Microtensile Bond Strength of a Repair Composite to Leucite-Reinforced Feldspathic Ceramic

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The purpose of this study was to evaluate the microtensile bond strength of a repair composite resin to a leucite-reinforced feldspathic ceramic (Omega 900, VITA) submitted to two surface conditioning methods: 1) etching with hydrofluoric acid + silane application or 2) tribochemical silica coating. The null hypothesis is that both surface treatments can generate similar bond strengths. Ten ceramic blocks (6x6x6 mm) were fabricated and randomly assigned to 2 groups (n=5), according to the conditioning method: G1- 10% hydrofluoric acid application for 2 min plus rinsing and drying, followed by silane application for 30 s; G2- airborne particle abrasion with 30 µm silica oxide particles (CoJet-Sand) for 20 s using a chairside air-abrasion device (CoJet System), followed by silane application for 5 min. Single Bond adhesive system was applied to the surfaces and light cured (40 s). Z-250 composite resin was placed incrementally on the treated ceramic surface to build a 6x6x6 mm block. Bar specimens with an adhesive area of approximately 1 ± 0.1 mm² were obtained from the composite-ceramic blocks (6 per block and 30 per group) for microtensile testing. No statistically significant difference was observed between G1 (10.19 ± 3.1 MPa) and G2 (10.17 ± 3.1 MPa) (p=0.982) (Student’s t test; α = 0.05). The null hypothesis was, therefore, accepted. In conclusion, both surface conditioning methods provided similar microtensile bond strengths between the repair composite resin and the ceramic. Further studies using long-term aging procedures should be conducted.

Key Words: ceramic, surface treatment, repair composite, bond strength.

INTRODUCTION

Intraoral porcelain might fracture due to factors namely occlusal forces, impact, internal defects and inadequate design (1). With the increasing interest in ceramic restorations, such as inlays, onlays, veneers and all-ceramic crowns, producing an esthetic and functional repair system that avoids over time-consuming and expensive remakes has become challenge (2). The earlier porcelain-repair systems relied on macromechanical retention, with preparation of grooves or undercuts. In contrast, the current generations of these systems are mostly based on micromechanical and chemical bond, by etching/silane application or grit blasting/silane application on the ceramic surface (3). Several resin-based materials have been used to repair porcelain restorations. It is suggested that the bond strength between these two types of materials is highly dependent on surface preparation (2). Acid etchings, air abrasion with aluminum oxides or silica coating by means of airparticle abrasion are the most commonly types of surface treatment (1). Acids work well on silica-based porcelains, although a strong one, typically hydrofluoric acid, is needed to promote micromechanical retention. Kupiec (4) evaluated three different ceramic surface treatments: a) aluminum ox-
ide (Al₂O₃) air abrasion (50 µm), b) 8% hydrofluoric acid, and c) air abrasion and hydrofluoric acid. The last combination recorded the most consistently effective bond strengths. Aluminum oxide abrasion provides a clean and reactive bonding surface in porcelains.

A tribochemical silica coating method has been successfully used. In this technique, the surfaces are abraded with 30 µm Al₂O₃ modified with silicon acid and then silica coating is performed. The silane is applied to react with the coated silica (5-7). The tribochemical silica coating provides fine mechanical retention, as well as physicochemical bonding between resin materials and ceramic, metal and composite restoratives (8).

However, silica coating has been suggested to aluminum-, aluminum/zirconium-, zirconium-based ceramic because etching of these ceramics does not degrade their compact surface of high crystal content, e.g., silica coating is indicated to acid-resistant ceramics with low silica content (7,9). Thus, bonding resin cements to feldspar-based porcelains with high silica content seems to be well established, by treating them with hydrofluoric acid and silane agents. Additionally, in theory, silica coating should not be the treatment of choice for silica-based ceramics because these materials already have silica to their chemomechanical bonding.

Adhesion between two substrates is traditionally studied using shear or tensile bond strength tests. It has been suggested that tensile testing may be better indicated to determine the ceramic-to-composite resin bond strength, as shearing testing evaluate the base material rather than the strength at the adhesive interface. It has also been stated that small-sized specimens have higher bond strength when submitted to microtensile testing (10,11). The effects of pretreatments on porcelain can also be examined by scanning electron microscopy.

The purpose of this study was to evaluate the microtensile bond strength of a repair composite resin to leucite-reinforced feldspathic ceramic submitted to two surface conditionings methods (etching with hydrofluoric acid + silane application or tribochemical silica coating). The null hypothesis is that both surface treatments can generate similar bond strengths.

