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**Effect of thermal treatment on ZnO-derived zinc-metal-organic framework
for detecting volatile organic compounds**

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RESUMO

Este trabalho apresenta a aplicação de ZnO, um óxido metálico semicondutor do tipo n, derivado de estrutura metal-orgânica (MOF), para detecção de compostos orgânicos voláteis microbianos (MVOCs). A detecção rápida, seletiva e em baixos limites de detecção é um dos principais desafios relatados. Além disso, estudos sobre a influência dos tamanhos de cristalitos do óxido metálico no desempenho do sensor ainda são escassos. A proposta e foco desta pesquisa consistem na aplicação de ZnO com diferentes tamanhos de cristalitos no desempenho como sensor de MVOCs. O precursor do ZnO (MOF-5) foi preparado utilizando o método solvotérmico assistido por micro-ondas, empregando o sal metálico $Zn(NO_3)_2 \cdot 6H_2O$ e ligante orgânico ácido tereftálico em dimetilformamida (DMF). O ZnO, aplicado como sensor, foi sintetizado pela degradação térmica (400 °C a 800 °C) da MOF-5, resultando em ZnO com tamanhos variados de nanopartículas: 17,8 nm a 450 °C (ZnO-450), 26,8 nm a 600 °C (ZnO-600) e 34,3 nm a 800 °C (ZnO-800). As caracterizações eletrônicas e estruturais evidenciam a formação de ZnO puros e fases cristalinas desejadas. Os sensores foram testados para 100 ppm dos MVOCs: acetona, 2-butanona, m-xileno, tolueno, benzeno e 1-Pentanol, em temperaturas variadas de 200 °C a 450 °C. Todos os sensores apresentaram boa detecção dos MVOCs 1-pentanol e 2-butanona a 400 °C, com destaque para o sensor ZnO-450, que exibiu a maior resposta ao MVOC 1-pentanol, atingindo um valor de resposta de 2680,0. Entretanto, observou-se uma alta sensibilidade e seletividade no sensor ZnO-600, com uma resposta de 2205,0 para 1-pentanol na mesma temperatura de operação, sendo que sua seletividade entre os MVOCs 1-pentanol e 2-butanona foi maior quando comparado ao sensor com tamanho de cristalito de 17,8 nm. Os mesmos parâmetros de seletividade do sensor ZnO-600 foram observados para o sensor ZnO-800, com tamanho de cristalito de 34,3 nm, embora os valores de resposta para os MVOCs 1-pentanol e 2-butanona tenham sido inferiores, 232,0 e 100,01, respectivamente. Portanto, demonstramos que o tamanho das nanopartículas de ZnO influencia diretamente a resposta e o desempenho do sensor, sendo mais seletivo para 1-pentanol a 400 °C o sensor de ZnO calcinado a 600 °C, com tamanho de cristalito de 26,8 nm. Destaca-se também o bom desempenho desse sensor na presença de uma atmosfera úmida e em diferentes faixas de umidade relativa controlada.

Palavras-chave: Semicondutores tipo n. MOF-5. Óxido de zinco. Sensor de gás. 1-Pentanol.

