

Nathália Viana de Souza Cruz

Efeito de dentifrícios com concentração reduzida de fluoreto associado ao trimetafosfato de sódio sobre a erosão do esmalte dentário

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Tese apresentada à Faculdade de Odontologia da Universidade Estadual Paulista "Júlio de Mesquita Filho", Campus de Araçatuba para obtenção de título de Mestre em Ciência Odontológica- Área de Concentração: Saúde Bucal da Criança.

Orientador: Profº Drº Alberto Carlos Botazzo Delbem

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Resumo

CRUZ, N.V.S. **Efeito de dentifrícios com concentração reduzida de fluoreto associado ao trimetafosfato de sódio sobre a erosão do esmalte dentário.** 2013. 58 f. Dissertação (Mestrado). Faculdade de Odontologia, Universidade Estadual Paulista, Araçatuba, 2013.

Resumo

O objetivo deste estudo foi avaliar *in vitro* o efeito de dentifrícios com baixa concentração de fluoreto (250 ppm), combinado com trimetafosfato de sódio (TMP) sobre a erosão do esmalte. Cento e quarenta e quatro blocos de esmalte bovino foram selecionados pela dureza de superfície inicial e submetidos a desafios erosivos (2 e 5 dias). Os tratamentos foram realizados com slurry de dentifrício (5 mL / bloco por 15 s): Sem Fluoreto e TMP (controle negativo), 250 ppm de F, 250 ppm F com 0,25, 0,5 e 1,0% de TMP e controle positivo (1425 ppm F) . O desafio erosivo foi produzido por imersão em Sprite (pH 2,8) 4 vezes / dia por 5 minutos, seguido por 1 hora em saliva artificial. A dureza de superfície, dureza transversal e desgaste da superfície foram analisados como variáveis de resposta. Os dados foram submetidos à análise de variância (2-way), seguido pelo teste de Student-Newman-Keuls. O aumento do tempo do desafio erosivo resultou em maior desgaste do esmalte ($p < 0,001$). Grupos com TMP apresentaram menor desgaste do esmalte do que os grupos com apenas flúor e o controle negativo ($p < 0,001$). A presença de maior quantidade de fluoreto (controle positivo), levou a uma superfície mais dura do esmalte ($p < 0,001$) em comparação com o controle negativo e o dentifrício de 250 ppm F. A adição de TMP ao dentifrício com 250 ppm de F teve um efeito direto sobre a dureza. O trimetafosfato de sódio (0,25%) pode aumentar o efeito contra a erosão do dentifrício de baixa concentração de fluoreto (250 ppm). Este efeito foi melhor do que o observado com a pasta de dente comercial.

Palavras-chave: Fluoretos. Cremes Dentais. Fosfatos.

Abstract

CRUZ, N.V.S. **Effect of low-fluoride toothpaste combined with sodium trimetaphosphate on enamel erosion.** 2013. 58 f. Dissertation (Master's degree) – Araçatuba School of Dentistry, UNESP—Univ. Estadual Paulista, 2013.

Abstract

The purpose of this study was to evaluate in vitro the effect of low-F toothpastes (250 ppm) combined with sodium trimetaphosphate (TMP) on enamel erosion, one hundred forty four bovine enamel blocks were selected by their initial surface hardness and subjected to a erosion challenges (2- and 5-days). Treatments were carried out using slurry toothpastes (5 mL/block for 15 s): non fluoride and TMP (negative control), 250 ppm F, 250 ppm F plus 0.25, 0.5 and 1.0% TMP toothpastes and positive control (1,425 ppm F). The erosive challenge was produced by immersion in soda (pH 2.8) 4 times/day, 5 minutes each time, followed by 1 hour in artificial saliva. The surface and cross-sectional hardness and surface wear were analyzed as response variables. The data were submitted to analysis of variance (2-way) followed by Student-Newman-Keuls's test. The rise in the time of erosive challenge resulted in a larger enamel wear ($p < 0.001$). TMP groups showed lower enamel wear than fluoride-only and negative control ($p < 0.001$). The presence of higher amounts of fluoride (positive control) led to a harder enamel surface ($p < 0.001$) compared to the negative control and 250 ppm F toothpaste. The addition of TMP to toothpaste with 250 ppm F had a direct effect on hardness. The results suggest that the sodium trimetaphosphate (0.25%) can enhance the effect against erosion of low fluoride toothpaste (250 ppm). This effect it was in higher level than the observed with the commercial toothpaste.

Keywords: Fluoride. Toothpastes. Phosphates.

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Lista de abreviaturas

% =	Porcentagem
μm =	Micrômetro
$^{\circ}\text{C}$ =	Graus Celsius
ΔKHN =	Perda integrada da dureza de subsuperfície
$\mu\text{g F/g}$ =	Micrograma de fluoreto por grama
CSH =	Cross- sectional hardness
F =	Fluoreto
FI =	Flúor iônico
FT =	Flúor total
g =	Aceleração da gravidade
h =	Hora
HCl =	Ácido clorídrico
HA =	Hidroxiapatita
KNO_3 =	Nitrato de Potássio
mL =	Mililitros
mm =	Milímetros
Mol/L =	Molaridade
NaF =	Fluoreto de Sódio
NaOH =	Hidróxido de Sódio
ppm =	Partes por milhão
pH =	Potencial hidrogeniônico
s =	Segundos
SrCl_2 =	Cloreto de Estrôncio
TMP =	Trimetafosfato de Sódio
TISAB =	Tampão de ajustador de força iônica total

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Effect of low-fluoride toothpaste combined with sodium trimetaphosphate on enamel erosion

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Summary

Background: The effect on enamel demineralisation of low-F toothpastes containing sodium trimetaphosphate (TMP) has been studied. However, there is limited literature is still scarce in giving evidence on the effect of these formulations on dental erosion.