MATERIAL AND METHODS

Ten ceramic blocks (6 x 6 x 6 mm) of leucite-reinforced feldspathic ceramic (Omega 900, Vita Zahnfabrik, Germany; Batch #5335; Dentine 4M2) were fabricated. One of the surfaces of each ceramic block was polished with #300-, 600-, 800-, 1000- and 1200-grit SiC papers. The ceramic blocks were randomly assigned to 2 groups (n=5), according to the surface treatment: Group 1 (HF) - 6% hydrofluoric acid (Porcelain Etchant; Bisco Inc., Schaumburg, IL, USA; Batch #0006) was applied for 60 s to porcelain surface, rinsed, dried and a coat of silane agent (Porcelain Primer; Bisco Inc.; Batch #0300012763) was applied and left to dry for 30 s; Group 2 (CoJet - tribochemical silica coating) - ceramic surface was submitted to airborne particle abrasion with 30 µm silica oxide particles (CoJet-Sand) (sandblasting parameters: a- position: perpendicular to the surface; b- distance: 10 mm; c- time: 20 s; d- pressure: 2.8 bars) using a chairside air-abrasion device (CoJet System; 3M/ESPE, St. Paul, MN, USA; Batch #68421), followed by silane application (Sil; 3M/ESPE) for 5 min.

After surface treatments, Single Bond adhesive system (3M/ESPE; Batch #1105) was applied to ceramic surface and light cured for 20 s (XL 3000, 3M/ESPE; light intensity = 500 mW/cm²; distance = 0). Z-250 composite resin (3M/ESPE; Batch #3CE) was accommodated incrementally on the treated ceramic surface to build a 6x6x6 mm block. The resin was used in 2-mm-thick increments, each polymerized for 40 s.

After a 7-day storage in distilled water at 37°C, the ceramic/resin blocks were taken to a precision cutting machine with a water-cooled diamond saw (Labcut 1010; Extec Corp., Enfield, CT, USA) (Figs. 1 and 2). In all blocks, the first cuts (±0.5 mm) were discarded because results could be influenced by excess or lack of adhesive at the interface. Next, 3 approximately 1-mm-thick slices were obtained. Other three approximately 1-mm-thick slices were obained. A total of nine specimens per ceramic/resin block and 45 specimens per group were obtained (Fig. 2). The specimens had the following characteristics (Fig. 2C): a- non-trimmed rectangular form (bar specimens); b- nearly symmetric square cross-section area of 1 ± 0.1 mm²; c- length of about 10 mm (7,11,12).

Microtensile Bond Strength Test

Before testing, the adhesive area of each bar was...
measured using a digital caliper (Mitutoyo, Tokyo, Japan). Each specimen was bonded with cyanoacrylate adhesive to a custom-made device perpendicular to the force applied, therefore avoiding sprain forces at the interface. Only the ends of the specimen were bonded. The device/specimen set was adapted to a universal testing machine (Emic DL-1000; Emic, São José dos Pinhais, PR, Brazil) and tested in microtensile strength at crosshead speed of 1 mm.min⁻¹ until fracture. Microtensile bond strength calculations were made using the following equation: \( \sigma = L / A \), where \( \sigma \) is the bond strength (MPa), \( L = \) test load (N), \( A = \) adhesive area (mm²). The results were analyzed statistically by Student’s t-test (\( \alpha = 0.05 \)). The debonded bar specimens were analyzed with a scanning electron microscope (JSM-T330A; Jeol, Tokyo, Japan) at ×150 to evaluate the failure modes (adhesive, cohesive or mixed).

**Micromorphological Analysis**

The surfaces of two ceramic specimens were submitted to tribochemical silica coating and analyzed topographically under scanning electron microscopy. **RESULTS**

G1 (10.19 ± 3.1 MPa) and G2 (10.17 ± 3.1 MPa) presented statistically similar microtensile bond strength means (\( p=0.982 \)). Regardless of the experimental group, all (100%) debonded specimens presented failure in the adhesive zone (Fig. 2D), either adhesive or mixed (adhesive and cohesive of the cement) (Fig. 3A, B).

The topographic analysis showed that the acid-etched ceramic surface (Fig. 4A) was rougher (more
irregularities) than those coated with silica oxide particles (Fig. 4B).

DISCUSSION

By using porcelain-repair systems, porcelain can be repaired without removal of the prosthesis. This study sought to determine the best surface treatment for a leucite-reinforced glass ceramic repaired with a composite resin. Under the tested experimental conditions, both ceramic surface treatments provided similar bond strengths between the leucite-reinforced feldspar ceramic and the repair composite resin. The null hypothesis was therefore accepted.

