

## In Vitro Safety Evaluation of Caffeic Acid

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*Phenolic compounds are abundant in the Brazilian plant kingdom and they are part of a large and complex group of organic substances. Cinnamic acids are part of this group of organic compounds, and caffeic acid is one of its representatives. Besides exhibiting a powerful antioxidant activity, increasing the collagen production and preventing the premature aging, caffeic acid has demonstrated antimicrobial activity and may be promising in the treatment of dermal diseases. One of the applications of caffeic acid is in emulsions, which are known to be widely used by consumers for pleasant and refreshing sensory, although few studies have reported the efficacy and safety of these products on the skin. The relevance of this study is based on evidence and to clarify the cytotoxic potential of this substance through preliminary studies in vitro. The cytotoxicity evaluation was carried out using the MTT method (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide), a colorimetric assay which determines the amount of insoluble violet crystals formed by the reduction of MTT in the mitochondria of living cells. A dose versus response curve was constructed, and it was possible to use the equation to determine the IC<sub>50</sub> of caffeic acid or the product concentration needed to cause 50% lethality of the cells. The results are promising since caffeic acid concentration that promoted 50% toxicity in HepG2 cells (IC<sub>50</sub>=781.8 µg/mL) is approximately 330 to 400 times greater than the concentration required to inhibit 50% of DPPH (IC<sub>50</sub> DPPH= 2.39 µg/mL) and ABTS (IC<sub>50</sub> ABTS= 1.96 µg/mL) radicals scavenging activity, respectively. The maximum concentration of caffeic acid tested (1140 mg/mL) did not reach 50% of cell death in HaCat cells. Thus, it was concluded that the caffeic acid does not cause toxicity in HepG2 and HaCat cells in the concentrations required to promote*

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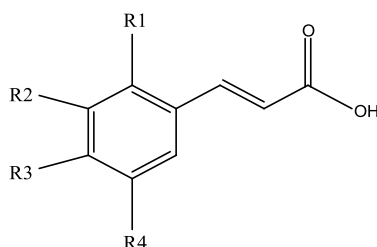
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*antioxidant activity in vitro, and it can be applied in topical products.*

## Introduction

Phenolic compounds are abundant in the Brazilian plant kingdom and they are part of a large and complex group of organic substances. Cinnamic acid derivatives are part of this group of organic compounds, also called phenylpropanoids, which contains nine carbons in its structures (Figure 1). Caffeic acid (3,4-dihydroxycinnamic) is a representative of this group (Soares, 2002).

**Figure 1.** Chemical Structure of the Main Cinnamic Acids. Cinnamic Acid  $R1 = R2 = R3 = R4 = H$ , *o*-coumaric acid:  $R2 = OH$ , *p*-coumaric acid:  $R3 = OH$ ; Caffeic Acid:  $R2 = R3 = OH$ ; Ferulic Acid:  $R2 = OCH_3$  and  $R3 = OH$



The phenolic antioxidants act as free radical scavengers and sometimes as metal chelators. They act both in the initiation step as well as in the propagation of the oxidative process. These substances are relatively stable owing to the resonance of the aromatic ring shown by them. The phenolic acids are characterized by the presence of a benzene ring, a carboxylic acid moiety and one or more hydroxyl or methoxy moieties in the molecule, which confer antioxidant properties to it (Aklillu et al.; 2014; Djordjevic et al. 2010; Dossie de Antioxidantes, 2014). The phenylpropanoids act as antioxidants by eliminating oxygen free radicals (Chen et al., 1997; Meyer et al., 1998) and chelating pro-oxidant metal ions, especially iron (Paiva-Martins et al., 2002; Hynes et al., 2004).

Besides exhibiting a powerful antioxidant activity, increasing collagen production and preventing premature aging, caffeic acid has demonstrated antimicrobial activity and may be promising in the treatment of dermal diseases. One of the applications of caffeic acid is in emulsions, which are known to be widely used by consumers for pleasant and refreshing sensory, although few studies have reported the efficacy and safety of these products on the skin (Magnani et al., 2014).