Aim: To evaluate the effect of low-F toothpastes (250 ppm) combined with TMP on enamel erosion in vitro.

Design: One hundred forty-four bovine enamel blocks were selected by initial surface hardness and subjected to erosion challenges (2 and 5 days). Treatments were carried out using toothpaste slurries (5 mL/block for 15 s): no fluoride or TMP (negative control); 250 ppm F; 250 ppm F plus 0.25, 0.5, or 1.0% TMP; and positive control (1,425 ppm F). The erosive challenge was produced by immersion in soda (pH 2.8) for 5 min four times/day, followed by 1 h in artificial saliva. The surface and cross-sectional hardnesses and surface wear were analysed as response variables. The data were submitted to analysis of variance (2-way) followed by Student–Newman–Keuls test.

Results: Increased duration of erosive challenge resulted in greater enamel wear ($p < 0.001$). TMP groups showed less enamel wear than fluoride-only and negative control groups ($p < 0.001$). Increased concentration of fluoride (positive control) led to a harder enamel surface ($p < 0.001$) compared with the negative control and 250 ppm F

toothpaste. The addition of TMP to toothpaste with 250 ppm F had a direct effect on hardness.

Conclusion: TMP (0.25%) enhances the anti-erosive effect of low-F toothpaste (250 ppm).

1 Introduction

Dental erosive wear has become more prevalent and its incidence have varied between 12.3% and 18% in 11 to 16-year-old children [1]. Tooth erosion is a multifactorial condition (patient-related and nutritional factors) and over time, the interaction of all these factors may lead to either progression or indeed protection of the surface [2]. The terms erosion and erosive tooth wear have been suggested to refer to the chemical and chemical-mechanical processes [3].

As is the case in caries and periodontal disease, changes in oral health and dietary habits, and oral products and toothpastes, are used to prevent or reduce the progression of dental erosion [4]. Fluoride has been used as a supplementary preventative measure to decrease dental erosion, and displays superior effect when applied in high concentration [5]. Topical fluoride is most frequently applied as a toothpaste but has limited beneficial effects on erosion and abrasion [6-8].

The addition of some actives to improve the effectiveness against erosion has been studied in toothpastes with 1,000 to 1,450 ppm F. A positive association exists between 10% xylitol and fluoride at 1,030 ppm F on the effectiveness of the toothpaste on erosion. This may be related to the ability of xylitol to facilitate calcium movement and availability in the demineralised areas [9]. Some commercial toothpastes have both fluoride and desensitising agents (e.g., SrCl_2 , potassium citrate, and KNO_3) and are claimed to prevent erosion [10]. However, the interaction between NaF and KNO_3 did not produce favourable results. Another possible combination is fluoride and inorganic phosphate, such as sodium trimetaphosphate (TMP) [11]; 3% TMP in a toothpaste with 500 ppm F showed greater erosion reduction than a toothpaste with 1,100 ppm F.

Utilisation of this TMP and F association has shown decreased enamel demineralisation [11-14] in caries and dental erosion. In addition, the fluoride concentration can be reduced in products functioning against

demineralisation. Considering the good results against dental caries (data not published - in vitro study) and which differs from erosion process, the present study evaluated, in vitro, the effect of low-F toothpastes (250 ppm) plus TMP on the erosion of tooth enamel.

2 Material and methods

Experimental design

One hundred forty-four bovine enamel blocks were assigned among the following toothpastes: no fluoride or TMP (negative control); 250 ppm F; and 250 ppm F plus 0.25, 0.5, or 1.0% TMP experimental toothpastes as well as a commercial toothpaste (positive control). The groups of blocks were additionally subdivided into 2 conditions according to duration of erosive challenge (2 or 5 days). A group size of 12 enamel blocks was calculated considering an α -error level of 5% and a β -error level of 10%, based in the results of Moretto et al. [11] and Manarelli et al. [14]. The erosive challenge was produced in all blocks by immersion in soda (Sprite Zero™, pH 2.8) for 5 min 4 times per day. The factors studied were type of dentifrice and duration of erosive challenge. The surface and cross-sectional hardnesses and surface wear were analysed as response variables.

Enamel sample preparation

Enamel blocks (4 × 4 mm) were harvested from bovine incisors kept in 2% formalin, pH 7.0, for 30 days prior to experimental procedures. The enamel surface of the blocks was ground flat with water-cooled silicon carbide paper discs (400, 600, and 1,200 grades; Buehler, Lake Bluff, Ill, USA) removing approximately 200 μ m of the enamel surface. After polishing, the baseline hardness of the enamel surface was determined and blocks with 340.0 to 370.0 kgf/mm² were selected. To maintain a

reference surface for lesion-depth determination by profilometry, 2 layers of nail varnish were applied to half of the surface of each block [14].

Toothpaste formulation and fluoride and pH analysis

The experimental toothpastes were manufacture and comprised the same basic formula (except for F and the presence or lack of TMP) including the following ingredients: carboxymethyl cellulose, sodium methyl-*p*-hydroxybenzoate, sodium saccharin, peppermint oil, glycerol, hydrated silica, sodium lauryl sulphate, and water. To compare and validate the results, the following toothpastes were assembled: negative control (without F or TMP); 250 ppm F (NaF; Merck, Darmstadt, Germany); and 250 ppm F with TMP (TMP; Sigma-Aldrich Co., St Louis, MO, USA) added at concentrations of 0.25, 0.5, or 1.0%. A commercial toothpaste was utilised as positive control (Sensodyne Pro-Enamel, 1,425 ppm F as NaF, 5% KNO₃, pH 7.0; GlaxoSmithKline Brazil Ltd., Rio de Janeiro, Brazil).

