

**SÃO PAULO STATE UNIVERSITY - UNESP  
CAMPUS OF JABOTICABAL**

**SELENIUM AND SULPHUR: MITIGATION IN PLANT  
STRESSES**

**Leonardo Warzea Lima**

Biologist

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**SÃO PAULO STATE UNIVERSITY - UNESP  
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STRESSES**

**Leonardo Warzea Lima**

**Advisor: Dr. Priscila Lupino Gratão**

**Co-Advisor: Dr. André R. dos Reis**

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Orientador: Priscila Lupino Gratão  
Co-orientador: André Rodrigues dos Reis  
Banca examinadora: Tiago Tezzoto, Tiago Santana Balbuena  
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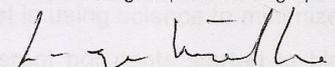
TÍTULO: SELENIUM AND SULPHUR: MITIGATION IN PLANT STRESSES

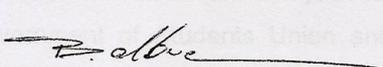
AUTOR: LEONARDO WARZEA LIMA

ORIENTADORA: PRISCILA LUPINO GRATÃO

Aprovado como parte das exigências para obtenção do Título de Mestre em AGRONOMIA (PRODUÇÃO VEGETAL), pela Comissão Examinadora:

  
 Profa. Dra. PRISCILA LUPINO GRATÃO  
 Departamento de Biologia Aplicada - Á Agropecuária / FCAV / UNESP - Jaboticabal

  
 Prof. Dr. TIAGO TEZOTTO  
 Centro Universitário Octávio Bastos / UNIFEOB - São João da Boa Vista/SP

  
 Pesquisador TIAGO SANTANA BALBUENA  
 Departamento de Tecnologia / FCAV / UNESP - Jaboticabal

Jaboticabal, 18 de maio de 2016.

## **AUTHOR'S CURRICULUM INFORMATION**

**Leonardo Warzea Lima** was born on March 01, 1987 in São Paulo, Brazil. His parents Eliana Regina Warzea Lima and Wladimir Godoy Lima. In 2005 he finished the high school in São Paulo. In 2010 his career path was clear and started to study for his bachelor degree in Biology at the São Paulo State University (UNESP), campus of Jaboticabal, which provided a rich and solid background for his research career. Among all the experiences on different departments, the one year training at the Plant Physiology Laboratory in 2012, working on the area of abiotic stress, was a watershed experience. Under the supervision of the Dr. Priscila Lupino Gratão, he could be surrounded by the routine of a laboratory with basic training and the specific methodology used to measure the level of the oxidative stress on different vegetable tissues. At the same year, he started his Undergraduate Research Mentorship Program, when received a scholarship to develop the project called "Selenium antioxidant effects on fruits of mutant tomato in response to the exposure to Cadmium". After one year studying Selenium (Se) and the heavy metal contamination, the interest in using science to minimize vegetable stress and environmental impact of different pollutants started to bloom. In 2013 he got involved with different academic activities at the university, occupying the position of public relations at the Government of Students Union and the student's representative to the Board of the Department of Biology, furthermore he created the Biological Sciences Academic Journal called "Conexão (Connection)", which became a powerful way to disseminate information and ideas through interviews and articles among students. As a curious Biologist, passionate about plant physiology, Leonardo was approved in 2014 to continue his studies on a Master of Science degree program in Agronomy, at the same University, in order to prepare for the long term goal of pursuing a career of teaching and research. As a Master's student, he became more independent, creative and proactive, able to conduct the research on the analysis of the antioxidants effects of Se and also to look forward to work on related areas of research. In March 2016, Leonardo was approved to continue his studies on a PhD program in Botany at the Colorado State University, Fort Collins-USA, under the supervision of the Dr. Elizabeth A. H. Pilon-Smits.

*“Discovery consists of seeing what everybody has seen, and thinking what nobody has thought”.*

**Albert von Szent-Györgyi**

*To my parents, Wladimir and  
Eliana, for the unconditional  
love and support.*

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## SELENIUM AND SULPHUR: MITIGATION IN PLANT STRESSES

**ABSTRACT** - Plants do not have specific defense mechanisms to counteract the diverse range of abiotic stresses and pollutants into the environment, and its survival depends on the flexibility and adaptability of its own natural defense mechanisms. Furthermore, the maintenance of cellular homeostasis depends on several interlinked and complex mechanisms, while the cellular defense system does not follow a specific pattern of action and may differ due to various factors such as plant species, exposure time to the stress, plant developmental stage, different organs and tissues analyzed. In the light of these considerations, this dissertation aimed to highlight and investigate the role of Sulfur and Selenium against different plant stresses, through the enzymatic and non-enzymatic plant responses and other related defense mechanisms. In the first chapter the author characterizes the general biochemical mechanisms of the antioxidant cell defense, specifically the reactive oxygen species (EROs) formation and its chemical singularities and the induced oxidative stress, the enzymatic antioxidant defense system, specifically the superoxide dismutase (SOD) and Catalase (CAT) enzymes, the non-enzymatic mechanisms against the stress, including the Ascorbate-Glutathione cycle, the GSH (reduced glutathione), the phytochelatins and also proline formation. The plant nutritional status during the stress is crucial in order to maintain a proper defense response. In view of this, the chapter two is a published review about the participation of Sulfur (S) on the stress defense. This nutrient has a role in fundamental processes such as electron transport, structure, regulation and it is also associated with photosynthetic oxygen production, abiotic and biotic stress resistance and secondary metabolism. Moreover, few chemical elements are considered benefic to plants, while Selenium (Se) is the most relevant. In the chapter three the author describes the role of Se to detoxify the stress induced by heavy metal contamination, its powerful antioxidant characteristics and the improvement of the antioxidant enzymes activity and overall defense mechanisms. The chapter four consists of a scientific project conducted by the author. The aim of this study was to investigate whether Selenium, under the form of selenite ( $\text{Na}_2\text{SeO}_3$ ), may avoid the uptake, translocation and concentration of Cadmium ( $\text{CdCl}_2$ ), in different tomato tissues, indicating possible mechanisms to counteract the stress, as well as to analyze the fruits overall status through the nutritional analyses, dry weight, pigments and proline concentration. The results demonstrate that alleviating effect of Se in tomato under Cd contamination could be related to restriction of  $\text{Cd}^{2+}$  uptake and translocation, enhancing micronutrient concentration in fruits and, finally, enhancing fruit proline concentration.

**Key-Words:** Heavy metal, acclimation, stress adaptations, oxidative stress

## CHAPTER 1 – GENERAL CONSIDERATIONS

### 1. Introduction

In order to facilitate the understanding and comprehension of the degenerative mechanisms triggered by different sources of abiotic stresses, the meaning of the word “stress” must be primarily defined under the vegetal perspective. Plants maintain a very intimate and necessary contact with the environment in which they are located, through the water and minerals absorption from the soil as well as the capture of the light energy from the sun, among other processes. Therefore, it is easy to assume that the environment exerts an important role to the physiology, anatomy and the plants biochemistry, influencing systemically their growth and development.

Consequently, any alterations of the adequate and specific environmental conditions for each vegetable species to develop accordingly and healthy can disrupt physical and chemical changes in the plant as a result of a primary reaction to the stress, such as water deficiency or excess of salts in the soil, which can cause the stomatal closure and an increased production of compatible osmolytes, like proline and glycine-betaine, for example (BANU, et al., 2009; SHEVYAKOVA et al., 2013). Such modifications are reversible and all plant metabolism regulates after the environmental conditions returns to the regular state. However, when the stress is strongly severe or persists for a long time an imbalance in the cellular redox state occurs, mainly due to the overproduction of reactive oxygen species (ROS) above the cellular antioxidant capacity, leading to the destruction of membrane lipids, proteins, nucleic acids and other cellular components, resulting in a secondary stress called oxidative (OPDENAKKER et al., 2012, IRFAN et al., 2013).

These ROS are unstable and partially reduced forms of the atmospheric oxygen ( $O_2$ ), formed during the aerobic cellular metabolism in all cell organelles that have the electron transport chain (the oxygen acts as the final electron acceptor in the chain) or a highly oxidized metabolic rate, such as mitochondria and peroxisomes (CUYPERS, et al., 2010). These ROS results from the transfer of one, two or three electrons to the  $O_2$  molecule, forming respectively the superoxide radical ( $O_2^{\cdot-}$ ), the hydrogen peroxide ( $H_2O_2$ ) or the hydroxyl

radical ( $\text{OH}^{\bullet}$ ) and also by the  $\text{O}_2$  excitation processes which generates the “Singlet” oxygen (SHIEBER and CHANDEL, 2014).

The  $\text{O}_2$  is a stable and not reactive molecule but, contrarily, these ROS are able to oxidize and denature many other molecules and structures due to the unpaired number of electrons in their final outer shell, leading to the cell destruction and thus to the death of tissues and organs such as roots, leaves, fruits or seeds. Consequently, the oxidative stress occurs when there is a serious imbalance between the ROS production and the antioxidant defense system, in any cellular compartment.

## **2. Literature Review**

### **2.1 Antioxidant defense mechanisms**

The cellular defense mechanisms corresponds to the enzymatic and non-enzymatic antioxidant responses capable to directly denature the ROS, such as the superoxide radical ( $\text{O}_2^{\bullet-}$ ) or the hydrogen peroxide ( $\text{H}_2\text{O}_2$ ), and also neutralize the stressor which is in excess in the cell environment (normally a metal ion), thus neutralizing the deleterious effects of the reactive processes.

### **2.2 Enzymatic defense mechanisms**

#### **2.2.1 Superoxide dismutase (SOD, EC 1.15.1.1)**

The superoxide dismutases (SODs) are metalloenzymes that constitute the first enzymatic barrier against the induced oxidative stress, catalyzing the dismutation reaction of the  $\text{O}_2^{\bullet-}$ , forming  $\text{O}_2$  (cell oxygen) and  $\text{H}_2\text{O}_2$  (hydrogen peroxide) (SHIEBER and CHANDEL, 2014) (Figure 1). It is important to mention that the antioxidant process triggered by the SODs can be considered as the primary defense mechanism of the plant cells against the ROS, mainly because it interferes negatively and directly in the Haber-Weiss reaction (Figure 2), which produces the highly reactive and not destructible hydroxyl radical ( $\text{OH}^{\bullet}$ ), from the combination of  $\text{O}_2^{\bullet-}$  and  $\text{H}_2\text{O}_2$  (CUYPERS et al., 2010).



**Figure 1.** Dismutation reaction of the  $\text{O}_2^{\cdot -}$ , forming  $\text{O}_2$  (cell oxygen) and  $\text{H}_2\text{O}_2$  (hydrogen peroxide), by the superoxide dismutases (SODs) metalloenzymes.



**Figure 2.** Haber-Weiss reaction: hydroxyl radical ( $\text{OH}^{\cdot}$ ) formation from the combination of  $\text{O}_2^{\cdot -}$  and  $\text{H}_2\text{O}_2$ .

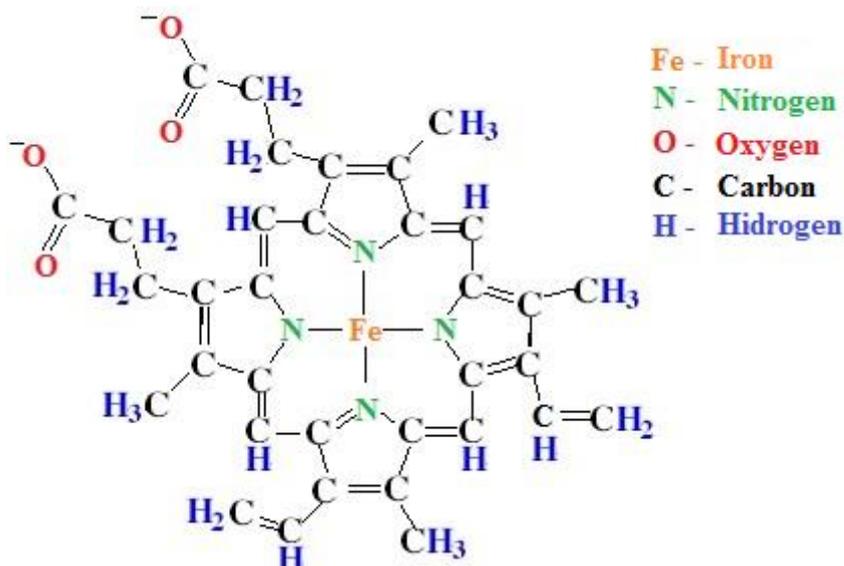
These SODs enzymes are present in different plant tissues and are mainly found in chloroplasts, the mitochondrial matrix and the cytoplasm of cells. Moreover, three different isoforms of this metalloenzymes are known, differing by the metal ion present in the molecule active site and also by the action location in the cells, while the most abundant isoform in vegetables is the SOD which have copper (Cu) and zinc (Zn) in its active site (Cu / Zn - SODs), found mainly in the stroma of chloroplasts and also in the cytosol. Another important isoform are those SODs which have manganese (Mn) on the active site, present in the mitochondrial matrix. On the other hand, the SOD which have Fe (iron) in its active site are rarely found in plants, however this isoform can be associated with chloroplasts (KUMAR et al., 2014). These SOD isoforms are very similar when are compared among different plant species, so that the differences are associated with the isoform found and the concentration in the tissues.

### 2.2.2 Catalase (CAT, EC 1.11.1.6)

The hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) is an unstable and highly reactive molecule and its concentration generally becomes higher in the cellular environment during a stressful condition and also as a result of the enzymatic reaction of the SOD, which can induce the oxidative damage of other molecules and tissues. Consequently, the  $\text{H}_2\text{O}_2$  can be quickly converted into  $\text{H}_2\text{O}$  (water) and  $\text{O}_2$  (cell oxygen) by the specific action of enzymes such as catalase (CAT,

EC 1.11.1.6) and other peroxidases in different cellular compartments (ROYCHOUDHURY et al. 2012).

Catalase was the first antioxidant enzyme discovered and characterized, and basically consists of a polypeptide with an approximate weight of 70 kDa, arranged in a tetrameric molecule, so that each monomer contains a prosthetic heme group with a central atom of Fe, as shown in figure 3 (MHANDI et al., 2010).

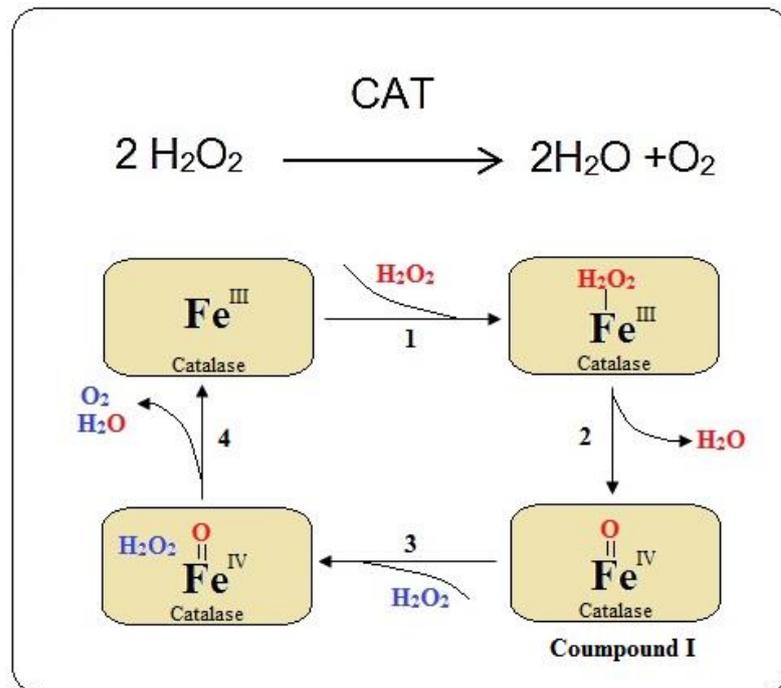


**Figure 3.** Chemical structure of Catalase (CAT).

This enzyme can be found in all living organisms and is the main route of  $\text{H}_2\text{O}_2$  degradation in order to form  $\text{H}_2\text{O}$  and  $\text{O}_2$ , according to the reaction shown in Figure 4. The dismutation reaction occurs in four distinct steps, so that the  $\text{O}_2$  molecule formed at the end of the reaction (step 4) is derived from a single molecule of  $\text{H}_2\text{O}_2$ , while the two  $\text{H}_2\text{O}$  molecules formed during the process (steps 2 and 4) are derived from the combination of the two  $\text{H}_2\text{O}_2$  substrates (Figure 4).

Additionally, both CAT and other peroxidases primarily reduces the  $\text{H}_2\text{O}_2$  molecule by breaking the  $\text{O}_2$  binding site, forming a molecule of water and one intermediate molecule called compound I, where the central atom of Fe (from the CAT enzyme) covalently binds to the liberated oxygen after the  $\text{H}_2\text{O}_2$  reduction (step 2). The CAT specific activity consists of the oxidation of a second  $\text{H}_2\text{O}_2$  molecule to form  $\text{O}_2$  and  $\text{H}_2\text{O}$  (step 4), while Fe is reduced again

to its original oxidation state and the process restarts (step 4 and 1). Finally, it is important to mention that CAT is unique among all the other enzymes responsible for the  $\text{H}_2\text{O}_2$  degradation (peroxidases), mainly because it does not use any equivalent reducing agent, acting directly and efficiently during the entire process (Figure 4).



**Figure 4.** Simplified scheme of the  $\text{H}_2\text{O}_2$  degradation catalyzed by the CAT enzyme, resulting in  $\text{H}_2\text{O}$  (water) and  $\text{O}_2$  (oxygen) formation. Both  $\text{H}_2\text{O}_2$  molecules used in the process as well as their products from their degradation (steps 2 and 4) are represented in the same color. Roman numbers represent the oxidation state of the Fe atom. Adapted from Mhamdi et al., 2010.