The cosmetic industry and researchers have sought to use *in vitro* experiments, replacing animal models for the study of the efficacy and safety of new products in recent times. Additionally, the policy of the European Community has determined that cosmetic products produced and marketed in

member countries should not use animals for conducting toxicological tests (Augustin et al., 1996).

As an alternative to animal testing, testing on cell cultures have been proposed in order to predict the toxic potential of chemicals and formulations (Livny et al., 2003; Viravaidya et al., 2004; Rabachini, 2007). Thus, the experiments in cell cultures enable the study of cellular behavior in a controlled microenvironment using non-tumor human epithelial cells (Bedoni et al., 2007; Heise et al., 2006). Culture of primary cells such as keratinocytes, skin explants, organs, or organotypic cultures of cells have been used in preclinical testing of natural products (Botham et al., 1998; Medina, 2000; Abu-Yousif et al., 2008).

As commensurate with the ongoing trend of replacing the *in vivo* tests of cosmetics, this study used *in vitro* techniques to assess the safety and efficacy of the product developed.

The aim of this study was to provide evidence and to clarify the cytotoxic potential of this substance through preliminary studies *in vitro*.

## Experimental Methods

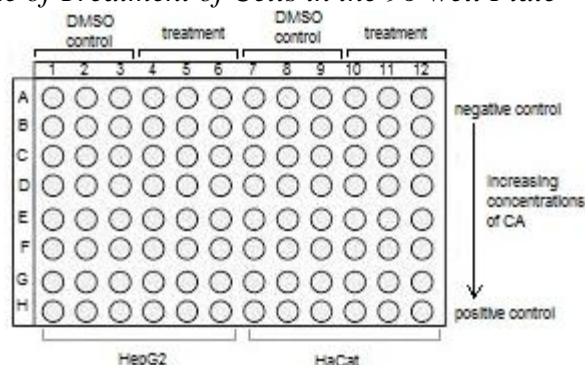
The cytotoxicity evaluation was carried out using the MTT method (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide), a colorimetric assay which determines the amount of insoluble violet crystal formed by the reduction of MTT in the mitochondria of living cells.

Two human cell lines were assessed: human hepatoma cells which are capable of metabolizing xenobiotics (HepG2) and metabolically incompetent keratinocytes (HaCat), grown in MEM (Minimum Essential Media) culture media with 10% of fetal bovine serum and antibiotics (penicillin 100 U / ml, streptomycin 0.1 mg / ml).

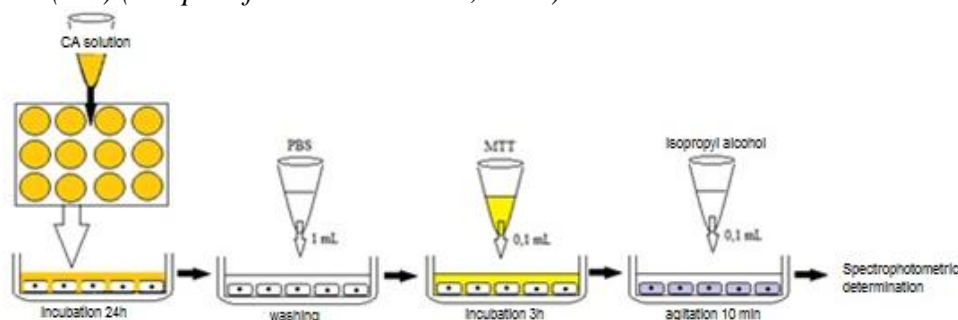
The cultures were maintained in an incubator at 37°C in an atmosphere of 5% CO<sub>2</sub> for, approximately, four weeks. The cells with confluence of 80 to 90% were trypsinized, adding MEM with fetal bovine serum to neutralize the trypsin, centrifuged at 1200 rpm for 3 minutes and, then, seeded in 96 well plates. Cell density required for cytotoxicity assays was  $1 \times 10^6$  cells/ml.