Total fluoride (TF) and ionic fluoride (IF) were determined using a specific electrode for F (Orion 9609-BN; Orion Research Inc., Beverly, MA, USA) according to the methods described by Delbem et al. [15]. Approximately 0.1 g of each toothpaste was combined with 10.0 mL deionised water in a polypropylene tube (15 mL). For the TF measurement, 0.25 mL dentifrice suspension, to which was added 0.25 mL 2 mol/L HCl, was agitated for 1 h at 45°C, and then 0.5 mL 1.0 mol/L NaOH was added to the mixture. For IF, the dentifrice suspension was centrifuged at $906 \times g$ for 20 min and 0.25 mL of the supernatant solution was treated with 0.25 mL 2 mol/L HCl and 0.5 mL 1.0 mol/L NaOH. TISAB II was added to the samples and the standards (total ionic strength adjustment buffer) and results were expressed in µg F/g.

The pH of the toothpaste slurries was analysed in triplicate with a pH electrode (FH 2A09E-Analyser; São Paulo, SP, Brazil) and ion analyser (Orion 720A^{plus}). The pH electrode was calibrated according to standard pH levels of 7.0 and 4.0.

Erosive challenge and toothpaste treatment

For 2 or 5 days, the enamel blocks were subjected to erosive challenges, which comprised immersion in soda (Sprite™, pH 2.8, 5 mL/block; Companhia de Bebidas Ipiranga, Ribeirão Preto, Brazil) during 5 min (unstirred), 4 times per day [16-19]. After each erosive immersion, the enamel blocks were washed with deionised water then exposed to one of the toothpaste slurries (toothpaste/deionised water, 1:3 w/w; 5 mL/block) for 15 s at room temperature. The enamel blocks were stored in artificial saliva (5 mL/block; unstirred; 37° C) for 1 h between acid challenges/treatments. The composition of the artificial saliva was as follows: 1.5 mmol/L $\text{Ca}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$, 0.9 mmol/L $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$, 150 mmol/L KCl, 0.1 mol/L Tris buffer, pH 7.0, 0.03 ppm F [20]. Following all 4 challenges/treatments, blocks remained in artificial saliva until the beginning of the next day erosive challenge. The solutions were changed once a day.

Surface hardness and profilometric analysis

The surface hardness for selection of the enamel blocks (baseline) was determined using a Micromet 5114 hardness tester (Buehler, Lake Bluff, Ill, USA) and the software Buehler OmniMet (Buehler). Five indentations were placed 100 μm apart using a Knoop diamond (25 g, 10 s). After the experimental period, the surface hardness was again measured in the area exposed to the erosive challenge. Five indentations, as used for baseline hardness determination, were made in a different part of the surface.

Enamel loss was determined in relation to the reference surface by profilometry (Surftest SJ 401; Mitutoyo American Corp.). The profilometric trace (length = 2 mm) included the reference surface (baseline) and the exposed surface. Five readings were performed on each block and the average enamel loss was calculated (μm) [14].

Cross-sectional hardness determination

Blocks were sectioned at the centre, and half of each block was incorporated in acrylic resin and subsequently polished. Cross-sectional hardness (CSH) was determined under a 5-g load for 10 s using a Micromet 5114 hardness tester. Nine indentations at different distances (5, 10, 15, 20, 25, 30, 40, 50, and 70 μm) from the external surface of the enamel were made in the centre of the blocks, for both the control and the eroded areas. The integrated area of hardness ($\text{kgf}/\text{mm}^2 \times \mu\text{m}$) of the demineralised and sound enamel was calculated using the trapezoidal rule (GraphPad Prism, version 3.02) and subtracted from the integrated area of hardness of sound enamel to yield integrated hardness (ΔKHN) [21].

Statistical analysis

The values of enamel wear (μm), surface hardness, and ΔKHN were considered as outcome measures. The toothpastes and periods of erosive challenge were the variation factors. The data from wear and surface hardness were submitted to analysis of variance (2-way) followed by Student–Newman–Keuls method test for multiple comparisons, considering the toothpastes and periods of erosive challenge as variation factors. In each periods erosive (2 or 5 days), the results from ΔKHN were subjected to analysis of variance (2-way) followed by Student–Newman–Keuls method, considering the toothpastes and depth in enamel as variation factors. A correlation test (Spearman's test) was applied to verify relationships between the variables. SigmaPlot software version 12.0 was used for the statistical analyses, and the significance limit was set at 5%.

3 Results

The concentrations of fluoride in the experimental and commercial toothpastes were as expected and showed a variation of less than 10% (Table 1). The addition of TMP to toothpastes did not affect the pH of the

formulations and proved to be neutral. Data surface hardness of the enamel before the erosion challenge showed good standardisation of the preparation of enamel blocks. The mean values (342.5–350.9 kgf/mm²) were similar among groups and the duration of the erosive challenge ($p = 0.960$).

Despite having more fluoride in the formulation, the positive control showed wear similar ($p \geq 0.682$) to those of the negative control and 250 ppm F toothpastes (Figure 1A). When TMP was combined with fluoride, the enamel wear was 2-fold lower than that of the negative control ($p < 0.001$). Increased duration of erosive challenge resulted in greater enamel wear ($p < 0.001$). Meanwhile, time did not influence the protective action of the toothpastes ($p = 0.224$). However, a reduction in the effect against erosion was observed by increasing the TMP concentration in the toothpaste.

The presence of a higher concentration of fluoride (positive control) led to a harder enamel surface ($p < 0.001$) compared with the negative control and 250 ppm F toothpastes (Figure 1B). Regards to positive control versus 250 ppm F toothpastes, this hardness was 40% greater after the 2-day challenge and 100% after the 5-day challenge. Because the toothpaste with 250 ppm F showed similar hardness values to the negative control ($p = 0.683$), the addition of TMP to toothpaste with 250 ppm F had a direct effect on hardness. By combining 0.5% and 1% TMP with 250 ppm F the surface hardness was similar to that of the positive control ($p \geq 0.100$). There was not correlation between surface hardness values and wear values (Spearman's $r = -0.148$; $p = 0.077$).