Furthermore, CAT has three different types of isoenzymes referred as CAT1, responsible for 80% of the total  $\text{H}_2\text{O}_2$  degradation in the cell (specifically in peroxisomes), formed during the photorespiration process in chloroplasts. CAT2, mainly found in vascular tissues and CAT3, located in the mesophyll of leaves (MHAMDI et al., 2010). This enzyme is essential for the  $\text{H}_2\text{O}_2$  denaturation process in plants that have higher rates of photorespiration, such as all the C3 plants.

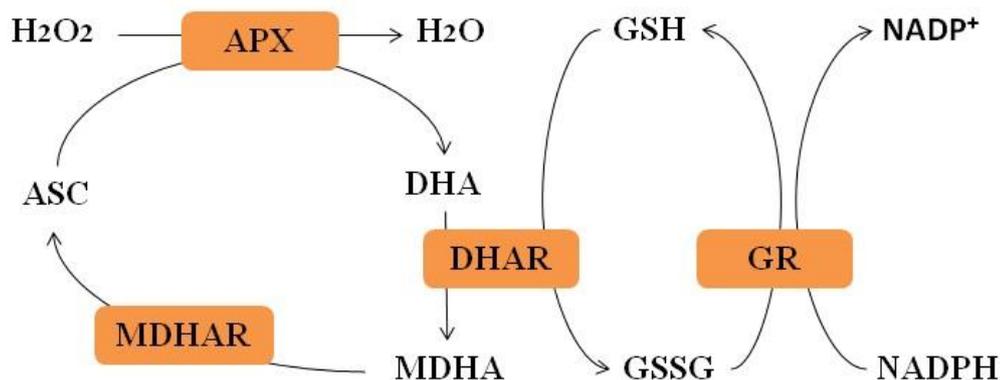
Other enzymes are also responsible for the  $\text{H}_2\text{O}_2$  degradation in the cell environment. Peroxidases are heme proteins (have a heme group in the

molecule, characterized as a central iron atom in an organic porphyrin ring) responsible for the H<sub>2</sub>O<sub>2</sub> molecule reduction concomitantly with the oxidation of a specific substrate, as the ascorbate peroxidase (APX, EC 1.11.1.11), glutathione peroxidase (GPX, EC 1.11.1.9) and guaiacol peroxidase (GPOX activity, EC 1.11.1.7), for example. These peroxidases participate in many essential metabolic processes such as the cell growth regulation, lignification, phenolic oxidation, defense against pathogens, antioxidant defense and protection against various stresses.

The guaiacol peroxidase (GPOX, EC 1.11.1.7) enzyme, for example, is also part of this group and presents basics and acids isoforms in plants. The acid isoform is directly related to the processes involved in the cell wall biosynthesis, including the lignin formation, while its basic isoform participates in the regulation of the AIA (indolylacetic acid, auxins) degradation. In vitro experiments the GPOX catalyze the hydrogen donors oxidation, in the absence of a specific substrate, however, in vivo, the detoxification can become the primary function of certain isoforms.

### **2.2.3 Glutathione (GSH)**

Glutathione (tripeptide formed from the amino acids glutamate, cysteine and glycine) is a major metabolite in the defense system against ROS and its degenerative effects during the enzymatic antioxidant defense, participating along with the ascorbate peroxidase (APX) enzyme in the Halliwell-Asada cycle or ascorbate-glutathione cycle (Figure 5), being necessary to the proper functioning of this cycle. This metabolite can be found in the plant under the oxidized form (GSSG) or also the reduced form (GSH), which is important for non-enzymatic antioxidant defense (see section 2.3), so that the system always prioritize an increased concentration of GSH over the GSSG in the cellular environment.



**Figure 5.** Ascorbate-Glutathione cycle (Halliwell-Asada). Enzymes: ascorbate peroxidase (APX, EC 1.11.1.11), dehydroascorbate reductase (DHAR, EC 1.8.5.1), monodehydroascorbate reductase (MDHAR, EC 1.6.5.4), glutathione reductase (GR, EC 1.6.4.2). Other compounds: hydrogen peroxide ( $\text{H}_2\text{O}_2$ ), water ( $\text{H}_2\text{O}$ ), ascorbate (ASC), dehydroascorbate (DHA), monodehydroascorbate (MDHA), glutathione (GSH), oxidized glutathione (GSSG). Adapted from Inzé and Montago (1995).

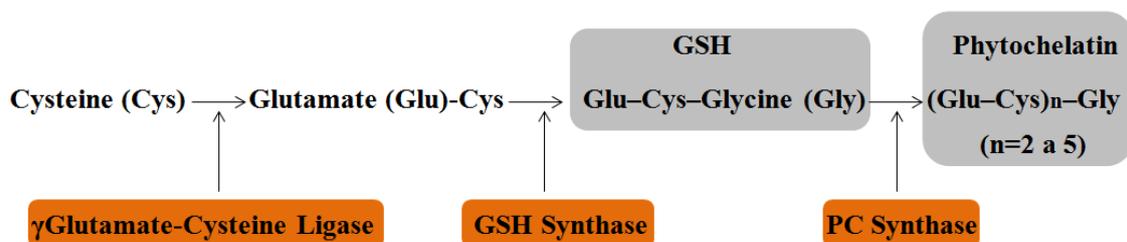
In this cycle, the antioxidant action occurs by the conversion of the  $\text{H}_2\text{O}_2$  molecule into  $\text{H}_2\text{O}$  through the APX enzyme; however, the process is completely dependent upon the glutathione conversion cycle (Figure 5). The enzyme glutathione reductase (GR, EC 1.6.4.2) is responsible for the conversion of GSSG into GSH using as an electron donor the NADPH, while the formed GSH corresponds to the substrate used by the dehydroascorbate reductase (DHAR) enzyme in the ascorbate (ASC) formation pathway, which is used as a substrate for the APX enzyme antioxidant activity (Figure 5) (INZÉ and MONTAGO, 1995).

The glutathione reductase (GR) enzyme is almost of universal occurrence, being found in eukaryotes and in prokaryotes, from heterotrophic and photosynthetic bacteria to higher plants. This enzyme has the prosthetic group FAD (flavin adenine dinucleotide), responsible for catalyzing the electron transfer reaction from NADPH to GSSG, forming the GSH (VOET and VOET, 1995).

### 2.3 Non-enzymatic defense mechanisms

The non-enzymatic mechanisms for cellular detoxification against the ROS are important and act jointly with the enzymatic antioxidant system in order to maintain the cellular redox state. Participate in these processes the phytochelatins (PCs), proline, flavonoids, alkaloids and carotenoids, among others (FOYER and NOCTOR, 2012). The phytochelatins represent the most important route against the metals, semimetals and heavy metals contamination, since these molecules have the ability to complex and inactivate these compounds, storing them into vacuoles in the cell.

The glutathione (GSH) plays an essential role in both the enzymatic antioxidant defense against ROS and the PCs formation, and determining a pathway of action will result in the inactivation of the other corresponding via (ROYCHOUDHURY et al., 2012). The GSH synthesis process involves glutamate, glycine and cysteine, and the PCs are synthesized from the GSH already formed, as shown in Figure 6. The GSH formation consists of two reactions triggered by  $\gamma$ Glutamate-cysteine ligase (EC 6.3.2.2) and GSH synthase (EC 6.3.2.3) enzymes, while the conversion of GSH in phytochelatins occurs by the specific action of the glutathione- $\gamma$ Glutamyl-cysteinyl-transferase (EC 2.3.2.15) enzyme or PC synthase (Figure 6).



**Figure 6.** GSH and Phytochelatin synthesis in plants. Adapted from Inouhe, 2005.

The amino acid proline is part of the compatible osmolytes group, which is responsible for the cell osmotic adjustment maintenance during the water and salt stress and also by the excess of nutrients in the soil, with an important role in plant protection against other different abiotic stresses, such as the exposure

to low temperatures, soil acidity and the heavy metals exposure (SHEVYAKOVA et al., 2013).

Furthermore, different studies demonstrate that this osmolyte is able to actively clean the ROS from the cell during the oxidative stress, besides conferring protection and stabilization of some cellular structures such as membranes, proteins and also enzymes during severe stresses (BANU et al., 2009). It is important to note that the products resulting from the proline molecule catabolism, after the alleviation of the stress condition, will be used in the oxidative phosphorylation process in mitochondria, generating molecules of ATP (adenosine 5'-triphosphate), which is important in the cellular recovery processes (ASHRAF and FOOLAD, 2007).

Additionally, the proline synthesis in plants occurs in chloroplasts and also in the cytoplasm, with the glutamic acid as a precursor. This acid is converted to glutamate  $\gamma$ -semialdehyde acid (GSA) by the  $\Delta$ 1-pyrroline-5-carboxylate synthetase (P5CS) enzyme, which in turn is spontaneously converted to the  $\Delta$ 1-pyrroline-5-carboxylate (P5C), which it is then reduced to proline by the P5C reductase enzyme (VERBRUGGEN and HERMANS, 2008).

In summary, the non-enzymatic antioxidant system operates through different mechanisms responsible for the direct elimination of the stressor, cellular homeostasis regulation or even with molecules that acts directly as antioxidants, such as flavonoids, which inactivate the ROS, and alkaloids or carotenoids which are capable to donate H<sup>+</sup>, which mitigates the negative effects of the oxidation processes caused by the reactive oxygen species.

## **2.4 Cadmium contamination and the induced stress**

Industrial activities involved in metal smelting and the increasing in mining processes on a global scale can be considered as the major sources of environmental contamination. Heavy metals are among the main pollutants generated by these sources and represents a major threat to living organisms because of its high toxicity, persistence in the environment and bioaccumulation (LIU et al., 2010, 2011).

However, several other anthropogenic activities are responsible to release into the environment a large amount of these pollutants, which accumulates in

the soil and limit the plant productivity (ZHOU et al., 2013). According to Xu et al. (2013), the presence of heavy metals in agricultural soil, such as Cadmium (Cd), occurs as a consequence to the excessive use of phosphate fertilizers and pesticides, which allow its assimilation in crops and the consequent transfer through the food chain, representing a great danger to human and animal health.

Cadmium is the most toxic of all heavy metals. It is responsible to cause numerous health problems to humans (XU et al., 2013), because of its accumulation in different organs such as kidneys, liver and central nervous system, with a half-life that can reach fifteen to thirty years in the organism (GONÇALVES et al., 2012). Due to its strong global demand, approximately 30000 tons of Cd are released into the atmosphere every year, while 4000 to 14000 tons of this total amount comes only from industrial activities, such as the rechargeable batteries, alloys for welding, pigments and paints industry (ATSDR , 2012).

The presence of Cd in high concentration in the soil inhibits the plant growth (CHENN et al., 2011; IRFAN et al., 2013) causes a decrease in the chlorophyll and carotenoids content (LIU et al., 2010, 2011), affect the proper function of chloroplasts and the CO<sub>2</sub> fixation, induces the antioxidant enzymes inactivation and the ROS overproduction, leading to the lipid peroxidation process and the consequent oxidative stress (XU et al., 2013).

It is important to mention that Cd is not capable to induce the ROS overproduction directly but reacts with the Sulphur present in the thiol group in the cysteine and other different proteins, which results in a lower GSH concentration in the cell environment. This process results to a deflected defense system against the oxidative stress (CUYPERS et al., 2010). Furthermore, the increased concentration of iron (Fe) ions in the cellular environment, due to its replacement by Cd in different proteins, may also induce the production of OH<sup>-</sup> radicals, which are highly reactive and resistant to cellular defense mechanisms, resulting in a severe oxidative stress (DORTA et al., 2003).

Great number of proteins and enzymes have metals in their active sites, the capacity of Cd to replace them due to chemical similarities molecules it is notable. Consequently, the antioxidant system metalloenzymes, such as the

SOD, can be inactivated by this replacement (GONÇALVES et al., 2008). According to Guimarães et al. (2008) CAT levels in peroxisomes are also drastically reduced in the presence of 50 mM of Cd due to its oxidation, which also impairs the oxidative stress management.

Leaf chlorosis is also a regular symptom of the Cd induced stress, which demonstrates that the cellular photosynthetic apparatus is severely affected by this heavy metal. The molecular structure of chlorophyll is characterized by the presence of a central Magnesium (Mg) atom and its replacement by Cd alters its stability, decreasing the absorption of the light energy and the consequent production of carbohydrates (CHENN, et al., 2011; XU et al., 2013). The leaf chlorosis appearance can also be explained by the decrease in the amount and replication of chloroplasts in the cell, deficiency of phosphorus (P), manganese (Mn) and a negative influence on the activity of the enzymes related to the production of chlorophyll (GUIMARÃES et al., 2008).

## **2.5 Selenium as a stress alleviation strategy in plants**

Selenium (Se) is an essential micronutrient for humans and animals and exerts important roles in the organism, participating on the antioxidant defense system (CATANIA et al., 2009). The absence or inadequate ingestion of this nutrient in the diet can cause health problems related to malnutrition. In vegetables, Se in low concentrations exerts positive physiological effects, such as growth improvement, reduction of ROS concentration and lipid peroxidation, improves the accumulation of starch and sugars and provides a delayed senescence (XUE et al, 2001; FENG and WEI, 2012). Despite not been recognized as a nutrient, some studies demonstrate the beneficial effects to the plants, such as the increased antioxidant capacity after Se application (XUE et al., 2001; ZEMBALA et al, 2010) and also its ability to reduce the heavy metals availability while alleviate its toxic effects (MUÑOZ et al., 2007) (see chapters 3 and 4 for more detailed information regarding selenium and heavy metals).

This mineral can be found under different forms in the soil, such as selenite ( $\text{SeO}_3^{2-}$ ), selenate ( $\text{SeO}_4^{2-}$ ) and also under its organic molecules like SeCys (selenium atom bonded to a cysteine molecule) and SeMet (selenium

atom bonded to a methionine molecule), which are assimilated by the roots (NOWAK, 2013).

Furthermore, the presence of Se in the intracellular environment induces an increased levels and activity of different antioxidant enzymes, thereby regulating the concentration of ROS (see chapter 3). The glutathione peroxidase (GSH-Px) enzyme, for example, demonstrates an enhanced activity after the Se application, eliminating effectively the H<sub>2</sub>O<sub>2</sub> and alleviating the oxidative stress (ZEMBALA et al, 2010; FENG and WEI, 2012). According to Feng and Wei (2012), Se can also stimulate the spontaneous dismutation of O<sub>2</sub><sup>-</sup> into H<sub>2</sub>O<sub>2</sub> radical, without the catalytic process exerted by the SOD enzyme. However, these authors report that excessive Se concentrations can lead to an imbalance in the levels of GSH, thiol radicals (-SH) and also the NADPH in the cell, which are important for the ROS elimination and stress defense mechanisms.

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## **CHAPTER 2 - SULFUR METABOLISM AND STRESS DEFENSE RESPONSES IN PLANTS\***

### **Abstract**

Sulfur management is an important issue in crop plant nutrition. Sulfur has a role in fundamental processes such as electron transport, structure and regulation. It is also associated with photosynthetic oxygen production, abiotic and biotic stress resistance and secondary metabolism. Sulfate uptake, reductive assimilation and integration into cysteine and methionine are the central processes that direct oxidized and reduced forms of organically bound S into their various functions. Sulfur-containing defense compounds that are crucial for plant survival during biotic and abiotic stress include elemental sulfur, hydrogen sulfide, glutathione, phytochelatins, S-rich proteins and various secondary metabolites. Formation of these compounds in plants is closely related to the supply, demand, uptake and assimilation of S. This review will highlight the role of S during the stress response in plants and the relationship between S metabolism and primary S nutrition.

### **Keywords**

Abiotic stress, Antioxidants, Oxidative stress, Plant nutrition, Sulfur uptake and metabolism

### **Introduction**

Environmental variation triggers plant acclimation, adaptation or death. Natural and anthropogenic activities induce biotic and abiotic stresses during agricultural and forestry operations. Plant metabolism is often damaged by toxic compounds and hazardous chemicals present in soils and air (Su et al. 2014; Iannone et al. 2015). Thus, acclimation and adaptation processes are crucial to plant survival, and the identification and understanding of plant tolerance mechanisms are of major importance. Stress can be defined as any alteration in normal plant growth conditions (Boaretto et al. 2014). Some of these alterations

are related to temperature, salinity, water supply, ozone, soil acidification and heavy metal toxicity, among others (Azevedo et al. 1998; Monteiro et al. 2011; Bulbovas et al. 2014; Nogueirol et al. 2015). Studies of plants have been conducted to evaluate the effects of these changes on growth and development. Plants possess very efficient defense pathways that allow the scavenging of reactive oxygen species (ROS), protecting the cells from oxidative damage (Gratão et al. 2005).

The primary function of regulatory mechanisms is to manage fluxes of sulfur (S) in response to developmental and environmental changing conditions. The goal for the plant is to optimize the use of available S to match the demands for growth and development, and resistance to stress (Hawkesford 2012). Sulfur assimilation starts from the uptake of external sulfate by the activity of sulfate transporter (SULTR) in roots. On the other hand, plants are able to use foliar absorbed  $H_2S$  as S source for growth, especially under conditions where the S to the roots is limited (Koralewska et al. 2008). Sulfate is activated by Adenosine-5'-triphosphate sulfurylase (ATPS, EC 2.7.7.4) and then catalyzed by Adenylyl-sulfate reductase (APR, EC 1.8.99.2) and sulfite reductase (EC 1.8.7.1) to produce sulfide. Regarding the primary metabolism in plants, nitrate and sulfate need to be reduced prior to their incorporation into various essential organic nitrogen (N) and S compounds. The uptake and assimilation of S and N are strongly interrelated, since the major proportion of the reduced N and S in plants is incorporated into amino acids and subsequently into proteins (Stulen and De Kok 2012). An important coordination with C/N metabolism occurs at the level of cysteine and methionine biosynthesis, with the cysteine synthase complex (serine acetyltransferase (SAT, EC 2.3.1.3) and O-acetylserine(thiol)lyase (OASTL, EC 4.2.99.8)) acting as both a sensor and a regulator, mediated by a reversible association/dissociation of the complex. SAT is active when associated with OASTL, but inactive when dissociated. As the dissociation is promoted by excess OAS, the complex effectively senses both OAS and S availability and self regulates further OAS production accordingly (Hawkesford 2012). Thus, OAS is a signal mediating between substrate availability and flux. OASTL, which is in excess, will always catalyze synthesis of cysteine given availability of OAS and sulfide. The formation of these compounds is closely related to the

supply, demand, uptake and assimilation of S in plants. In this review, we provide essential information about the role of these S-containing compounds, particularly with regard to abiotic stress acclimation and plant tolerance adaptations.