ABSTRACT

This work presents the application of ZnO, an n-type semiconductor metal oxide derived from a metal-organic framework (MOF), for the detection of microbial volatile organic compounds (MVOCs). Rapid, selective detection with low limits of detection is one of the main reported challenges. Furthermore, studies on the influence of metal oxide crystallite sizes on sensor performance are still scarce. The proposal and focus of this research are the application of ZnO with different crystallite sizes to evaluate its performance as an MVOC sensor. The ZnO precursor (MOF-5) was prepared using a microwave-assisted solvothermal method, employing the metal salt $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ and the organic ligand terephthalic acid in dimethylformamide (DMF). ZnO, applied as a sensor, was synthesized by thermal degradation (400 °C to 800 °C) of MOF-5, resulting in ZnO with varied nanoparticle sizes: 17.8 nm at 450 °C (ZnO-450), 26.8 nm at 600 °C (ZnO-600), and 34.3 nm at 800 °C (ZnO-800). Electronic characterizations and structural analyses confirm the formation of pure ZnO and the desired crystalline phases. The sensors were tested for 100 ppm of the MVOCs: acetone, 2-butanone, m-xylene, toluene, benzene, and 1-pentanol, at varying temperatures from 200 °C to 450 °C. All sensors showed good detection of the MVOCs 1-pentanol and 2-butanone at 400 °C, with particular emphasis on the ZnO-450 sensor, which exhibited the highest response to the MVOC 1-pentanol, reaching a response value of 2680.0. However, high sensitivity and selectivity were observed in the ZnO-600 sensor, with a response of 2205.0 for 1-pentanol at the same operating temperature, and its selectivity between the MVOCs 1-pentanol and 2-butanone was higher compared to the sensor with a crystallite size of 17.8 nm. The same selectivity parameters of the ZnO-600 sensor were observed for the ZnO-800 sensor, with a crystallite size of 34.3 nm, although the response values for the MVOCs 1-pentanol and 2-butanone were lower, at 232.0 and 100.01, respectively. Therefore, we demonstrate that the size of the ZnO nanoparticles directly influences the response and performance of the sensor, with the ZnO sensor calcined at 600 °C and with a crystallite size of 26.8 nm being more selective for 1-pentanol at 400 °C. The good performance of this sensor in the presence of a humid atmosphere and across different ranges of controlled relative humidity is also noteworthy.

Keywords: n-type semiconductors. MOF-5. Zinc oxide. Gas sensor. 1-Pentanol.

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ABBREVIATIONS AND ACRONYMS LIST

BET	Brunauer-Emmett-Teller
EDX	Energy-dispersive X-ray
FESEM	Field emission scanning electron microscopy
FTIR	Fourier transform infrared spectroscopy
HRTEM	High-resolution transmission electron microscopy
MOF	Metal-organic framework
MVOC	Microbial volatile organic compound
RH	Relative humidity
SMO	Semiconducting metal oxide
TEM	Transmission electron microscope
TEY	Total electron-yield
TFY	Total fluorescence-yield
TGA	Thermogravimetric analyses
XAS	X-ray absorption spectroscopy
XPS	X-ray photoelectron spectroscopy
XRD	X-ray Diffraction

SYMBOLS LIST

Å	Angstrom
°C	Celsius degree
cm	Centimeter
nm	Nanometer
cm³/g	Cubic centimetre per gram
eV	Electron volt
g	Gram
h	Hour
keV	Kilo électron volt
kV	Kilovolt
mA	Milliampère
mg	Milligram
mL	Milliliter
mmol	Millimol
min	Minute
ppm	Parts per million
s	Second
m² g⁻¹	Square meter per gram

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1 INTRODUCTION AND BIBLIOGRAPHIC REVIEW

The development of conductometric sensors based on semiconductor metal oxides (SMOs) has gained prominence in the detection of microbial volatile organic compounds (MVOCs) (BARSAN; KOZIEJ; WEIMAR, 2007; BARSAN; WEIMAR, 2020) compared to other detection methods, such as optical and mechanical ones (SOHRABI et al., 2023), due to their great multifunctionality, morphological diversification, physical, chemical and thermal stability, in addition to low costs (CHATTERJEE; MITRA; MUKHOPADHYAY, 1999; WANG et al., 2020; WAN et al., 2004).