Regarding depth of demineralisation of remaining enamel (Figure 1C), the formulation with higher capacity of remineralisation than the negative control ($p < 0.001$). This analysis revealed that erosion time influenced the action of the toothpaste ($p < 0.001$). For a longer period of erosion, it is possible to observe a greater differences between: groups with and without fluoride, with lower and higher concentrations of fluoride and lower and higher percentage of TMP in the formulations ($p < 0.001$). This

is also evident in depth analysis in the enamel (Figure 2). The demineralisation at 5 days of the remaining enamel as a function of depth showed profile more different compared to 2 days. Nevertheless, the addition of TMP (0.5% and 1%) reduced demineralisation of the remaining enamel to a depth similar to that of the positive control toothpaste ($p \geq 0.473$). There was correlation between ΔKHN values and wear values (Spearman's $r = 0.566$; $p < 0.001$).

The demineralisation profile of the remaining enamel as a function of depth showed the mineral loss after 5 days of erosion was 15 μm greater than that after 2 days (Figure 2). At a depth of 5 μm after a 2-day erosion challenge, a concentration of 250 ppm F produced decreased softening of the remaining enamel when compared with the negative control.

4 Discussion

In recent decades, studies have focused on alternatives that may help improve effectiveness against enamel demineralisation (caries and erosion). However, there are limited reports on the association between phosphate and fluoride; this study aimed to evaluate the effectiveness of this combination in toothpaste formulations against enamel erosion. The choice of sodium trimetaphosphate (TMP) was due to its ability to reduce the solubility of hydroxyapatite (HA) in acidic medium and to remain bound to the HA longer than other polyphosphates [22]. In this study, formulations with TMP without fluoride were not tested because, despite its action against demineralisation, its effect is lower than when associated with fluoride [12].

The addition of TMP to the 250 ppm F formulation enabled verification of an effect against erosion not present with a concentration of 1,425 ppm F. Notably, increasing the TMP concentration from 0.25% to 1.0% did not improve the anti-erosive effect of the toothpaste, but the remaining enamel was less demineralised. The adsorption of TMP onto the enamel surface [22,23] facilitates the diffusion of calcium ions [13,23], as well as

fluoride ions, according to Takeshita et al. [12]. This may explain the lower demineralisation of remaining enamel with increasing TMP concentration in a low-fluoride formulation (250 ppm). However, as noted earlier, enamel with decreased softening does not have decreased wear. After a 5-day period of erosion, the enamel was less demineralised with increasing TMP concentration in the 250 ppm F formulation; however, there was no change in the effect against erosion. Thus, the lower wear in the presence of the TMP/F combination may be related to the ability of TMP to block acid diffusion into the deep layers of the enamel. In addition, TMP interacts with the enamel, which may produce a protective layer on the enamel surface thereby hindering acid diffusion [13,23]. Thus, the major effect of TMP during the erosive challenge is to prevent acid diffusion into the enamel.

Acid accumulation over 5 days increased enamel wear and affected the efficacy of the TMP/F combination against erosion. This effect was seen mainly at high TMP concentrations in association with fluoride. These results agree with those of Manarelli et al. [14], which revealed increased enamel wear with increased TMP concentration in the formulation, together with an increase in enamel hardness as in the present study. Because polyphosphates may be adsorbed in greater quantities and more rapidly at acid pH [24], TMP might have displayed a disaggregating effect [22]. TMP has a smaller disaggregating effect than other polyphosphates [22], but when applied to demineralised enamel at high concentrations, this effect seems more apparent thereby facilitating a greater loss of enamel. The following sequence of events is possible: (1) TMP reacts with demineralised enamel, especially at higher concentration, and (2) promotes disaggregation of outer enamel with resultant (3) increased enamel erosion, which exposes the inner enamel—which is harder. According to Takeshita et al. [13], TMP enhanced the reduction into the demineralisation area in the deeper layers.

As the blocks were kept stored in artificial saliva after the last erosive/treatment process would be more correct to state that the

hardness testing (superficial and in depth) analyzed the remineralizing ability of the formulations. Surface hardness may be related to enamel wear [11] but does not indicate how much structure was lost. In the present study, the surface hardness was not related to wear. Nevertheless, the cross-sectional hardness can provide great potential to explain the results obtained after erosive challenge mainly the data from 5 days. The comparison between 250 ppm F and positive control (1,425 ppm F) toothpastes clearly demonstrates this situation. With regard to the positive control toothpaste, the greater available fluoride in the formulation explains the lower enamel softening after a long period of remineralization. The erosive challenge is a process that occurs at a pH of <4.0 and a surface with less softening will not prevent demineralisation and acid diffusion into the enamel. The pH of the dissolution of calcium fluoride, fluorapatite or any precipitate that might be formed in the presence of fluoride is exceeded during the erosive challenge. The fluoride-rich precipitate inside the lesion does not necessarily block acid diffusion to deeper layers 10 and fluoride alone does not prevent enamel wear in a challenge of this level. Addition to pH, the erosive medium is not saturated with calcium and phosphate [25]. These data strengthen the effect of TMP against erosion and its mechanism of action described previously.

The in vitro studies can estimate the role of new anti-erosive compounds [26] and to screening tests [25]. However this is a chemical model and the data obtained using this model (or any other in vitro protocol) should be considered carefully due to limitations in reproducing the oral environment with all of the biological variations known to influence erosion [27].