### **Sulfur uptake and assimilation**

Most soils currently used for agricultural and forest crops are naturally low in fertility, and chemical fertilization should be implemented to provide the crop requirements for essential nutrients such as nitrogen (N), phosphorus (P), potassium (K) and S. S uptake is directly driven by demand. Inadequate S nutrition can cause the inefficient use of other nutrients, such as carbon (C) and N, leading to deficiencies and decreases in protein biosynthesis, chlorophyll content and eventually crop yield (Lunde et al. 2008; Mazid et al. 2011; Iqbal et al. 2013). On the other hand, environmental pollution from sulfur dioxide (SO<sub>2</sub>), H<sub>2</sub>S, sulfite (SO<sub>3</sub><sup>2-</sup>) and sulfate (SO<sub>4</sub><sup>2-</sup>) is a serious global problem and can be toxic to plants (Krischan et al. 2012).

S is a component of proteins, the amino acids cysteine (Cys) and methionine (Met), vitamins (biotin and thiamin), cofactors (Co-A and S-adenosyl methionine, SAM) and a range of secondary metabolites (Mazid et al. 2011). S is an essential macronutrient for living organisms and has multiple roles in plant development, including catalytic, regulatory and structural functions (such as in protein disulfide bonds, cellular membrane SO<sub>4</sub><sup>2-</sup> esters and electron transport through Fe-S groups). S-containing compounds such as PCS and GSH also have a role in trace element homeostasis (Na and Salt 2011). S is an important substrate/reductant in reactions during abiotic stress processes; GSH, a major antioxidant in plant stress defense and the major non-protein S source in plants (Kopriva and Rennenberg 2004; Ghelfi et al. 2011; Rennenberg and Herschbach 2012; Seth et al. 2012) is present in all root and leaf cell compartments, with the exception of the apoplast in the absence of stress (Josefczak et al. 2012).

S is taken up from the soil solution predominantly as SO<sub>4</sub><sup>2-</sup> in an energy-dependent process mediated by specific membrane-bound SO<sub>4</sub><sup>-</sup> transporters (Buchner et al. 2004; Davidian and Kopriva 2010). Plants can also obtain

organic forms of S, such as S-containing amino acids, organic  $\text{SO}_4^{2-}$  and elemental S, from the soil solution. Although of less significance, S in the form of atmospheric  $\text{SO}_2$  can be absorbed by plant leaves and fruits (Mazid et al. 2011), and atmospheric  $\text{H}_2\text{S}$  can be absorbed through leaf stomata (Riemenschneider et al. 2005).

The translocation of  $\text{SO}_4^{2-}$  into plastids for assimilation, storage in vacuoles, and long-distance transport among organs requires specific transporters; the mechanism of plasma membrane transport is proton-coupled co-transport (Buchner et al. 2004). There are approximately 12 to 16 reported genes encoding  $\text{SO}_4^{2-}$  transporters (SULTR) in plant species. SULTR proteins can be classified according to their protein sequence similarities into SULTR 1 to 5 (for a review: Buchner et al. 2004; Davidian and Kopriva 2010). These transporters can move  $\text{SO}_4^{2-}$  into the plant when soils are deficient in S; SULTR1;1 (skilled in trace  $\text{SO}_4^{2-}$  uptake) and SULTR1;2 (major component) have been identified on root hairs and root epidermal and cortical cells of knockout mutants of *Arabidopsis* (Takahashi et al. 2011). From the structural perspective, SULTR are members of a family of membrane-bound solute transporters provided to 12 domains that cross the plasma membrane (Takahashi et al. 2012). SULTR2;1 is a low-affinity SULTR that appears to be involved in  $\text{SO}_4^{2-}$  translocation from roots to shoots. The initial uptake by the root epidermis and cortical cells depends on high-affinity transport, as well as on the displacement of  $\text{SO}_4^{2-}$  into tissues and/or organs, which can be very specialized. Low-affinity transporters (in *Arabidopsis*: AtSULTR2;1 and AtSULTR2;2) were shown to be more related to vascular transport of  $\text{SO}_4^{2-}$  and to the regulation of the cytoplasmic  $\text{SO}_4^{2-}$  concentration during accumulation in the vacuole (Buchner et al. 2004; Davidian and Kopriva 2010).

The coordination and the dynamics between the pathways of short- and long-distance transporters require specific signaling mechanisms to control and regulate a range of genes encoding specific proteins involved in S uptake, transport and assimilation (Davidian and Kopriva 2010). Expression of these transporters is regulated by internal and external sulfate signals and by the N, C and S reductive assimilation pathways, including phytohormones and variable metabolites (Gojon et al. 2009; Davidian and Kopriva 2010).

After uptake,  $\text{SO}_4^{2-}$  is assimilated into Cys, an amino acid at the intersection of primary metabolism, protein synthesis and the formation of low molecular weight S-containing defense compounds (Rausch and Wachter 2005; Gotor et al. 2014). Excess  $\text{SO}_4^{2-}$  transported to leaves is stored in vacuoles and constitutes a large S reserve for plant metabolism (Iqbal et al. 2013).

Cys synthesis is required for GSH biosynthesis and occurs in plastids, mitochondria and the cytosol. Activated  $\text{SO}_4^{2-}$  also forms 3'-phosphoadenosine 5'-phosphosulfate (PAPS, EC 1.8.99.2), the S-donor for sulfonation, sulfation or sulfuryl-transfer reactions; the reaction is catalyzed by sulfotransferases (STs) that play important roles in cell communication, plant development and defense (Negishi et al. 2001). In parallel with S assimilation, some 'dissimilatory' reactions, such as the release of  $\text{H}_2\text{S}$  from Cys 1 and 2, might contribute to defense processes (Rausch and Wachter 2005; Lisjak et al. 2011; for more information on  $\text{H}_2\text{S}$  see Lisjak et al. 2010 and Lisjak et al. 2013).

$\text{SO}_4^{2-}$  assimilation is critical for providing reduced S for various cellular redox processes (for review: Jacob and Anwar 2008; Takahashi et al. 2011) and for the synthesis of GSH (Kopriva and Rennenberg 2004; Ghelfi et al. 2011; Seth et al. 2012). GSH has many distinct functions in plant cell metabolism, including controlling gene expression linked to the redox state of cells or subcellular compartments; being an important reducing cofactor of many enzymes related to ROS detoxification (for review: Noctor 2006; Foyer and Shigeoka 2011); and directly controlling the S assimilation pathway. Reduced forms of S decrease significantly during S uptake and assimilation (Kopriva 2006; Chan et al. 2013).

Many studies have emphasized the importance of N in S assimilation and plant stress defenses (Kopriva and Rennenberg 2004; Siddiqui et al. 2008; 2012; Salvagiotti et al. 2009; Carfagna et al. 2011).  $\text{SO}_4^{2-}$  assimilation declines under nitrate ( $\text{NO}_3^-$ ) deficiency, and the capacity to reduce  $\text{NO}_3^-$  and the activity of nitrate reductase (NR, EC 1.6.6.1-3) are diminished in plants that are starved for  $\text{SO}_4^{2-}$  (Kopriva and Rennenberg 2004; De Bona et al. 2011). In tobacco plants, for example,  $\text{SO}_4^{2-}$  uptake by the roots was drastically reduced when NR was inactivated (Kruse et al. 2007; Siddiqui et al. 2012). Moreover, in N-starved plants, the activities of enzymes responsible for S assimilation and the mRNA levels associated with related genes decreased, but the addition of two distinct

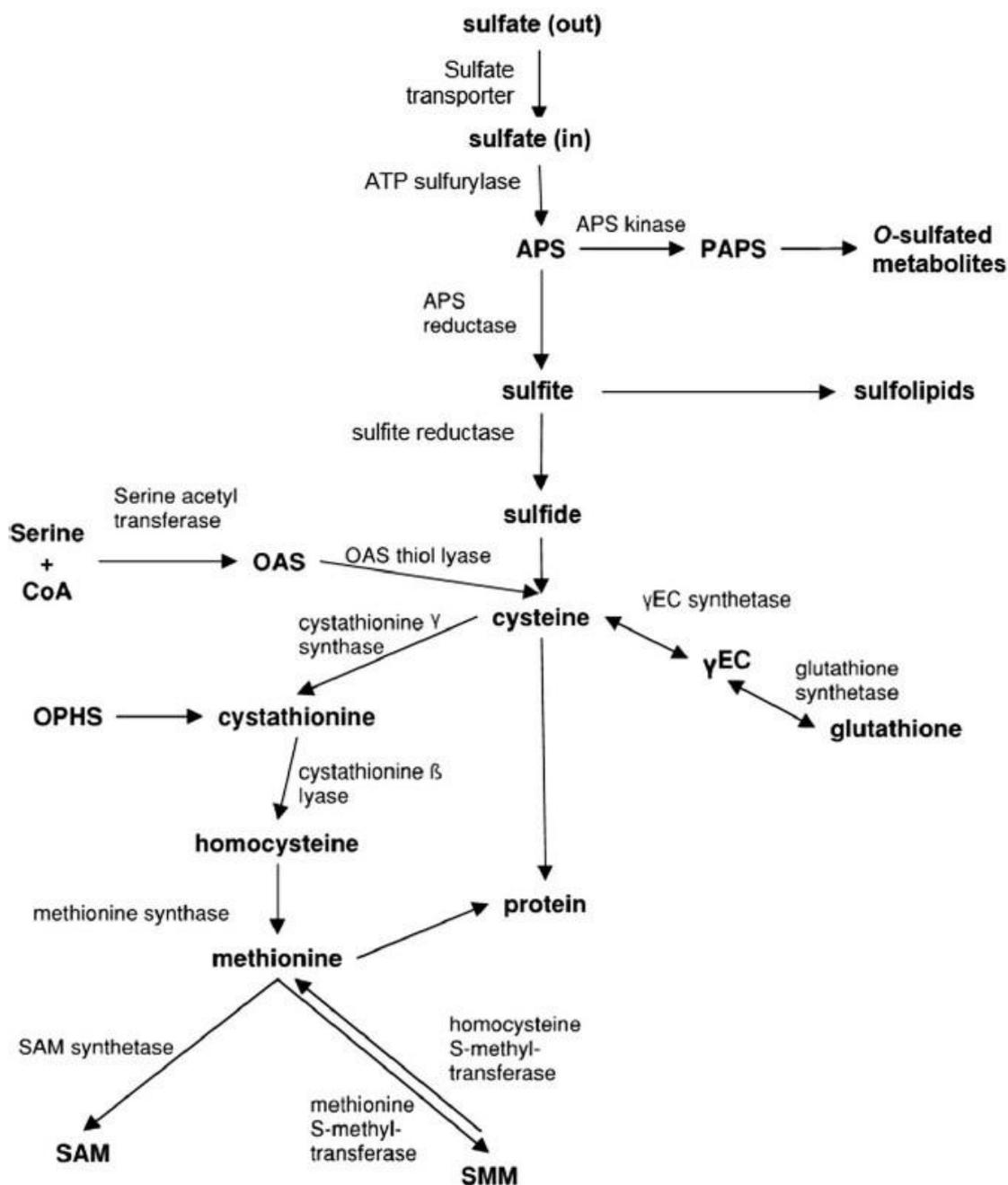
N sources ( $\text{NO}_3^-$  and ammonium,  $\text{NH}_4^+$ ) quickly restored the enzymatic function (Koprivova et al. 2000).

In cereal plant species, increases in S fertilization can enhance the efficiency of N uptake and use because S is a constituent of some enzymes involved in N metabolism (Salvagiotti et al. 2009; De Bona et al. 2011). Wheat plants exposed to distinct N and S levels revealed an important relationship between N and S. When N was less limiting, N uptake was high at the highest S concentration. This increase in uptake was more directly correlated with recovery efficiency than with internal use efficiency (Salvagiotti et al. 2009). Nonetheless, it has been demonstrated that adenosine 5' phosphosulfate reductase (APR, EC 1.8.99.2), the key enzyme for  $\text{SO}_4^{2-}$  assimilation, is regulated by carbohydrates (Lewandowka and Sirko 2008; Chan et al. 2013). The availability of Cys is another crucial factor in GSH synthesis, but an adequate supply of glutamate and glycine is also important (Kopriva and Rennenberg 2004).

The activity and expression of  $\text{SO}_4^{2-}$  transporters and APR in plants are modulated by their S status and the demand for growth (Koralewska et al. 2008). A key enzyme of plant S metabolism, O-acetylserine(thiol) lyase (OAS-TL, EC 2.5.1.47, also named cysteine synthase), catalyzes the formation of Cys from the sulfide ion ( $\text{S}^{2-}$ ) and O-acetylserine, as illustrated in Fig. 1 (Youssefian et al. 2001). Cys biosynthesis can be regarded as the exclusive function of S reduction in plants and is a key limiting step in the production of GSH and in tolerance to biotic and abiotic stresses (Youssefian et al. 2001; Mera et al. 2014). OAS-TL plays a key role in the synthesis of Cys and GSH, which are required for regulation of plant responses in response to oxidative stress (Youssefian et al. 2001; Gotor et al. 2014).

Studies of barley plants demonstrated that N or S deficiency altered GSH levels in leaves. In N- and S-starved plants, GSH levels doubled, and the Cys concentration was shown to increase by 50% (Carfagna et al. 2011). In *Brassica juncea*, N and S enhanced the activity of adenosine triphosphate-sulfurylase (ATP-S, EC 2.7.7.4), a key enzyme in the S assimilation pathway, which activates  $\text{SO}_4^{2-}$  via an ATP-dependent reaction. In response to an environmental N deficit, the addition of  $\text{NO}_3^-$  or  $\text{NH}_4^+$  rapidly improved ATP-S and OASTL function (Siddiqui et al. 2012). N addition positively affected OASTL

activity in plant roots, and a precise sequence of N metabolism and S assimilation is necessary to provide the N precursors for Cys biosynthesis (Carfagna et al. 2011). Thus, S assimilation is significantly related to assimilation of  $\text{NO}_3^-$  and C (Yoshimoto et al. 2007). Some transcriptional factors responsible for  $\text{SO}_4^{2-}$  uptake and assimilation have been identified, demonstrating a relationship between mRNA levels (to APR and ATP-S), protein biosynthesis and enzyme activity (Koprivova et al 2000; Hesse et al. 2003; Davidian and Kopriva 2010).



**Figure 1.** Biosynthetic pathways for S-containing amino acids and their derivatives. A key enzyme of plant S metabolism, OAS-TL, also named cysteine synthase, catalyzes the formation of Cys from the sulfide ion ( $S^{2-}$ ) and O-acetylserine. APS: adenosine-5'-phosphosulfate; PAPS: 3'-phosphoadenosine-5'-phosphosulfate;  $\gamma$ -EC:  $\gamma$ -glutamyl-cysteine; OAS: O-acetylserine; CoA: acetyl coenzyme A; SAM: S-adenosylmethionine (S-AdoMet); SMM: S-methylmethionine (modified from Hawkesford 2005; Koprivova and Kopriva 2014).

## Phytohormones in S assimilation

Some studies have reviewed the importance of the relationship between S assimilation and phytohormones (Maruyama-Nakashita et al. 2004; Maruyama-Nakashita et al. 2005; Kopriva 2006; Khan et al. 2013). Phytohormones are essential for plant acclimation and adaptation to environmental changes (Peleg and Blumwald 2011). The signaling pathway of phytohormones is linked to efficient nutrient use, plant defense pathways and plant developmental processes and metabolism (Fatma et al. 2012). Phytohormones such as cytokinins (CK), gibberellins (GA), auxins (AU), ethylene (ET), jasmonates (JA) and salicylic acid (SA) can interact with mineral nutrients under both normal and stress conditions, playing an essential role in salt stress control and affecting plant growth recovery, cell division, germination and seed production, even when applied exogenously (Fatma et al. 2012).

CK is known to be related to the N cycle and metabolism, being involved in N and P assimilation. Therefore, it seems that CK has a general role in the assimilation of nutrients, including S (Kopriva 2006).

The expression of indole-3-acetic-acid-amido synthetase (IAA-amido synthetase, EC 6.3.2.-) in rice seedlings was correlated with an increase in expression of LEA (late embryogenesis abundant) genes, which have been shown to promote drought stress tolerance (Zhang et al. 2009). The expression of many other genes related to auxin synthesis and to enzyme biosynthesis, transporters and activity can also be regulated by ET, whilst auxin seems to affect ET biosynthesis (Peleg and Blumwald 2011). ET has an important role in improving N use efficiency, photosynthetic rates and plant growth in N-optimal and N-deficient *Brassica juncea* plants (Khan et al. 2008). ET signaling increases with SA and/or JA, resulting in expression of a wide range of genes related to plant defense.