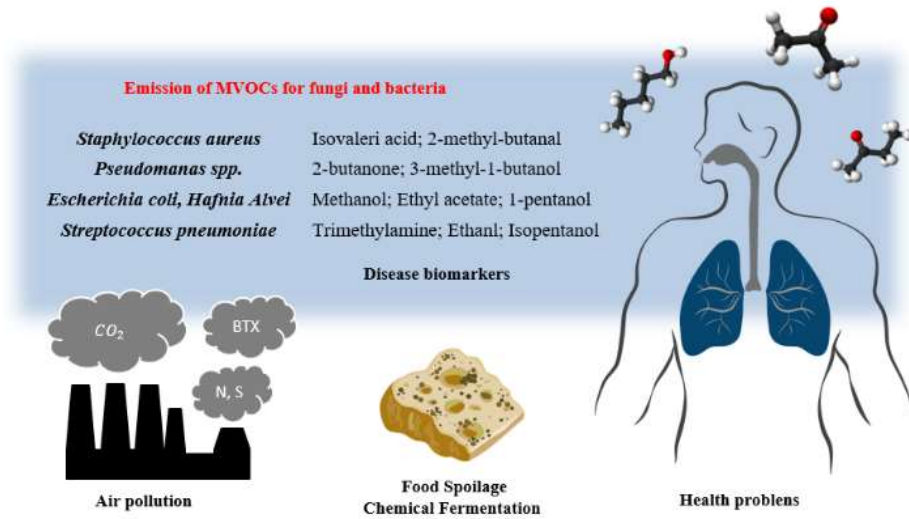
The importance of detecting different chemical classes of these compounds, produced by microorganisms, arises from several factors that directly impact society (ANAND; PHILIP; MEHENDALE, 2014; KAMPA; CASTANAS, 2008; KORPI; JÄRNBERG; PASANEN, 2009; PASANEN et al., 1998; SCHULZ; DICKSCHAT, 2007). Among these factors are food spoilage (WANG et al., 2016) and the danger to human health caused by pathogenic microorganisms (PASQUARELLA; PITZURRA; SAVINO, 2000; SCHÜTZE et al., 2017). Furthermore, there are volatile organic compounds exhaled through respiration that are directly associated with diseases such as cancer and can act as biomarkers, aiding in the diagnosis of potentially undetectable diseases (MACHADO et al., 2015; SCHMIDT; PODMORE, 2015; SUN; SHAO; WANG, 2016).

In this context, this research aimed to synthesize and apply ZnO-based SMO sensors for detecting MVOCs. SMOs are materials with unique and highly versatile electrical characteristics, mainly when derived from metal-organic frameworks (MOFs) (WANG et al., 2020), ensuring high surface area, porosity, absorptivity, and crystallinity (KUMAR; MASRAM, 2021). These properties have great potential for application in toxic gas detection (SOHRABI et al., 2023). The sensor properties are directly linked to their morphology, particularly the crystallite size of the material (REN et al., 2022), which directly influences the material's active sites and gas adsorption capacity (XU et al., 2000), thereby altering the sensor performance.

1.1 Microbial volatile organic compounds (MVOCs)

Identifying and controlling the emission of atmospheric pollutants known as volatile organic compounds (VOCs) is of great importance due to their impact on society (HAO et al., 2024). VOCs correspond to organic compounds with boiling points below 250 °C and high vapor pressures, evaporating quickly (ANAND; PHILIP; MEHENDALE, 2014; HE et al., 2019). They are responsible for the reduction of air quality (KAMPA; CASTANAS, 2008), and

Figure 1 – Overview of volatile organic compounds and their interactions.



Source: adaptation from literature (KAMPA; CASTANAS, 2008; SCHÜTZE et al., 2017; SUN; SHAO; WANG, 2016).

Furthermore, some MVOCs are produced by specific microorganisms, which, when exhaled by the respiratory system, can be directly associated with diseases, acting as biomarkers and aiding in the diagnosis of potentially undetectable diseases (DUFFY; MORRIN, 2019; MACHADO et al., 2015; SCHMIDT; PODMORE, 2015). Currently, conventional methods cannot meet monitoring and safety requirements as they require expensive and time-consuming processes to identify microorganisms and MVOCs (ZHU et al., 2019), such as the use of gas chromatography coupled with mass spectrometry, as well as chemiluminescent and fluorescent sensing techniques. Therefore, the application of semiconductor metal oxide sensors can address these issues in a simple and cost-effective way, enabling the rapid, practical, efficient, and selective detection of MVOCs, while also facilitating the integration of information that can assist in disease diagnosis and food safety risk assessment.

