolão, black plum, jamun or java plum, belongs to the Myrtaceae family. This plant possesses fruits that are oblong berries, deep purple or bluish in colour with a pinkish pulp. These fruits are widely consumed and also used for the treatment of various diseases as an astringent, antiscorbutic, diuretic, antidiabetic, anti-Leishmania (Rodrigues et al., 2015), and a treatment for chronic diarrhoea (Ayyanar and Subash-Babu, 2012). In addition to all of the above-described properties, the fruit from the Jambolan (*Syzygium cumini*

(L.) Skeels) plant has also been reported to have strong antioxidant and anti-genotoxic potential (Baliga et al., 2011).

*S. cumini* fruit pulp is a source of phenolic compounds such as flavonoids and phenolic acids (Reynertson et al., 2008; Faria et al., 2011; Tavares et al., 2016). It also contains hydrolysable tannins, which may be the main phenolic compounds responsible for the astringency of the edible parts of fruit (Tavares et al., 2016), and anthocyanins: 3,5-diglucosides of delphinidin, petunidin and malvidin (Faria et al., 2011; Tavares et al., 2016), which are responsible for fruit pigment and are also thought to have antioxidant and anticancer bioactivity (Clifford, 2000; Nile and Park, 2014). Consuming these anthocyanin-rich fruits are described as a strategy to prevent cardiovascular diseases, cancer and neurodegenerative diseases (Oliveira et al., 2012).

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However, most studies published to date have essentially focused on raw foods, despite the fact that the human diet includes mainly cooked and processed foods. The domestic consumption of fruits may be either fresh or processed, including handmade frozen pulp, juice, jams, jellies and others. Industrial processing typically includes a heat treatment, such as drying, sterilisation and/or pasteurization, in order to destroy pathogenic microorganisms and inactivate enzymes, in order to increase shelf-life and availability of the product for consumption. However, such processing can drastically affect phenolic content and, as a consequence, the antioxidant capacity and anticancer bioactivities of foods formulated from these raw materials. For these reasons, anthocyanin-rich fruits, with few exceptions, have been poorly explored by the food industry, because they are with changes in temperature, pH, oxygen and light (Turfan et al., 2011). Anthocyanins are the most sensitive, since they are also affected by the presence of metal ions, enzymes, ascorbic acid, and sulphur dioxide (Cavalcanti et al., 2011; Turfan et al., 2011).

The majority of studies found in the literature correlate the anti-proliferative activities and extracts of fruits *in vitro* with their polyphenolic compounds, but such studies have not examined the role of thermal treatment (Medina et al., 2011; Leite-Legatti et al., 2012; Neri-Numa et al., 2013; Zhang et al., 2013; Rascón-Valenzuela et al., 2015). Therefore the functional properties of jambolan and the growing market demand for fruit pulps with flavor and colour, the pulp of this fruit shows high potential for use in the food processing industry, since this potential remains under-explored. Pasteurization is a heat treatment widely used in the fruit juice industry and the evaluation of changes/losses during this thermal processing is very important, because the use of high temperatures may result in a significant decrease in the concentration of bioactive compounds and consequently reduction of biological activity. In this context, considering the therapeutic potential of *S. cumini* pulp and the potential benefits for the human diet, we aimed to evaluate the influence of pasteurization during pulp-fruit processing on polyphenol levels in *S. cumini* pulp content. We then sought to correlate our findings with antioxidant and *in vitro* anti-proliferative activity.

## 2. Materials and methods

### 2.1. Chemical reagents

The following reagents were purchased from Sigma Chemical Co.: 2,2-diphenyl-1-picrylhydrazil (DPPH), Folin-Ciocalteu and quercetin. Aluminium chloride hexahydrate, potassium acetate and gallic acid salts were obtained from Synth, as were solvents and acids.