5 Conclusion

Based on results and considering the limitations of this in vitro study, it was concluded that sodium trimetaphosphate (0.25%) can enhance the effect against erosion of low-fluoride toothpaste (250 ppm). This effect was greater than that observed with the commercial toothpaste claimed to prevent erosion. However, additional in situ and clinical studies must be conducted to confirm these in vitro results.

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Table and figure legends

Table 1. Mean values (SD, n = 3) of total (T) and ionic (I) fluoride (F) and pH in the toothpastes

Figure 1. Mean values of wear, surface hardness, and Δ KHN after the erosive challenges (2 and 5 days). Distinct letters indicate statistic difference between treatments as well as period of challenge. Vertical bars denote standard deviation.

Figure 2. Depth profiles of mean hardness for each treatment according to time of erosion challenge

(A) Erosion period of 2 days

5 μ m: 250 TMP 0.25% = 250 TMP 0.5% ($p = 0.617$); 250 TMP 0.5% = 250 ppm F ($p = 0.223$); and 250 TMP 1% = positive control ($p = 0.591$)

10 μ m: negative control = 250 ppm F ($p = 0.299$); 250 TMP 0.25% = 250 TMP 0.5% ($p = 0.627$); and 250 TMP 1% = positive control ($p = 0.796$)

15 μ m: negative control = 250 ppm F ($p = 0.647$); 250 ppm F = 250 TMP 0.5% ($p = 0.507$); and 250 TMP 0.25% = 250 TMP 1% = positive control ($p = 0.623$)

20 µm: sound = 250 TMP 0.25% = 250 TMP 1% = positive control ($p = 0.896$); negative control = 250 ppm F ($p = 0.166$); negative control = 250 TMP 0.5% ($p = 0.234$); and 250 TMP 0.25% = 250 TMP 0.5% ($p = 0.245$)

* no difference among groups ($p = 0.169$; Student–Newman–Keuls, $p < 0.001$)

(B) Erosion period of 5 days

5 µm: 250 TMP 1% = positive control ($p = 0.217$)

10 µm: 250 TMP 0.5% = positive control ($p = 0.660$)

15 µm: 250 TMP 0.5% = 250 TMP 1% = positive control ($p = 0.659$); 250 ppm F = 250 TMP 0.25% ($p = 0.893$)

20 µm: 250 TMP 0.5% = 250 TMP 1% = positive control ($p = 0.569$); 250 ppm F = 250 TMP 0.25% ($p = 0.265$)

25 µm: sound = 250 TMP 0.5% = 250 TMP 1% = positive control ($p = 0.725$); 250 ppm F = 250 TMP 0.25% ($p = 0.754$)

30 µm: sound = 250 ppm F = 250 TMP 0.25% = 250 TMP 0.5% = 250 TMP 1% = positive control ($p = 0.725$)

* no difference among groups ($p = 0.754$; Student–Newman–Keuls, $p < 0.001$)

Table 1. Mean values (SD, n = 3) of total (T) and ionic (I) fluoride (F) and pH in the toothpastes

Groups	TF (µg F/g)	IF (µg F/g)	pH
Negative control	11.1 (1.4)	11.5 (0.8)	7.7 (0.02)
250 ppm F	268.7 (3.1)	272.3 (1.1)	7.1 (0.02)
250 TMP 0.25%	282.0 (2.8)	273.4 (6.8)	7.5 (0.01)
250 TMP 0.5%	270.6 (3.5)	277.3 (2.1)	7.3 (0.01)
250 TMP 1%	271.2 (0.9)	270.3 (6.7)	7.0 (0.02)
Positive control	1,578.2 (14.0)	1,432.2 (230.3)	7.3 (0.04)

