



Research article

Strain engineering in molybdenite: A theoretical insight into the 2D phase of α -MoO₃

José A.S. Laranjeira^a, Fredy M. Gonzalo^a, Victor J.R. Rivera^a, Luis A. Cabral^b,
Pablo A. Denis^c, Julio R. Sambrano^{a,*}

^a Modeling and Molecular Simulation Group, São Paulo State University, School of Sciences, Bauru, SP, Brazil

^b Department of Physics and Meteorology, São Paulo State University, School of Sciences, Bauru, SP, Brazil

^c Computational Nanotechnology, DETEMA, Facultad de Química, UDELAR, Montevideo 1157, 11800, Uruguay

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ABSTRACT

This first-principles study explores the structural, thermal, mechanical, electronic, optical, and vibrational properties of molybdenite, the 2D phase of α -MoO₃ with P2₁/m symmetry. The structure is stable at near room temperature and meets the Born–Huang mechanical criteria. Molybdenite shows anisotropic mechanical properties, with Young's modulus between 84.33 and 138.48 N/m, and Poisson's ratio from 0.14 to 0.23. It has an indirect band gap with a valence band maximum at the S point and a conduction band minimum at Γ , varying from 1.75 eV (PBE) to 2.95 eV (HSE06). Light polarized along x shows visible range absorption and a peak around 3.50 eV, while y-polarized light is mainly active in the ultraviolet. Strain engineering shows a significant band gap change under the y-strain, from 1.10 to 2.30 eV, while the gap remains stable under the x-strain. Infrared and Raman spectra identify key vibrational modes at 713.06 cm⁻¹ and 708.39 cm⁻¹, respectively. Notably, the strain also induces substantial changes in the optical response, including enhanced anisotropy and redshifts in the absorption edge, further expanding the potential of molybdenite for applications in flexible electronics, strain-tunable sensors, and polarization-sensitive optoelectronic devices.

1. Introduction

Molybdenum trioxide (MoO₃) is a transition metal oxide that has attracted significant attention due to its chemical stability, low toxicity, abundance, and multifunctional properties [1–3]. It is widely explored in applications such as catalysis [4,5], gas sensing [6–8], electrochromic devices [9,10], and energy storage systems [11,12]. MoO₃ exists in several polymorphic forms, including the orthorhombic (α -MoO₃), monoclinic (β -MoO₃), and hexagonal (h -MoO₃) phases [13]. Among these, α -MoO₃ is the most thermodynamically stable at room temperature and is the most extensively studied [14]. The β and h phases are metastable and typically transform into the α -phase upon heating or prolonged storage.

α -MoO₃ features a layered structure composed of distorted [MoO₆] octahedra, which contributes to its high anisotropy and enables inter-layer diffusion of ions or molecules [3]. As an n-type semiconductor, MoO₃ exhibits a wide band gap and a layered crystal structure that facilitates intercalation and ion transport, making it advantageous for

various technological applications [15,16]. To tailor the structural, optical, and mechanical properties of MoO₃, various synthesis methods have been developed, including sol–gel processing [17–19], hydrothermal synthesis [20–22], microwave-assisted synthesis [23–25], solvothermal methods [26], and sonochemical approaches [27,28].

In this framework, several methods have been established for synthesizing few-layered MoO₃, such as mechanical exfoliation [29], pulsed laser deposition [30], thermal evaporation [31], sputtering [32], spray pyrolysis [33], chemical vapor deposition [34], hydrothermal synthesis [35] and electrodeposition [36]. Among these, liquid-phase exfoliation has emerged as a widely used approach for obtaining single- and few-layered structures, typically involving solvent-assisted grinding followed by sonication, either in pure solvents [37–42] or in aqueous solutions containing surfactants.

Liu et al. [43] described an efficient method for exfoliating 2D MoO₃ nanosheets through a combination of grinding and sonication. They identified a 50:50 ethanol–water mixture as the most suitable solvent system, despite each solvent being less effective on its own. The

* Corresponding author.

E-mail address: jr.sambrano@unesp.br (J.R. Sambrano).

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produced 2D MoO₃ nanosheets notably improve sensor performance by providing greater sensitivity and quicker response and recovery times, due to their expanded surface area and numerous active sites.

Jeon and colleagues [44] synthesized 2D MoO₃ nanosheets by intercalating and exfoliating MoS₂, followed by oxidation with nitric acid. These nanosheets retain the 2D shape of MoS₂, with a thickness of about 70 % thinner than 9 nm, allowing for better ion diffusion than bulk MoO₃. In electrochromic devices, they provide superior optical contrast and faster switching, approaching the performance of advanced WO₃ nanostructures, thus demonstrating great potential for electrochromic applications.

Employing first-principles density functional theory (DFT) simulations, Liu and coworkers [45] predicted the high-performance hydrogen gas sensing of 2D MoO₃ doped with Pt or Rh. Upon exposure to oxygen in the air, the band gap of Pt- and Rh-doped MoO₃ decreases significantly by 0.62 eV and 0.51 eV. The sensor exhibits substantial charge variations—1.487 e for oxygen adsorption and 0.394 e for hydrogen adsorption—indicating high sensitivity and a low detection limit.

DFT calculations and deformation potential theory were utilized to demonstrate that few-layer MoO₃ exhibits exceptionally high acoustic-phonon-limited carrier mobility exceeding 3000 cm² V⁻¹ s⁻¹ [46]. Moreover, a pronounced in-plane transport anisotropy is observed, with mobility ratios between crystallographic directions reaching 20–30. This anisotropy arises from the strong directional dependence of both the elastic modulus and the deformation potential, which are closely linked to the valence charge density distribution and the states near the band edges.

Tong et al. [47] employed first-principles calculations and the Boltzmann transport equation to investigate the phonon-limited thermal conductivity of MoO₃. Remarkably, they predicted an ultralow room-temperature in-plane thermal conductivity of 1.57 W/mK and 1.26 W/mK along the two principal crystallographic directions—values significantly lower than those of most known 2D materials. This behavior is attributed to soft flexural and acoustic phonon modes coupled through the finite thickness, as well as to pronounced anharmonic lattice interactions that enable strong three- and four-phonon scattering. These findings position MoO₃ as a promising material for thermoelectric applications.

The present work presents a characterization of the 2D phase of molybdenum trioxide, which we will refer to as molybditene, through first-principles DFT calculations. The study includes an examination of the electronic structure via band structure analysis and density of states. Additionally, mechanical and optical properties will be investigated, along with an evaluation of the effects of uniaxial strain on the structural and electronic features of molybditene.

2. Computational Setup

In this study, all computational analyses were executed using the Vienna ab initio simulation package (VASP) [48]. The generalized gradient approximation (GGA), based on the Perdew, Burke, and Ernzerhof (PBE) functional [49,50], was employed to calculate the exchange-correlation energies, while the projector augmented wave (PAW) method [50] accounted for interactions between core and valence electrons. Electronic states were described using plane waves, and the plane wave cutoff energy was chosen at 520 eV. To avoid interlayer interactions, a 15 Å vacuum layer was added along the c-axis. For the Brillouin zone, structural optimization (partial density of states) was performed on a 5 × 5 × 1 (18 × 18 × 1) Gamma-centered k-point grid. Structural optimizations were carried out using the conjugate-gradient algorithm until the convergence criteria were satisfied, which involved energy deviations for atomic positions and lattice parameters being below 10⁻⁵ eV, and Hellmann-Feynman forces per atom not exceeding 0.015 eV/Å. To comprehensively evaluate the thermodynamic stability of molybditene, ab initio molecular dynamics (AIMD) simulations were performed [51,52]. To a more accurate

description of the electronic band gap, the HSE06 functional [53] was utilized. For this functional, the structure was optimized employing a coarser 3 × 3 × 1 k-point mesh. Subsequently, a denser mesh of 9 × 9 × 1 mesh was considered for band structure calculations.

