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Decoding disorder signatures of AuCl_3 and vacancies in MoS_2 films: from synthetic to experimental inversion

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Abstract

This study investigates the scope of application of a recently designed inversion methodology that is capable of obtaining structural information about disordered systems through the analysis of their conductivity response signals. Here we demonstrate that inversion tools of this type are capable of sensing the presence of disorderly distributed defects and impurities even in the case where the scattering properties of the device are only weakly affected. This is done by inverting the DC conductivity response of monolayered MoS_2 films containing a minute amount of AuCl_3 coordinated complexes. Remarkably, we have successfully extracted detailed information about the concentration of AuCl_3 by decoding its signatures on the transport features of simulated devices. In addition to the case of theoretically generated Hamiltonians, we have also carried out a full inversion procedure from experimentally measured signals of similar structures. Based on experimental input signals of MoS_2 with naturally occurring vacancies, we were able to quantify the vacancy concentration contained in the samples, which indicates that the inversion methodology has experimental applicability as long as the input signal is able to resolve the characteristic contributions of the type of disorder in question. Being able to handle more complex, realistic scenarios unlocks the method's applicability for designing and engineering even more elaborate materials.

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1. Introduction

Research on bidimensional materials design, especially involving 2D transition metal dichalcogenides (TMDs), has been in focus since the last decade for their characteristic thermal, mechanical, and electronic properties [1–9]. Along with it, methods for enhancing their properties, such as chemical functionalization, are of immense interest and are driven by the possibility of tuning their electronic, optical, and catalytic capabilities [10–18]. Such doping techniques rely on the fabrication of lattice defects; that is, they rely on systematic breakage of the crystalline symmetry through vacancies, chemical adsorptions, or atom substitutions, to mention but a few [19–24]. These defects modify the transport properties of materials but have the drawback of drastically elevating the complexity of any attempt to theoretically describe and accurately predict their detailed responses to externally applied fields.

For example, it is well established that adding non-periodic inclusions to an otherwise crystalline system will always make *ab initio* calculations significantly more costly. For the same reason, once disorder comes into play, even greater becomes the cost for solving the inverse problem, i.e. the recovery of the system's composition from its spectral measurements. Driven by the ultimate goal of designing materials that behave on our own accord, we must be able to accurately determine the effect that the concentration of a given set of defects has on the properties of a system. Consequently, finding the solution to such inverse problems is paramount.

One approach for designing a material with a specific measurement response is determining what configuration of elements and impurities a system needs to generate a desired signal. This is normally referred to as an inverse problem, and by definition, problems of that type are convoluted by nature. Similarly to the endeavor of determining an object from its shadow projected on a wall, this problem consists of determining the components of a system from a measurement of one of its spectral function projections on the space of a significant quantity. An inversion method capable of decoding any signature in this way has to be able to accurately detect the presence and quantity of different types of disorder in the system. To make matters worse, defects and impurities often have very distinct signatures on how they impact a system's response. While substitutional impurities are expected to have a stronger impact on the transport properties, adsorbed atoms and molecules or coordinated complexes tend to cause weaker changes to the transport features. Being able to extract structural information from a system composed of substitutional and adsorbed impurities can be seen as a way to attest to the sensitivity of one's inversion procedure.

With that in mind, we apply a recently formulated methodology capable of extracting the information about the number of impurities on various systems based on the minimization of a functional that draws out information from a measurement signature [25–28]. This inversion method has demonstrated reliable accuracy when applied to a variety of systems and input quantities. The method has been successfully implemented to acquire information regarding vacancies as well as multiple substitutional impurities in various different materials. Furthermore, the method has been used to obtain information not only of global quantities, e.g. the total impurity concentration, but also information of a local nature such as the spatial mapping of the underlying disorder [27]. However, all cases studied thus far have involved fully synthetic cases in which a Hamiltonian associated with a disordered configuration is employed to generate the physical response of that specific system. Using this response as the sole input of the methodology, the inversion task consists of obtaining as much information as possible about that specific Hamiltonian without availing of any prior knowledge about that synthetically generated system. Such a synthetic case is supposed to serve as a proxy for the realistic situation of using experimentally obtained signals as the inversion input function.

In this article, we further expand the aforementioned inversion methodology by applying it to a novel, more complex system—thin films of molybdenum disulfide (MoS₂) with adsorbed gold coordination complexes (AuCl₃) in the presence of vacancies. Besides testing the accuracy of the inversion method with coexisting substitutional and adsorbed impurities, we also consider the case of experimentally generated signals as the sole input function, in addition to the usual synthetic approach.