The plates were incubated for 24 hours for effective adhesion of cells to the wells. For positive control, was added 1 ml of 10% DMSO to the cells. The negative control was prepared using MEM. Finally, various concentrations of caffeic acid (solubilized in MEM with 0.5% DMSO) were added to the wells. To determine the effect of DMSO used to solubilize the CA in the cells, a control with DMSO was used for each treatment concentration (Figure 2).

The plates were then incubated for 24 hours.

**Figure 2.** Scheme of Treatment of Cells in the 96-well Plate

After this time, the treatment was removed, the plates were gently washed with Phosphate Buffer Saline (PBS) (Bruggisser et al., 2002), and 100  $\mu$ L of MTT (1 mg / ml PBS) was added to each well. The presence of purple formazan crystals was observed after 3 to 5 hours of incubation at 37°C and protected from UV light. The method can be seen in the scheme of Figure 3.

**Figure 3.** Scheme of the Procedure used to Assess the Cytotoxicity of Caffeic Acid (CA) (Adapted from Chiari et al., 2012)

To solubilize the formazan crystals, 100  $\mu$ L of isopropyl alcohol were added to each well. Then, the spectrophotometric determination of absorbance was performed at a wavelength of 595 nm in a plate reader.

The percentage of dead cells was calculated in relation to the negative control, representing the cytotoxicity of each treatment, as proposed by Zhang et al. (2004), and the percentage of living cells was also subsequently determined. A dose *versus* response curve was constructed, where it was possible to use the equation to determine the IC<sub>50</sub> of caffeic acid or the concentration of it needed to cause 50% lethality of the cells.

Cytotoxicity assays were performed on at least 3 independent experiments.

## Results and Discussion

The cytotoxicity of caffeic acid was carried out using two cell lines, HepG2 and HaCat cells. HepG2 cells were chosen since they act as a parameter of the primary route of metabolism in the human body, thus making it possible

to evaluate the toxicity of the product through its metabolite formed in these cells.

Since the focus of this study is the cosmetic area, HaCaT keratinocytes were selected as an analogue of the epidermis, thus being possible to verify the toxicity to the skin when the product is applied on it.

According to Bruggisser et al. (2002), an interaction between the MTT reagent and caffeic acid may occur, possibly due to the oxidizing action of caffeic acid that promotes the reduction of the MTT salt to form formazan crystals in the same manner as occurs in living cells, leading to a false positive result. To avoid this problem, the 96-well plates were washed gently with PBS to remove the caffeic acid and a solution of MTT was then added (Figure 4).

**Figure 4.** Culture of HepG2 Cells (left) and HaCaT Cells (right) after the Addition of MTT Solution

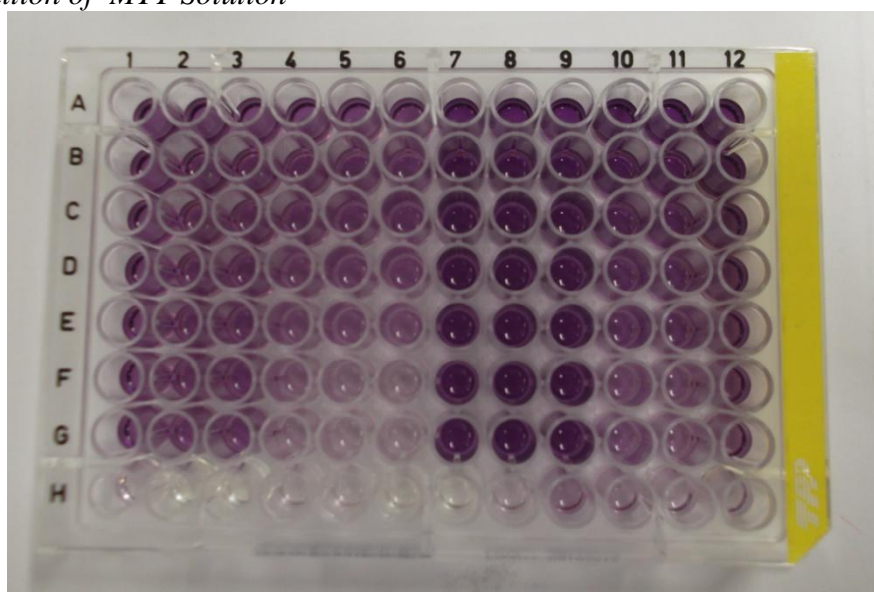
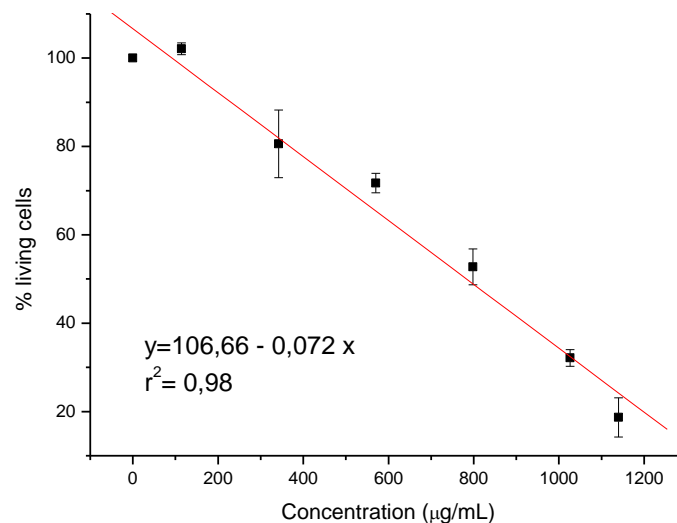