### 2.2. Processing and pasteurization of *S. cumini* pulp

Jambolan fruits (*Syzygium cumini* (L.) Skeels) were collected at Paulista State University (UNESP), Campus of Assis—São Paulo State, during January and February 2011. Five kilograms of jambolan fruit were harvested from three plants (clones), then sanitised with chlorinated water (25 ppm). The pits were removed, and the pulp was processed in a domestic processor. The pulp obtained was distributed in polyethylene bags and subsequently stored at  $-18^{\circ}\text{C}$  until the time of application. The jambolan pulp (100 g each) was submitted to pasteurization by heating in a water bath at  $70^{\circ}\text{C}$  for 5 min with stirring, followed by immediate cooling in an ice bath and storage at  $-18^{\circ}\text{C}$ . The pulp, with or without thermal treatment, was submitted to analyzes of pH, total acidity, moisture and soluble solids as described by AOAC (2000).

### 2.3. Extract preparation

Extracts were prepared according to Barreto et al. (2009), with some modifications. For every 10 g of jambolan pulp, 40 mL of methanol/water (8:2) was added. The solution was stirred for one hour. The mixture was then filtered in a Buchner funnel and washed using small portions of methanol. This step was performed two more times, using 20 mL of an 8:2 methanol/water solution. The filtrates were transferred to a volumetric flask, and the final volume was adjusted to 100 mL. Extract levels of DPPH, flavonoids and phenolic compounds were analyzed. For HPLC and anti-proliferative activity analyzes, the extract obtained was then concentrated through a rota-evaporator at  $40^{\circ}\text{C}$  and lyophilised for 72 h.

### 2.4. Quantitative analysis of phenolic content

Total phenolic content was estimated using the Folin-Ciocalteu colorimetric method, based on the procedure of Singleton and Rossi (1965). Quantitative measurements were performed based on a standard calibration curve generated with 100, 200, 300, 400 and 500 mg/L of gallic acid in 80% methanol. Total phenolic content was expressed as gallic acid equivalents (GAE) in grams per 100-g sample (mg GAE/100 g).

Total flavonoid content was determined using the methodology proposed by Chang et al. (2002). The extracts (0.5 mL) were mixed with 1.5 mL of 95% ethanol, 0.1 mL of 10% aluminium chloride hexahydrate, 0.1 mL of 1 M potassium acetate and 2.8 mL deionised water. After 40-min incubation at room temperature, the absorbance of the reaction mixture was measured at 415 nm. Results were expressed as milligrams of quercetin equivalents (QE) per 100 g of sample (mg QE/100 g). The standard curve for quercetin was obtained with a concentration range of 0–50 mg/L.

Radical scavenging activity was performed according to Rufino et al. (2010). First, a DPPH methanolic solution (0.06 mM) was prepared. Then, 100  $\mu\text{L}$  of fruit extract were added to 3.9 mL of DPPH solution. The decrease in absorbance at 515 nm was monitored once each minute until stabilisation. Antioxidant capacity was expressed as the sample concentration required to reduce the original amount of DPPH radicals by 50% ( $\text{EC}_{50}$ ), and values were expressed as g fruit/g DPPH.

Total anthocyanin content was determined according to Fuleki and Francis (1968a,b): extraction with ethanol and HCl (1.5 N) (85:15) solution, followed by mixture filtration and analysis in a spectrophotometer at 535 nm. Results were expressed as milligrams of cyanidin 3-glucoside (cyd 3-glu) equivalent per 100 g of fresh weight.

### 2.5. Analysis by HPLC

High-performance liquid chromatography coupled with ultraviolet-visible diode-array detection (HPLC-DAD/UV-vis) was performed in a Dionex UltiMate 3000HPLC system (Dionex, Idstein, Germany) equipped with an UltiMate 3000 Pump, Ulti-Mate 3000 Autosampler Column Compartment, UltiMate 3000 Photodiode Array Detector and Chromeleon software. A reversed phase Acclaim<sup>®</sup> 120 column (C18, 5  $\mu\text{m}$ , 120 Å, 4.6 mm  $\times$  250 mm) was used for these experiments. The column temperature was held at  $40^{\circ}\text{C}$  for chromatographic separation. After 10 min of re-equilibration, the column was ready for a new injection. Phenolic acids and flavonols are usually detected at wavelengths ranging from 210 to 320 nm. The injection volume was 10  $\mu\text{L}$  (250 mg/mL in 40% ethanol). The mobile phase was a gradient of 1% phosphoric acid aqueous solution (A) and methanol (B). The gradient was as follows: 0–2 min, 0–15% B; 2–5 min, 15–25% B; 5–10 min, 25–30% B; 10–15 min, 30–35% B; 15–25 min, 35–50% B; 25–30 min, 50–60%