GA enhances the effects of salt stress in soybean, perhaps by regulating the availability of other phytohormones. Abscisic acid (ABA) reduced Na<sup>+</sup> and Cl<sup>-</sup> content and, consequently, the Na<sup>+</sup>/K<sup>+</sup> ratio and increased the Ca<sup>2+</sup>, K<sup>+</sup>, soluble sugar and proline contents in rice crops (Khorshidi et al. 2009; Iqbal et al. 2013). Phytohormones can alleviate the effects of salt stress and improve

plant tolerance by influencing proline metabolism. N and/or Ca<sup>2+</sup> accumulation are altered, and a link between Ca<sup>2+</sup> signaling and SA content improves proline content (Du et al. 2009; Al-Whaibi et al. 2012). Proline accumulation seems to be regulated by ABA-dependent and ABA-independent pathways (Iqbal et al. 2013).

ABA appears to be the phytohormone that responds most rapidly to plant stress. ABA synthesis and expression of ABA-inducible genes cause stomatal closure in plants under drought stress (Peleg and Blumwald 2011). Many genes associated with ABA biosynthesis and ABA receptors were identified in *Arabidopsis* (Brocard-Gifford et al. 2004) and maize (Peleg and Blumwald 2011).

Plants exposed to SA and ABA exhibited higher GSH concentrations and glutathione reductase activity (GR, EC 1.8.1.7, also named glutathione-disulfide reductase, GSR) (Kopriva 2006; Nazar et al. 2011; Pál et al. 2014). This result confirms the relationship among GSH content, S assimilation and stress defense. ABA is related to environmental stress adaptation and SA plays a key role in plant stress tolerance, expression of genes that encode chaperones, heat-shock proteins and antioxidants, and genes related to secondary metabolites (Kopriva 2006; Pál et al. 2014; Peleg and Blumwald 2011; for a review on the role of phytohormones in stress tolerance, see Carvalho et al. 2011). ABA increases the levels of mRNA encoding cytosolic OAS-TL, a key enzyme in S assimilation and metabolism (Kopriva 2006).

The increased demand for GSH can be met by activation of pathways involved in S assimilation and Cys biosynthesis. Microarray analyses have indicated that the messenger ribonucleic acid (mRNA) transcript level of OAS-TL was upregulated in response to zinc stress in *A. thaliana* (Becher et al. 2004) and was constitutively elevated in its metal-tolerant relative, *A. halleri* (Weber et al. 2004).

Many studies address the production and influence of GSH and Cys in relieving biotic and abiotic stress in plants, such as in the protective response to oxidative stress resulting from various factors (Ruiz and Blumwald 2002; Rahoui et al. 2014; Zhang et al. 2014). Under abiotic stress, GSH demand increases (to promote stress tolerance), activating enzymes in the S assimilation pathway. Enzymatic activity, genetic manipulation of enzymes involved in S assimilation

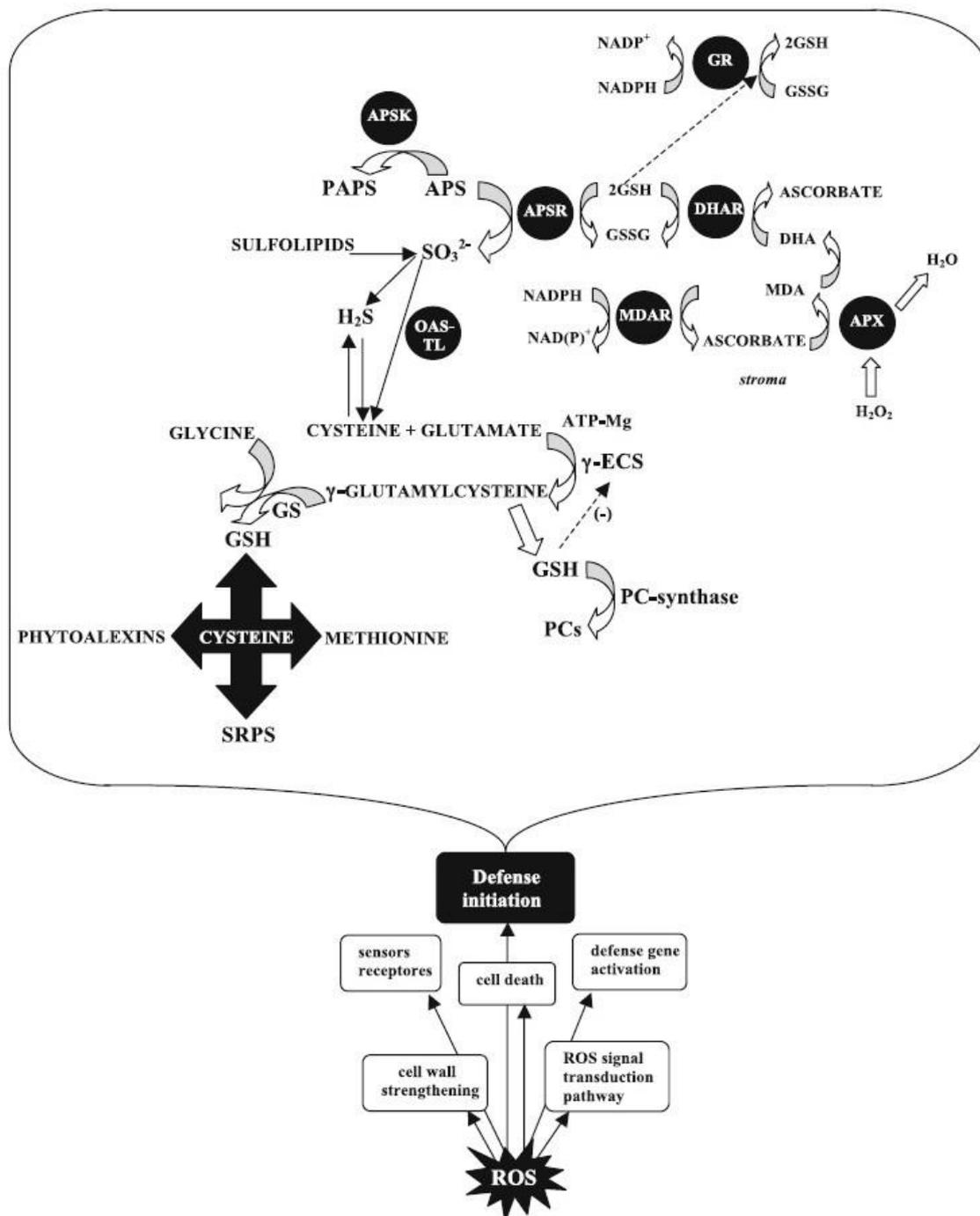
and external S supply can lead to abiotic stress tolerance in plants (Rennenberg et al. 2007; Nazar et al. 2011).

Cys content and GSH content and biosynthesis at the transcriptional level are regulated by JA (known to participate in the transduction of stress response) and methyl jasmonate (MeJA) (Shan and Liang 2010; Gfeller et al. 2011). Treatment of *Arabidopsis* with MeJa increased the mRNA levels corresponding to many genes involved in S assimilation and GSH synthesis, without affecting the content of S metabolites or of mRNA levels associated with  $\text{SO}_4^{2-}$  transporters (Kopriva 2006). These signaling compounds also increased the accumulation of mRNA associated with genes that are involved in S metabolism (Takahashi et al. 2011) and associated with S deficiency. This suggests that JA has a signaling role for inducing S assimilation under S deficiency (Sasaki-Sekimoto et al. 2005; Srivastava et al. 2013).

Several limitations of the suggested stress response have to be emphasized because of the importance of the GSH system with respect to other components of the photoprotective and antioxidative defense systems. The roles of JA, SA, GA and ABA in regulating S assimilation enzymes are crucial for acquired abiotic stress tolerance in plants.

### **Sulfur oxidation states in the cell**

Under stressful conditions, the induction or increase in S compounds related to plant defense is also crucial for detoxification of excessive ROS (Noctor et al. 2012), resulting in diverse modes of action for S-containing secondary metabolites (Fig. 2). In the cell, S can be found in several oxidation states mediated by different enzyme families. For example, the sulfotransferase protein family (SOT, EC 2.8.2) catalyzes the transfer of sulfonate molecules in the highest oxidation state to an appropriate hydroxyl group of many substrates using 3'-phosphoadenosine 5'-phosphosulfate (PAPS) as the sulfuryl donor. The SOT also catalyzes the sulfonation of a wide range of compounds and produces  $\text{SO}_4^{2-}$  esters and conjugates (Klein and Papenbrock 2004).



**Figure 2.** Sulfur-containing defense compounds. GR: Glutathione Reductase, APX: Ascorbate Peroxidase, GPX: Glutathione Peroxidase, MDHAR: Monodehydroascorbate Reductase, DHAR: Dehydroascorbate Reductase, APS: 5'-adenylylsulfate, PAPS: 3'-phosphoadenylylsulfate, APSK: APS kinase, APSR: APS reductase, OAS-TL: O-acetylserine thiol lyase (modified from Rausch and Wachter 2005; Mendoza-Cózatl et al. 2005).

Cys is synthesized in the last stage of photosynthetic assimilation of sulfate in plant cells and is the first organic compound containing reduced S. Cys has essential roles in the function, structure and regulation of proteins, being the precursor of many important S-containing compounds involved in plant defense signaling and plant development (Gotor et al. 2014).

Macroarray analysis revealed an integrated signaling pathway in plant defense gene expression in *Arabidopsis*, with upregulation by MeJA of several genes related to S metabolism, GSH, Cys and Met biosynthesis and S-rich defense proteins involved in GS metabolism (Jost et al. 2005; Guo et al. 2013).

### **Methionine biosynthesis**

Met synthesis links Cys biosynthesis to the aspartate-derived amino acid biosynthetic pathway (for review, see Hawkesford and Kok 2006; see also other papers published by the group of M. Hawkesford; for reviews on the aspartate pathway, see Azevedo et al. 1997; 2006). Biosynthesis of Met from Cys involves three enzymatic steps. O-phosphohomoserine (OPHS, EC 2.7.1.39) derived from the aspartate pathway is a common substrate for both threonine and Met synthesis, catalyzed by threonine synthase (TS, EC 4.2.3.1) and methionine synthase (MS, EC 2.1.1.13), respectively (Azevedo et al. 1997). Cystathionine  $\gamma$ -synthase (CgS, EC 2.5.1.48) catalyzes the synthesis of cystathionine from Cys and OPHS by trans-sulfuration (Hawkesford 2005). Cystathionine is then converted to homocysteine (a  $\beta$ -cleavage reaction) by cystathionine  $\beta$ -lyase (CbL, EC 4.4.1.8). Homocysteine is exported from chloroplasts and converted (by methylation) into Met through MS activity. The activity of CgS and TS will influence biosynthesis of Met and threonine, respectively. CgS activity almost certainly has a large effect on flux and is most likely feedback-regulated by Met or a derivative (Azevedo et al. 1997; 2006). Similarly, TS activity is regulated by S-adenosylmethionine (SAM, also known as S-AdoMet), which is a derivative of Met (Azevedo et al. 1997; Wang and Frey, 2007). These controls effectively maintain the Met pool within close constraints. Rather small gene families encode the proteins of this pathway (CgS: 2 genes, CbL: 1 gene, MS: 3 genes). Furthermore, Met is a gateway to

many other important S-containing metabolites, including S-methylmethionine (SMM), SAM and dimethylsulfonio-propionate (DMSP). SMM is a transportable derivative of Met. It can revert to Met by donating a methyl group to homocysteine in a reaction catalyzed by homocysteine S-methyltransferase (HMT, EC 2.1.1.10). Under some circumstances, SMM may be the major S constituent of the phloem sap, and it has a role in delivering S to sink tissues such as seeds (Hawkesford 2005).

SAM is one of the most important S-compounds in plant metabolism (Azevedo et al. 2006); it is involved in many processes and is the main methyl donor involved in transmethylation of proteins, nucleic acids, polysaccharides and fatty acids (Ma et al. 2003).

SAM is also a precursor of the polyamine (PA) synthetic pathway (spermidine/spermine biosynthesis pathway) and of nicotinamide biosynthesis (important for Fe nutrition in plants). SAM is known as the 'activated Met form' (Bürstenbinder and Sauter 2012), and up to 80% of the Met pool may be converted to SAM at the expense of adenosine triphosphate (ATP) utilization (Ravanel et al. 1998; Iqbal et al. 2013) by SAM synthetase (SAMS, EC 2.6.1.6, five genes in the family). Spermidine and spermine have multiple proposed roles, including stress response, pH regulation, DNA replication and senescence processes. Consumption of SAM may increase S demands to meet these needs, although ultimately Met is recycled. SAM is also the precursor for ET, a potent modulator of plant growth and development that is involved in stress signaling (Wang et al. 2002). The synthesis of ET from SAM is catalyzed by 1-aminocyclopropane-1-carboxylic acid synthase (ACCS, EC 4.4.1.14) and ACC oxidase (ACCO, EC 1.14.17.4). Met is not consumed in this reaction but is recycled, resulting in no net S demand. A side product of the final biosynthetic step for ET is cyanide, which is detoxified to  $\beta$ -cyanoalanine by  $\beta$ -cyanoalanine synthase (CAS, EC 4.4.1.9), an isoform of OAS-TL (Hatzfeld et al. 2000). DMSP is produced in high concentrations in many marine algae and in some higher plants, such as marsh grasses in the genus *Spartina*, sugar cane and *Wollastonia biflora*. It is synthesized in higher plants via SMM, but it is generally present in low concentrations in other plant species. Several roles have been proposed, including salt tolerance and herbivore deterrence.

SAM, in decarboxylated form and catalyzed by S-adenosylmethionine decarboxylase (SAMDC, EC 4.1.1.50), provides 5'-desoxy-(5'-),3-aminopropyl-(1), a methylsulfonic salt required for PA biosynthesis (Roy and Wu 2002; for a review in PA: Alcázar et al. 2010; Hussain et al. 2011; Bitrián et al. 2012). PA I metabolites are essential to plant survival and have been correlated with biotic and abiotic stress resistance in many plant species; studies have employed exogenous PA application and genetic manipulations of different plant species (Bitrián et al. 2013). Increased biosynthesis of putrescine and spermidine in transgenic tobacco plants that had human SAMDC inserted into their genomes resulted in greater resistance to salt, drought and biotic stress (Waie and Rajam 2003). Microarray, transcriptomic and proteomic studies have demonstrated the role of PA in signaling cascades that increase plant tolerance or resistance to biotic and abiotic stress (Hussain et al. 2011).

### **Sulfur compounds related to plant defense**

Environmental stress usually affects plant cell homeostasis and development, increasing ROS production and leading to oxidative stress (Arruda and Azevedo 2009; Azevedo et al. 2011; Cia et al. 2012; Boaretto et al. 2014). Abiotic stress is a consequence of the effects of a wide variety of distinct external agents on plants, such as temperature (heat or chilling), water (drought or flooding), salinity, proton toxicity, heavy metals, overexposure to ultraviolet rays, ozone and others (Azevedo et al. 1998; Gratão et al. 2005; Monteiro et al. 2011; Bulbovas et al. 2014; Nogueirol et al. 2015). All living organisms have a series of pathways to combat environmental stress. In plant species, changes in photorespiration, enzymatic and non-enzymatic antioxidant pathways, regulation and responsive gene expression, and morphological and anatomical adaptations have been identified and investigated (Foyer and Shigeoka 2011). Excessive ROS generation has been considered a negative process for many years, but it is an essential component of signaling processes that prompt adjustments in gene expression and cellular structure in response to environmental changes (Shao et al. 2008; Foyer and Shigeoka 2011; Monteiro et al. 2011).

Plants can respond to abiotic stresses in a number of ways. These include, as a primary step, the induction of a network of signaling pathways and, at later stages, the response by specific proteins, metabolites and other compounds triggered by the signal transduction of the first step (Shulaev et al. 2008). Molecular analysis revealed that both short-term and long-term responses are important for understanding the progression of signaling events when the external and then the internal nutrient supply become depleted (Schachtman and Shin 2007). Similarly, it is critical to understand how experiments are designed because chronic and acute treatments with a stressor can produce completely different responses (Gratão et al. 2008). Furthermore, these distinct responses may improve the understanding or identification of the mechanisms involved in stress tolerance or of plants that are tolerant to the induced stressful condition. Signal transduction and detection networks that control plant responses to nutrient deprivation are not characterized for N and S to the extent it should considering how important these elements are. As already emphasized in this review, the S assimilation pathway is related to plant responses to abiotic stress and to defense mechanisms. It is a source of reduced S for many cellular processes and for synthesis of Cys, which is used in Met synthesis and/or incorporated into proteins or GSH (Siddiqui et al. 2012).

It is important to bear in mind that ROS is naturally produced by the cell metabolism. Oxidative stress occurs when the redox balance is disturbed and excess ROS induces a range of stress defense mechanisms (Gratão et al. 2005). ROS is an upstream mediator of nutrient signaling and increases rapidly after mineral nutrient deprivation, as indicated by Schachtman and Shin (2007). This is a major problem because the literature concerning the effect of an element/nutrient on a plant is extensive, but studies on the secondary responses and effects on the uptake and translocation of other essential elements are more limited. The resulting knowledge gap causes uncertainties regarding the full effects of the stressful conditions to which the plants were subjected. More integrated studies must be conducted.

Research on oxidative stress induced by heavy metals is increasing dramatically. The number of studies being published is astonishing, although the majority confirms known information. Nonetheless, it appears that the stress

induced by metals increases demand for reduced S, activating the expression of  $\text{SO}_4^{2-}$  transporters and enzymes of the assimilatory pathway (Hawkesford 2005; Hawkesford and Kok 2006; see also other papers published by the group of M. Hawkesford). Recent studies indicate that  $\text{SO}_4^{2-}$  transport in the plant vascular system, its assimilation in leaves and the recycling of S-containing compounds are related to drought stress signaling and response (Takahashi et al. 2011; Hawkesford 2012).