1.2 Semiconductor metal oxides based on ZnO for MVOC detection

Semiconductor metal oxides (SMOs) comprise a class of materials with excellent electrical characteristics and are widely used in several areas, such as catalysis and photocatalysis (DJURIŠIĆ; LEUNG; CHING NG, 2014; ZHANG; LIN, 2014), supercapacitors, energy storage, and conversion (WU et al., 2011), and primarily as sensors for detecting toxic gases (KRISHNA et al., 2022; SOHRABI et al., 2023; WANG et al., 2020). SMOs are classified as n-type or p-type according to the type of receptor and conduction mechanisms (KIM; LEE, 2014). N-type and p-type semiconduction arise from extrinsic defects.

3 CONCLUSION AND NEXT CHALLENGES

This research aimed to develop ZnO sensors for the detection of MVOCs under various operating conditions. In line with this objective, Chapter 2 provides a review of key topics, specifically MVOCs and conductometric sensors based on semiconductor metal oxides (SMOs). Detecting MVOCs is crucial due to the environmental and social problems caused by various classes of bacteria and fungi. The rapid and selective identification of these compounds, using low-cost materials, drives the search for new alternatives. The synthesized ZnO proved to be an excellent material for MVOC detection, especially for 1-pentanol, when the crystallite size of the nanoparticles was modified. Chapter 3 describes the fabrication of ZnO through the thermal treatment of a MOF-5 precursor, synthesized via a microwave-assisted solvothermal method. Varying the crystallite size enhanced sensor performance. At the optimal operating temperature, ZnO with intermediate crystallite size demonstrated increased selectivity, improved detection of 1-pentanol, and the shortest response time among all the sensors studied.

Future challenges include better characterization of the synthesized material using Transmission Electron Microscopy (TEM) to obtain more detailed structural information, as well as a more in-depth investigation of the electronic and surface properties of ZnO through X-ray absorption spectroscopy and X-ray photoelectron spectroscopy analyses. These analyses will provide clearer insight into the detection mechanism. Furthermore, this study aims to support future research into the development of simple ZnO sensors derived from MOF-5, while demonstrating the importance and influence of nanoparticle size on sensor performance, whether for 1-pentanol or other MVOCs produced by specific bacteria and fungi. This will contribute to the development and expansion of the MVOC database. Improving selectivity and sensor applications under real-world temperature and humidity conditions remains a significant challenge in this field.

REFERENCES

ANAND, S. S.; PHILIP, B. K.; MEHENDALE, H. M. Volatile Organic Compounds. Em: **Encyclopedia of Toxicology: Third Edition**. [s.l.] Elsevier, 2014. p. 967–970.

ARAÚJO, E. A. et al. Synthesis, growth mechanism, optical properties and catalytic activity of ZnO microcrystals obtained via hydrothermal processing. **RSC Advances**, v. 7, n. 39, p. 24263–24281, 2017.

BARSAN, N.; KOZIEJ, D.; WEIMAR, U. Metal oxide-based gas sensor research: How to? **Sensors and Actuators, B: Chemical**, v. 121, n. 1, p. 18–35, 30 jan. 2007.

BARSAN, N.; WEIMAR, U. **Conduction Model of Metal Oxide Gas Sensors** *Journal of Electroceramics*. [s.l.: s.n.].

BARSAN, N.; WEIMAR, U. **7.3.3 Fundamentals of Metal Oxide Gas Sensors**. AMA Service GmbH, 18 dez. 2020.

CHATTERJEE, A. P.; MITRA, P.; MUKHOPADHYAY, A. K. **P1: FJL/FIX P2: FJS/LMQ/FIA P3: FJM/FGD Chemically deposited zinc oxide thin film gas sensor**. [s.l.: s.n.].

CHEN, T. et al. Correlation between electronic structure and magnetic properties of Fe-doped ZnO films. **Journal of Applied Physics**, v. 111, n. 12, 15 jun. 2012.

DA TRINDADE, L. G. et al. Influence of ionic liquid on the photoelectrochemical properties of ZnO particles. **Ceramics International**, v. 44, n. 9, p. 10393–10401, 15 jun. 2018.