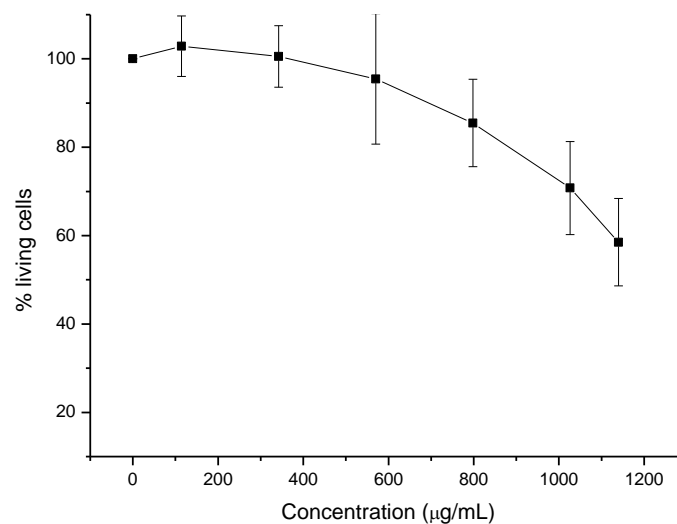
Figure 5 shows the results obtained for HepG2. A dose *versus* response curve was constructed where it was possible to use the equation generated by linear regression to determine the  $IC_{50}$  of caffeic acid, i.e. the product concentration needed to cause 50% lethality of the cells.

**Figure 5.** Dose versus Response Curve of HepG2 Cells treated with various Concentrations of Caffeic Acid



The same process was conducted for HaCat cells and results are seen in Figure 6 below.

**Figure 6.** Dose versus Response Curve of HaCat Cells treated with various Concentrations of Caffeic Acid



The results obtained are found to be promising since the caffeic acid concentration that was toxic for 50% of HepG2 cells ( $IC_{50}=781.8 \mu\text{g/mL}$ ) is approximately 330 to 400 times greater than the concentration required to inhibit 50% DPPH ( $IC_{50} \text{ DPPH}= 2.39 \mu\text{g/mL}$ ) and ABTS ( $IC_{50} \text{ ABTS}= 1.96 \mu\text{g/mL}$ ) free radicals scavengers, respectively. The maximum concentration of

caffeic acid tested (1140 µg / mL) did not reach 50% of cell death in HaCat cells.

## Conclusion

For the cytotoxicity assay, it was concluded that the caffeic acid does not promotes toxicity in HepG2 and HaCat cells in the concentration needed to exhibit antioxidant activity *in vitro* and, thus, it can be applied in topical products.

## Acknowledgments

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