Different chemical functionalization techniques have been carried out on MoS₂ such as organohalides functionalization on liquid exfoliated flakes [29, 30], functionalization on MoS₂ grown by chemical vapor deposition with ligands that passivate the chalcogen vacancies [14, 31] and the recently developed route of single Au atoms functionalization via the formation of S-Au-Cl₃ coordination complexes, which allow for Fermi level tuning via p-type doping, high exciton-to-trion ratios on the photoluminescence spectrum and enhanced thermal boundary conductance [16]. The latter comprises a kind of functionalization that does not depend on the presence of lattice defects for passivation or substitution, so we can straightforwardly model it using a combination of Density Functional Theory (DFT) and Tight Binding (TB) techniques [32–35]. The independence between the occupancy of S-Au-Cl₃ complexes and the presence of S vacancies further allows for an uncomplicated combination of both types of disorder in our simulations. Since the chalcogen vacancies are commonly

formed defects due to the types of atomic bonding in monolayer TMDs [36–40], it is essential that such defects are considered during simulations and while designing materials with TMD as hosts. That justifies our choice of host and dopant since the adsorbed AuCl_3 coexist with vacancies, which ultimately are equivalent to substitutional impurities.

2. Methods

2.1. Theoretical calculations

The inversion method in question consists of minimizing the functional $\chi(n)$ that computes the misfit between a measurement signal and the calculated average of that response over different configurations containing disorder in a concentration n given by

$$\chi(n) = \frac{1}{\varepsilon_+ - \varepsilon_-} \int_{\varepsilon_-}^{\varepsilon_+} dE [\Gamma(E) - \langle \Gamma(E, n) \rangle]^2, \quad (1)$$

where $\Gamma(E)$ is the input signal of choice evaluated at energy E . To avoid the ill-definition of the inversion problem, we assume such an input signal belongs to a definite system containing an unknown amount of disorder. Alternatively, we can interpret this as the signal we want to mimic by implementing the appropriate amount of disorder to the pristine defined system. One obtains such correct concentration $n_{\min}$ by minimizing the functional $\chi(n)$. $\varepsilon_{\pm}$ are the extremity values of the selected energy spectrum range, and $\langle \Gamma(E, n) \rangle$ is the Configuration Average (CA) of that quantity over N_c configurations containing a concentration of disorder n calculated as [26]

$$\langle \Gamma(E, n) \rangle = \frac{1}{N_c} \sum_{m=1}^{N_c} \Gamma_m(E, n). \quad (2)$$

The function $\Gamma_m(E, n)$ on the righthand side of equation (2) is then calculated from a TB model for the different configurations considered. The TB Hamiltonian, in turn, is generated with parameters obtained from DFT calculations. Here, we considered two situations, pristine monolayer MoS_2 , and a AuCl_3 complex adsorbed on the TMD chalcogen. The primitive 2D MoS_2 unit cell has hexagonal symmetry and is formed by a triple atomic layer with the Mo layer sandwiched by two S ones [41]. For our calculations, we use a rectangular cell replicated 3×4 times (36 atoms), forming a supercell with dimensions $9.77 \text{ \AA} \times 11.28 \text{ \AA}$ and with $3.22 \text{ \AA}$ of thickness after the DFT relaxation process. To avoid spurious interactions between mirror cells along the normal to the plane direction, $15 \text{ \AA}$ of vacuum is included. For the AuCl_3 -adsorbed case, one AuCl_3 complex was included with the gold atom positioned $2.69 \text{ \AA}$ (after relaxation) above a S atom, as shown in figure 1(a). The total-energy and electronic-structure results were obtained via DFT as implemented in the SIESTA code [42]. Exchange and correlation are described by a combination of a revised version of the Perdew-Wang 86 (rPW86) [43] exchange functional, with a semilocal correlation energy functional in the generalized gradient approximation of Perdew, Burke, and Ernzerhof (PBE) [44] and a nonlocal van der Waals