Plants are known to synthesize and release  $\text{H}_2\text{S}$  in a process catalyzed by L-cysteine desulfhydrase (LCD, E.C. 4.4.1.1) and involving conversion of L-Cys to  $\text{H}_2\text{S}$ , pyruvate and ammonia (García-Mata and Lamattina 2010).  $\text{H}_2\text{S}$  was formerly considered toxic to plant development, inducing excessive production of ROS. However, as understanding of its role in metabolic stress responses has increased, several studies have shown that its main function in plants is as a signaling molecule that controls physiological and biochemical processes (Jin et al. 2011; Li et al. 2012). For example, a study of spinach plants fumigated with  $\text{H}_2\text{S}$  gas demonstrated that approximately 40% of the  $\text{H}_2\text{S}$  was converted into GSH in plant leaves (Lisjak et al. 2011).  $\text{H}_2\text{S}$  had an essential role in alleviating the stress damage caused by aluminum chloride ( $\text{AlCl}_3$ ) in germinating wheat seedlings, increasing esterase and amylase activity and maintaining low malondialdehyde (MDA) and  $\text{H}_2\text{O}_2$  levels (Zhang et al. 2010). Pre-treatment with sodium hydrosulfide (NaHS, a  $\text{H}_2\text{S}$  donor) resulted in increased activity of guaiacol peroxidase (GPX, EC 1.11.1.7), superoxide dismutase (SOD, EC 1.15.1.1), ascorbate peroxidase (APX, EC 1.11.1.11) and catalase (CAT, EC 1.11.1.6) and decreased aluminum (Al) uptake in Al pre-treated seeds of wheat; this confirmed  $\text{H}_2\text{S}$  as a signaling molecule in response to abiotic stress (Zhang et al. 2010). In *Vicia faba*, *A. thaliana* and *Impatiens walleriana*,  $\text{H}_2\text{S}$  also induced stomatal closure and ABA-dependent signaling, possibly through the regulation of ABC transporters in guard cells under drought, and enhanced tolerance of water stress (García-Mata and Lamattina 2010). Tobacco cells pretreated with NaHS also exhibited heat tolerance and regrowth after stress exposure (Li et al. 2012). *A. thaliana* exposed to the same pretreatment produced more  $\text{H}_2\text{S}$ , exhibited drought tolerance by limiting stomatal aperture and increased the production and expression of drought marker genes (Jin et al. 2011).

In plants, incorporation of O-acetylserine sulfhydrylase (OASS, EC 2.5.1.65) and serine acetyltransferase (SAT, EC 2.3.1.30) into the cysteine synthase (CysK, EC 2.5.1.47) complex plays a regulatory role in S assimilation and Cys biosynthesis (Francois et al. 2006). The molecular mechanisms for the coordination of S, nitrogen and carbon assimilation are not yet known in detail. O-acetylserine, a precursor of Cys, was proposed as the signal regulating  $\text{SO}_4^{2-}$  assimilation, but it is not likely to be the outgoing signal for N and C metabolism. In S-deprived plants, for example, the level of glucose, fructose and phosphoenolpyruvate decreased, while starch concentration increased (Lunde et al. 2008). The reduction in photosynthetic rate, increase in oxidation of the ferredoxin:thioredoxin system, changes in starch synthesis and degradation, and the lower use of carbohydrates as an energy source can explain the changes in carbohydrates content (Lunde et al. 2008).

Complementary deoxyribonucleic acid (cDNA) array analysis revealed that expression of genes involved in auxin synthesis is induced upon S-starvation, suggesting a possible role of phytohormones (Kopriva and Rennenberg 2004). Clearly, and despite significant progress in understanding the regulation of  $\text{SO}_4^{2-}$  assimilation and GSH synthesis, their coordination with N and C metabolism is not yet fully understood, and the several potential signal molecules identified are still far from being sufficiently explanatory (Kopriva and Rennenberg 2004).

Stressed plants usually exhibit decreased rates of cellular division and elongation and, consequently, reduced or inhibited growth. This response may not only be a means of preserving energy for the defense process but may also function as protection against hereditary damage (Li et al. 2014). The chloroplasts play a major role in modulating the plant response, being both sensitive to abiotic stress factors and a major site for S assimilation (Biswal et al. 2008). These organelles are also important for ROS production because of the interaction between electrons escape from the photosynthetic electron transport chain and molecular oxygen (Foyer and Noctor 2012). Chloroplasts can therefore coordinate the C, N and S metabolic pathways, providing essential precursors for the synthesis of S-containing compounds (Jamal et al. 2006; Biswal et al. 2008). Another interesting aspect is that Fe-deficiency can result in severe disruption to the thylakoid lamellae, with loss of grana, but such

damage to the photosynthetic apparatus can be diminished by S nutrition. Photosynthetic activity and sucrose synthase and ribulose-1,5-bisphosphate carboxylase/oxygenase (RuBisCO) activity are also closely related to S status (Muneer et al. 2014).

### **Glutathione (GSH), metallothioneins (MTs) and phytochelatins (PCs)**

The stress response involves enzymatic antioxidant and other defense systems, including sulfur-containing compounds such as the essential macronutrient sulfur (S); glutathione (GSH: a S-containing thiol tripeptide,  $\gamma$ -L-glutamyl-L-cysteinyl-glycine); a class of phytochelatins [PCS: (g-Glu-Cys) $_n$ -Gly,  $n = 2$  to 5 usually]; S-rich proteins; S-amino acids; hydrogen sulfide ( $H_2S$ ); and a range of secondary metabolites (Gratão et al. 2012).

The GSH pool determines the degree of expression of genes linked to defense. It is controlled by many signaling pathways before and during stress, establishing a direct link between stress defense gene expression and GSH biosynthesis. Microarray, reverse transcription polymerase chain reaction (RT-PCR) and high performance liquid chromatography (HPLC) analyses of *Arabidopsis* plants exposed to cadmium (Cd) revealed that plants activate the S assimilation pathway by increasing transcription of specific genes that enhance the supply of GSH for PCS synthesis (Kawashima et al. 2011; Jobe et al. 2012). Moreover, roots and leaves have also been shown to exhibit distinct responses to Cd stress (Herbette et al. 2006; Gallego et al. 2012).

Metallothioneins are proteins with two structural domains (Cys-rich and metal-binding) involved in metal homeostasis and detoxification (Majic et al. 2008; Choppala et al. 2014; Gu et al. 2014). But in the case of metal stress, the action of PCS is essential. PCS are Cys-rich peptides related to GSH and most likely are synthesized in the same pathway (Qureshi et al. 2007; Zagorchev et al. 2013). Plants can inactivate the toxic effects of excessive ROS by intracellular chelation of the metallic ion by GSH and/or PCS in the cytosol. Depending on the plant species, these complexes can be transported into the vacuole by a specific metal-requiring enzyme (Pál et al. 2006; Yadav 2010; Hossain and Komatsu 2013). Increased GS can be considered a sulfate reserve for PCS synthesis, however the identity of the S-compounds that are reduced

when GS increases and the impact of metals on their metabolism are unknown, despite the hypothesis that GS provides an additional S source under metal stress condition (Ernst et al. 2008; Bell and Wagstaff 2014). Additionally, a common response by metal-stressed plants may be the activation of the ascorbate–glutathione cycle, either for the removal of  $H_2O_2$  or to ensure the availability of GSH for the synthesis of these metal-binding proteins (Vitória et al. 2001; Jozefczak et al. 2012).

## Glucosinolates

Glucosinolates are amino acid-derived secondary metabolites consisting of a thioglucose moiety, a sulfonated aldoxime, and a side chain derived from either aliphatic or aromatic amino acids (Halkier and Gershenzon 2006). When plant tissues are disrupted the glucosinolates are hydrolyzed by the highly active plant enzyme myrosinase, a thioglucosidase. The cleavage of the glucose thioester linkage produces an unstable intermediate that rearranges into biologically active thiocyanates, isothiocyanates, nitriles and oxalidine-2-thiones, depending on reaction conditions and presence of additional proteins (Hell and Kruse 2006). These products are chemically very reactive and may interfere with proteins and free amino acids. They are generally caustic and potentially toxic, hence their antimicrobial activities in plant defence. Although the glucosinolates also occur in a number of other plant families, the economic importance of oilseed rape, mustard, and the cabbage subspecies raised further interest in the biochemistry and molecular biology of glucosinolates biosynthesis and degradation (Hell and Kruse 2006).

In the reductive S assimilation pathway, the activities of enzymes in the pathway are influenced by S supply, with some (*in vitro*) evidence that ATPS and APR can form a complex to by-pass a branch point in the S assimilation pathway catalyzed by adenosine-5'-phosphosulfate kinase (APSK). APSK forms 3'-phosphoadenosine 5'-phosphosulfate (PAPS), an important substrate for the formation of the secondary S-containing metabolites including glucosinolates, and is therefore a significant enzyme in members of *Brassicaceae* (Leung et al. 2006). Four APSK genes have been cloned from *Arabidopsis*, for example, APSK1, APSK2, and APSK4, all localized in the

plastid, while APSK3 is a cytoplasmic isoform (Mugford et al. 2009). The use of mutants has shown that APSK1 is able to produce sufficient PAPS to maintain normal plant growth (Mugford et al. 2010), while disruption of APSK1 and APSK2 expression reduces the biosynthesis of glucosinolates (Mugford et al. 2009) demonstrating that the expression of APSK genes are strongly linked to the biosynthesis of glucosinolates. However, in non-glucosinolate accumulating species, including for example onion, the secondary (APSK-mediated) pathway must also operate to generate important pools of sulfate esters.

In order to elucidate the synthesis and degradation of glucosinolates in plants, (Mugford et al. 2009) observed highly significant alterations in the levels of glucosinolates and their desulfo-precursors in *A. thaliana* apk1, apk2, apk3, apk4 and wild-type mutants. The levels of each individual glucosinolates were reduced in the leaves of the mutant so that total glucosinolate levels reached only 15% of that in wild-type mutant. The reduction was accompanied by a massive increase in desulfo-precursors, which reached a ten-fold higher concentration than the mature glucosinolates in wild-type leaves. A similar reduction in glucosinolate levels was detected in the seeds of the mutant plants, however, the sulfo-precursors did not accumulate in the seeds. These results confirm that glucosinolates are not synthesized in seeds but are transported in the mature sulfate form (Magrath and Mithen 1993).

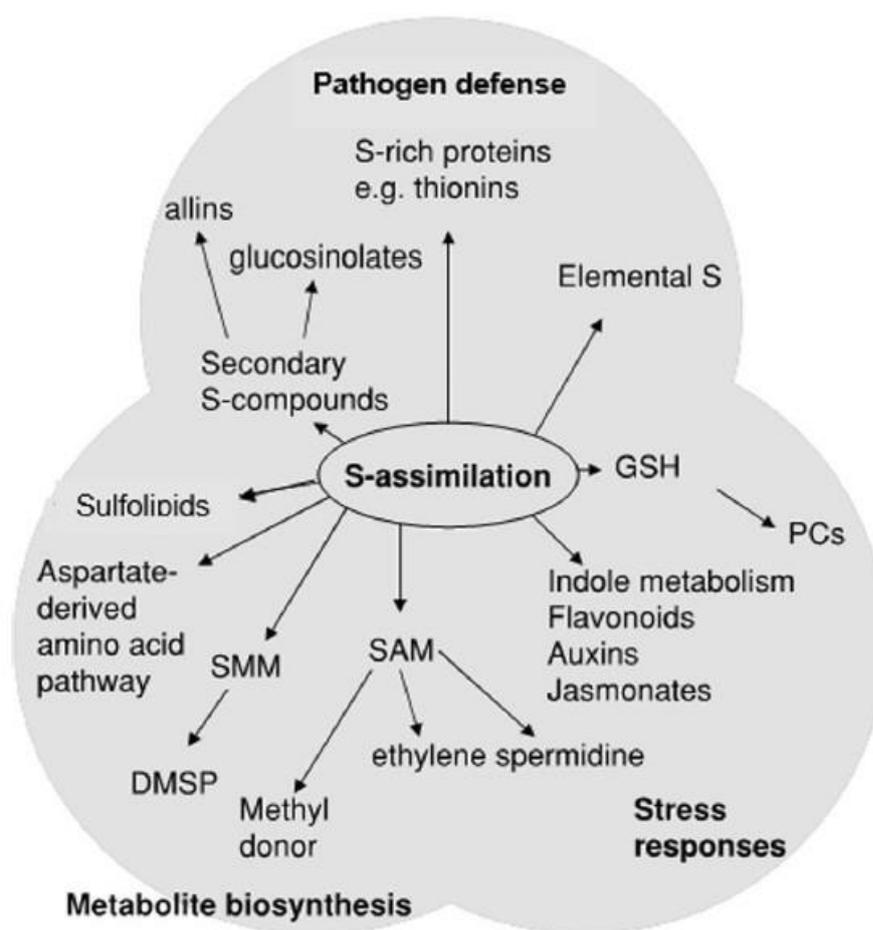
## Final Considerations

S is an essential chemical nutrient for plant growth and survival. S-containing defense compounds play significant roles in plant metabolism, stress response, cellular acclimation and adaptation (Fig. 3). They have gained much attention worldwide as biochemical genetics, plant physiology and breeding have been integrated to produce stress-tolerant plants and to identify preferential tolerance mechanisms. The development of transgenic plants with multi-stress tolerance should be the focus of research in the near future because stress factors rarely occur singly. Yet, the use of natural occurring or induced mutants cannot be left.

Our knowledge of the effects of transgenic S-containing compounds is still limited, partially because of the complex regulatory mechanisms described

previously. Significant advances in metabolic analyses with the “omics” techniques can improve identification of non-target gene products, enabling simultaneous consideration of gene expression, enzyme activity and metabolites.

In conclusion, this literature review is expected to improve our understanding of the essential mechanisms involved in oxidative stress in plants and the induction of S-containing compounds. The increased understanding may aid researchers in overcoming problems that occur in contaminated environments.



**Figure 3.** Sulfur assimilation is linked to multiple metabolic pathways responsible for a diverse range of physiological functions. Three major areas include primary metabolite biosynthesis, stress responses and pathogen defense. GSH: glutathione; PCs: phytochelatin; SAM: S-adenosylmethionine (S-AdoMet); SMM: S-methylmethionine; DMSP: dimethylsulfoniopropionate (Modified from Hawkesford 2005).

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## APPENDIX

**Abbreviations**

ABA	Abscisic acid
ACCO	1-aminocyclopropane-1-carboxylic acid oxidase
ACCS	1-aminocyclopropane-1-carboxylic acid synthase
Al	Aluminum
AlCl <sub>3</sub>	Aluminum chloride
APR	Adenosine 5' phosphosulfate reductase
APX	Ascorbate peroxidase
ATP	Adenosine triphosphate
ATP-S	Adenosine triphosphate-sulfurylase
AU	Auxins
C	Carbon
CAS	$\beta$ -cyanoalanine synthase
CAT	Catalase
CbL	Cystathionine $\beta$ -lyase
Cd	Cadmium
CgS	Cystathionine $\gamma$ -synthase
cDNA	Complementary deoxyribonucleic acid
CK	Cytokinins
Cys	Cysteine
CysK	Cysteine synthase
ET	Ethylene
GA	Gibberellins
DMSP	Dimethylsulfonio-propionate
GLU	Glutamate or glutamic acid
GLY	Glycine
GPX	Guaiacol peroxidase
GR	Glutathione reductase
GS	Glucosinolates
GSH	Glutathione

H <sub>2</sub> O <sub>2</sub>	Hydrogen peroxide
H <sub>2</sub> S	Hydrogen sulfide
HMT	Homocysteine S-methyltransferase
HPLC	High-performance liquid chromatography
JA	Jasmonates
K	Potassium
LCD	L-cysteine desulfhydrase
MDA	Malondialdehyde
MeJA	Methyl jasmonate
Met	Methionine
mRNA	Messenger ribonucleic acid
MS	Methionine synthase
MTs	Metallothioneins
N	Nitrogen
NaHS	Sodium hydrosulfide
NR	Nitrate reductase
OASS	O-acetylserine sulfhydrylase
OASTL	O-acetylserine(thiol) lyase
OPHS	O-phosphohomoserine
PA	Polyamines
PAPS	3'-phosphoadenosine 5'-phosphosulfate
PCS	Phytochelatins
ROS	Reactive oxygen species
RT-PCR	Reverse transcription polymerase chain reaction
RuBisCO	Ribulose-1,5-bisphosphate carboxylase/oxygenase
S	Sulfur
S <sup>2-</sup>	Sulfide
SA	Salicylic acid
SAM	S-adenosyl methionine
SAMDC	S-adenosylmethionine decarboxylase
SAMS	S-adenosyl methionine synthetase, SAM synthetase
SAT	Serine acetyltransferase
SLC13	Sodium/ SO <sub>4</sub> <sup>2-</sup> co-transporter
SLC26	SO <sub>4</sub> <sup>2-</sup> /anion exchanger in animals

SMM	S-methylmethionine	
SO <sub>2</sub>	Sulfur dioxide	
SO <sub>3</sub> <sup>2-</sup>	Sulfite	
SO <sub>4</sub> <sup>2-</sup>	Sulfate	
SOD	Superoxide dismutase	
SOT	Sulfotransferase protein family	
STs	Sulfotransferases	
SUL	Proton/ SO <sub>4</sub> <sup>2-</sup> -co-transporter in yeast	
SULTR	Proton/ SO <sub>4</sub> <sup>2-</sup> -co-transporter in plants	
TS	Threonine	synthase

## CHAPTER 3 - SELENIUM AND HEAVY METALS\*

### 1. Selenium: a powerful antioxidant

Selenium (Se) is an important micronutrient for humans and animals, participating as a constituent atom of some antioxidant enzymes, including glutathione peroxidase (GPOX), which is involved in hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) detoxification inside cells (Fairweather-Tait, 2011). The absence or inadequate intake of this element in the diet can cause a number of Se-related health diseases (Mehdi et al., 2013).