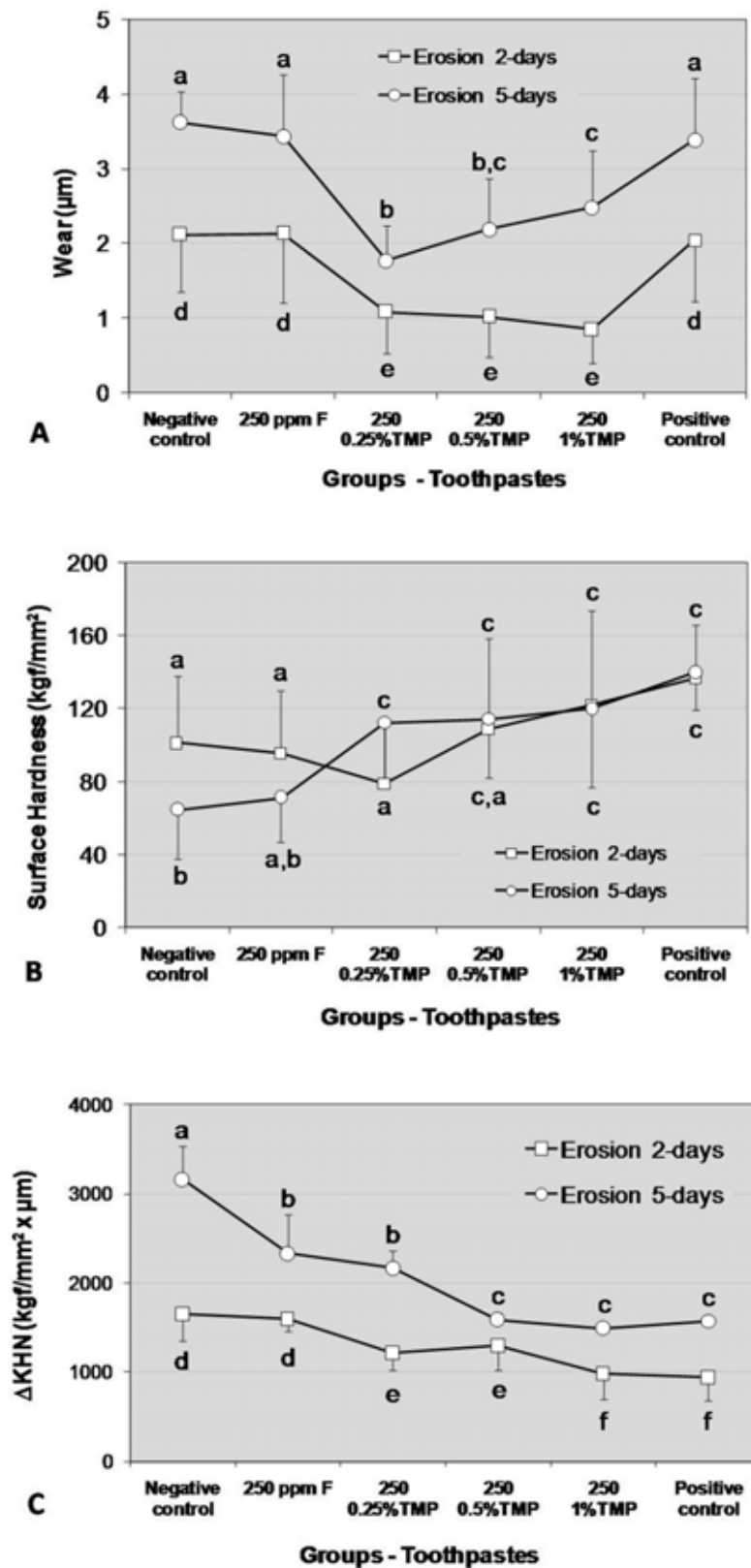


Figure 1. Mean values of wear, surface hardness, and ΔKHN after the erosive challenges (2 and 5 days). Distinct letters indicate statistic difference between treatments. Vertical bars denote standard deviation.

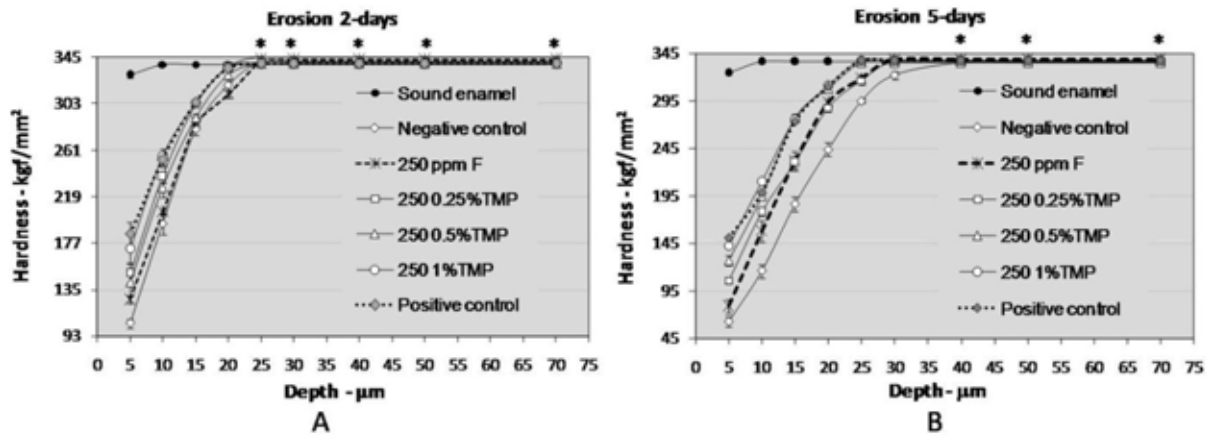


Figura 2. Depth profiles of mean hardness for each treatment according to time of erosion challenge

Anexos

ANEXO A

INSTRUÇÕES AOS AUTORES



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Original Articles: Divided into: Summary, Introduction, Material and methods, Results, Discussion, Bullet points, Acknowledgements, References, Figure legends, Tables and Figures arranged in this order. The summary should be structured using the following subheadings: Background, Hypothesis or Aim, Design, Results, and Conclusions and should be less than 200 words. A brief description, in bullet form, should be included at the end of the paper and should describe Why this paper is important to paediatric dentists.

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- Please provide up to 3 bullet points for your manuscript under the heading: 1. Why this clinical report is important to paediatric dentists. Bullet points should be added to the end of your manuscript, before the references.

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Introduction should be brief and end with a statement of the aim of the study or hypotheses tested. Describe and cite only the most relevant earlier studies. Avoid presentation of an extensive review of the field.

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 2003; 61: 347-355.
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 consequences of premature birth on palatal morphology, dental occlusion,
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A maximum of 30 references should be numbered consecutively in the order in which they appear in the text (Vancouver System). They should be identified in the text by bracketed Arabic numbers and listed at the end of the paper in numerical order. Identify references in text, tables and legends. Check and ensure that all listed references are cited in the text. Non-refereed material and, if possible, non-English publications should be avoided. Congress abstracts, unaccepted papers, unpublished observations, and personal communications may not be placed in the reference list. References to unpublished findings and to personal communication (provided that explicit consent has been given by the sources) may be inserted in parenthesis in the text. Journal and book references should be set out as in the following examples:

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2. Ministry of Health, Department of Planning. Annual Statistical Report.

- Abu Dhabi: Ministry of Health, 2001.
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If more than 6 authors please, cite the three first and then et al. When citing a web site, list the authors and title if known, then the URL and the date it was accessed (in parenthesis). Include among the references papers accepted but not yet published; designate the journal and add (in press). Please ensure that all journal titles are given in abbreviated form.