DE PERES, M. L. et al. Zinc oxide nanoparticles from microwave-assisted solvothermal process: Photocatalytic performance and use for wood protection against xylophagous fungus. **Nanomaterials and Nanotechnology**, v. 9, 2019.

DE SÁ, B. S. et al. Microwave-assisted solvothermal synthesis of In-MIL-68 derived hollow In₂O₃ microrods for enhanced 1-pentanol sensing performance. **Materials Today Communications**, v. 37, 1 dez. 2023.

DJURIŠIĆ, A. B.; LEUNG, Y. H.; CHING NG, A. M. **Strategies for improving the efficiency of semiconductor metal oxide photocatalysis**. **Materials Horizons** Royal Society of Chemistry, , 1 jul. 2014.

DONG, C.-L. Soft-x-ray spectroscopy probes nanomaterial-based devices. **SPIE Newsroom**, 2007.

DUFFY, E.; MORRIN, A. **Endogenous and microbial volatile organic compounds in cutaneous health and disease**. **TrAC - Trends in Analytical Chemistry** Elsevier B.V., , 1 fev. 2019.

EWALD, C. et al. Role of potassium loading in ZnO-based gas sensors under NO₂ exposure – Operando diffuse reflectance infrared Fourier transform spectroscopic study. **Sensors and Actuators B: Chemical**, v. 393, p. 134321, 15 out. 2023.

FENNELL, J. F. et al. Nanodrähte in Chemo- und Biosensoren: aktueller Stand und Fahrplan für die Zukunft. **Angewandte Chemie**, v. 128, n. 4, p. 1286–1302, 22 jan. 2016.

FRANKE, M. E.; KOPLIN, T. J.; SIMON, U. **Metal and metal oxide nanoparticles in chemiresistors: Does the nanoscale matter?** *Small*, jan. 2006.

GE, M. et al. Controllable synthesis of hierarchical assembled porous ZnO microspheres for acetone gas sensor. **Sensors and Actuators, B: Chemical**, v. 220, p. 356–361, 10 jul. 2015.

GUO, L. et al. Gas sensor based on MOFs-derived Au-loaded SnO₂ nanosheets for enhanced acetone detection. **Journal of Alloys and Compounds**, v. 906, 15 jun. 2022.

HAO, R. et al. Emission characteristics, environmental impact, and health risk assessment of volatile organic compounds (VOCs) during manicure processes. **Science of the Total Environment**, v. 906, 1 jan. 2024.

HE, C. et al. **Recent Advances in the Catalytic Oxidation of Volatile Organic Compounds: A Review Based on Pollutant Sorts and Sources.** *Chemical Reviews*American Chemical Society, , 10 abr. 2019.

HERNÁNDEZ-MACEDO, M. L. et al. Gases and volatile compounds associated with microorganisms in blown pack spoilage of Brazilian vacuum-packed beef. **Letters in Applied Microbiology**, v. 55, n. 6, p. 467–475, 2012.

HE, Y. et al. ZIF-8 derived ZnWO₄ nanocrystals: Calcination temperature induced evolution of composition and microstructures, and their electrochemical performances as anode for lithium-ion batteries. **Electrochimica Acta**, v. 367, 20 jan. 2021.

JAISWAL, J.; SINGH, P.; CHANDRA, R. Low-temperature highly selective and sensitive NO₂ gas sensors using CdTe-functionalized ZnO filled porous Si hybrid hierarchical nanostructured thin films. **Sensors and Actuators, B: Chemical**, v. 327, 15 jan. 2021.

JEONG, S. Y.; KIM, J. S.; LEE, J. H. **Rational Design of Semiconductor-Based Chemiresistors and their Libraries for Next-Generation Artificial Olfaction.** *Advanced Materials*Wiley-VCH Verlag, , 1 dez. 2020.

JING, Z.; ZHAN, J. Fabrication and gas-sensing properties of porous ZnO nanoplates. **Advanced Materials**, v. 20, n. 23, p. 4547–4551, 2 dez. 2008.

