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## Status of Rifaximin: A Review of Characteristics, Uses and Analytical Methods

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### ABSTRACT

Rifaximin, an oral antimicrobial, has many advantages because it is selective intestine, has minimal adverse effects and is used for the treatment of some diseases such as hepatic encephalopathy, irritable bowel syndrome, travelers' diarrhea, ulcerative colitis, *Clostridium difficile* and acute diarrhea. Rifaximin in the form of 200 mg tablets is commercially available. The crystalline  $\alpha$  form is therapeutically safe and effective. In most of the official compendia, rifaximin has no monograph and in none of them is there a monograph for rifaximin tablets. The literature, however, contemplates this gap with varied methods. The literature presents some methods for evaluation of rifaximin in both biological fluid and pharmaceutical product. High performance liquid chromatography stands out for the evaluation of rifaximin. Most of the methods reported in the literature are for pharmaceuticals products. They use (1) toxic organic solvents, harmful to the operator and the environment, and/or (2) buffer solution, which has a shorter service life and requires time-consuming washes of the chromatographic system generating more waste. So, this work aims to discuss (i) properties; (ii) applications; (iii) polymorphism and (iv) analytical methods of rifaximin by the look of green chemistry. This review shows an extremely current topic of great importance to the chemical-pharmaceutical area and everything it involves, since the analytical process until the impact on the environment in which it is embedded.

### KEYWORDS

Analytical methods; antimicrobial; green chemistry; HPLC; polymorphism; rifaximin

### Introduction

Rifaximin is an oral antimicrobial with broad spectrum of action non-absorbable. It acts in the gastrointestinal tract and presents minimal adverse effects.<sup>[1,2]</sup> It is practically not absorbed and achieves high concentrations in the human intestine, where it is active against many enteropathogens.<sup>[3]</sup>

It is a non-systemic antimicrobial derived from rifamycin.<sup>[4]</sup> Its high tolerability can be compared to placebo.<sup>[2,5]</sup> It is used safely for the treatment of many different diseases by children, the elderly and debilitated people. Thus, rifaximin reaches a large target audience and therefore analytical methods for the evaluation of its quality must be effective and reliable. However, in the current scenario, they must also be environmentally friendly.

The use of non-toxic reagents, the careful choice of the analytical method as well as the apparatus to be used, the speed of the analysis, the residues generated and the amount of sample used are examples of parameters relevant to green chemistry. Currently, analytical processes must be viewed in a multi-dimensional way and appropriate to the world reality, where there is a growing concern with the health of the environment and everything that surrounds it.<sup>[6–9]</sup>

Therefore, the objective of this review is to discuss about (i) properties, (ii) applications, (iii) polymorphism and (iv) analytical methods of rifaximin by the look of green chemistry.

### Rifaximin

Rifaximin, **Figure 1**, has structure analogous to rifampicin and is a derivative of rifamycin. Rifampicin and rifabutin are equally effective in tuberculosis schemes for HIV-positive and HIV-negative patients. Rifabutin is more active than rifampicin against *Mycobacterium avium*, and is preferentially used in multiple drug regimens for these infections, but otherwise rifampicin is the drug of choice. Resistance develops rapidly with rifampicin, so it is rarely used alone.

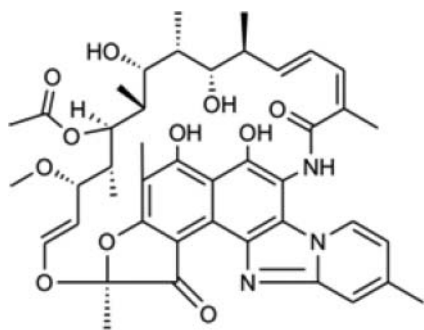
Rifaximin binds to the  $\beta$  subunit of the bacterial enzyme RNA polymerase DNA-dependent and inhibits bacterial RNA synthesis.<sup>[2]</sup>