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Anexo B

Preparo dos blocos de esmalte bovino (4x4mm)



1. Coroa do dente bovino incisivo central inferior, separada da raiz por meio de disco diamantado de duas faces (KG Sorensen D 91), montado em motor de bancada (Nevoni), mantido sob refrigeração (água destilada/deionizada).
-



2. Secção da coroa utilizando disco diamantado (série 15 HC Diamond - n. 11-4244 Buehler) separando a superfície vestibular da lingual.
-



3. Face vestibular fixada na placa de acrílico.
-



4. Secção da face vestibular no sentido longitudinal, na porção mais plana, utilizando-se 2 discos diamantados (série 15 HC Diamond –n. 11-4244 Buehler), montados em cortadeira sob refrigeração com água destilada/deionizada e separados por um disco espaçador de alumínio com 4 mm de espessura. Em seguida, foi realizado o corte no sentido transversal.
-



5. Fragmento vestibular do dente bovino, fixado sobre placa de resina. Ao lado, bloco de esmalte dentário.
-

Anexo c

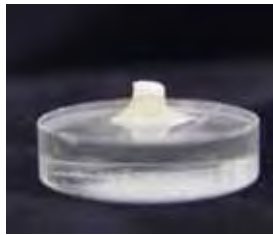
Planificação da dentina e polimento do esmalte



1. Bloco de esmalte fixado em disco de resina acrílica pré-fabricada (± 3 cm de diâmetro por ± 11 mm de espessura), com auxílio de cera pegajosa (Kota Ind. e Com. LTDA), com a superfície dentinária voltada para cima.

-
2. Ajuste da dentina para obtenção de superfícies paralelas entre esmalte e dentina, utilizando Politriz Beta – Grinder – Polisher e Vector Power Head (Buehler, Lake Bluff, IL, USA) e lixas de granulação 320 (Carbimet Paper Discs, 30-5108-320, Buehler), durante 30 segundos sob baixa rotação e refrigeração.
-





3. Blocos fixados com a superfície do esmalte voltada para cima para serem polidos.

Seqüência do polimento de esmalte:

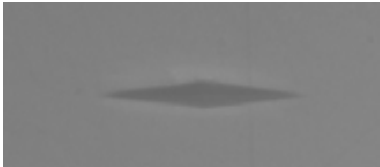
- ✓ Pedra-pomes, água deionizada e taça de borracha montada em contra-ângulo em baixa-rotação.
- ✓ Lixas de granulação 400 (30 segundos), 600 (20 segundos) e 1200 (30 segundos) e refrigeração a água. Limpeza em lavadora ultrassônica e água destilada/deionizada por 2 minutos, entre cada lixa;
- ✓ Acabamento final com disco de papel feltro TEXMET 1000 (Buehler Polishing Cloth) durante 1 minuto com suspensão de diamante 1 micron base-água (Buehler);
- ✓ Lavagem durante 30 segundos com jato de água deionizada;
- ✓ Limpeza em lavadora ultrassônica Modelo 2110 (Branson, Danbury CT,USA) com água destilada/ deionizada (2minutos).

Anexo D

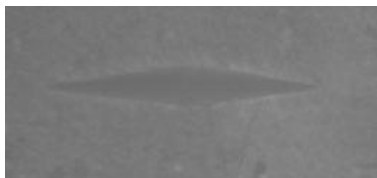
Análise da dureza de superfície do esmalte



1. Microdurômetro Micromet 5114 Hardness Tester (Buehler, Lake Bluff, USA e Mitutoyo Corporation, Kanagawa, Japan), com penetrador tipo Knoop, acoplado ao Software para análise de imagem Buehler OminMet (Buehler, Lake Bluff, USA).
-



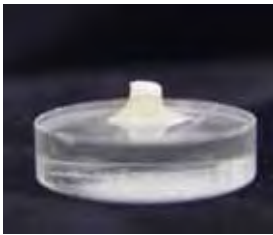
2. Fotomicrografia da impressão inicial (SH_i). (Aumento: 50x)
-



3. Fotomicrografia da impressão final (SH_f). (Aumento: 50x)
-

Anexo E

Proteção dos blocos de esmalte



1. Seleção prévia dos blocos realizada através de microdureza e observação a olho nu da superfície de cada bloco.
-



2. Esmalte cosmético foi utilizado para proteger uma área dos fragmentos necessária para realização dos testes de perfilometria.
-

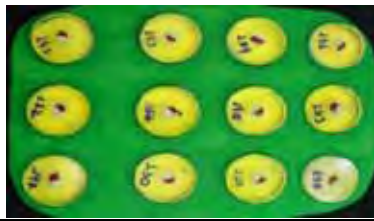
Anexo F

Esquema de desafio erosivo e tratamento



Anexo G

Sequência de erosão



Dispositivo em E.V.A., com capacidade para 12 blocos, colocado em recipiente próprio com solução remineralizadora em intervalos de 2 horas.



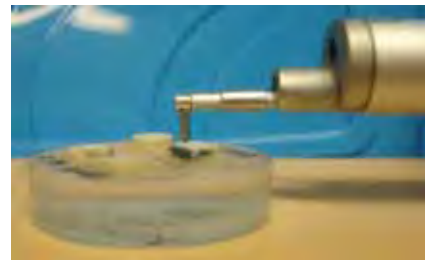
Dispositivo em E.V.A., com capacidade para 12 blocos, colocado em recipiente com sprite zero® durante 5 minutos.