KAMPA, M.; CASTANAS, E. **Human health effects of air pollution.** *Environmental Pollution*, jan. 2008.

KAZEMIAN, M. et al. X-ray imaging and micro-spectroscopy unravel the role of zincate and zinc oxide in the cycling of zinc anodes in mildly acidic aqueous electrolytes. **Journal of Power Sources**, v. 524, 15 mar. 2022.

KHATIB, M.; HAICK, H. **Sensors for Volatile Organic Compounds.** *ACS Nano*American Chemical Society, , 24 maio 2022.

KIM, H. J.; LEE, J. H. **Highly sensitive and selective gas sensors using p-type oxide semiconductors: Overview.** *Sensors and Actuators, B: Chemical*, 1 mar. 2014.

KIM, I. D.; ROTHCHILD, A.; TULLER, H. L. Advances and new directions in gas-sensing devices. *Acta Materialia*, v. 61, n. 3, p. 974–1000, fev. 2013.

KORPI, A.; JÄRNBERG, J.; PASANEN, A. L. **Microbial volatile organic compounds.** *Critical Reviews in Toxicology*, fev. 2009.

KRISHNA, K. G. et al. **Nanostructured metal oxide semiconductor-based gas sensors: A comprehensive review.** *Sensors and Actuators A: Physical* Elsevier B.V., , 1 jul. 2022.

KUMAR, G.; MASRAM, D. T. Sustainable Synthesis of MOF-5@GO Nanocomposites for Efficient Removal of Rhodamine B from Water. *ACS Omega*, v. 6, n. 14, p. 9587–9599, 13 abr. 2021.

LANGFORD, J. I.; WILSON, A. J. C. **Seherrer after Sixty Years: A Survey and Some New Results in the Determination of Crystallite Size.** *J. Appl. Cryst.* [s.l.: s.n.].

LEE, I. et al. Hollow Metal-Organic Framework Microparticles Assembled via a Self-Templated Formation Mechanism. *Crystal Growth and Design*, v. 15, n. 11, p. 5169–5173, 29 set. 2015.

LEE, J. et al. Precise control of surface oxygen vacancies in ZnO nanoparticles for extremely high acetone sensing response. *Journal of Advanced Ceramics*, v. 11, n. 5, p. 769–783, 1 maio 2022.

LI, P. Z.; ARANISHI, K.; XU, Q. ZIF-8 immobilized nickel nanoparticles: Highly effective catalysts for hydrogen generation from hydrolysis of ammonia borane. *Chemical Communications*, v. 48, n. 26, p. 3173–3175, 29 fev. 2012.

LÜ, Y. et al. MOF-templated synthesis of porous Co₃O₄ concave nanocubes with high specific surface area and their gas sensing properties. *ACS Applied Materials and Interfaces*, v. 6, n. 6, p. 4186–4195, 26 mar. 2014.

MACHADO, M. M. et al. Construção de um reator de plasma descarga corona para eliminação de compostos orgânicos voláteis. *Química Nova*, v. 38, n. 1, p. 128–131, 1 jan. 2015.

MANDAYO, G. G. et al. **Strategies to enhance the carbon monoxide sensitivity of tin oxide thin films.** *Sensors and Actuators, B: Chemical. Anais...* 15 out. 2003.

MATSUNAGA, N. et al. Formulation of gas diffusion dynamics for thin film semiconductor gas sensor based on simple reaction-diffusion equation. *Sensors and Actuators, B: Chemical*, v. 96, n. 1–2, p. 226–233, 15 nov. 2003.

Micromachined metal oxide gas sensors opportunities to improve sensor performance (Simon, Barsan, 2001). [s.d.].

MITTAL, A.; ROY, I.; GANDHI, S. **Drug Delivery Applications of Metal-Organic Frameworks (MOFs)**. [s.l: s.n.]. Disponível em: <www.intechopen.com>.

MOKOENA, T. P.; SWART, H. C.; MOTAUNG, D. E. A review on recent progress of p-type nickel oxide based gas sensors: Future perspectives. **Journal of Alloys and Compounds**, v. 805, p. 267–294, 15 out. 2019.