Rifaximin is used in cases of hepatic encephalopathy,<sup>[1–5,10]</sup> ulcerative colitis,<sup>[10]</sup> irritable bowel syndrome,<sup>[4,2]</sup> *Clostridium difficile*,<sup>[2,3]</sup> diarrhea of travelers<sup>[2–4,11,12]</sup> and acute diarrhea.<sup>[13]</sup>

### Hepatic encephalopathy

Hepatic encephalopathy is a complex liver disease. It is considered as a metabolic disorder or neurophysiological.<sup>[10]</sup>

Normal patients remove nitrogen residues produced by gastrointestinal bacteria. In this process, ammonia is metabolized in urea and excreted. Patients with hepatic encephalopathy cannot convert ammonia to urea and consequently it accumulates in the blood. This generates accumulation in the central nervous system and affects neurotransmission. Hepatic



**Figure 1.** Chemical structure of rifaximin (CAS 80621-81-4,  $C_{43}H_{51}N_3O_{11}$  and molecular weight  $785.9 \text{ g mol}^{-1}$ ).

encephalopathy is associated with portal hypertension in patients with cirrhosis and chronic liver disease.<sup>[2]</sup>

### Treatment

The neuropsychiatric symptoms and neuromuscular dysfunction associated with hepatic encephalopathy are accounted for the clinical and socioeconomic burden of chronic liver disease for patients and their caregivers. Hepatic encephalopathy often results in hospitalization and decreased survival in patients with cirrhosis. Prevention of hepatic encephalopathy episodes can improve the results for patients while they wait for transplantation and improve post-transplant function. A long-term therapeutic intervention to prevent the recurrence of hepatic encephalopathy is necessary to decrease the burden of health care, enhance quality of life and improve results for patients with chronic diseases.<sup>[1]</sup>

Current strategies as removal of precipitating factors, increase the elimination of ammonia or reduce ammonia production are used for hepatic encephalopathy. For this, lactulose was chosen, but its continued use is problematic and there are patients who do not tolerate or do not respond to therapy with lactulose.<sup>[1,2,5]</sup> Rifaximin has been an option for this case.<sup>[2,5]</sup>

### Irritable bowel syndrome

Irritable bowel syndrome, a chronic gastrointestinal disease, affects up to 20% of the general population. The diagnosis is often an exclusion diagnosis.<sup>[2]</sup>

It can be characterized by change in bowel habits, abdominal pain, diarrhea, constipation and swelling. The etiology of irritable bowel syndrome may be due to changes in peripheral and central sensory.<sup>[2]</sup> In addition to this, data show that there is a relationship between intestinal bacterial overgrowth and irritable bowel syndrome.<sup>[2,4]</sup> In it the normal intestinal flora of the proximal intestine is altered in quantity and quality; flora adopts the characteristics of large bowel flora of aerobic and anaerobic and aerobic and anaerobic coliform predominate at densities  $> 10^3 \text{ CFU mL}^{-1}$  and often greater than  $10^5 \text{ CFU mL}^{-1}$ .<sup>[4]</sup>

### Clostridium difficile

Rifaximin can be a treatment option for patients with multiple recurrent episodes of *C. difficile* according to the Society for

Healthcare Epidemiology of America (SHEA) and Infectious Diseases Society of America (IDSA).<sup>[2,3]</sup>

### Traveler's diarrhea

Traveler's diarrhea is a disease very common for those returning from tropical areas and it is usually caused by *Escherichia coli* (enterotoxigenic and enteroaggregative mostly), which may be identified in 50% of cases,<sup>[11,12]</sup> followed by *Shigella* spp. The traveler's diarrhea is a self-limited disease and normally treatment with antibiotics is not required, unless the circumstances become worse.<sup>[12]</sup>

### Treatment

The most common treatment for travelers' diarrhea with antibiotics uses fluoroquinolones. However, amoxicillin + clavulanic acid, azithromycin, erythromycin or cotrimoxazole are often used.<sup>[12]</sup>