Vibrational analysis at the Γ point was carried out via numerical second derivatives of the total energies estimated with the coupled perturbed HF/Kohn–Sham algorithm [54] implemented in CRYSTAL17 software [55]. The structural parameters, including average bond lengths (l_{av}), distortion index (D), polyhedral volume (V), and effective coordination number (ECoN), were collected using the VESTA software [56].

The elastic constants (C_{ij}) were determined by taking the second derivative of the energy (E) with respect to the strain components (ϵ_i and ϵ_j), as described by the following equation:

$$C_{ij} = \partial^2 E / \partial \epsilon_i \partial \epsilon_j \quad (1)$$

The orientation-dependent behavior of Young's modulus (Y), shear modulus (G), and Poisson's ratio (ν) was obtained through the following expressions [57]

$$1/E(\theta) = S_{11}c^4 + S_{22}s^4 + 2S_{16}c^3s + 2S_{26}cs^3 + (S_{66} + 2S_{12})c^2s^2 \quad (2)$$

$$\nu(\theta)/E(\theta) = (S_{66} - S_{11} - S_{22})c^2s^2 - S_{12}(c^4 + s^4) + (S_{26} - S_{16})(cs^3 - c^3s) \quad (3)$$

$$1/4G(\theta) = (S_{11} + S_{22} - 2S_{12})c^2s^2 + S_{66}(c^2 - s^2)^2 / 4 - (S_{16} - S_{26})(c^3s - cs^3) \quad (4)$$

where $s = \sin\theta$, $c = \cos\theta$, and $\theta \in [0, 2\pi]$ is the angle with respect to the +x axis. $S_{ij} = C_{ij}^{-1}$ are the elastic compliance constants.

To quantify the mechanical anisotropy, we evaluated the angular dependence of the in-plane Young's modulus (Y), shear modulus (G), and Poisson's ratio (ν), the degree of anisotropy was quantified by the ratio (η_p)

$$\eta_p = \frac{P_{\max}}{P_{\min}} \quad (5)$$

where P denotes Y , G or ν . This dimensionless ratio provides a straightforward metric to assess the directional variation of each property and the extent of anisotropy in the 2D material.

The energetic stability of the molybditene was evaluated by computing its cohesive energy (E_{coh}), which is defined by:

$$E_{coh} = (E_{\text{Molybditene}} - \sum_i n_i E_i) / \sum_i n_i, \quad (6)$$

where $E_{\text{Molybditene}}$ is the total energy of the molybditene structure, E_i is the energy of an isolated atom i (Mo and O), and n_i is the number of i atoms in the sheet.

The optical properties were calculated within the independent particle approximation (IPA) using the LOPTICS = .TRUE. flag in the VASP code. In this approach, the imaginary part of the frequency-dependent dielectric tensor is computed from direct interband transitions, following the formalism outlined by Gajdoš et al. [58]. The summation involves transitions between occupied valence and unoccupied conduction bands at each k-point, requiring a sufficiently dense k-point mesh (18 × 18 × 1) and an increased number of empty bands (typically two to three times the number of occupied bands) to ensure convergence.

In this approach, the optical absorption coefficient $\alpha(\omega)$ was determined from the complex dielectric function $\epsilon(\omega)$, which describes the frequency-dependent response of the electronic system to an external electromagnetic field. The dielectric function is defined as

$$\epsilon(\omega) = \text{Re}[\epsilon(\omega)] + i\text{Im}[\epsilon(\omega)], \quad (7)$$

where $\text{Re}[\varepsilon(\omega)]$ represents the dispersive part associated with energy storage, and $\text{Im}[\varepsilon(\omega)]$ accounts for optical losses due to absorption. The absorption coefficient can then be expressed as

$$\alpha(\omega) = \sqrt{2 \left[(\text{Re}[\varepsilon(\omega)])^2 + (\text{Im}[\varepsilon(\omega)])^2 \right]^{1/2} - \text{Re}[\varepsilon(\omega)]} \quad (8)$$

This formulation highlights the direct relationship between the frequency-dependent dielectric response and the absorption of electromagnetic radiation in the material, providing valuable insights into its optoelectronic performance.

3. Results and Discussion

3.1. Structure and Stability

Molybditene adopts a monoclinic structure that crystallizes in the $P2_1/m$ space group (No. 11), as illustrated in Fig. 1. The optimized lattice parameters are $a = 3.95 \text{ \AA}$ and $b = 3.70 \text{ \AA}$. These values are closer to the reported by Zhang and Lai [46] for MoO_3 monolayer ($a = 3.721 \text{ \AA}$ and $b = 3.919 \text{ \AA}$) at the optB88-vdW level. Its structure is composed of distorted $[\text{MoO}_6]$ octahedra, with an average Mo–O bond length of approximately $2.00 \text{ \AA}$ and an effective coordination number (ECoN) of 3.42. The deviation between the ideal coordination number (6 for octahedral geometry) and the calculated ECeN indicates a significant distortion within the coordination polyhedron, reflecting the pronounced structural distortion of the $[\text{MoO}_6]$ units. This two-dimensional phase can be derived from the exfoliation of a single layer of $\alpha\text{-MoO}_3$. The computed cohesive energy (E_{coh}) is -5.60 eV/atom , confirming the energetic stability of this 2D material with respect to its isolated atomic constituents.

To evaluate the thermal stability of monolayer MoO_3 , we performed *ab initio* molecular dynamics (AIMD) simulations at 300 K using a $3 \times 3 \times 1$ supercell over a timescale of 10 ps . The total energy profile, shown in Fig. 2a, exhibits only moderate fluctuations around a well-defined average value, with no signs of structural degradation or bond breaking throughout the simulation. This energy behavior indicates thermodynamic equilibrium and good structural resilience at room temperature.

Fig. 2b and c compare the atomic configurations before and after the simulation, respectively. The final structure maintains the overall crystallographic arrangement, confirming the structural integrity of the

monolayer under thermal perturbation. However, small distortions are observed, particularly among the oxygen atoms, which exhibit slight displacements due to thermal motion. These results collectively demonstrate that monolayer MoO_3 possesses excellent thermal stability, with only minor structural distortions upon thermal excitation.

The phonon dispersion was calculated to analyze the dynamical stability of the novel structure, as represented in Fig. 3. One negative phonon mode with a minimum frequency of -0.27 THz (-9.00 cm^{-1}) at $\Gamma \rightarrow X$ direction. Such small imaginary frequencies could be an artifact of limited supercell size considered ($3 \times 3 \times 1$ supercell), k -points, or reflect the lattice instability over large wave undulations. Wang et al. [59] define -2 THz in the phonon vibration frequency as the limit for free-standing monolayers.