OLIVEIRA, T. N. T. et al. Improved triethylamine sensing properties by designing an In₂O₃/ZnO heterojunction. **Sensors and Actuators Reports**, v. 3, 1 nov. 2021.

OPREA, A. et al. Basics of semiconducting metal oxide-based gas sensors. Em: **Gas Sensors Based on Conducting Metal Oxides: Basic Understanding, Technology and Applications**. [s.l.] Elsevier, 2018. p. 61–165.

PASANEN, A.-L. et al. **CRITICAL ASPECTS ON THE SIGNIFICANCE OF MICROBIAL VOLATILE METABOLITES AS INDOOR AIR POLLUTANT** *Environment International*. [s.l: s.n.].

PASQUARELLA, C.; PITZURRA, O.; SAVINO, A. **The index of microbial air contamination**. *Journal of Hospital Infection* W.B. Saunders Ltd, , 2000.

PERFECTO, T. M.; ZITO, C. A.; VOLANTI, D. P. Effect of NiS nanosheets on the butanone sensing performance of ZnO hollow spheres under humidity conditions. **Sensors and Actuators, B: Chemical**, v. 334, 1 maio 2021.

REN, X. et al. Conductometric NO₂ gas sensors based on MOF-derived porous ZnO nanoparticles. **Sensors and Actuators B: Chemical**, v. 357, 15 abr. 2022.

ROSO, S. et al. Synthesis of ZnO nanowires and impacts of their orientation and defects on their gas sensing properties. **Sensors and Actuators, B: Chemical**, v. 230, p. 109–114, 1 jul. 2016.

SALUNKHE, R. R.; KANETI, Y. V.; YAMAUCHI, Y. **Metal-Organic Framework-Derived Nanoporous Metal Oxides toward Supercapacitor Applications: Progress and Prospects**. *ACS Nano* American Chemical Society, , 27 jun. 2017.

SANTOS, G. S. M. et al. MOF-derived Co₃O₄-ZnO heterostructure for 3-methyl-1-butanol detection. **Sensors and Actuators B: Chemical**, v. 408, 1 jun. 2024.

SCHMIDT, K.; PODMORE, I. Current Challenges in Volatile Organic Compounds Analysis as Potential Biomarkers of Cancer. **Journal of Biomarkers**, v. 2015, p. 1–16, 30 mar. 2015.

SCHULZ, S.; DICKSCHAT, J. S. **Bacterial volatiles: The smell of small organisms**. *Natural Product Reports*, 2007.

SCHÜTZE, A. et al. Highly sensitive and selective VOC sensor systems based on semiconductor gas sensors: How to? **Environments - MDPI**, v. 4, n. 1, p. 1–13, 1 mar. 2017.

SOHRABI, H. et al. Metal-organic frameworks (MOF)-based sensors for detection of toxic gases: A review of current status and future prospects. **Materials Chemistry and Physics**, v. 299, 15 abr. 2023.

SUN, X.; SHAO, K.; WANG, T. Detection of volatile organic compounds (VOCs) from exhaled breath as noninvasive methods for cancer diagnosis Young Investigators in Analytical and Bioanalytical Science. **Analytical and Bioanalytical Chemistry**, v. 408, n. 11, p. 2759–2780, 1 abr. 2016.

Synthesis and characterization of a new hybrid material (MOF5)(BENNABI; BELBACHIR, 2017). [s.d.].

THOMMES, M. et al. Physisorption of gases, with special reference to the evaluation of surface area and pore size distribution (IUPAC Technical Report). **Pure and Applied Chemistry**, v. 87, n. 9–10, p. 1051–1069, 1 out. 2015.

THORN, R. M. S.; GREENMAN, J. **Microbial volatile compounds in health and disease conditions**. **Journal of Breath Research**, jun. 2012.

TURGUT, E. et al. Oxygen partial pressure effects on the RF sputtered p-type NiO hydrogen gas sensors. **Applied Surface Science**, v. 435, p. 880–885, 30 mar. 2018.