The mean microtensile bond strength data recorded in the present investigation are consistent with those of previous studies (2,6) and are much lower than the reasonable composite-to-dentin bond strength goal stated in the literature (20 MPa). Matsumura et al. (13) showed that all systems used to produce mechanical or chemical adhesion between substrates should provide at least 10 MPa of bond strength to be indicated for clinical situations. The low values might be attributed to the fact that contemporary porcelain repair systems provides mostly chemical bonding to porcelain. The former repair systems offer a type of treatment that can lead to lower bond strength under water storage and/or thermocycling, which, however, was not performed in this study.

Regarding the topographic patterns produced by the treatments on ceramic surface, the specimens etched with HF (Fig. 4A) presented a rougher surface (more irregularities) than the ones abraded with silica oxide particles (Fig. 4B). Chemical and mechanical treatments of porcelain surfaces increase the total surface area and the surface energies as well. According to Phoenix and Shen (14), this mechanical interlocking exerts significant effects upon the formation and maintenance of ceramic-to-resin bond. To make leucite-based ceramic more wettable, preparation of numerous small and uniform pits or grooves are preferred (16).

On the other hand, the ceramic surface coated by silica oxide presents high chemical reactivity to resin materials. (1,3,7,8,17). Bonding to ceramic substrates seems to be dependent on the presence of silica in their surfaces, which has great affinity for silane agents (5). As the silica layer is well incorporated to ceramic surface, silane application enhances the resin bond. In view of these results, there is a trend to choose the silica coating for repairing porcelain substrates (16). Kern and Thompson (9) stated that silica coating can improve bonding of resin to glass-infiltrated aluminum oxide ceramic (In-Ceram), but they also stated that sandblasting of all-ceramic restorations with feldspar glass materials should be avoided because of the great volume loss produced by sandblasting. Therefore, sandblasting of feldspar-based ceramic with aluminum oxide particles can damage ceramic surface and hence it should not be indicated in such cases.

HF etching, which consists of a preferential

Figure 4. SEM micrograph of the ceramic surface abraded with 30 µm silica oxide particles (×5000) (A). SEM micrograph of the ceramic surface etched with 6% hydrofluoric acid for 2 min (×5000) (B).
attack of the leucite phase resulting in a retentive surface, can also provide reasonable bond strengths between composite and ceramic, especially when followed by silane application (4). The variety of concentrations of commercially available hydrofluoric acids indicates that the optimal concentration and duration of their application are not well established (17).

However, all ceramic restorations are prone to crack propagation through the glass matrix, leading to failures. Intrinsic microscopic defects are also responsible for cracks and catastrophic fractures of these materials. Thus, the following relations are possible: a) the higher the amount of glass, the lesser the ceramic strength (acid sensitive-ceramics); b) the lower the amount of glass and the higher the content of crystals, the higher the ceramic strength (acid resistant-ceramics).

In this sense, traditional adhesive procedures can contribute to strengthen less resistant ceramics. HF etching and adhesive cementation can limit crack propagation, especially because the cement can infiltrate and seal the failures created by acid attack (18). Silane agents improve the chemical bonding between ceramic and resin cement and also fill intra-ceramic defects.

There is a noteworthy difference from the experimental design of many studies to the present one. Shear strength test has been the most commonly employed test modality to study the performance of porcelain repair systems. But it is believed that this test geometry causes high tensile surface stresses within the porcelain, close to the area of load application, initiating fracture at the porcelain surface. Tensile strength test would more likely provide adhesive failures between materials because the load is applied perpendicularly to the long axis of the sample, avoiding the occurrence of sprain forces and shearing stresses on the adhesive area (10). The dimensions of the specimens in the present study were based on the inverse relationship between bond strength and bond surface area, as published elsewhere (12).

Thus, for a test to reproduce the real bond strength between an adhesive (resin cement) and an adherent (dental, metallic, ceramic or polymeric-substrate), it is essential that the interfacial zone is the most stressed area, notwithstanding the mechanical test used (12). According to a stress distribution study (19), some mechanical tests do not actually stress the interface zone. The shear test, for example, is criticized because the stress is non-homogeneously distributed at the adhesive interface, requiring more substrate. Thus, the stresses are concentrated in a restricted zone distant from the adhesion zone and hence most fractures occur in the substrate. This phenomenon prevents the measurement of true interfacial bond strength and limits further improvements in the bonding systems (under estimated and misinterpreted results). The analysis of the failure mode and fractography reduce the risk of data misinterpretation, such as stating that “The bond strength was higher than the cohesive strength of the substrate” (11).

Within the limitations of this investigation, it may be concluded both surface conditioning methods provided similar microtensile bond strengths between the repair composite resin and the ceramic. Further studies using long-term aging procedures should be conducted.

**REFERENCES**

Composite microtensile bond strength to ceramic


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