3.2. Mechanical Properties

The mechanical properties of molybditene were systematically investigated through first-principles calculations, and the elastic constants were obtained as $C_{11} = 142.99 \text{ N/m}$, $C_{12} = 19.81 \text{ N/m}$, $C_{22} = 87.08 \text{ N/m}$, and $C_{66} = 41.05 \text{ N/m}$. These values fulfill the Born–Huang mechanical stability criteria for two-dimensional materials with rectangular symmetry, specifically the conditions $C_{11} > 0$, $C_{66} > 0$, and $C_{11}C_{22} > C_{12}^2$, which confirms that molybditene is mechanically stable under small elastic deformations. To further analyze the anisotropic mechanical response of this monolayer, we evaluated the direction-dependent Young's modulus (Y), shear modulus (G), and Poisson's ratio (ν), whose polar plots are presented in Fig. 4. The calculated η_Y , η_G , and η_ν were 1.64, 1.09, and 1.64 respectively, indicating a moderate degree of mechanical anisotropy. Compared to other well-known 2D materials, monolayer MoO_3 exhibits moderate in-plane stiffness. Its maximum Young's modulus of 138.48 N/m is higher than that of MoS_2 [60] (127.7 N/m), but significantly lower than that of graphene [61] (348 N/m) and MXene $\text{Ti}_3\text{C}_2\text{O}_2$ [62], which shows values as high as $247\text{--}319 \text{ N/m}$ depending on the surface termination. In terms of shear modulus, MoO_3 reaches a maximum value of 44.71 N/m , which is lower than those of graphene (150 N/m) and MXene $\text{Ti}_3\text{C}_2\text{O}_2$ ($105\text{--}133 \text{ N/m}$), but still in the range of MoS_2 (52 N/m). The Poisson's ratio of MoO_3 , ranging from 0.14 to 0.23 , aligns with values reported for graphene (0.17) and MoS_2 (0.22). These results position MoO_3 as mechanically competitive, particularly when compared to MoS_2 , while offering anisotropic behavior potentially useful for strain-engineered applications.

3.3. Electronic Analysis

The electronic band structure and projected density of states (PDOS) are presented in Fig. 5 and Fig. 6, respectively. The results reveal that molybditene exhibits an indirect band gap, with the valence band maximum (VBM) located at the S point and the conduction band minimum (CBM) at the Γ point in the Brillouin zone. The band gap energy was determined to be 1.75 eV within the generalized gradient approximation (PBE functional) and increases to 2.95 eV when using the more accurate hybrid functional (HSE06), highlighting the importance of exchange-correlation treatment for predicting electronic behavior.

The valence band edge near the S point shows noticeable curvature and dispersion of approximately 1 eV , suggesting significant band delocalization. Similarly, the conduction band near the Γ point displays a nearly parabolic dispersion, characteristic of low effective masses. These features point to lightweight charge carriers at both band edges, which are highly desirable for optoelectronic applications due to enhanced carrier mobility and efficient optical absorption.

From the projected density of states (PDOS), it is possible to analyze the composition of the electronic states near the Fermi level and gain insights into the bonding nature in molybditene. The valence band is mainly composed of oxygen p -orbitals (O_p), with significant contributions from the p_x , p_y , and p_z orbitals at the VBM. This indicates that the

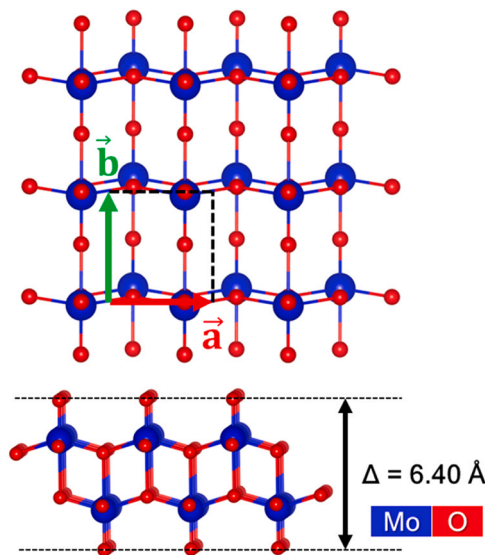


Fig. 1. Top and side views of a representative unit cell of molybditene, with the lattice vectors indicated. The structure exhibits a thickness of approximately $6.40 \text{ \AA}$.

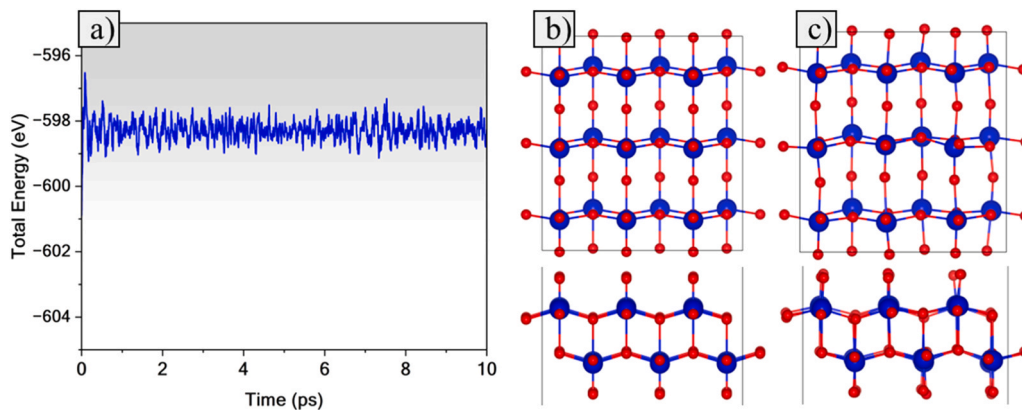


Fig. 2. Thermal stability analysis of monolayer MoO₃ via *ab initio* molecular dynamics (AIMD) at 300 K. (a) Total energy fluctuations over a 10 ps simulation using a $3 \times 3 \times 1$ supercell. (b) Top and side views of the initial atomic structure. (c) Top and side views of the final configuration after the AIMD run, revealing minor distortions and local rippling due to thermal motion.

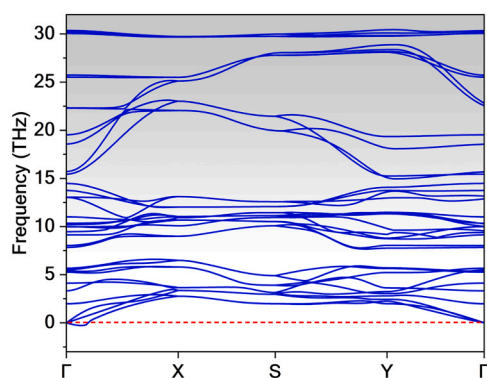


Fig. 3. Phonon dispersion along the high-symmetry pathway in the Brillouin zone obtained for MoO₃ monolayer.

highest occupied states are mainly derived from O_p orbitals, with only minor hybridization from Mo_d orbitals. In contrast, the CBM is dominated by molybdenum *d*-orbitals, especially the *d_{xy}* component, with an additional notable contribution from the oxygen *p_x* orbital. This composition suggests that the states at the CBM arise primarily from the hybridization of Mo and O orbitals in the Mo–O bonds. Furthermore, the band structure reveals degeneracies at the VBM and CBM along certain high-symmetry directions; however, these degeneracies are lifted in other regions of the Brillouin zone. This partial symmetry breaking is attributed to distortions in the [MoO₆] octahedra, which locally perturb the crystal field and lift the degeneracy of the electronic levels.