WANG, Y. et al. **Microbial volatile organic compounds and their application in microorganism identification in foodstuff**. **TrAC - Trends in Analytical Chemistry** Elsevier B.V., , 1 abr. 2016.

WANG, Y. et al. Microwave hydrothermally synthesized metal-organic framework-5 derived C-doped ZnO with enhanced photocatalytic degradation of Rhodamine B. **Langmuir**, v. 36, n. 33, p. 9658–9667, 25 ago. 2020.

WAN, Q. et al. Fabrication and ethanol sensing characteristics of ZnO nanowire gas sensors. **Applied Physics Letters**, v. 84, n. 18, p. 3654–3656, 3 maio 2004.

WEN, W.; WU, J. M.; WANG, Y. DE. Large-size porous ZnO flakes with superior gas-sensing performance. **Applied Physics Letters**, v. 100, n. 26, 25 jun. 2012.

WILLIAMS, D. E. **Semiconducting oxides as gas-sensitive resistors** **Sensors and Actuators B**. [s.l: s.n.].

WOO, H. S. et al. Co-doped branched ZnO nanowires for ultraselective and sensitive detection of xylene. **ACS Applied Materials and Interfaces**, v. 6, n. 24, p. 22553–22560, 24 dez. 2014.

WU, W. et al. Controlled synthesis of monodisperse sub-100 nm hollow SnO₂ nanospheres: A template-and surfactant-free solution-phase route, the growth mechanism, optical properties, and application as a photocatalyst. **Chemistry - A European Journal**, v. 17, n. 35, p. 9708–9719, 22 ago. 2011.

XU, C. et al. **Grain size effects on gas sensitivity of porous SnO₂-based elements** **Sensors and Actuators B**. [s.l: s.n.].

XU, H. et al. A novel method for improving the performance of ZnO gas sensors. **Sensors and Actuators, B: Chemical**, v. 114, n. 1, p. 301–307, 30 mar. 2006.

XU, J. et al. **Grain size control and gas sensing properties of ZnO gas sensor****Sensors and Actuators B**. [s.l: s.n.]. Disponível em: <www.elsevier.nl/locate/sensorb>.

ZAKI, S. E. et al. Role of oxygen vacancies in vanadium oxide and oxygen functional groups in graphene oxide for room temperature CO₂ gas sensors. **Sensors and Actuators, A: Physical**, v. 294, p. 17–24, 1 ago. 2019.

ZHANG, D. et al. Room-temperature high-performance acetone gas sensor based on hydrothermal synthesized SnO₂-reduced graphene oxide hybrid composite. **RSC Advances**, v. 5, n. 4, p. 3016–3022, 2015.

ZHANG, S. B.; WEI, S. H.; ZUNGER, A. Intrinsic n-type versus p-type doping asymmetry and the defect physics of ZnO. **Physical Review B - Condensed Matter and Materials Physics**, v. 63, n. 7, 31 jan. 2001.

ZHANG, T.; LIN, W. **Metal-organic frameworks for artificial photosynthesis and photocatalysis**. **Chemical Society Reviews** Royal Society of Chemistry, , 21 ago. 2014.

ZHANG, X. et al. ZIF-8 derived hierarchical hollow ZnO nanocages with quantum dots for sensitive ethanol gas detection. **Sensors and Actuators, B: Chemical**, v. 289, p. 144–152, 15 jun. 2019.

ZHAO, J. et al. Microwave Hydrothermal Synthesis of In₂O₃-ZnO Nanocomposites and Their Enhanced Photoelectrochemical Properties . **Journal of The Electrochemical Society**, v. 166, n. 5, p. H3074–H3083, 2019.

ZHU, Z. et al. Multichannel pathway-enriched mesoporous NiO nanocuboids for the highly sensitive and selective detection of 3-hydroxy-2-butanone biomarkers. **Journal of Materials Chemistry A**, v. 7, n. 17, p. 10456–10463, 2019.

ZITO, C. A. et al. Bicone-like ZnO structure as high-performance butanone sensor. **Materials Letters**, v. 223, p. 142–145, 15 jul. 2018.