In higher plants, selenium has not been recognized as an essential nutrient so far, but at suitable levels it can induce a variety of beneficial effects, such as enhanced photosynthesis, growth improvement, higher accumulation of starch and sugars and delayed senescence (Zembala et al., 2010; Feng and Wei, 2012). In addition, in plants that grow in the presence of heavy metals, selenium can stimulate the cell antioxidant capacity through the enhancement of the activity of antioxidant enzymes and the synthesis of non-enzymatic molecules, like glutathione (GSH) and phytochelatin (PCs), and may induce the spontaneous dismutation of the superoxide radical ( $\text{O}_2^{\cdot-}$ ) into  $\text{H}_2\text{O}_2$  (Feng et al., 2013). The lower concentration of reactive oxygen species (ROS) due to Se would lower the lipid peroxidation process caused by metal-induced oxidative stress (Feng and Wei, 2012) (Figure 1).

### 2. Antioxidant defense mechanisms

Reactive oxygen species are the unstable and partially reduced forms of the atmospheric oxygen ( $\text{O}_2$ ), which shows a great capacity to oxidize other cell compounds. These molecules are formed from the transfer of one, two or three electrons to the  $\text{O}_2$  molecule, thus forming the superoxide radical ( $\text{O}_2^{\cdot-}$ ), hydrogen peroxide ( $\text{H}_2\text{O}_2$ ), hydroxyl radical ( $\text{OH}^{\cdot}$ ) and also the "Singlet" oxygen ( $\text{O}_2^1$ ), respectively, during the aerobic cellular metabolism in organelles like mitochondria and peroxisomes (Shieber and Chandel, 2014). The cellular defense responses are important to maintain low concentrations of ROS, and involve enzymatic and non-enzymatic antioxidant mechanisms. The superoxide dismutase (SOD) constitutes the first enzymatic barrier against the oxidative stress, catalyzing the dismutation reaction of  $\text{O}_2^{\cdot-}$ , in order to form  $\text{O}_2$  and  $\text{H}_2\text{O}_2$  (Shieber and Chandel, 2014). Subsequently,  $\text{H}_2\text{O}_2$  can be quickly

\*Further publication (2017). Book: "Selenium in plants: Molecular, Physiological, Ecological and Evolutionary Aspects". Chapter 20: Interactions of Se with other elements and organic nutraceuticals. Springer, USA.

converted into H<sub>2</sub>O and O<sub>2</sub> by the action of specific enzymes such as catalase (CAT) and peroxidases (POX). The non-enzymatic molecules implied in ROS detoxification are also important to maintain the cellular redox state, and mainly include glutathione (GSH), phytochelatins (PCs), proline, flavonoids, alkaloids, carotenoids (Foyer and Noctor, 2012).

### **3. Role of selenium against heavy metal stress in plants**

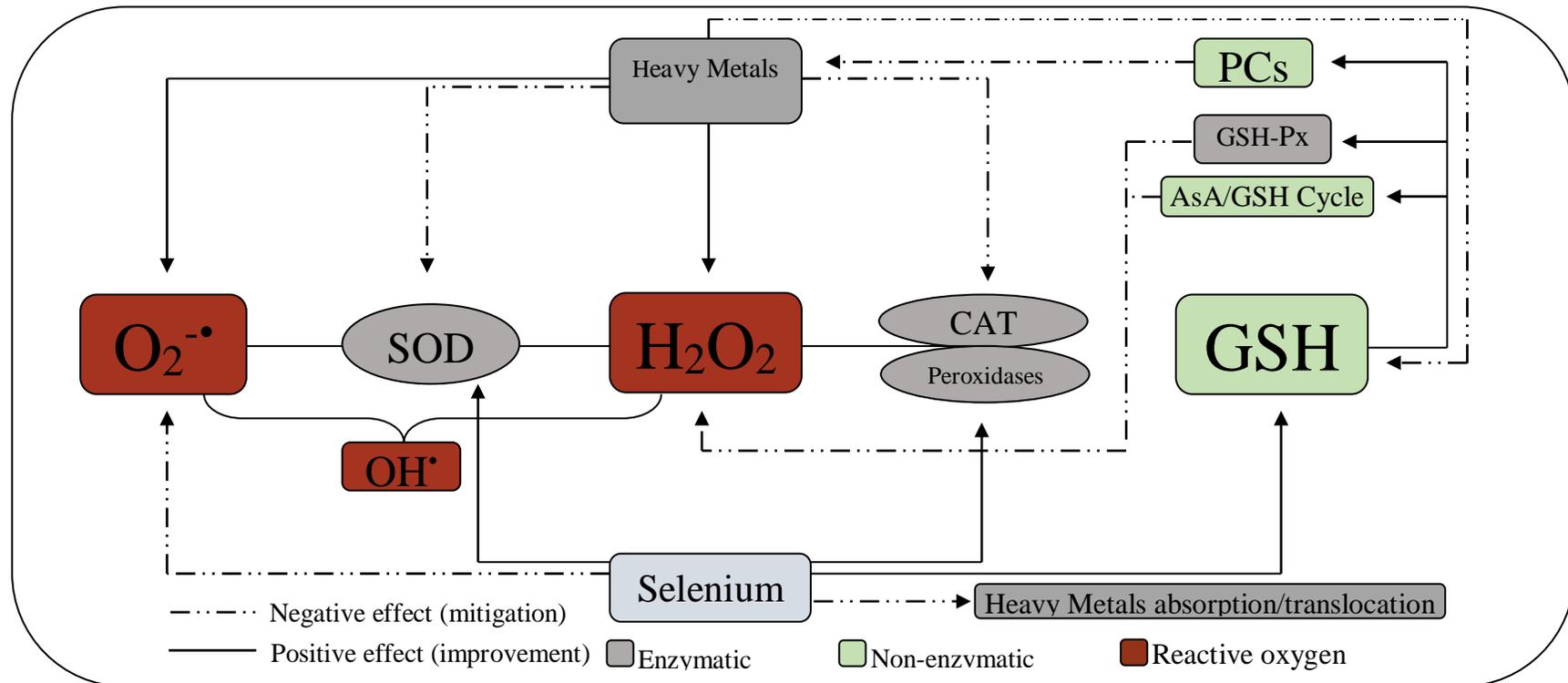
High concentrations of heavy metals in soils can lead, directly or indirectly, to plant death and losses in crop yields. The reason is that metals can cause an imbalance in the cellular redox status due to the metal-induced overproduction of ROS over the cell antioxidant capacity (Cuypers et al., 2010; Hasanuzzaman et al., 2012). The result of elevated levels of ROS is the damage to membrane phospholipids, proteins, nucleic acids and other cellular components. However, a number of studies has shown that Se can improve the enzymatic and non-enzymatic antioxidant systems of plants, thus enhancing their capacity to overcome the oxidative stress (Malik et al., 2012; Lin et al., 2012; Feng et al., 2013) (Figure 1).

#### **3.1 Cadmium (Cd)**

Cadmium (Cd) is one of the most toxic among heavy metals. This metal can be complexed with the organic fraction of soil, and be released as Cd<sup>2+</sup>, which is easily assimilated by plants through membrane transporters involved to the uptake of some nutrients, like Ca<sup>2+</sup>, Fe<sup>2+</sup>, Mg<sup>2+</sup>, Cu<sup>2+</sup> and Zn<sup>2+</sup> (Qin, et al., 2013). The presence of high concentrations of Cd<sup>2+</sup> in soil can cause a decrease of the plant capacity to accumulate these and other nutrients and molecules, such as chlorophyll, carotenoids and a broad spectrum of proteins, including antioxidant enzymes, that contain some of these nutrients in their structure (Cuypers et al., 2010; Hasanuzzaman et al., 2012). Indeed, due to high metal chemical similarity, Cd<sup>2+</sup> can replace other metals in the active sites of antioxidant metalloenzymes, e.g. SOD and CAT, therefore causing its inactivation (Cuypers et al., 2010).

Furthermore, the free Cd<sup>2+</sup> ions shows high affinity to the thiol groups (R-SH) present in different molecules, such as reduced glutathione (GSH), proteins and cysteine (Cuypers et al., 2010). Selenium can also be incorporated in cysteine by replacing sulfur (S) into the thiol group to form the analog Se-amino acid Se-cysteine (SeCys) (Malagoli

et al., 2015). Consequently,  $\text{Cd}^{2+}$  and Se may compete for the same binding site into the thiol group of cysteine. As a result, Cd uptake, translocation and accumulation in plants may be reduced (Lin et al., 2012).



**Figure 1.** Selenium effects against the oxidative stress induced by the heavy metal contamination. AsA (Ascorbate), CAT (Catalase), GSH (reduced glutathione), GSH-Px (Glutathione peroxidase), PCs (Phytochelatin), SOD (Superoxide dismutase), Reactive Oxygen Species: superoxide radical ( $O_2^{\cdot-}$ ); hydrogen peroxide ( $H_2O_2$ ) and hydroxyl radical ( $OH^{\cdot}$ ) formed from the combination of  $O_2^{\cdot-}$  and  $H_2O_2$  (Haber-Weiss reaction).

Recent studies reported the positive effect of Se on the activity of antioxidant enzymes in response to Cd stress. Lin et al. (2012) showed that the application of 3  $\mu\text{M}$  Se to rice (*Oryza sativa*) plants can increase the activity of SOD, peroxidase or guaiacol peroxidase (POD/GPOX) enzymes in roots and leaves, respectively. Fifty  $\mu\text{M}$  Se were proved to improve the activity of CAT, GPOX, glutathione reductase (GR), ascorbate peroxidase (APX) and enzymes related to the ascorbate-glutathione cycle, like monodehydroascorbate reductase (MDHAR) and dehydroascorbate reductase (DHAR), as well as non enzymatic compounds of this cycle like GSH, especially in the oxidized form (GSSG), in rape (*Brassica napus*) (Hasanuzzaman et al., 2012). Moreover, 5 and 10  $\mu\text{M}$  of Se increased the activity of CAT, APX and GR in leaves of sunflower (*Helianthus annuus*) (Saidi et al., 2014a), and concomitantly decreased ROS production, lipid peroxidation, oxidative stress, and recovered the membrane physicochemical characteristics.

### 3.2 Arsenic (As)

Arsenic (As) is a metalloid, which can be mainly found in the forms of arsenate ( $\text{AsO}_4^{3-}$ ) or arsenite ( $\text{AsO}_3$ ) in soils and waters. As contamination in soils occurs mainly in response to anthropogenic activities, like the use of pesticides and sewage sludge in crop fields, as well as to other activities not directly related to agriculture, such as mining and metal melting. The detrimental effect triggered by this metal in plants is related to a reduction of growth and development caused by photosynthesis inhibition, inefficient nutrition and oxidative stress (Malik et al., 2012; Han et al., 2015).

Similar to  $\text{Cd}^{2+}$ , As bind to the sulfur presented in the T-SH group of GSH. Whether selenium is provided to plants, it can compete with As for the binding to the thiol group, thus actively reducing As absorption (Han et al., 2015). Likewise, 5  $\mu\text{M}$  Se are reported to reduce As uptake in mungbean (*Phaseolus aureus* Roxb.), and alleviate oxidative stress by enhancing the activity of SOD, POD, APX enzymes and the synthesis of GSH and ascorbic acid (ASC) (Malik et al., 2012). Similar results were found in tobacco (*Nicotiana tabacum*) plants treated with  $0.1 \text{ mgL}^{-1}$  selenite (Han et al., 2015). Srivastava et al. (2009) found that 5  $\mu\text{M}$  and 10  $\mu\text{M}$  selenate decreased the lipid peroxidation process, likely because of higher production of the non-protein thiol GSH in *Pteris vittata* L.

### 3.3 Lead (Pb)

Lead is one of the most dangerous pollutants worldwide, where the main sources are fertilizers, pesticides, mining, metal melting, automobiles fumes and industrial waste or discharge. This heavy metal is considered carcinogenic to humans as it causes DNA damages and its synthesis inhibition, while in plants Pb can disrupt the membrane structure and permeability causing dehydration and decrease of the electron transportation in photosynthesis, this metal also binds to the thiol groups of amino acids, enzymes and proteins and, as a final result, induces the overproduction of ROS and oxidative stress (Mroczek-Zdyrska and Wojcik, 2012).

The Se beneficial effects against Pb in plants have been described by few authors recently and are directly related to the ROS scavenging in cells (Mroczek-Zdyrska and Wojcik, 2012; Yuan et al., 2013; Hu et al., 2014). For instance, 1.5  $\mu\text{M}$  selenite concentration lowered the superoxide radical ( $\text{O}_2^{\cdot-}$ ) production and concentration in the apical part of the root in *Vicia faba* L. *minor*, increased the activity of POD/GPOX enzymes and caused the content of thiol groups (T-SH) (Mroczek-Zdyrska and Wojcik, 2012). In addition, 1  $\mu\text{M}$  selenite concentration improved the leaf biomass of coleus (*Coleus blumei* Benth.) and decreased the rate of lipid peroxidation, likely because of the higher GSH level in roots (Yuan et al., 2013). Hu et al. (2014) demonstrate that 0.5  $\text{mg kg}^{-1}$  selenite could induce low Pb concentration in rice (*Oryza sativa*) tissues (shoot and husk).

### 3.4 Other heavy metals

Excess of some nutrients (metalloids) in plants can increase the production of ROS and cause decreased, denaturation and inactivation of antioxidant enzymes. Manganese (Mn) is an important microelement for plants, but at high concentration it can be toxic. The toxicity is related to photosynthesis suppression, membrane integrity disruption, lower protein metabolism and oxidative stress. As showed by Saidi et al. (2014b), 5  $\mu\text{M}$  of selenate can effectively counteract the Mn detrimental effects in sunflower (*Helianthus annuus*) by improving the CAT, APX and GPOX activity.

On the contrary, chromium (Cr) has no biological function in plants and can be toxic at any concentration in soil, especially near areas with industrial activities. Se concentrations as high as 3  $\mu\text{M}$  can improve SOD activity in rice roots alleviating the

toxic effects of Cr on growth, and increase H<sup>+</sup>-ATPase activity, thus protecting the plants from Cr-induced oxidative stress (Cao et al., 2013).

Mercury (Hg) is also a harmful environmental pollutant, and soil contamination by this metal comes from mining, metal smelting, and industrial activities. Its presence in plants causes growth inhibition, oxidative stress, lipid peroxidation, deficient chlorophyll production and photosynthesis (Zhao et al., 2013). Selenite and selenate can improve growth in garlic (*Allium sativum*) and also avoid Hg absorption, translocation and accumulation in roots and leaves, when they are applied at doses higher than 1 mg L<sup>-1</sup> (Zhao et al., 2013).

#### **4. Final considerations**

The maintenance of cellular homeostasis under heavy metals contamination depends on several interlinked and complex mechanisms. The antioxidant defense does not follow a specific pattern of action, even under Se application, and may differ due to various factors such as plant species, concentration, exposure time, nutrient concentration in soil, plant developmental stage, organs, form of Se and tissues analyzed. By this way, the plant defense against heavy metals and other different abiotic stresses is a dynamic and adaptive system, in which the Se is extensively investigated to improve the enzymatic and non-enzymatic antioxidant responses to counteract the heavy metal induced stress.

In this contest, Se can detoxify the oxidative stress directly, as a response to its powerful antioxidant characteristics, converting the superoxide radical (O<sub>2</sub><sup>•-</sup>) into hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>). Indirectly, Se improves the antioxidant enzymes activities such as superoxide dismutase (SOD), catalase (CAT) and other peroxidases (POX), and the non-enzymatic antioxidant compounds concentration like the reduced glutathione (GSH), proline, flavonoids, alkaloids carotenoids and phytochelatins (PCs). Se is also expected to actively reduce the heavy metal absorption and translocation, mainly as a response to the production of PCs and competition for the same binding sites into the cell environment. All this processes, as a final analysis, provides better maintenance of the ROS production and concentration in cells and, as a consequence, decrease the induced oxidative stress and avoid the destruction of the cell membranes and structures.

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## CHAPTER 4 - SELENIUM ALLEVIATES CADMIUM TOXICITY IN TOMATO FRUITS\*

*Abstract.* Selenium (Se) is a beneficial element for plants and essential to mammals, being increasingly used as a fertilizer. Cadmium (Cd) is toxic to crops and consumers. In order to identify the possible effect of Se in alleviating tomato stress from Cd<sup>2+</sup> toxicity, the fruits dry weight, chlorophyll, carotenoids, proline concentration, macronutrients, micronutrients, Cd<sup>2+</sup> and Se concentration, as well as Cd<sup>2+</sup> and Se translocation index were analyzed in two different tomato genotypes, Micro-Tom (MT) and the *high pigment-1 (hp1)*. Tomato plants were submitted to Cd<sup>2+</sup> (0 and 0.5 mM CdCl<sub>2</sub>) and Se (0 and 50 µM Na<sub>2</sub>SeO<sub>3</sub>), individually applied or in combination. Fruit growth was not affected by Se application neither when applied alone or with Cd<sup>2+</sup>, nor were pigment concentrations in fruits of both genotypes. Contrarily, fruit proline concentration increased in genotype *hp1* when Se was applied alone and in the MT genotype when Se was applied with Cd<sup>2+</sup>. Fruit micronutrient concentration were negatively affected by Cd<sup>2+</sup> and by the mixture of Se and Cd<sup>2+</sup>, while the Se-alone treatment enhanced the concentration of Mn and Zn in *hp1* and Fe in MT. Importantly, Se demonstrated a strong capability to reduce Cd<sup>2+</sup> concentration in the roots of both genotypes and Cd<sup>2+</sup> translocation to above ground parts only in the MT genotype. Thus, the alleviating effect of Se on Cd<sup>2+</sup> stress in tomato could be related to restriction of Cd<sup>2+</sup> uptake and translocation, enhancing micronutrient concentration in fruits and, finally, enhancing fruit proline concentration. Selenium fertilization may therefore be considered a promising way to alleviate Cd toxicity in tomato.