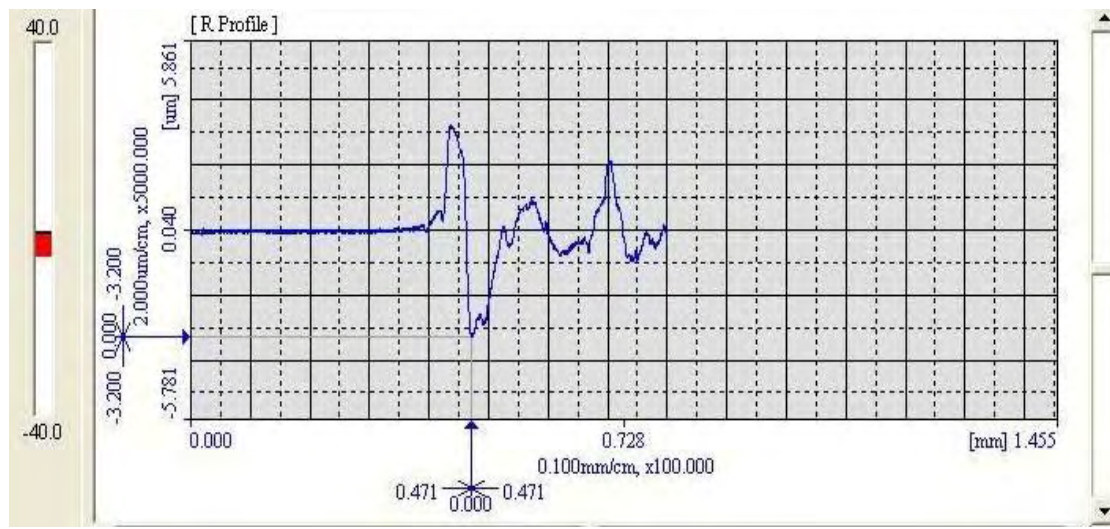
Dispositivo em E.V.A., com capacidade para 12 blocos, colocado em recipiente com slurry de dentífrício (1:3) durante 15 segundos.

Anexo H

Análise do desgaste



Rugosímetro – Mitutoyo (Mitutoyo American Corporation)



Surfpack-SJ versão 1.6

Anexo I

Análise da dureza em secção longitudinal



1. Embutir – utilizada para inclusão dos blocos de esmalte em 5 gramas de resina acrílica auto polimerizante (JET, Campo Limpo Paulista, Brasil). Os blocos foram fixados em posição com cola adesiva (Super Bonder – Loctite).



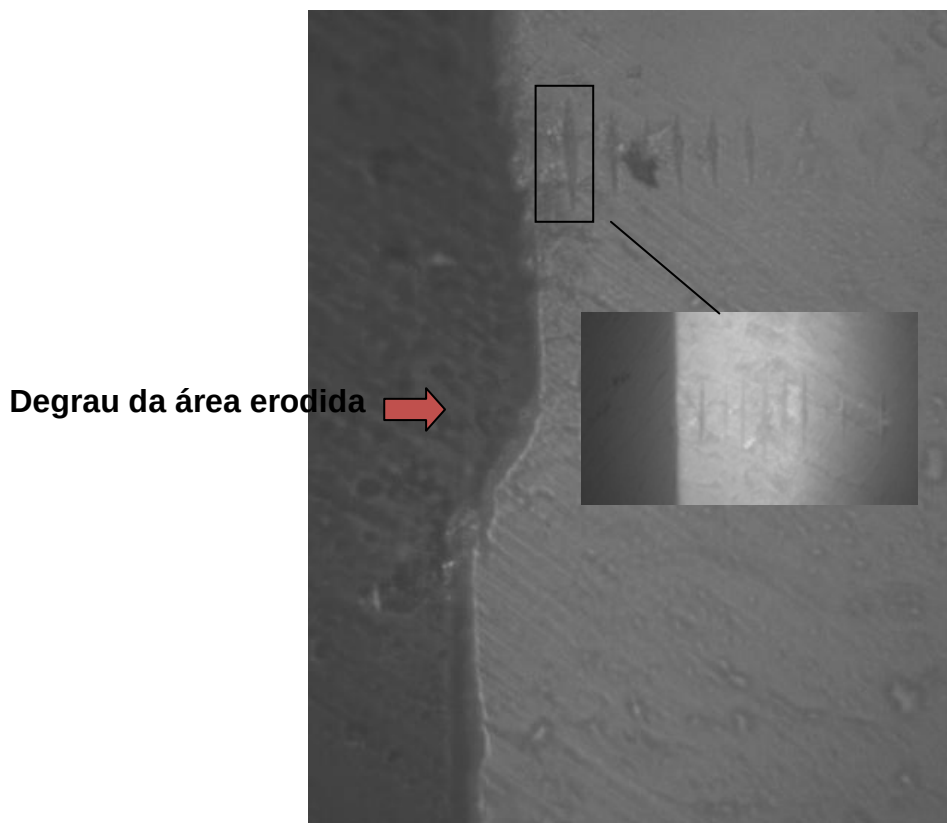
2. Corpo de prova – plano longitudinal voltado para a superfície da resina acrílica.



3. Microdurômetro Micromet 5114 Hardness Tester (Buehler, Lake Bluff, USA e Mitutoyo Corporation, Kanagawa, Japan), com penetrador tipo Knoop, acoplado ao Software para análise de imagem Buehler OminMet (Buehler, Lake Bluff, USA).

Seqüência do polimento de esmalte:

- ✓ Lixas de granulação 320 (1 minuto), 600 e 1200 (2 minutos) e refrigeração a água. Limpeza em lavadora ultrassônica e água destilada/ deionizada por 2 minutos, entre cada lixa;
- ✓ Acabamento final com disco de papel feltro Microcloth Supreme PSA (Buehler) durante 2 minutos com suspensão de diamante 1/4 micron base-água (Buehler);
- ✓ Lavagem durante 30 segundos com jato de água deionizada;
- ✓ Limpeza em lavadora ultrassônica utilizando água destilada/deionizada (2 minutos).



Fotomicrografia das impressões. (Aumento: 1000x)

Anexo J**Esquema representativo da dosagem de F nos
dentifrícios**