B; 30–35 min, 60–80% B; 35–45 min, 80–100% B. The flow rate was set at 1 mL/min. The phenolic acids (gallic acid, chlorogenic acid, caffeic acid, *p*-coumaric acid, ferulic acid) and flavonols (rutin, myricetin, quercetin and kaempferol) standards (Sigma Chemical Co.) were applied. Stock solutions of all standards were prepared in methanol. For the HPLC analysis, phenolic compounds were identified by comparing their retention times with those of pure standards (Granato et al., 2011).

## 2.6. In vitro anti-proliferative activity

Anti-proliferative activity was screened using the methodology described by the Developmental Therapeutics Program at NCI/NIH (Monks et al., 1991; available at <http://dtp.nci.nih.gov/>). A non-tumour cell line, VERO (epithelial cells of monkey kidney), and the following human tumour cell lines were obtained from the National Cancer Institute (Frederick, MA, USA): U251 (glioma), UACC-62 (melanoma), MCF-7 (breast), NCI-ADR/RES (ovarian-expressing phenotype multiple-drug resistance), 786-0 (renal), NCI-H460 (lung, non-small cell), PC-3 (prostate), OVCAR-03 (ovarian), and K562 (leukaemia). Stock cultures were grown in medium containing 5 mL RPMI 1640 (GIBCO BRL) supplemented with 5% foetal bovine serum (FBS, GIBCO) at 37 °C with 5% CO<sub>2</sub>. A penicillin:streptomycin mixture (1000 µg/L:1000 U/L, 1 mL/L) was added to the experimental cultures. Cells in 96-well plates (100 µL cells × well<sup>-1</sup>) were exposed to extracts in DMSO (Merck)/RPMI (0.25, 2.5, 25, and 250 µg × mL<sup>-1</sup>) at 37 °C, 5% CO<sub>2</sub> in air for 48 h. The final DMSO concentration (less than 0.2%) did not affect cell viability. Before (T<sub>0</sub> plate) and after sample addition (T<sub>1</sub> plates), cells were fixed with 50% trichloroacetic acid. The level of cell proliferation was determined by spectrophotometric quantification (540 nm) of cellular protein content using the sulphorhodamine B assay. Using the concentration-response curve for each cell line, the GI<sub>50</sub> (concentration that produces 50% growth inhibition) was determined through non-linear regression analysis using ORIGIN 8.5<sup>®</sup> software (OriginLab Corporation) (Vendramini-Costa et al., 2010).

## 2.7. Statistical analysis

The results were expressed as mean values ± standard deviation from two extraction replicates, each run in triplicate. The analysis of variance (ANOVA) was determined using SAS software (version 9.2, SAS Institute, Cary, N.C., USA). The means were compared using Tukey's test. Statistical significance was set at a level of *p* < 0.05. Correlation and regression analyzes were performed using ORIGIN 8.5<sup>®</sup> software (OriginLab Corporation).

## 3. Results and discussion

Our results regarding the pH and total soluble solid levels of the *Syzygium cumini* pulp are compatible with those reported in the literature (Mussi et al., 2015; Tavares et al., 2016). Previous reports described *S. cumini* as an acidic fruit with pH ranging from 3.29 to 4.04 and soluble solid content ranging from 11.4 to 11.76 Brix. Acidity is an advantage in *S. cumini* pulp, because intensive heat treatment is not necessary, as bacteriological development does not take place at these pH levels. In this study, we found higher levels of soluble content (about 18 Brix) than those described by Mussi et al. (2015). Highly soluble content is important, because less sugar will need to be added and less time will be required for the water to evaporate. These factors require the expenditure of less energy, resulting in a more economical process with increased product yield (Pereira et al., 2012). Therefore, *S. cumini* is a good candidate for industrial processing (Table 1).