The effective masses of electrons and holes at the CBM and VBM were calculated along the main symmetry directions of the Brillouin zone

using the PBE functional. These values were obtained by fitting a second-order polynomial to the band edges, ensuring reliable results. The data are summarized in the Table 1.

As shown, the effective masses are highly anisotropic and depend strongly on the direction. Along the $\Gamma \rightarrow X$ direction, the electron effective mass is relatively low (0.70 *m₀*), indicating favorable transport for electrons, while the hole effective mass is much larger in magnitude (3.00 *m₀*), suggesting lower mobility for holes in this direction. Along $X \rightarrow S$, electrons are heavier (3.37 *m₀*), but holes become considerably lighter (0.59 *m₀*). The largest electron mass appears along $S \rightarrow Y$ (10.04 *m₀*), which points to sluggish electronic transport in this direction, while the hole mass here is 1.71 *m₀*. Along $Y \rightarrow \Gamma$, the electrons possess the lowest effective mass (0.66 *m₀*), whereas the holes exhibit the highest effective mass (5.24 *m₀*).

These results indicate that the material exhibits strong directional (anisotropic) transport properties. The relatively small electron effective masses along $\Gamma \rightarrow X$ and $Y \rightarrow \Gamma$ directions could be advantageous for applications that require efficient electron conduction, while the heavier masses in other directions may lead to interesting anisotropic carrier dynamics.

3.4. Optical Properties

The optical properties of molybditene were evaluated through its absorption and reflection spectrum, as illustrated in Fig. 7, revealing a pronounced anisotropic behavior characteristic of its low symmetry 2D structure. In the infrared region, the material exhibits negligible optical activity for both polarization directions, indicating transparency in this range. However, distinct differences emerge between the *xx*- and *yy*-polarized components as the photon energy increases. For the *x*-component, corresponding to the in-plane *xx* direction, the first

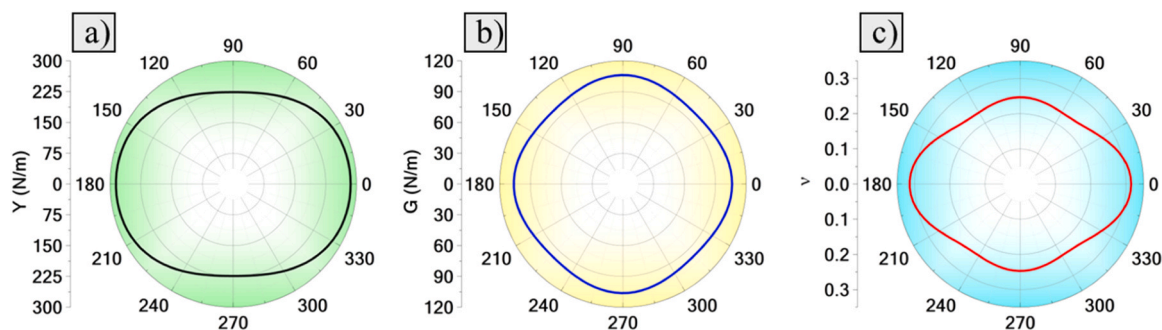


Fig. 4. Directional dependence of the mechanical properties of molybditene. Panels (a)–(c) present (a) Young's modulus (*Y*), (b) shear modulus (*G*), and (c) Poisson's ratio (*ν*), respectively.

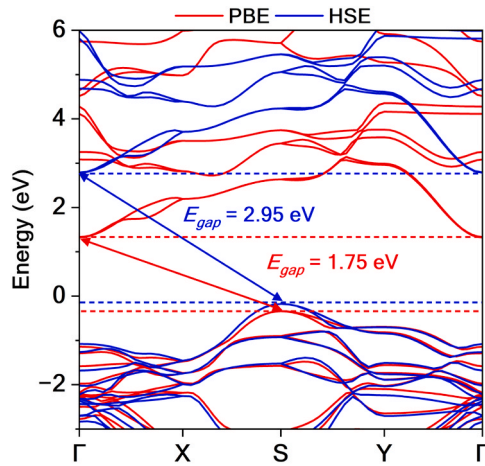


Fig. 5. Electronic band structure of molybditene calculated using the PBE (red) and HSE06 (blue) functionals. The band gap energy (E_{gap}) obtained from each method is indicated.

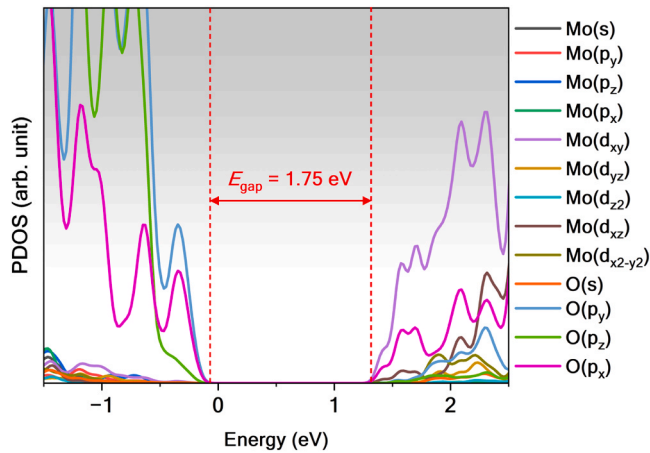


Fig. 6. Projected density of states (PDOS) of molybditene determined at PBE/GGA level. Color-coded curves distinguish the orbital contributions.

Table 1

Effective masses of electrons and holes (in units of the free electron mass, m_0) calculated along high-symmetry directions in the Brillouin zone for unstrained molybditene using the PBE functional.

Path	Electron (m^*/m_0)	Hole (m^*/m_0)
$\Gamma \rightarrow X$	0.70	3.00
$X \rightarrow S$	3.37	0.59
$S \rightarrow Y$	10.04	1.71
$Y \rightarrow \Gamma$	0.66	5.24

absorption and reflection peaks appear in the green-to-blue region of the visible spectrum (approximately 2.4–2.6 eV). The strongest absorption peak is centered around 3.5 eV.