Quinolones presents a good activity against traveler's diarrhea pathogens, but they are expensive for some countries. However, currently pathogens causing travelers' diarrhea have been shown to be resistant to quinolones. A reduction in the effectiveness of quinolones highlights the urgency of finding alternative therapies. For example, ciprofloxacin is very effective against *E. coli* and *Shigella sonnei*, but it is associated with toxicity and drug interactions. Thus, non-absorbable antimicrobials should overcome the limitations of the systemic antimicrobials and rifaximin, a derivative of rifamycin characterized by a broad antibacterial spectrum, is an excellent option.<sup>[11,12]</sup>

### Acute diarrhea

Acute diarrhea, severe and fatal in some cases, hits children most of the time. Children with genitourinary malformations or anal dysfunctions often experience diarrhea outbreak due to modification of the normal bacterial flora.<sup>[13]</sup>

### Treatment

Diarrhea frequently occurs in children. Some of them had experience with acute gastroenteritis because they used nitrofurantoin or trimethoprim/sulfamethoxazole to prevent urinary tract infections. Fast antimicrobial treatment of diarrhea episodes in these children is necessary to avoid complications. In this situation, rifaximin is a good choice. It has a wide range of antimicrobial activity, is selective bowel and therefore has minimal adverse effects.<sup>[13]</sup>

### Polymorphism

Rifaximin is marketed in hydrate form, but the environmental conditions may trigger their conversion to the amorphous form.<sup>[14]</sup> On the market tablets of 200 mg are found. The dose is 600 mg (1 tablet 3 times daily) or 800 mg (2 tablets 2 times a day).

Rifaximin, an antimicrobial for local action in the gastrointestinal tract, can exist in form of crystals and amorphous.

Rifaximin formulation contains  $\alpha$  form that has limited systemic bioavailability. However, the amorphous form was found in formulation of generic products. This does not guarantee the great tolerability of rifaximin, since the amorphous form has a higher systemic bioavailability.<sup>[14]</sup>

The regulatory authorities are increasingly attentive to the polymorphisms. Generic products contain the same active principle, same dose, same pharmaceutical form, administered in the same way, in the same dosage and used for the same therapeutic indication of the reference product. This context, regulatory agencies recommended guides to contemplate trials capable of recognizing the interchangeability of branded and generic products.<sup>[14–16]</sup>

Therapeutic problems can exist in the administration of different crystalline forms of products containing rifaximin, but the problem is greater when the exchange of crystalline forms and amorphous forms happens.<sup>[14]</sup>

The polymorphism of drug depends of the synthesis, solvents used, purification and crystallization. Normally, the productive process generates a crystalline form. Studies of the literature show that the reference rifaximin has a crystalline form in the form of a hydrate and that changes in the solvents used in its process can lead to a mixture of crystalline forms and even an amorphous form.<sup>[14]</sup>

This mixture of different crystalline forms of rifaximin ( $\alpha + \beta$ ,  $\beta$ ,  $\beta +$  amorphous) and even the amorphous form have different values of potency compared to the potency of rifaximin form  $\alpha$ . The different combinations of rifaximin forms,  $\alpha + \beta$ ,  $\beta$ ,  $\beta +$  amorphous, present melting point of approximately 217°C. The amorphous form presents greater solubility than the form  $\alpha$ ,  $\beta$  and the combinations  $\alpha$  and  $\beta$ . However, it does not have the same excellent tolerability as the reference drug, form  $\alpha$ .<sup>[17,18]</sup>

## Analytical methods

Even with all its importance, all its uses and its superiority to other drugs, rifaximin lacks of the analytical methods in the literature and in most official compendia for your quality control, because the drug's efficiency is not enough if it does not present the quality required for its safe and effective use.