Differences in phenol, anthocyanin and flavonoid levels are usually related to variations in the methodologies employed in

**Table 1**

Physical-chemical, phenolic compounds and anti-radical activity determined in jambolan pulp with or without pasteurization (fresh matter).

	Without Pasteurization	With Pasteurization
pH	4.15 <sup>a</sup> ± 0.20	4.12 <sup>a</sup> ± 0.30
Moisture (%)	86.07 <sup>a</sup> ± 0.003	84.76 <sup>b</sup> ± 0.001
Soluble Solids (°Brix)	17.80 <sup>a</sup> ± 0.20	19.00 <sup>b</sup> ± 0.17
Total phenolics (mg GAE/100 g)	206.95 <sup>a</sup> ± 1.11	221.83 <sup>b</sup> ± 0.60
Total anthocyanin (mg cyd 3-Glu/100 g sample)	213.00 <sup>a</sup> ± 1.02	195.00 <sup>b</sup> ± 0.72
Total flavonoids (mg QE/100 g)	25.29 <sup>a</sup> ± 0.22	29.45 <sup>b</sup> ± 2.15
Antioxidant capacity EC <sub>50</sub> (gfruit/gDPPH)	53.25 <sup>a</sup> ± 2.93	83.16 <sup>b</sup> ± 9.78

Results expressed as mean ± standard deviation; Different letters indicate significant difference between means in the same line (Tukey's test *p* ≤ 0.05); GAE: gallic acid equivalent; cyd 3-glu: cyanidin 3-glucoside; QE: quercetin equivalent; EC<sub>50</sub>: Concentration of antioxidant required to reduce the original amount of free radicals by 50%.

sample extraction, reaction conditions, and other factors that affect fruit composition. Total phenolic content was higher in this study as compared to the results reported by Rufino et al. (2010) (185 mg/100 g fresh matter) and Faria et al. (2011) (148.3 mg/100 g of fresh matter). Similarly, total flavonoid content (Table 1) was much higher in this study as compared to the results reported by Benherlal and Arumughan (2007) (7 mg/100 g fresh matter) but lower when compared to the results reported by Faria et al. (2011) (91.2 mg of catechin equivalent/100 g sample). Sultana and Anwar (2008) found total flavonol (kaempferol, quercetin, myricetin) values of 0.24 mg/100 g dry matter and quercetin levels of 0.12 mg/100 g dry matter in jambolan pulp. The anthocyanin content in our experiments was higher than the results obtained by Benherlal and Arumughan (2007) (134 mg cyanidin-3-glucoside/100 g fresh weight) and by Rufino et al. (2010) (93.3 mg/100 g fresh matter) but lower than that obtained by Mussi et al. (2015) and similar to levels reported by Faria et al. (2011) (211 mg/100 g).

The next step will be to analyze differences in the phenolic composition profile induced by the pasteurization of *S. cumini* pulp. These values should then be compared to those obtained without thermal treatment. In this study, pasteurization did not affect *Syzygium cumini* pulp pH and increased total soluble solid content (about 6.7%), as shown in Table 1.

Total phenolic and flavonoid levels showed an increase after pasteurization (about 7.2% and 16.4%, respectively) with a reduction in anthocyanin content (about 8.5%). Decreases in total phenolic and flavonoid levels (especially anthocyanins) induced by changes in temperature are well described in the literature. Brownmiller et al. (2008) reported a 43% decrease in total anthocyanin content in blueberry purees after pasteurization. Turfan et al. (2011) evaluated the stability of anthocyanins after pasteurization (95 °C/10 min) during the processing of pomegranate anthocyanins (*Punica granatum* L., cv. Hicaznar) and verified a loss of 8–14%. Patras et al. (2009) reported that anthocyanin and phenol levels in strawberry and blackberry purees were also affected by thermal processing (70 °C/2 min), which showed that pasteurization reduced flavonoid levels in fruit purées. Zoric et al. (2014) noted that the duration of heating as well as the temperature used impacted anthocyanin degradation more than that of other phenols.