In contrast, the y-polarized response (for yy direction) shows markedly different behavior. The first absorption peak for this component occurs in the ultraviolet region, around 3.2 eV, suggesting that molybditene presents a lower optical activity for this polarization within the visible range. The main absorption peak for the yy component appears at approximately 4.0 eV, significantly higher in energy than that of the xx component. This strong directional dependence of the optical response highlights the intrinsic optical anisotropy of molybditene, which stems from its reduced lattice symmetry and distinct electronic transitions along different crystallographic directions. Such anisotropy is closely

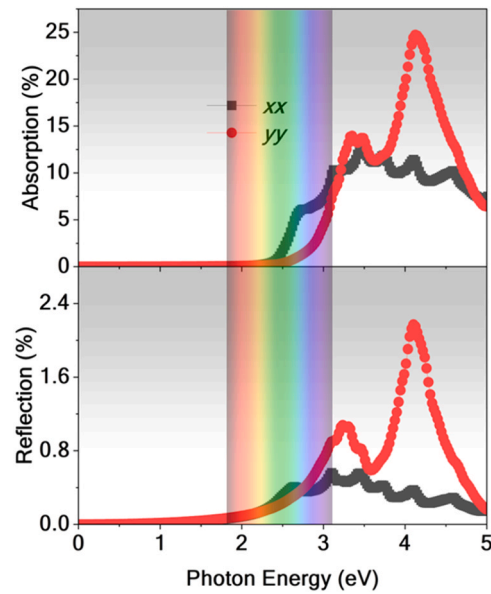


Fig. 7. Optical properties of monolayer molybditene. Panels (a) and (b) show the absorption and reflectivity spectra, respectively, for the in-plane xx (black) and yy (red) components.

related to directional variations in the optical dipole matrix elements, which govern the strength and polarization dependence of optical transitions [63].

Fig. 8 displays the real and imaginary parts of the frequency-dependent dielectric function for monolayer MoO₃, considering incident light polarized along the x (black curve) and y (red curve) crystallographic directions. The shaded region corresponds to the visible spectrum, which facilitates the interpretation of optical activity relevant to technological applications.

The real part of the dielectric function, shown in panel (a), describes the dispersive response of the material. For both polarization directions, $\text{Re}[\epsilon(\omega)]$ increases significantly near the onset of absorption (approximately 2.2–2.5 eV), reaching prominent peaks between 2.5 and 3.2 eV. The static dielectric constant (as ω approaches zero) is moderate (closer to 4) and demonstrates optical anisotropy, with the yy component slight higher than the xx component. At higher photon energies (above 4 eV), the real part of the dielectric function becomes negative.

The imaginary part of the dielectric function, presented in panel (b), is directly associated with optical absorption due to interband electronic transitions. $\text{Im}[\epsilon(\omega)]$ is nearly zero below the band gap for both polarization directions, confirming the absence of optical transitions in this range. The sharp absorption onset occurs around 2.4–2.6 eV, consistent with the calculated absorption, followed by prominent peaks in the visible and ultraviolet range. The most intense absorption peak for the yy polarization appears close to 4.0–4.2 eV, while for the xx direction it is lower, around 3.5 eV. These peaks are signatures of strong interband transitions and underline the highly anisotropic optical response of the material.

The differences in intensity and position of the main optical features between the xx and yy polarizations underscore the intrinsic optical anisotropy of MoO₃, which stems from its lattice structure and the directional character of the dipole-allowed electronic transitions.

3.5. Strain Response

Strain engineering was used to investigate the structural and electronic response of molybditene under mechanical deformations along the x and y directions, with uniaxial deformations ranging from −3 % to +3 %. Key structural descriptors such as the average Mo–O bond length (l_{av}), the volume of the [MoO₆] octahedra (V), the distortion index (D),

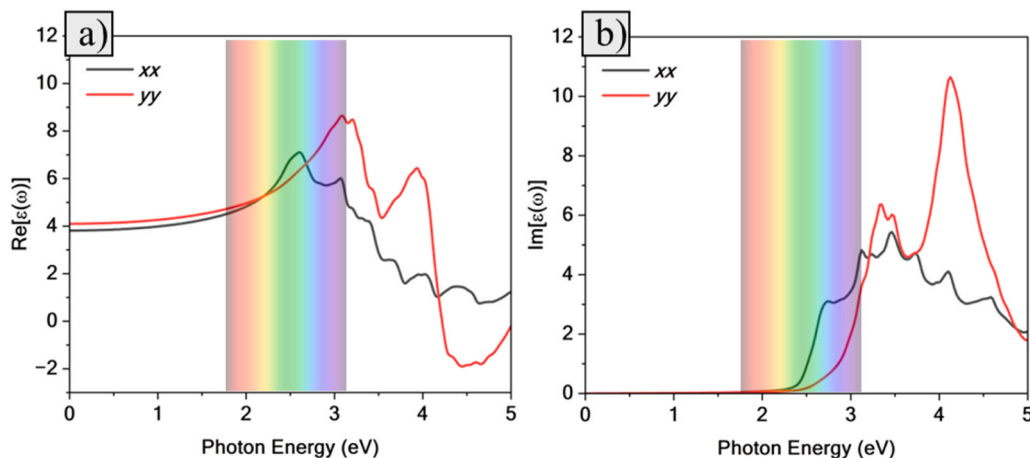


Fig. 8. Real (a) and imaginary (b) parts of the frequency-dependent dielectric function for monolayer MoO₃ under in-plane polarized light, computed using the independent particle approximation (IPA). The black and red curves represent the xx and yy polarization directions, respectively.

and the effective coordination number (*ECoN*) along with the band gap energy (E_{gap}), were evaluated under these mechanical deformations. The behavior of these parameters is summarized in Fig. 9.

Anisotropic behavior is clearly observed in the evolution of l_{av} , D , and E_{gap} . The l_{av} increases with tensile strain along the y-direction (ϵ_y). Conversely, for strain along the x-direction (ϵ_x), a slight reduction in l_{av} is observed, indicating a subtle bond compression. Despite these directional differences, the total variation of l_{av} remains within a narrow window (~ 0.015 Å). The distortion index D , which quantifies the deviation of the bond lengths on [MoO₆] octahedra from l_{av} , exhibits a similar trend. Notably, deformation along the x-axis leads to a decrease in D as the ϵ_x increases, while for the y-axis, notice that D increases with ϵ_y .

Interestingly, the E_{gap} shows a remarkable anisotropic response to strain. Along the x-direction, E_{gap} shows a moderate variation, fluctuating slightly between 1.63 and 1.83 eV. However, under y-axis strain, E_{gap} exhibits a much more significant variation, ranging from 1.10 eV to 2.30 eV approximately. This implies that uniaxial strain along the y-direction can be effectively used to tune the optical and electronic properties of molybdenite, potentially enabling its use in flexible

electronics, tunable optoelectronic devices, or strain sensors.

In contrast, the V and $ECoN$, which represent the spatial occupation and local bonding environment around Mo atoms, exhibit more isotropic behavior. For both ϵ_x and ϵ_y , $ECoN$ decreases with increasing strain, indicating that tensile deformation leads to weaker coordination and less compact local environments. Specifically, $ECoN$ ranges from 3.57 to 3.23 along the x-direction and from 3.72 to 3.25 along the y-direction, highlighting a progressive loss of octahedral coordination. Similarly, the volume V of the [MoO₆] units increases slightly under strain, with values varying from 10.08 Å³ to 10.23 Å³ (for ϵ_x) and from 10.06 Å³ to 10.24 Å³ (for ϵ_y).