Reliable and effective analytical methods are extremely important for the promotion of true and inducible results of the analyzed products. However, for this, the validation of analytical methods is necessary as it ensures the capability of the method.<sup>[19,20]</sup>

Rifaximin is not present in the investigated pharmacopoeias.<sup>[21–23,72]</sup> The British Pharmacopoeia<sup>[24]</sup> and European Pharmacopoeia<sup>[25]</sup> only contemplate the monograph for rifaximin in raw material by high performance liquid chromatography (HPLC).

The pharmaceutical product of rifaximin tablets is not yet present in any official compendium, which makes the specification of the final product indefinite. This can cause damage to the quality of medicines containing rifaximin released on the market, since the chemical-pharmaceutical industries do not have a quality control parameter established in official documents.

The literature shows methods for detection of rifaximin in biological fluid as rat urine, mouse serum, human plasma and

dried blood spots and milk by HPLC or high performance liquid chromatography coupled to mass spectrometry (HPLC-MS).<sup>[10,26–32]</sup>

The literature also shows the detection of rifaximin in pharmaceutical product as tablets by spectrophotometry in the ultraviolet (UV),<sup>[33–36]</sup> visible (Vis)<sup>[37]</sup> and infrared region (IR),<sup>[38]</sup> capillary electrophoresis (CE),<sup>[39]</sup> thin layer chromatography (TLC)<sup>[40]</sup> high-performance thin-layer chromatographic (HPTLC)<sup>[41]</sup>; HPLC<sup>[42–50]</sup> and HPLC-MS.<sup>[40]</sup>

The literature also reports microbiological method for the calculation of the power of rifaximin by turbidimetry,<sup>[44]</sup> which is a more dynamic method than the traditional method by diffusion in agar. In the case of antimicrobials, the simultaneous evaluation of the material by physico-chemical and microbiological methods are extremely important. Physico-chemical methods do not always faithfully assess the potency of drugs such as antimicrobials made by the microbiological method. Care in working with these two types of techniques is fundamental to ensure quality medicines for the population.<sup>[51–53]</sup>

Dissolution method for tablets is important for the verification of dissolution profile and pharmacotechnical formula, since an excellent medicine is not sufficient if it does not dissolve properly. Rifaximin presents a dissolution method with reading in spectrophotometer in the ultraviolet region.<sup>[54]</sup>

The study of the stability of raw material and pharmaceutical products is required throughout the world. The conditions used in this type of study, which includes short- and long-term stability studies, are recommended by International Conference on Harmonisation,<sup>[55]</sup> by National Agency of Sanitary Surveillance<sup>[56]</sup> and by guide of World Health Organization.<sup>[57]</sup> The monitoring of drug stability helps reveal the behavior and possible product problems within the validity period. The literature shows the behavior of tablets of rifaximin during 6 months subjected to simultaneous conditions of temperature ( $40 \pm 2^\circ\text{C}$ ) and humidity ( $75 \pm 5\%$ ).<sup>[58]</sup>

All these studies are essential, especially because rifaximin can present polymorphs with characteristics and therapeutic totally different and unwanted. Analytical differences can indicate the presence of different polymorphs.

Universities have performed a fundamental role serving as research centers for the development and validation of analytical methodologies, contributing to sanitary control activities and scientific enrichment in the area.<sup>[59]</sup>

Table 1 shows the conditions of the methods for evaluation of rifaximin in biological fluids and pharmaceutical product, and Figure 2 illustrates their distribution.

The use of methods, considered green, makes unnecessary the remediation of environmental impacts often observed today.<sup>[6,60–63,70,764–66]</sup> Some methods for evaluation of rifaximin contemplate this thought because they use only ethanol and water, instead of methanol in the UV method, for example, or simply just potassium bromide, as shown in Table 1.

The thinking of green chemistry begins with the choice of analytical method to be used followed by apparatus and diluents. In this choice, suitability for the intended purpose must be sufficient and it cannot be based on the method that everyone uses or because it is fashionable. In a routine analysis, for example, is HPLC-MS really necessary? Is not UV or even TLC enough? Is the pre-column chromatographic necessary?