Mussi et al. (2015) described a reduction in anthocyanin content (60–70%) when processing *S. cumini* residue, but antioxidant activity levels still ranged from 93 to 97%. The authors did not correlate this trend with activity levels for other phenolic compounds. In this study, total anthocyanin and antioxidant activity levels were reduced after pasteurization (about 8.5% and 56%, respectively)

**Table 2**  
Quantification and chromatographic and spectroscopic characteristics of non-anthocyanic phenolic compounds from jambolan pulp obtained by HPLC-DAD.

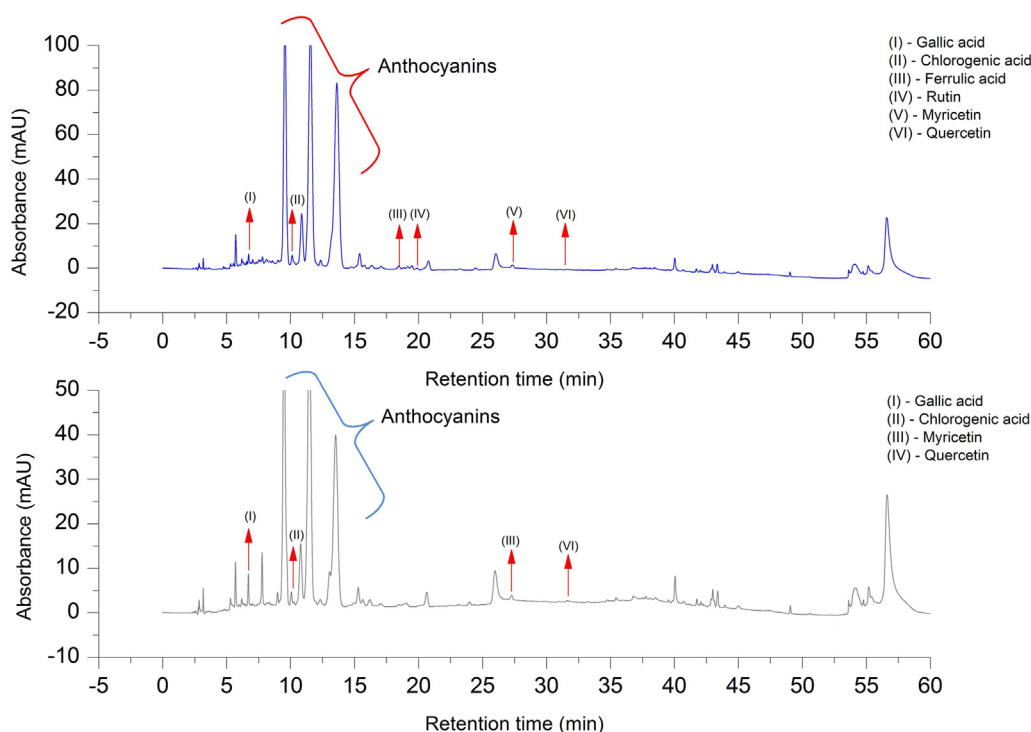
Compounds	t <sub>R</sub> (min)	λ (nm)	Without pasteurization (mg/Kg of fresh weight)	With pasteurization (mg/Kg of fresh weight)
Gallic acid <sup>a</sup>	6.71	272	28.04 <sup>a</sup> ± 0.93	44.04 <sup>b</sup> ± 1.90
Chlorogenic acid <sup>a</sup>	10.12	280	205.60 <sup>a</sup> ± 6.01	93.08 <sup>a</sup> ± 2.51
Ferulic acid <sup>a</sup>	18.45	280	7.81 ± 0.23	n.d.
Rutin <sup>a</sup>	19.88	370	5.96 ± 0.34	n.d.
Myricetin	27.33	215, 338	35.40 <sup>b</sup> ± 0.85	23.56 <sup>a</sup> ± 0.79
Quercetin	31.60	370	4.52 <sup>a</sup> ± 0.22	5.16 <sup>b</sup> ± 0.28
Total			287.33 <sup>b</sup> ± 5.98	165.84 <sup>b</sup> ± 4.78