*Keywords.* Tomato, Abiotic stress, *High-pigment 1*, Micro-Tom, Heavy metal.

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## 1. Introduction

Among the numerous sources of abiotic stress, the heavy metal contamination in agricultural soil is a worldwide concern for crop production (Gonzales et al., 2015), while the metal mining, smelting and subsequent industrial processes are important sources of this environmental pollution. The presence of heavy metals, specifically in cultivated soil, represent a great concern due to the high toxicity and persistence in the environment (Liu et al., 2011) and may be related to the excessive use of phosphate fertilizers, which lead to the metal uptake by crops and consequent food chain bioaccumulation, representing a major threat to living organisms (Xu et al., 2013).

Cadmium (Cd) is one of the most toxic among the heavy metals. It can be complexed with the organic fraction of the soil, and be released as  $\text{Cd}^{2+}$  (Gallego et al., 2012), which is easily assimilated by plants through the same membrane transporters involved to the uptake of nutrients like  $\text{Ca}^{2+}$ ,  $\text{Fe}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{Cu}^{2+}$  and  $\text{Zn}^{2+}$  (Qin et al., 2013), leading to a nutrient deficiency, while its presence in soil can result in a commercial rejection of certain cultivars and may become dangerous to human health, especially if it is accumulated in tissues assimilated into the food chain, leading to considerable losses in plant productivity and hazardous health effects to humans (Gallego et al. 2012; Qin et al. 2013).

Several studies have demonstrated beneficial effects of supplying plants with low Selenium (Se) concentrations, which appear to enhance photosynthesis (Wang et al., 2012), cause a lower lipid peroxidation and improve the restoration of the membrane and overall structure of chloroplasts (Feng et al., 2013), stimulate the cellular antioxidant systems (Djanaguiraman et al., 2010; Walaa et al., 2010; Feng et al., 2013) and also improve the Proline concentration (Hawrylak-Nowak, 2009; Walaa et al. 2010), which plays an important role in cellular osmotic regulation and plant protection

against different abiotic stresses, therefore its quantification is used as a plant stress gauge (Shevyakova et al., 2013). However, the specific effect of Se to the Cd uptake, translocation and concentration in tomato plants, in order to avoid the Cd bioaccumulation in the food chain, have not been elucidated.

Tomato (*Lycopersicon esculentum* Mill.) cv. Micro-Tom (MT) and *high pigment-1 (hp1)* genotypes are considered excellent models for biological studies by virtue of their short life cycle and reduced size. The *hp1* genotype is characterized by high concentrations of antioxidants such as carotenoids (particularly lycopene), flavonoids, vitamin C and chlorophyll which improves the antioxidant defense (Calvenzani et al., 2010; Kilamb et al., 2013), which could help provide a better understanding of the interplay between Se and antioxidants in relation to Cd stress in tomato plants.

The aim of this study was to investigate whether selenite ( $\text{Na}_2\text{SeO}_3$ , henceforth referred to as Se), may avoid the uptake, translocation and concentration of Cd ( $\text{CdCl}_2$ ), henceforth referred to as  $\text{Cd}^{2+}$ , in different tomato tissues as well as to analyze the fruits overall status through the nutritional analyses, dry weight, pigments and proline concentration.

## **2. Material and methods**

### *2.1 Plant material and experimental treatments*

Seeds of tomato (*Lycopersicon esculentum* Mill.) cv. Micro-Tom (MT) and the genotype *high pigment-1 (hp1)*, were cultivated in separate boxes containing a mixture of 1:1 (by volume) of commercial pot mix (Plantmax HT-Eucatex<sup>®</sup>, Brazil) and medium size vermiculite, supplemented with 1 g.L<sup>-1</sup> of 10:10:10 NPK and 4 g.L<sup>-1</sup> of lime and maintained in a greenhouse. After the first true leaves appeared, seedlings were transplanted to 1 L Leonard pots (Gratão et al., 2012) (1 seedling per pot) filled with

sand and polystyrene (4:3) and supplied with Hoagland's nutrient solution (Hoagland and Arnon, 1950). Subsequently, twenty-one-day old plants were selected and further grown in the same solution spiked with either 0 mM (control) or 0.5 mM CdCl<sub>2</sub> (cadmium chloride, CAS: 10108-64-2) and 0 μM or 50 μM Na<sub>2</sub>SeO<sub>3</sub> (anhydrous sodium selenite, CAS: 10102-18-8) applied either individually or simultaneously. The Hoagland solution, with or without CdCl<sub>2</sub> and Na<sub>2</sub>SeO<sub>3</sub> was changed weekly and the total volume was maintained at a constant level by using distilled water. The experiment was designed to have the following treatments: T1: Control; T2: Se 50 μM (MT, *hp1*); T3: Cd 0.5 mM (MT, *hp1*); T4: Cd 0.5 mM + Se 50 μM (MT, *hp1*). After a period of 95 days post germination, corresponding to 74 days of exposure to CdCl<sub>2</sub> and/or Na<sub>2</sub>SeO<sub>3</sub>, samples of fruits were harvested, rinsed and frozen immediately in liquid N<sub>2</sub>, and stored at -80°C for further analyses.

## 2.2 *Fruits dry weight*

Samples of fruits (total fruit weight per plant) were placed in a thermal convection laboratory oven (Fanem<sup>®</sup> SP, BR – model 330) at 70°C for a week. Later, these fruits were ground in an electric mill (Marconi<sup>®</sup>, model 048) and the dry measurements were performed on an analytical balance (Denver instruments company<sup>®</sup>, model AA-200) accurate to 1.10<sup>-18</sup> mg. Fruit production was quantified as biomass (g).

## 2.3 *Fruits concentration of chlorophyll and carotenoids*

Samples of fruit pulp were subjected to spectrophotometric pigment quantification in triplicate. Pigment extraction was carried out in 80% acetone and the extract was filtered through filter paper using a vacuum pump. Measurement of pigments was performed in a spectrophotometer (Beckman<sup>®</sup>, model DU-640) at the following wavelengths: chlorophyll-*a* in 663 nm, chlorophyll-*b* in 647 nm and

carotenoids (carotene[c] + xanthophyll [x]) in 470 nm. The total chlorophyll (chl) and carotenoids (car) contents were calculated as described by Lichtenthaler (1987):

$$\text{Chl } a = 12.25 \times A_{663} - 2.79 \times A_{647} \text{ Eq. (A.1)}$$

$$\text{Chl } b = 21.50 \times A_{647} - 5.10 \times A_{663} \text{ Eq. (A.2)}$$

$$\text{Chl } a+b = 7.15 \times A_{663} + 18.71 \times A_{647} \text{ Eq. (A.3)}$$

$$\text{Car } c+x = (1000 A_{470} - 1.82 \text{ Chl}a - 85.02 \text{ Chl}b)/198 \text{ Eq. (A.4)}$$

Pigment concentration was expressed in micrograms of pigment per gram of tissue fresh weight ( $\mu\text{g g}^{-1}$  FW).

#### *2.4 Fruits proline concentration*

Free proline concentration in fruits was measured as described by Gratão et al. (2012). Samples (0.5 g) of fruit pulp were extracted in 3% sulphosalicylic acid. The homogenate was centrifuged at 10,000 g for 15 minutes at 4°C, and 2 mL of the supernatant was held for 1 hour in boiling water by adding 2 mL ninhydrin acid ( $\text{C}_9\text{H}_6\text{O}_4$  – CAS: 485-47-2) and 2 mL of glacial acetic acid ( $\text{CH}_3\text{CO}_2\text{H}$  – CAS: 64-19-7), to which cold toluene (4 mL) was added. The absorbance was read at 520 nm and proline concentration calculated as  $\text{mmol g}^{-1}$  FW against a proline standard curve.

#### *2.5 Fruits $\text{Cd}^{2+}$ and Se quantification and nutritional analyses*

Quantitative  $\text{Cd}^{2+}$ , Se and nutrient analysis was carried out using energy dispersive X-ray fluorescence spectrometry (EDXRF) as described by Tezzoto et al. (2013). Samples of fruits were dried at 70°C for 7 days, and 0.2 g DW of fine powder, obtained following grinding with mortar and pestle, were microwave digested with 2 mL of 70%  $\text{HNO}_3$ , 2 mL of  $\text{H}_2\text{O}_2$  and 2 mL of Milli-Q water (18.2  $\text{M}\Omega$  cm at 25°C) at a controlled pressure of 2 MPa, concentrated acids on a digestion block heated gradually to 203 °C. Biological samples with increased concentrations of  $\text{Cd}^{2+}$ , Se and nutrients

were used to establish standard calibration curves. The samples were irradiated in triplicate for 300s using a Shimadzu® EDX 720 system (São Paulo, Brazil). The samples were irradiated using an Rh X-ray tube operated at 15 kV (Na to Sc) and 50 kV (Ti to U) applied on the rhodium tube. The detection was carried out using the Si (Li) detector cooled with liquid nitrogen. For all studies, matrices with 2048 dependent variables (columns) were constructed (energy values), and the independent variable was the Cd<sup>2+</sup>, Se or nutrient concentration. The modeling and prediction tools for data exploration were performed with the use of the chemometrics package pirouette, version 3.11 (Infometrix®, Bothell, WA, USA).

## 2.6 *Cd<sup>2+</sup> and Se Translocation index*

The Cd<sup>2+</sup> and Se translocation index (TI, %) was calculated by dividing their concentration in the shoot (fruits, leaves) by their total plant concentration (root, leaves, fruits), and multiplying the quotient by 100.

## 2.7 *Statistical analyses*

The experimental design was randomized with five replication pots, being the results expressed as mean and standard error of mean ( $\pm$  SD) of three independent replicates of each extract for dry weight, pigments, proline concentration, Cd<sup>2+</sup>, Se and nutrients concentration analyses. The statistical analysis was performed using the Assistat software version 7.7 (Silva and Azevedo, 2009). ANOVA followed by a multiple comparison between means by the Tukey test for each character, at a 0.05 level of significance was performed.

### 3. Results and Discussion

#### 3.1 Nutritional analyses.

The application of  $\text{Cd}^{2+}$  caused a decrease in zinc (Zn) and iron (Fe) concentration in the MT fruits, while the *hpl* genotype contained less manganese (Mn) and Zn in the same situation compared to the control plants (Table 1). Furthermore, Se applied alone induced a decrease in Zn in the MT fruits, while an increase of Fe was noted (Table 1). The same treatment also induced an increase of Mn and Zn in *hpl* fruits, but, contrarily, the Fe concentration in fruits of this genotype was lower compared to the control (Table 1). When Se was supplied together with  $\text{Cd}^{2+}$ , the concentration of Fe and Zn decreased in fruits for both genotypes (Table 1), while Mn decreased only in *hpl* (Table 1) compared to the control plants.

Some reports have shown the synergetic effect of  $\text{Cd}^{2+}$  by stimulating the absorption of specific nutrients, whereas the concentration of Fe, for example, increased after  $\text{Cd}^{2+}$  contamination (Kumar et al., 2014). In our study, the application of 0.5 mM of  $\text{CdCl}_2$  caused an increase in potassium (K) and Fe concentration specifically in the *hpl* fruits (Table 1), which could indicate that the *hpl* suffered less membrane damage, from lipid peroxidation or other damaging processes, which could stimulate mineral transportation during the oxidative stress caused by the heavy metal exposure. In contrast, the Zn concentration decreased in fruits of both genotypes when 0.5 mM of  $\text{CdCl}_2$  was applied, the Fe concentration decreased specifically in the MT fruits and Mn concentration decreased in *hpl* under the same condition, in comparison to the control plants (Table 1).

The negative effect on nutrients concentration in response to the  $\text{Cd}^{2+}$  stress can be related to various deleterious mechanisms, due to increased lipid peroxidation, decreased plasmalema fluidity, and, as a consequence, harm to the entire cell integrity,

which can directly reduce absorption and translocation of nutrients (Zembala et al., 2010; Kumar et al., 2014). Furthermore, the form  $Cd^{2+}$ , usually concentrated in the organic fraction of the soil, can actively compete with the uptake of those nutrients that have the same valence number, such as  $Ca^{2+}$ ,  $Mg^{2+}$ ,  $Fe^{2+}$ ,  $Mn^{2+}$ ,  $Cu^{2+}$  and  $Zn^{2+}$  (Qin, et al., 2013; Kumar et al., 2014).

Combined supply of  $Cd^{2+}$  and Se resulted in a reduction in the Fe and Zn concentration in both genotypes and of Mn specifically in *hpl*, while no increases in nutrient absorption were observed (Table 1). However, the Se alone application increased the Mn and Zn concentration in the fruits of the *hpl* genotype and also Fe in the fruits of the MT, which could confer an efficient mechanism to alleviate the oxidative stress in this tissue, since this micronutrients are in the main active site of different isoforms of the superoxide dismutase enzyme (SOD, EC 1.15.1.1), the first enzymatic defense against the ROS in the cell environment (For review see Gratão et al., 2005).

Treatment	Macronutrients (g kg DW)					Micronutrients (mg kg DW)				
	Mg	P	S	K	Ca	Mn	Fe	Cu	Zn	
MT	1	1.49 ± 0.14 a	3.81 ± 0.06 ab	2.14 ± 0.04 a	24.00 ± 1 bc	3.60 ± 0.10 a	33.20 ± 0.56 bc	88.00 ± 1 c	4.64 ± 0.33 a	25.30 ± 0.06 c
	2	1.47 ± 0.03 a	3.16 ± 0.08 ab	2.19 ± 0.10 a	25.20 ± 0.36 b	3.40 ± 0.40 a	36.00 ± 2 b	113.00 ± 1.76 b	5.72 ± 0.20 a	21.30 ± 0.38 de
	3	1.35 ± 0.11 a	3.52 ± 0.31 ab	2.60 ± 0.10 a	25.20 ± 0.95 b	3.70 ± 0.10 a	34.70 ± 0.16 b	43.70 ± 0.11 f	5.36 ± 0.16 a	18.10 ± 0.22 f
	4	1.25 ± 0.11 a	2.70 ± 0.51 b	2.09 ± 0.02 a	22.20 ± 0.10 c	3.10 ± 0.10 a	30.40 ± 0.20 c	41.90 ± 0.05 f	4.79 ± 0.20 a	19.60 ± ef
<i>hp1</i>	1	1.44 ± 0.07 a	3.61 ± 0.26 ab	2.44 ± 0.12 a	24.20 ± 1.15 bc	3.80 ± 0.17 a	33.60 ± 0.13 b	110.60 ± 0.46 b	5.24 ± 0.11 a	34.30 ± 1.09 b
	2	1.18 ± 0.08 a	4.69 ± 0.28 ab	2.36 ± 0.10 a	24.00 ± 1 bc	4.00 ± 0.87 a	46.00 ± 1 a	66.70 ± 0.58 d	5.70 ± 0.06 a	39.50 ± 0.52 a
	3	1.32 ± 0.02 a	5.81 ± 0.05 a	2.79 ± 0.38 a	29.60 ± 0.44 a	3.00 ± 0.50 a	17.70 ± 0.84 e	141.80 ± 0.96 a	4.26 ± 0.04 a	24.00 ± 1 cd
	4	1.27 ± 0.08 a	2.96 ± 0.21 b	2.05 ± 0.02 a	22.20 ± 0.17 c	3.60 ± 0.30 a	21.50 ± 0.47 d	54.60 ± 0.67 e	3.80 ± 0.29 a	18.60 ± 0.54 ef

**Table 1.** Nutrients concentration in fruits. 1. Control / 2. Na<sub>2</sub>SeO<sub>3</sub> 50 μM / 3. CdCl<sub>2</sub> 0.5 mM / 4. CdCl<sub>2</sub> 0.5 mM with Na<sub>2</sub>SeO<sub>3</sub> 50 μM. Values are the mean of three replicates. Lower case letters indicate statistically significant differences (P < 0.05).

### 3.2 $Cd^{2+}$ and Se concentration and translocation index

The tomato genotypes (MT and *hp1*) showed differences in  $Cd^{2+}$  concentration in roots (Table 2), and also showed evidence of a Se effect to restrict  $Cd^{2+}$  translocation from root to shoot (Table 2). The  $Cd^{2+}$  concentration were much higher in roots when compared to the other tissues analyzed, for both genotypes. Furthermore, the roots of the MT genotype concentrated about 35% more  $Cd^{2+}$  compared to the *hp1* roots when exposed to the treatment in which the plants received only the heavy metal (Table 2). This difference between the genotypes could be directly related to plant growth, since the *hp1* fruits dry weight was much lower compared to MT under the Cd-alone treatment (Fig.1). However, the  $Cd^{2+}$  translocation index (TI) was considerably higher in the *hp1* genotype compared to the cv. MT (Table 2, 11.8% and 9% respectively) in the  $Cd^{2+}$  alone treatment, indicating that this heavy metal could be easily and readily translocated to the upper parts of the plant by the *hp1* compared to MT.

When  $Cd^{2+}$  was supplied simultaneously with Se, the heavy metal concentration in the root system and also in the leaves showed a very noticeable decrease for both genotypes compared to those treatments which received only  $Cd^{2+}$  (Table 2). This heavy metal can enter into root system through the cortical tissue, reaching the xylem through the apoplastic and symplastic pathway (Lux et al., 2011).

Additionally,  $Cd^{2+}$  could primarily be stored at the site of metal uptake, i.e. in the roots of most tomato varieties, which can be related to the restriction of translocation by the sequestration of Cd-chelates in vacuoles (Gallego et al., 2012), which can be improved by Se and ROS scavenging in the root cells (Clemens et al., 2013; Gratão et al., 2015). Likewise, when Se was applied simultaneously with the heavy metal, the

Cd<sup>2+</sup> T.I. demonstrated to be much lower when compared to the Cd-alone treatment in the MT genotype, 7.9% and 9% respectively (Table 2).