Table 1. Methods for evaluation of rifaximin in biological fluids and pharmaceuticals.

Method	Condition	LOD, LOQ and/or range	Recovery (%)	Detection	Matrix	Time (min)	Reference
LC	C18 column (5 $\mu\text{m}$ , 150 mm $\times$ 4.6 mm). The mobile phase was a mixture of methanol-acetonitrile-0.05 mol L <sup>-1</sup> mono potassium phosphate-0.5 mol L <sup>-1</sup> citric acid (50:25:20:5)	50–200 $\mu\text{g mL}^{-1}$	99.9	254 nm	Related substances	—	[50]
LC-MS/MS	Restek Pinnacle C18 column (50 $\times$ 2.1 mm, 5 $\mu\text{m}$ ) with a mobile phase consisted of ammonium acetate solution (15 mM, pH 4.32) and methanol	0.5–10 ng mL <sup>-1</sup>	98.2–109	Quantification was performed in positive mode using SRM	Human plasma	2.3	[10]
LC	Supelco LC-Hisep (150 $\times$ 4.6 mm, 5 $\mu\text{m}$ ) using acetonitrile: water: acetic acid (18:82:0.1 v/v/v) as a mobile phase	0.03, 0.10 $\mu\text{g mL}^{-1}$ and 0.10–20 $\mu\text{g mL}^{-1}$	—	233 nm	Rat serum and urine	13.5	[29]
2D-LC-MS/MS	Waters C18 column (150 $\times$ 4.6 mm, 5 $\mu\text{m}$ ) using 0.1% aqueous acetic acid: acetonitrile as mobile phase in a gradient elution mode	0.5–10 ng mL <sup>-1</sup>	—	Positive mode of ion detection was used	Rat serum	11.0	[30]
LC	Luna Phenomenax, C18 (150 mm $\times$ 4.6 mm i.d., particle size 5 $\mu\text{m}$ ) column, methanol:10 mM phosphate buffer (70:30 v/v pH adjusted to 3.0 by using orthophosphoric acid) as mobile phase and flow rate of 1.2 mL min <sup>-1</sup> at ambient temperature	0.042, 0.127 $\mu\text{g mL}^{-1}$ and 5–30 $\mu\text{g mL}^{-1}$	100.31	293 nm	Bulk and tablets	5.1	[47]
Vis	Rifaximin was dissolved in methanol and distilled water. Oxidative coupling reaction with ferric chloride and 3-methyl 1, 2-benzothiazoline hydrazone hydrochloride (MBTH) reagent were used.	0.469, 1.422 $\mu\text{g mL}^{-1}$ and 5–25 $\mu\text{g mL}^{-1}$	100.03	637 nm	Tablets	—	[36]
UV	Rifaximin was dissolved in methanol. Alkaline borate buffer (pH-12) was used for the dilutions and readings	0.5588, 1.6934 $\mu\text{g mL}^{-1}$ and 5–25 $\mu\text{g mL}^{-1}$	100.10	296 nm	Tablets	—	[36]
LC-MS	Zorbax SB C18, 4.6 $\times$ 75 mm, 3.5 $\mu\text{m}$ column, 10 mM ammonium formate (pH 4.0) and acetonitrile (20:80 v/v) as mobile phase at a flow rate of 0.3 mL min <sup>-1</sup>	20–20,000 pg mL <sup>-1</sup>	95.7–104.2	m/z 786.4 to 754.4 in SRM	Human plasma	3.4	[26]