<sup>a</sup> Note: Results in mean ± standard deviation. Means on the same line followed by different letters are significantly different (p < 0.05); n.d.: not detected; t<sub>R</sub>: Retention time, λ: Wavelength.

**Table 3**  
Antiproliferative activity (GI<sub>50</sub>, μg/mL) of doxorubicin and jambolan extracts against human cell lines.

	2	u	m	a	7	4	o	h	q
doxorubicin (positive control)	0.05 ± 0.04	<0.025	<0.025	0.18 ± 0.01	0.025	<0.025	<0.025	0.22 ± 0.01	<0.025
without pasteurization	>250	<0.25	>250	1.21 ± 0.17	>250	>250	>250	>250	>250
with pasteurization	>250	>250	>250	193.5 ± 7.71	3.8 ± 2.2	>250	53.7 ± 39.0	>250	>250

2 = U251 (glioma, Central Nervous System); u = UACC-62 (melanoma); m = MCF-7 (breast adenocarcinoma); a = NCI-ADR/RES (ovary, multidrug resistance phenotype); 7 = 786-0 (kidney); 4 = NCI-H460 (lung, non-small cells adenocarcinoma); o = OVCAR-3 (ovary); h = HT-29 (colon); q = HaCaT (keratinocyte human, immortalized normal cell).



**Fig. 1.** Chromatographic profile of jambolão pulp without (top) and with (bottom) pasteurization at 280 nm.

(Table 1). Therefore, we suggest that flavonoids and phenolic acids are as important as anthocyanins for the free radical scavenging of *S. cumini* pulp. Some studies about the antioxidant potential of phenolic compounds in fruits or foods have demonstrated the existence of synergistic or antagonistic effects among the various antioxidants present in the food matrix; thus, activity not only depends on flavonoids and/or phenolic structure but also on the environment in which said compounds are found (Palafox-Carlos et al., 2012; Ioannou et al., 2012).

Even so, *S. cumini* pulp pasteurization remains advantageous, because the process induces only limited anthocyanin loss, which allows the pulp to be used as a functional ingredient in several processed products that require the incorporation of fruit pulp, such as preserves, juices, nectar, ice cream, and yogurt. However, the

medium must be slightly acidic to preserve the characteristic colour of the fruit.

For the HPLC-DAD analysis, phenolic assessments were carried out by UV-vis analyzes, by comparing the retention times (t<sub>R</sub>) for available standards and by considering information from the literature. Three major peaks on the chromatogram shown in Fig. 1 were identified as the anthocyanins. The first peak, with a retention time of 9.4 min (UV absorption at 275 and 522.2 nm), may represent delphinidin 3-diglucoside. The second peak, with a retention time of 10.3 min (UV absorption at 274 and 524 nm), likely indicates the presence of delphinidin 3,5-diglucoside. The third peak at 13.4 min (UV absorption at 275 and 525 nm) suggests the existence of petunidin-3,5 diglucoside. These results are similar to those described by Faria et al. (2011), who found five types of anthocyanin

aglycones in *S. cumini* pulp: delphinidin, cyanidin, petunidin, malvidin and peonidin. Comparisons of jambolan pulp measurements with the retention times of authentic standards allowed us to identify phenolic acids (gallic acid: 6.71 min, chlorogenic acid: 10.1 min, ferulic acid: 18.4 min) and flavonoid (rutin: 19.9 min). Myricetin (27.3 min) and quercetin derivatives (31.6 min) were also detected and confirmed by UV absorption (215 and 338 nm, 370 nm, respectively). These results (Table 2) are similar to those described by Faria et al. (2011).