Fig. 10 represents the band structures for molybdenite under several strains at x- and y- directions. As previously discussed, deformations along the x-direction do not significantly alter the electronic structure of molybdenite. The band dispersion remains nearly unchanged, and the overall topology of the valence and conduction bands is preserved, indicating a relatively rigid electronic response to strain applied in this direction. In contrast, for ϵ_y , a pronounced impact is noticed on the electronic band structure. With increasing tensile strain, the VBM becomes progressively flatter, especially along the S $\rightarrow$ Y path. This

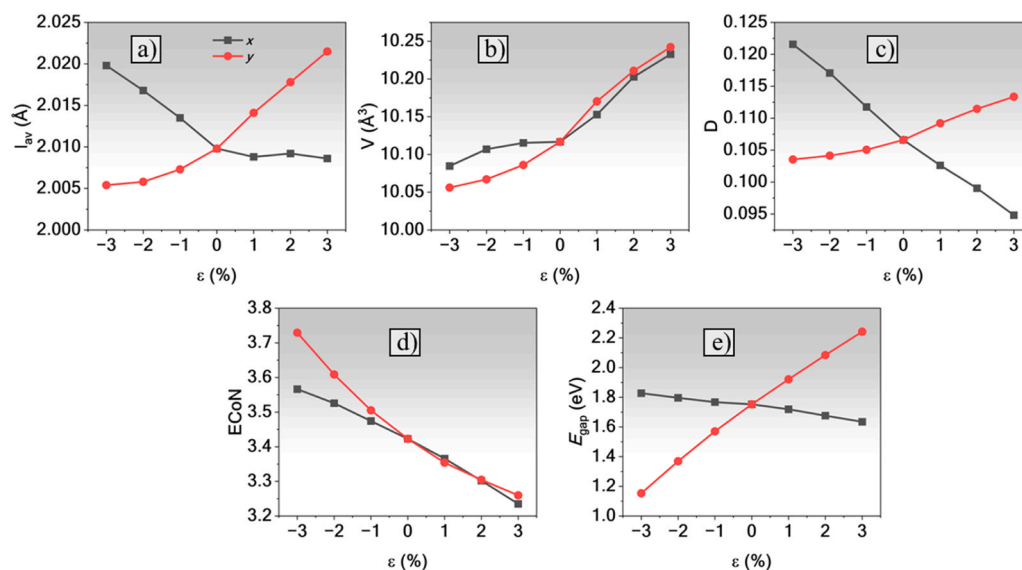


Fig. 9. Examination of structural and electronic parameters of molybdenite under uniaxial strain. Panels (a)–(e) present the evolution of (a) average Mo–O bond length (l_{av}), (b) [MoO₆] octahedral volume (V), (c) distortion index (D), (d) effective coordination number ($ECoN$), and (e) band gap energy (E_{gap}) as functions of the applied strain. The results are shown for each strain condition considered along both the x- and y-directions.

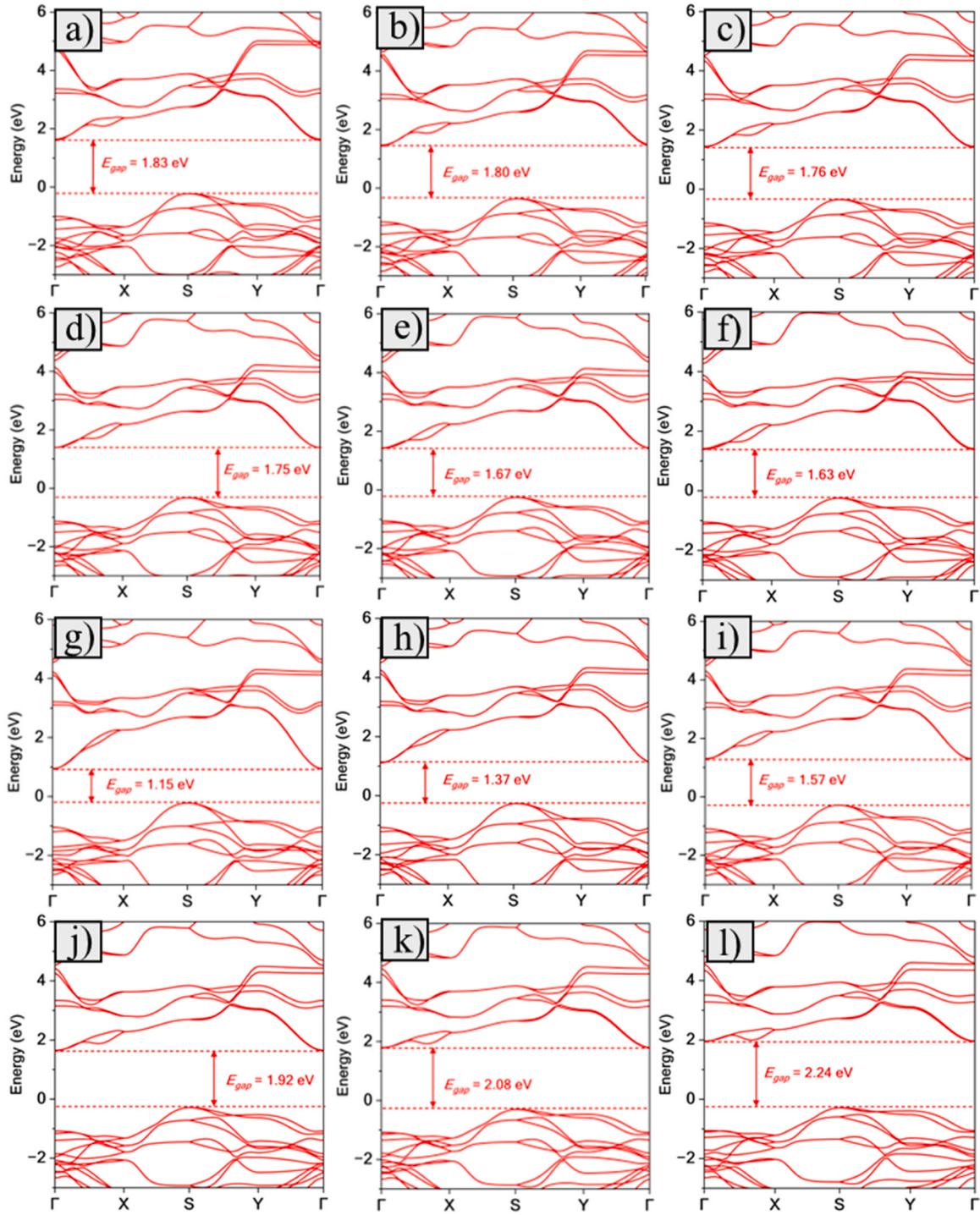


Fig. 10. Electronic band structures of molybditene under uniaxial strain. Panels (a)–(f) show the band structures for strain applied along the x-direction, with $\epsilon_x = -3\%$, -2% , -1% , $+1\%$, $+2\%$, and $+3\%$, respectively. Panels (g)–(l) correspond to strain along the y-direction, with $\epsilon_y = -3\%$, -2% , -1% , $+1\%$, $+2\%$, and $+3\%$. The band gap energy values (E_{gap}) are highlighted in each panel.

flattening is indicative of localized electronic states.

Under compressive strain along the y-axis, a different behavior is observed. The bands near the VBM and CBM become more dispersive, particularly those defining the CBM, which show a dispersion exceeding 2 eV. This suggests that compressive ϵ_y enhances band curvature, potentially improving electronic transport properties due to the lower effective mass of the charge carriers.