LC	End-capped octadecylsilyl silica gel (25 mm $\times$ 4.6 mm) column at 40°C using 37 volumes of solution of ammonium formate (pH 7.2) and 63 volumes of a mixture of equal volumes of acetonitrile and methanol as mobile phase. Flow rate at 1.4 mL min <sup>-1</sup>	—	—	276 nm	Raw material	—	[24,25]
LC	Symmetry C18 column, acetonitrile: ammonium acetate 85:15 (v/v) as mobile phase and flow rate of 1 mL min <sup>-1</sup> . The diluent of the samples was methanol	0.2, 0.5 ppm and 5–50 ppm	99.77	236 nm	Tablets	4.3	[43]
LC-MS	Chromolith Flash RP18e, (25 $\times$ 4.6 mm) column using mobile phase consisting of 0.05% formic acid: acetonitrile (55:45 v/v)	0.1–10 ng mL <sup>-1</sup>	98.42	Positive mode of ion was used	Dried blood spots	2.3	[31]
LC	C18 (150 mm and 4 $\mu\text{m}$ particle size) column, methanol and ammonium acetate as mobile phase	30, 150 ng mL <sup>-1</sup> and 128–513 ng 50 $\mu\text{L}^{-1}$	54.17	455 nm	Milk	0.5–0.6	[27]
LC	Waters XTerra RP18 column (250 $\times$ 4.6 mm, 5 $\mu\text{m}$ ) with 0.1% aqueous acetic acid: acetonitrile (45:55 v/v) as mobile phase	0.01, 0.03 $\mu\text{g mL}^{-1}$ and 0.03–10.0 $\mu\text{g mL}^{-1}$	99.5	239 nm	Rat serum	6.2	[32]
LC	C18 column using an aqueous tetra butyl ammonium hydrogen sulphate: methanol (10:90, v/v), with flow rate 1.0 mL min <sup>-1</sup>	0.786, 0.238 $\mu\text{g mL}^{-1}$ and 1–200 $\mu\text{g mL}^{-1}$	98.70–99.71	454 nm	Tablets	3.6	[45]
HPTLC	Aluminium plates precoated with silica gel 60 F254 and n-hexane : 2-propanol : acetone : ammonia (5 : 4 : 1, v/v/v/v) as mobile phase	61.25, 185.59 ng band <sup>-1</sup> and 400–3200 ng band <sup>-1</sup>	100.35–103.12	Rf of 0.59 $\pm$ 0.03 at 443 nm	Bulk and tablets	—	[41]
LC	Chromosil Symmetry C18 (150 $\times$ 4.6 mm, 5 $\mu\text{m}$ ) column, phosphate buffer pH 4.0 and acetonitrile (40:60 v/v) as mobile phase and flow rate 1.0 mL min <sup>-1</sup>	0.02, 0.10 $\mu\text{g mL}^{-1}$ and 10–60 $\mu\text{g mL}^{-1}$	100.6–101.4	292 nm	Bulk and tablets	3.0	[46]
LC	C18 column using an aqueous tetra butyl ammonium hydrogen sulphate (10 mM) (pH 3.37): acetonitrile (40:60, v/v) as mobile phase, flow rate 1.2 mL min <sup>-1</sup>	0.079, 0.024 $\mu\text{g mL}^{-1}$ and 0.1–200 $\mu\text{g mL}^{-1}$	98.49–98.83	441 nm	Bulk and tablets	5.7	[42]
LC	Symmetry C18 column (150 $\times$ 4.6 mm, 5 $\mu\text{m}$ ), flow rate of 1 mL min <sup>-1</sup> and methanol: phosphate buffer pH 3 (pH was adjusted with orthophosphoric acid), 65:35 v/v, as mobile phase	20–100 $\mu\text{g mL}^{-1}$	100.30	454 nm	Bulk and tablets	2.3	[49]