Treatment			Cd <sup>2+</sup> and Se concentration (mg kg <sup>-1</sup> DW)			TI (%)
			Fruits	Leaves	Roots	
Cadmium	MT	Cd	30.2 ± 0.9 c	596.3 ± 0.5 a	6340.4 ± 0.7 a	9
	<i>hpl</i>	Cd	44.8 ± 0.8 a	503.9 ± 0.9 b	4092.08 ± 0.05 b	11.8
	MT	Cd + Se	9.4 ± 0.7 d	258.9 ± 0.9 d	3117.8 ± 0.3 c	7.9
	<i>hpl</i>	Cd + Se	41.8 ± 0.3 b	324.5 ± 0.6 c	2002.7 ± 0.9 d	15.4
Selenium	MT	Se	4.3 ± 0.7 a	2.8 ± 0.9 b	358.34 ± 0.6 a	1.9
	<i>hpl</i>	Se	5.4 ± 0.6 a	5.32 ± 0.4 a	210.44 ± 0.5 c	4.9
	MT	Cd + Se	0	0	310.9 ± 0.2 b	0
	<i>hpl</i>	Cd + Se	0	0	193.02 ± 0.05 d	0

**Table 2.** Selenium (Se) and Cadmium (Cd<sup>2+</sup>) concentration in fruits, leaves, roots and translocation index (TI). Values are the mean of three replicates and standard deviation (±SD). Lower case letters indicate statistically significant differences (P < 0.05).

Furthermore, the free Cd<sup>2+</sup> radical shows high affinity to the thiol groups (-SH) presented in different molecules such as the glutathione (GSH), proteins and also the amino acid cysteine (Cuypers et al., 2010). Additionally, Se can be assimilated by the sulfate pathway, due to its chemical similarity with S, which may cause the incorporation of Se into proteins, through the replacement of S by Se into the thiol group in the selenocysteine (SeCys) amino acid (for a review see Malagoli et al., 2015). Consequently, the competition for the same binding site into the amino acid cysteine and proteins, thiol group, by Cd<sup>2+</sup> and Se, result in a lower uptake and translocation of the heavy metal (Lin et al., 2012).

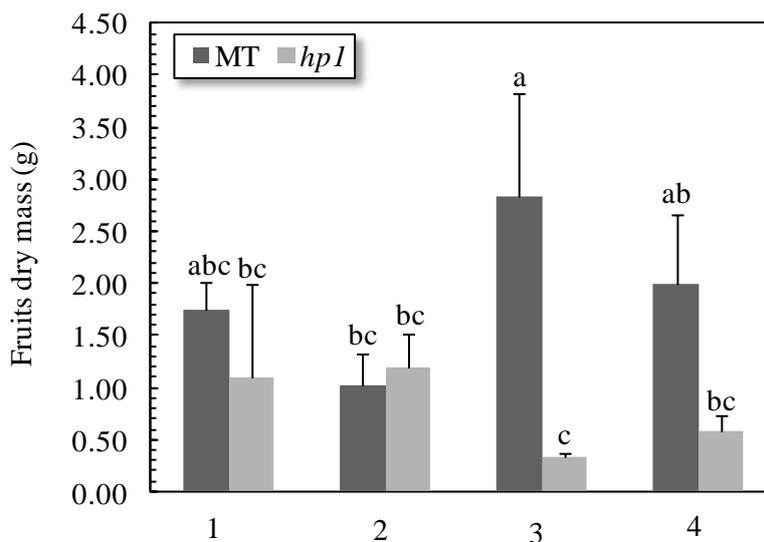
The Se concentration data showed similarities with those from Cd<sup>2+</sup>. The highest values were found in the roots compared to the other tissues analyzed for both genotypes (Table 2). Similarly, the MT genotype demonstrated almost 41% more Se in the root system compared to the *hpl* in the Se-alone treatment, 358 mg kg<sup>-1</sup> DW vs. 210

mg kg<sup>-1</sup> DW respectively (Table 2). Moreover, *hpl* showed a higher Se T.I. compared to MT, 4.9% and 1.9% respectively, which indicates that Se was readily translocated to the upper parts of the plant by the *hpl* compared to the MT genotype.

When Se and Cd<sup>2+</sup> were applied simultaneously, Se concentrations decreased in the roots of both genotypes (Table 2). Interestingly, the presence of Se in the shoot was not detectable and the T.I. was zero in the same condition for both genotypes. As already stated, the competition for the same binding sites into amino acids and proteins can be related to the lower Se and also Cd<sup>2+</sup> concentration in the root system. Other results in the literature have demonstrated that Se is not well translocated to shoot when applied in selenite form, maybe in response to its rapid assimilation into the organic forms of SeCys or SeMet and local incorporation into proteins in the root system (Malagoli et al., 2015).

### 3.3 Fruit dry weight

There were no significant differences in fruit biomass production between the treatments for either of the two tomato genotypes, Micro-Tom (MT) and *high pigment-1* (*hpl*) (Fig. 1). Therefore, at this concentration and chemical form, Se did not positively or negatively affect fruit production. It is known that when supplied at low levels to the plant, Se can attenuate oxidative stress, enhance membrane stability, mainly in response to lower lipid peroxidation, and could also increase the concentration of starch and sugars (Pezzarosa et al., 2014; Qing et al., 2015). On the other hand, excess Se can induce reactive oxygen species (ROS) overproduction and thus oxidative stress, and can compete with S (sulfur) transport and metabolism, culminating in damage to plants and growth reduction (Capaldi et al., 2015).



**Fig. 1.** Fruits dry weight. 1. Control / 2. Na<sub>2</sub>SeO<sub>3</sub> 50 μM / 3. CdCl<sub>2</sub> 0.5 mM / 4. CdCl<sub>2</sub> 0.5 mM with Na<sub>2</sub>SeO<sub>3</sub> 50 μM. Values are the mean of three replicates and standard deviation (±SD). Lower case letters above bars indicate statistically significant differences ( $P < 0.05$ ).

The supply of 0.5 mM of CdCl<sub>2</sub> also did not reduce fruit dry weight of both genotypes as compared to the control plants (Fig. 1). Current reports have demonstrated the detrimental effects of Cd<sup>2+</sup> on plant growth and development, mainly as a result of an imbalance of ROS production and the antioxidant system, reduction in water content, altered membrane permeability, lower photosynthesis and mineral deficiency (Clemens, et al., 2013; Irfan et al., 2014; Gratão et al., 2015).

The Se supply necessary to promote growth under an abiotic stress may differ depending on tissue type, plant species, stress type as well as Se speciation, concentration and application manner (for review: Feng et al., 2013). When Se was applied simultaneously with Cd, the fruit dry weight was also similar to the control plants for both genotypes (Fig. 1). Since CdCl<sub>2</sub> 0.5 mM alone did not significantly affect fruit production, the effect of Se on any negative effect of Cd on this parameter could not be tested.

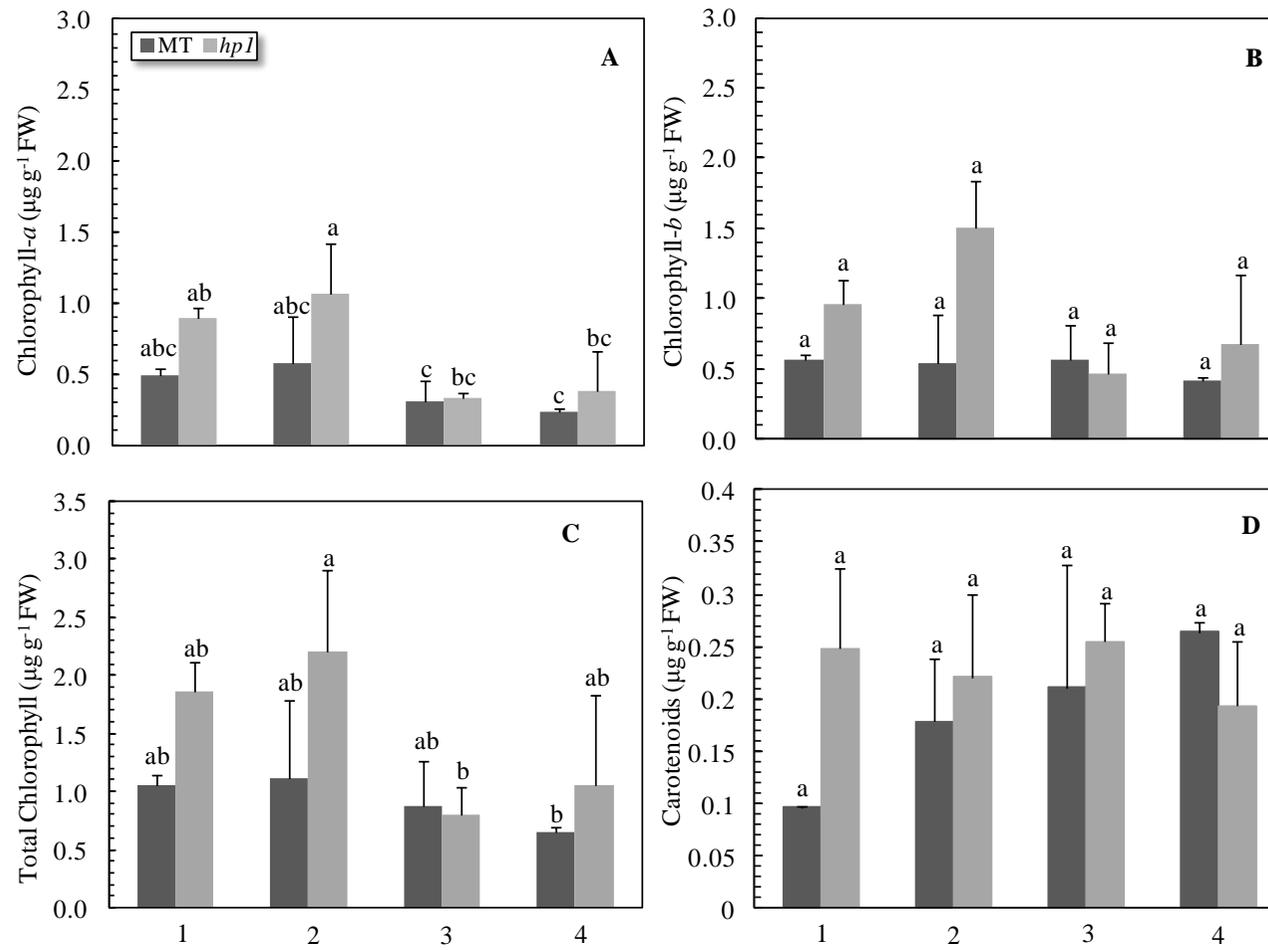
### 3.4 Chlorophyll and carotenoids concentration

A quantification of carotenoids and chlorophyll in tomato fruit was carried out in order to investigate whether Se could protect fruit development and maturation during Cd<sup>2+</sup> stress. Damaging effects of Cd<sup>2+</sup> on photosynthesis and pigment concentration have been reported, predominantly through modifications of the chloroplast structure and the photosynthetic apparatus, as a response to the higher oxidative stress and lipid peroxidation (Gallego, et al., 2012). Additionally, the negative effect of Cd<sup>2+</sup> specifically to the chlorophyll concentration can be a result of damages in pigments by ROS and a lower chlorophyll biosynthesis (Zhong et al., 2015).

Selenium (Na<sub>2</sub>SeO<sub>3</sub>) at the concentration of 50 µM did not significantly affect pigment concentration in fruits of both genotypes when it was applied alone or jointly with Cd<sup>2+</sup>, as compared to the control plants (Fig. 2). There were in fact no significant differences in pigment concentrations between the treatments (Fig. 2) and thus Cd supplied at this concentration did not significantly affect pigment levels. However, it is noteworthy that Se-treated plants had slightly higher levels and Cd-treated plants slightly lower levels (NS) of various pigments, causing the chlorophyll-*a* concentration in fruits of the *hpl* genotype to be higher when Se was applied alone in comparison to *hpl* plants which received only Cd<sup>2+</sup> (Fig. 2A, C). This effect cannot be considered a Se beneficial effect against Cd<sup>2+</sup> stress in fruits due to the fact that the pigment concentration were similar in the Cd-treated plants and in plants where Se was applied jointly with Cd.

The *hpl* genotype demonstrates an exaggerated light responsiveness as well as a natural increased fruit pigmentation and vitamin C, which can confer an antioxidant capacity to deal with the increased radiation absorption, energy transformation and the oxidized environment in chloroplasts (Carvalho et al., 2011). Further analyzes on other

tissues of this genotype would help to elucidate the Se effect to pigments concentration and photosynthesis under the oxidative stress.



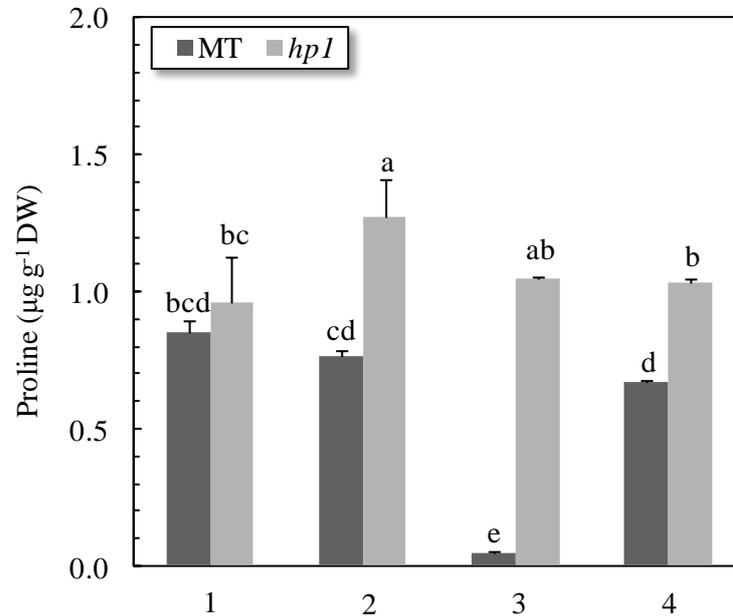
**Fig. 2.** Fruits pigments concentration. 1. Control / 2.  $\text{Na}_2\text{SeO}_3$  50  $\mu\text{M}$  / 3.  $\text{CdCl}_2$  0.5 mM / 4.  $\text{CdCl}_2$  0.5 mM with  $\text{Na}_2\text{SeO}_3$  50  $\mu\text{M}$ . Values are the mean of three replicates and standard deviation ( $\pm\text{SD}$ ). Lower case letters above bars indicate statistically significant differences ( $P < 0.05$ ).

### 3.5 Proline concentration

The proline concentration in control plants was similar for both genotypes (Fig. 3). Interestingly, when Se was applied alone, the proline concentration increased in the *hpl* fruits compared to the control (Fig. 3). The Cd-alone treatment, on the other hand, showed strongly decreased proline concentration in MT fruits in comparison to the control (Fig. 3). This result could be a response for a localized proline synthesis in the root instead of the shoot, related to the higher Cd<sup>2+</sup> concentration in this tissue (Table 2). The simultaneous application of Se with Cd<sup>2+</sup> increased the proline concentration for the MT genotype, compared to the Cd-alone treatment (Fig. 3). Proline accumulation in Se-supplied plants has been reported before. Walaa et al. (2010) observed that lipid peroxidation, induced by NaCl, was effectively minimized when the seedlings were pretreated with Se, showing an enhanced antioxidant activities and proline concentration. In addition, Hawrylak-Nowak (2009) found that supplementation of 5, 10 or 20 µM of selenite on cucumber seedlings improved the growth rate, photosynthetic pigments, proline concentration, and enhanced the antioxidant capacity of seedlings. However, the mechanisms related to the proline accumulation in Se-supplied plants have not been very well elucidated.

Proline concentration in a specific tissue under a stressful condition is widely varied among different plant species and is highly dependent on the type and the level of the stress whereupon the plant was submitted (Shevyakova et al., 2013). Proline overproduction can function as a mechanism to actively clear the cell ROS during oxidative stress, culminating in protection and stabilization of cell membranes and structures, proteins and enzymes during severe stresses (Ashraf and Foolad, 2007). Se application may induce proline synthesis through an improved activity of glutamyl kinase (GK EC 2.7.2.11), the first enzyme of the

proline biosynthetic pathway, and decreased activity of proline oxidase (PROX EC 15.5.2), responsible for the proline molecule denaturation (Khan et al., 2015).



**Fig. 3.** Fruits proline concentration. 1. Control / 2. Na<sub>2</sub>SeO<sub>3</sub> 50 µM / 3. CdCl<sub>2</sub> 0.5 mM / 4. CdCl<sub>2</sub> 0.5 mM with Na<sub>2</sub>SeO<sub>3</sub> 50 µM. Values are the mean of three replicates and standard deviation (±SD). Lower case letters above bars indicate statistically significant differences (P < 0.05).

The products resulting from the catabolism of proline molecules, after attenuation of the stress condition, are used in the oxidative phosphorylation process in mitochondria in an effort to synthesize molecules of ATP (adenosine 5'-triphosphate), important upon the cell recovery processes (Ashraf and Foolad, 2007). In summary, the results demonstrate that 50 µM of Se increased the proline concentration in fruits of the *hp1* genotype, and also increased the proline concentration in MT fruits when it was applied simultaneously with Cd<sup>2+</sup>, which could protect the tissue from the induced stress.

#### 4. Conclusions

Overall, the data show an alleviating effect of 50  $\mu\text{M}$  selenite against Cd toxicity in both tomato genotypes analyzed. The Se benefic effects included reduction in Cd uptake and translocation, particularly in the MT genotype, improvement of fruit Mn and Zn concentration in genotype *hp1* and Fe in the fruits of genotype MT and enhanced fruit proline concentration in both genotypes. This study suggests that Se fertilization may not only create nutritionally enhanced food but also be a means to alleviate Cd toxicity in tomato plants.

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