The analysis of DOS of molybditene under various uniaxial strain conditions (Fig. 11) reveals notable modifications in the electronic

structure. As expected, the valence band is predominantly derived from O states, while the conduction band is mainly composed of Mo states. Under strain along the y-direction (ϵ_y), a clear reduction in the DOS at the CBM is observed, suggesting an increase in band dispersion and a delocalization of electronic states. This trend is consistent with previous band structure analyses, where the CBM becomes more dispersive with tensile strain along y. This mechanical perturbation introduces a short range disorder in the local environment modify the overlap between Mo_d and O_p orbitals that dominate the valence and conduction band

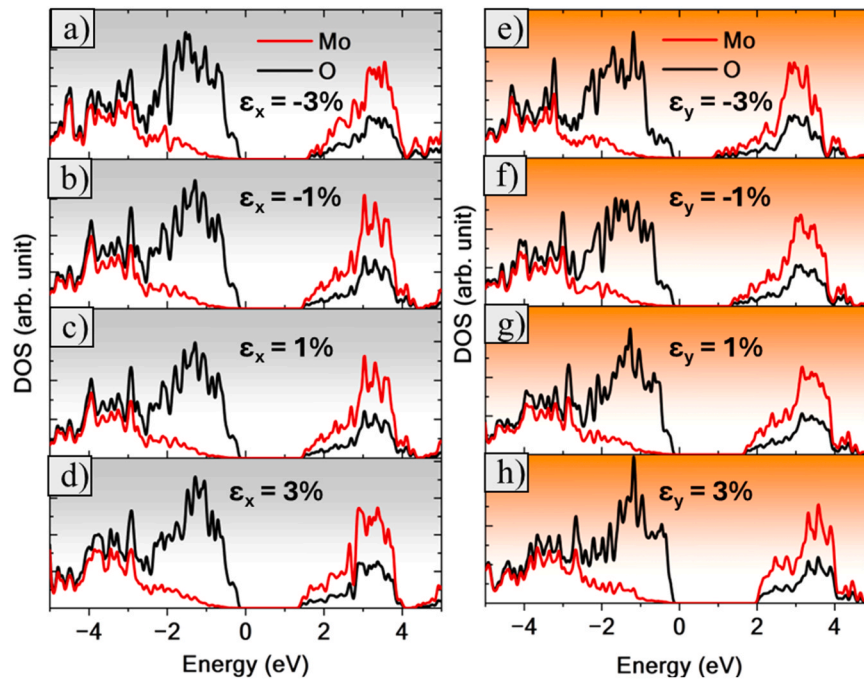


Fig. 11. Density of states (DOS) of molybdenite under uniaxial strain. Panels (a)–(d) show the DOS for strain applied along the x-direction, with $\epsilon_x = -3\%$, -1% , $+1\%$, and $+3\%$, respectively. Panels (e)–(h) correspond to strain along the y-direction, with $\epsilon_y = -3\%$, -1% , $+1\%$, and $+3\%$. The Fermi level was shifted for 0 eV.

edges. As strain alters the bond angles and lengths, it tunes the degree of orbital overlap, leading to anisotropic shifts in the band gap. These changes in the electronic structure are directly correlated with the observed variations in optical absorption, as they modify the interband transition probabilities, offering a viable route for band structure engineering in optoelectronic applications.

Fig. 12 presents the optical absorption spectra of monolayer MoO₃ under uniaxial strain along the x and y directions. The intrinsic anisotropy observed in the pristine form is preserved under all strain conditions, with the absorption along the yy polarization (red curves) consistently stronger than along xx (black curves), indicating direction-dependent optical behavior.

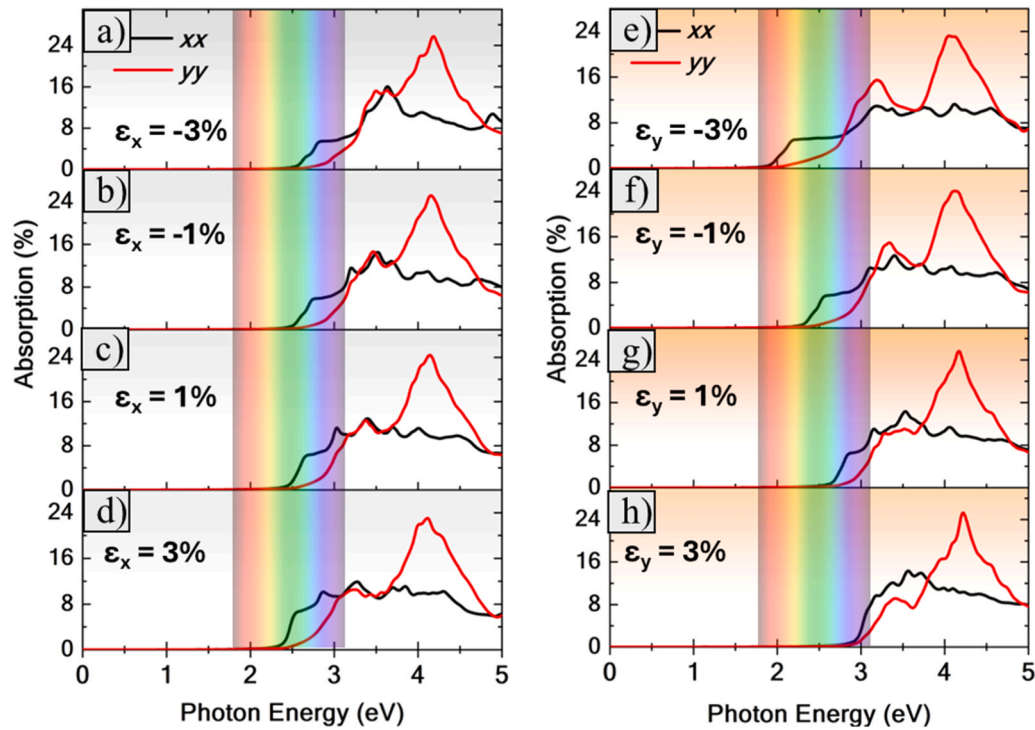


Fig. 12. Optical absorption of molybdenite under uniaxial strain. Panels (a)–(d) show the DOS for strain applied along the x-direction, with $\epsilon_x = -3\%$, -1% , $+1\%$, and $+3\%$, respectively. Panels (e)–(h) correspond to strain along the y-direction, with $\epsilon_y = -3\%$, -1% , $+1\%$, and $+3\%$. The red and black line indicate the absorption for xx and yy components, respectively.

For strain applied along the x-axis (panels a–d), the absorption onset remains relatively stable, varying slightly from 2.4 to 2.5 eV as the strain increases. The position of the first absorption peak in the yy component exhibits a mild redshift, shifting from 3.4 eV at compressive strain ($\varepsilon_x = -3\%$) to 3.2 eV at tensile strain ($\varepsilon_x = 3\%$). The main absorption peak in the yy channel appears at approximately 4.1 eV and remains essentially unaffected by the applied strain. For the xx component, the most intense peak also undergoes a redshift, varying from 3.4 eV ($\varepsilon_x = -3\%$) to 3.2 eV ($\varepsilon_x = 3\%$), occurring nearly at the same energy as the first absorption peak observed in the yy component. These results show that the mechanical perturbation along the x-direction only introduces moderate variations in the absorption spectra of molybditene.