UV	Rifaximin was dissolved in ethanol and distilled water The solutions were made in 0.01 M NaOH	1.39, 4.22 $\mu\text{g mL}^{-1}$ and 10–30 $\mu\text{g mL}^{-1}$ 0.301, 0.912 $\mu\text{g mL}^{-1}$ and 10–50 $\mu\text{g mL}^{-1}$	100.17 99.52	290 nm First derivative at 292.80 nm	Tablets Synthetic mixture	[33] [34]
UV	The solutions were made in 0.01 M NaOH	0.214, 0.648 $\mu\text{g mL}^{-1}$ and 10–50 $\mu\text{g mL}^{-1}$ 50–98 $\mu\text{g mL}^{-1}$	>99.0 100.70	292 nm 530 nm	Synthetic mixture Tablets	[35] [44]
Turbidimetric	Rifaximin was dissolved in ethanol and distilled water. The concentrations used were 50, 70 and 98 $\mu\text{g mL}^{-1}$					
LC	Eclipse Plus™ C18 5 $\mu\text{m}$ as column, mixture of water + 0.1% glacial acetic acid and ethyl alcohol in the ratio 52:48 (v/v) as mobile phase, flow rate 0.9 $\text{mL min}^{-1}$ and injection volume at 20 $\mu\text{L}$			290 nm	Tablets	[44]
LC-MS	Gemini C18 column (50 $\times$ 2.0 mm, 5 $\mu\text{m}$ ), acetonitrile: 10 mM ammonium formate (in 0.1% formic acid) (80:20, v/v) as mobile phase at a flow rate of 0.20 $\text{mL min}^{-1}$ and positive ionization mode	10 $\text{pg mL}^{-1}$ and 10–5000 $\text{pg mL}^{-1}$	>90.64	SRM mode and m/z 786.4 to 754.3	Human plasma	[28]
LC	Inertsil C18 (250 $\times$ 4.6 mm, 5 $\mu\text{m}$ ) column, sodium acetate buffer: acetonitrile (60:40 v/v pH adjusted to 5.0 by using NaOH) as mobile phase with flow rate 1.0 $\text{mL min}^{-1}$ at 25°C	0.71, 2.160 $\mu\text{g mL}^{-1}$ and 20–80 $\mu\text{g mL}^{-1}$		293 nm	Bulk and tablets	[48]
Vis	Rifaximin was dissolved in ethanol and distilled water	0.3075, 0.9317 $\mu\text{g mL}^{-1}$ and 15–50 $\mu\text{g mL}^{-1}$	99.12	477 nm	Tablets	[37]
IR	Rifaximin was diluted in potassium bromide (KBr), previously dried	0.20, 0.61 mg and 1.0–3.5 mg	100.13	1767–1701 $\text{cm}^{-1}$	Tablets	[38]
CE	Rifaximin was dissolved in ethanol and distilled water. The separation was performed using a silica capillary, borate buffer 25 mM pH 9.5 and 20 kV	10.72, 32.48 $\mu\text{g mL}^{-1}$ and 50–500 $\mu\text{g mL}^{-1}$	100.24	290 nm	Tablets	[18]
TLC	Silica gel as stationary phase and ethyl acetate: ethyl alcohol, 90:10 (v/v), as mobile phase			Rf of 0.62 at 254 nm	Tablets	[54]
LC-MS	Column Eclipse Plus™ C18 5 $\mu\text{m}$ , mixture of water + 0.1% glacial acetic acid and ethyl alcohol in the ratio 52:48 (v/v) as mobile phase, flow rate 0.9 $\text{mL min}^{-1}$ and injection volume at 20 $\mu\text{L}$ . Ion trap mass spectrometer, AmaZon SL Bruker™			m/z 784 in the negative mode	Tablets	[54]
Dissolution	Paddle apparatus at 50 rpm and 900 mL of acetate buffer of pH 5.0 + 0.2% SLS as dissolution medium		>98.92	290 nm	Tablets	[54]
Stability	Conditions of 40 $\pm$ 2°C and 75 $\pm$ 5% relative humidity				Tablets	[58]