In Table 2, the results of HPLC quantification show that pasteurization increased gallic acid levels by 57% and quercetin levels by 14%, but there was a significant decrease in chlorogenic acid and myricetin levels (about 54% and 33%, respectively). Neither ferulic acid nor rutin was detected after pasteurization. The increase in gallic acid after pasteurization may be directly related to the cleavage of covalent bonds and the release of this molecule to the medium. However, the mechanism underlying the decrease in rutin content that accompanied the increase in quercetin remains elusive. At first, the increase in quercetin might seem to be related to the decrease in rutin glucose, but quercetin is less resistant to increased temperature than is rutin (Ioannou et al., 2012).

Ioannou et al. (2012) also described, in their revision, the fact that it is not easy to dissociate the effect of thermal processing from food matrix effects, because the degradation of flavonoids is not only a function of temperature and the duration of heating; it may also depend on other parameters such as pH, the presence of oxygen and the presence of other phytochemicals in the medium. Regarding pH, the degradation of rutin and quercetin is enhanced under weakly alkaline and neutral reaction conditions. Notably, *S. cumini* pulp is acidic. Alternately, the process could be oxidative, because the presence of high concentrations of oxygen induces quercetin and rutin degradation, while the absence of oxygen has the opposite effect. However, Ioannou et al. stated that other phytochemicals such as chlorogenic acid in the medium may play a protective role, which could explain the medium presence of quercetin and maybe myricetin, because the latter compound is found in *S. cumini* pulp.

In this work, the anti-proliferative activity of the tested compounds was expressed as the concentration that produced 50% cell growth inhibition or a cytostatic effect ( $GI_{50}$ ,  $mg/mL^{-1}$ ) for each cell line (Table 3).

For *S. cumini* pulp, optimal results with regard to selectivity and cytostatic effects on growth inhibition among the cells tested before pasteurization were found for the melanoma cell line (UACC-62,  $GI_{50} < 0.25 mg/mL$ ), followed by the resistant ovarian cell line (NCI-ADR/RES,  $GI_{50} = 1.21 \mu g/mL$ ), where significant cytostatic activity was observed. After pasteurization, *S. cumini* presented a weak inhibitory effect on the growing ovary cell line (OVCAR-3,  $GI_{50} = 53.7 \mu g/mL$ ) and significant selectivity and cytostatic effects on the inhibition of growing kidney cell lines (786-0,  $GI_{50} = 3.7 \mu g/mL$ ) (Table 3). These differences in the anti-proliferative profile may be explained by chemical changes promoted by heat treatment as indicated by total phenolic, anthocyanin and flavonoid levels (Table 1) and HPLC analysis (Table 2, Fig. 1). Although the pulp had high selectivity for those cell lines, it showed weak anti-proliferative activity against other cancer cell lines and HaCat cells; i.e., pulp did not affect the proliferation of normal cells, making it an excellent candidate nutraceutical food.

#### 4. Conclusion

These findings suggest that pasteurization can be applied successfully to *S. cumini* pulp. The pH as well as total soluble solid, total phenol and flavonoid levels were increased. Despite the overall reduction in anthocyanin content and antioxidant activity in pasteurised as compared to non-pasteurised pulp, pasteurised pulp

demonstrated robust *in vitro* anti-proliferative activity in kidney cells (786-0) and weak *in vitro* anti-proliferative activity in ovarian cells (OVCAR-3), without affecting the proliferation of normal cells (HaCat cells).

Therefore, *S. cumini* pulp pasteurization is advantageous: with limited anthocyanin loss, fruit colour was preserved. *S. cumini* pulp could therefore be used as an additive in medicines and foods, provided that they are slightly acidic. Moreover, the pulp preserves a portion of the initial antioxidant capacity and anti-proliferative activity present prior to the pasteurization process, making it a good candidate for a nutraceutical food. Further studies evaluating other pulp-processing conditions will be required to increase the product's shelf-life, thus preserving its health benefits and contributing to the intake of anthocyanins and other polyphenolic compounds.

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