In contrast, the application of strain along the y-axis (panels e–h) results in a marked redshift in the absorption onset for both polarization directions. Specifically, for $\varepsilon_y = +3\%$ (panel h), the onset occurs around 3.0 eV, while for $\varepsilon_y = -3\%$ (panel e), it drops to approximately 1.9 eV. This significant shift indicates that compressive strain along the y-direction effectively narrows the optical gap, enhancing light absorption in the lower energy region. The first absorption peak for the xx component follows this redshift trend, varying from 3.2 eV ($\varepsilon_y = +3\%$) to 2.2 eV ($\varepsilon_y = -3\%$). In contrast, for the yy component, the first absorption peak remains relatively stable, shifting only from 3.4 eV to 3.1 eV across the same strain range. A similar strain-induced modulation of the optical response has been reported for other 2D materials. For example, in BAs monolayer [64], application of tensile strain leads to a blueshift in the first optical peak and a redshift in the second peak of the imaginary part of the dielectric function ($\text{Im}[\varepsilon(\omega)]$), with the magnitude of the shift being more pronounced for the second peak. It can be considered also the results for GeSe monolayers [65], both puckered and buckled phases. When strain is applied along the x direction, the intensity of the main absorption peak decreases and shifts toward higher energies. In the buckled structure, the first absorption peak disappears under compressive strain and moves to higher energies as tensile strain is applied. In contrast, for the puckered structure, increasing tensile strain results in a shift of the absorption edge to lower energies.

In particular, such strain-engineered control can be exploited in next-generation smart windows, strain-sensitive photodetectors [66,67], or flexible solar cells [68,69], where the optical absorption range can be

dynamically adjusted in response to mechanical deformation. Moreover, the reversible and low-energy nature of the strain-induced tuning makes molybditene a suitable candidate for integration into wearable electronics and sensors that operate under variable mechanical stress conditions.

3.6. Vibrational Features

Infrared and Raman spectroscopy are powerful, complementary techniques for probing the vibrational dynamics of materials. The vibrational properties of molybditene at the Γ point were analyzed using the coupled-perturbed Hartree-Fock/Kohn-Sham (CPHF/KS) formalism, with the resulting infrared (IR) and Raman spectra presented in Fig. 13. The irreducible representation of the vibrational modes is given by $\Gamma_{\text{vib}} = 4A_u + 7B_u + 8A_g + 4B_g$. Among these modes, three correspond to acoustic vibrations ($A_u + 2B_u$), while 11 are IR-active ($4A_u + 7B_u$) and 12 are Raman-active ($8A_g + 4B_g$). The most prominent IR peaks appear at 240.90 cm^{-1} (A_u), 537.08 cm^{-1} (A_u), and 713.06 cm^{-1} (B_u). In the Raman spectrum, the most intense peaks are observed at 708.39 cm^{-1} (A_g), with additional peaks at 221.97 cm^{-1} (A_g) and 1014.01 cm^{-1} (A_g).

4. Conclusions

In this work, we conducted a thorough first-principles investigation of the molybditene monolayer, a novel two-dimensional phase of MoO_3 characterized by monoclinic symmetry (space group $P2_1/m$). Thermal stability was confirmed through ab initio molecular dynamics simulations at 300 K, which revealed no structural breakdown or phase transitions over the simulation time. Structurally, molybditene maintains its layered integrity with only minor distortions in the $[\text{MoO}_6]$ octahedral units.

From a mechanical perspective, the elastic constants satisfy the Born-Huang criteria, and the direction-dependent analysis revealed moderate anisotropy in Young's modulus, shear modulus, and Poisson's ratio, pointing to directional mechanical response, which is advantageous for strain engineering.

The electronic band structure shows an indirect band gap with the

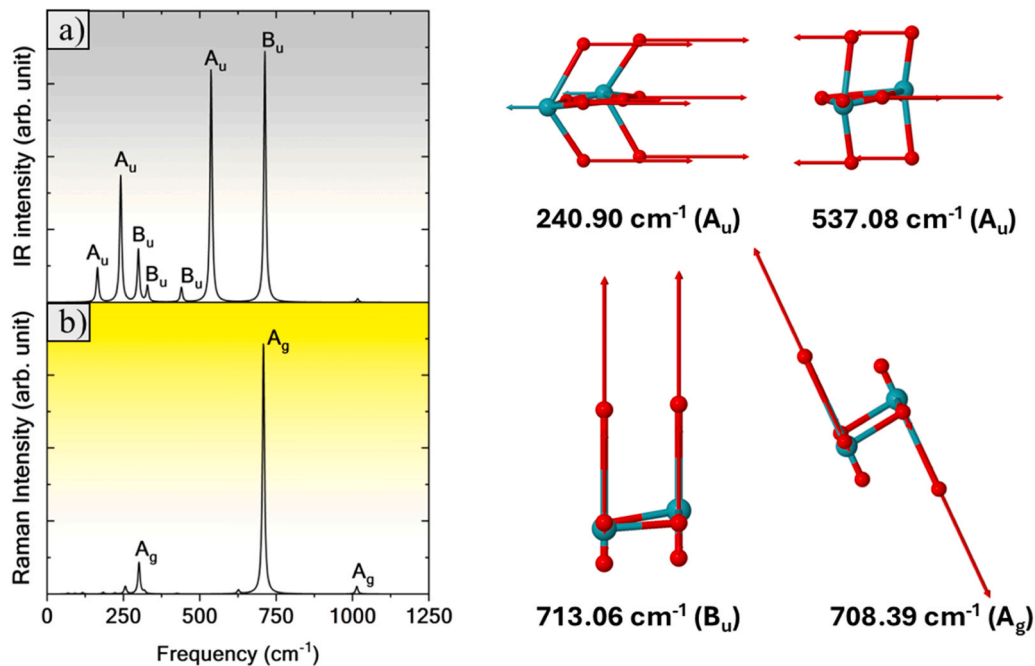


Fig. 13. Vibrational properties of molybditene. The upper panel presents the (a) infrared (IR) spectrum, while the (b) lower panel shows the corresponding Raman spectrum. The vibrations associated with the most intense peaks are showed at right.

valence band maximum located at the S point and the conduction band minimum at the Γ point. The calculated band gap varies from 1.75 eV (PBE) to 2.95 eV (HSE06), demonstrating a strong dependency on the exchange-correlation functional used.

Optical calculations reveal directionally dependent absorption and reflection spectra, with strong anisotropy between the x- and y-polarized components. The in-plane xx polarization shows strong absorption in the visible range (notably around 3.5 eV), while the yy component shows activity primarily in the ultraviolet region, reinforcing the material's potential as a polarization-sensitive optical material.

Strain engineering analyses reveal that structural and electronic properties respond distinctly to both tensile and compressive strains. While the average Mo–O bond lengths and the distortion index vary with direction and strain type, the band gap response is markedly anisotropic: moderate under ϵ_x , but highly tunable under ϵ_y , ranging from 1.10 eV to 2.30 eV. In addition, mechanical strain significantly modifies the optical properties, leading to increased anisotropy and a redshift in the absorption onset. These findings highlight the promise of molybdenite for flexible electronics, strain-sensitive sensors, and polarization-selective optoelectronic applications. This tunability could be harnessed to design devices with strain-modulated optical or electronic responses, such as piezoresistive sensors or flexible photodetectors.

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Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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