LC = liquid chromatography; LC-MS = liquid chromatography coupled to mass spectrometry; 2D-LC-MS/MS = two-dimensional reversed-phase liquid chromatography coupled to mass spectrometry; TLC = thin-layer chromatographic; HPTLC = high-performance thin-layer chromatographic; Rf = retention factor; Vis = spectrophotometry in the visible region; UV = spectrophotometry in the ultraviolet region; IR = spectrophotometry in the infrared region; CE = capillary electrophoresis; TLC = thin layer chromatography; SRM = selected reaction monitoring.



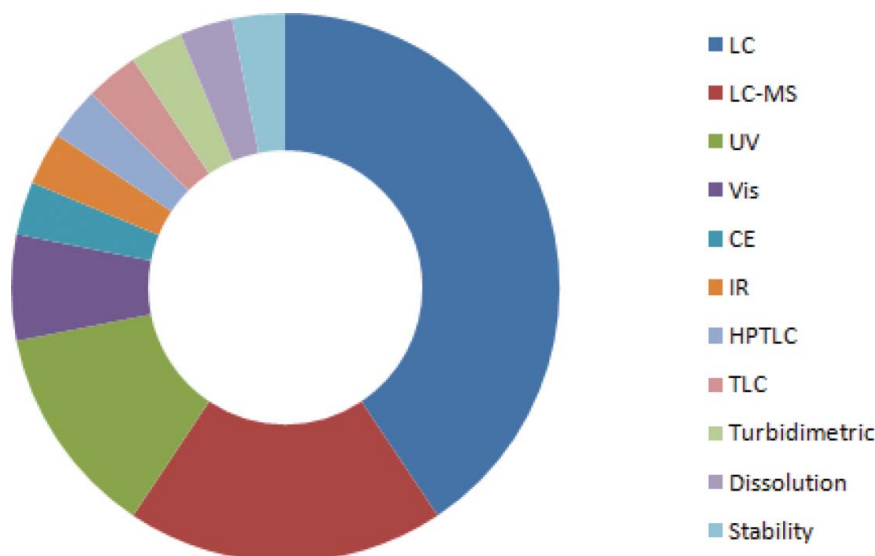


Figure 2. Current distribution of analytical methods for evaluation of rifaximin.

Buffer solutions have a short half-life and always require new preparations, which make the process more expensive. Acetonitrile, methanol and buffer solution are the diluents of choice for numerous analytical methods, but are they really needed? In the case of rifaximin, the answer is “no.” Ethanol and purified water were enough.

Faster, low cost and without toxic solvents methods must be preferred for the quantification of rifaximin, focusing on the multi-dimensional impact of analytical decisions. The correct analytical choices can provide cheaper medicines on the market and more accessible to the population. This conscious and mature attitude relieves the public health system.

The use of toxic solvents such as acetonitrile and methanol,<sup>[67]</sup> in addition to damaging the health and quality of life of the operator who is in direct and daily contact with the product, it also requires adequate waste treatment to avoid compromising the water, aquatic life, soil, plantations and animals.<sup>[68,69]</sup>

Analysts are now called upon to think about the analytical decision itself, targeting all that it impacts<sup>[71]</sup>. They must also take into consideration the development and choice of cleaner, greener, faster and lower cost methods with less steps and consumables.

Green analytical chemistry expects analysts who think beyond of the result of an analysis at any cost.

## Conclusion

Rifaximin,  $\alpha$  crystalline form, is an oral antimicrobial with minimal adverse effects. It is used since hepatic encephalopathy until acute diarrhea. Its wide use reaches from children to debilitated people and the evaluation of the quality of products containing rifaximina is fundamental. Thus, analytical methods are needed and currently they are seen by the look of green chemistry. Rifaximin presents in the literature options of clean, green, efficient, cheap and fast analytical methods by HPLC, UV, Vis, IR, CE and TLC. However, it still lacks a monograph in official compendium for evaluation of the final product.

## Conflicts of interest

The authors declare no conflicts of interest.

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