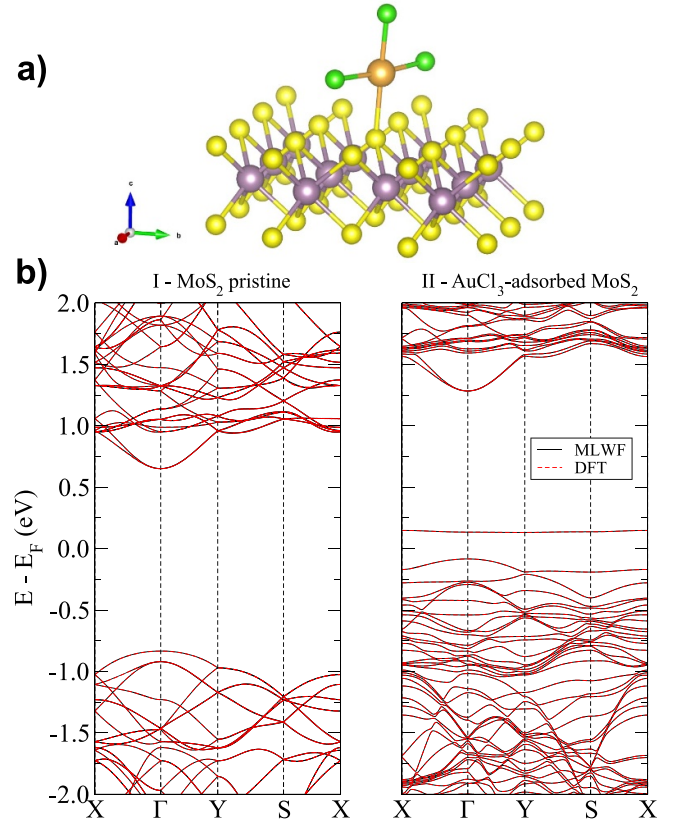


Figure 1. (a) Standard orientation of crystal shape for the $(\text{MoS}_2)\text{AuCl}_3$ disorder unit cell (purple sphere = Mo, yellow sphere = S, gold sphere = Au, green sphere = Cl). (b) Electronic band structure of I - pristine MoS_2 and II - AuCl_3 -adsorbed MoS_2 calculated via DFT (red dashed lines) and the corresponding MLWF one (black solid lines). The agreement between the curves is such that it makes them hardly distinguishable.

contribution proposed by Vydrov *et al* [45]. The inclusion of van der Waals corrections is important in this case to correctly describe the interaction between the adsorbed molecule and the monolayer MoS_2 [46]. A double- ζ -polarized (DZP) basis set with a real-space grid cutoff of 200 Ry is used. The BZ is sampled using a $6 \times 6 \times 1$ Monkhorst–Pack k-point mesh.

To obtain the TB parameters from the DFT results, we employed the Maximally localized Wannier functions (MLWF) method [47, 48] as implemented in the Wannier90 code [49]. We considered the projections of p_x , p_y and p_z orbitals of the chalcogen atoms and the five d orbitals of Mo, as well as the s and all the p orbitals of Au and Cl, respectively. The disentanglement procedure reached a convergence of the order of 10^{-8} , resulting in a very good agreement of the MLWF bands in comparison with the DFT ones, as can be viewed in figure 1(b). Experimental results for MoS_2 , fabricated in its single layer form, by means of mechanical exfoliation, exhibit a direct band gap of 1.8–1.9 eV [50]. We found a direct band gap of 1.5 eV for monolayer MoS_2 , which is a slightly lower value. However, G0W0 corrections overestimate the band gap to 2.6–2.7 eV [51]. In such a scenario we chose to keep the PBE result.

To tackle the memory and computational costs of calculating the configurational averages, we solve Kubo's formulas for DC conductivity using the Chebyshev polynomials Green function method via the quantum transport software for real space TB simulations KITE [52–54], where we can set the simulated temperature at $T = 300$ K by altering the modulating Fermi–Dirac distribution. Once we are dealing with large lattices with no translational symmetry due to defects, this real-space high-performance software proves exceptionally useful for performing spectral calculations. The DFT-derived parameters corresponding to a pristine MoS₂ unit cell (all the onsite potentials for each orbital as well as all the second-nearest-neighbors inter-orbital hoppings of the 3×4 set of primitive MoS₂ cells) serve to establish the repetition base for a pristine description of the MoS₂ system. This unit cell comprises 132 orbitals—three **p** orbitals for each sulfur atom and five **d** orbitals for each molybdenum atom. In addition, ten more empty, disconnected orbitals are reserved in the modelling of the unit cell to later house the adsorbed impurities in case the given unit cell is selected to have such an impurity. Now, we use the DFT-derived TB parameters relative to the unit cell containing 3×4 MoS₂ cells plus one coordinated AuCl₃ complex composed by one **s** orbital for gold and three **p** orbitals for each chlorine that define the orbitals of the adsorbed impurity in our model (illustrated in figure 1(a)). In accordance with the disorder concentration value, unit cells containing impurities will replace various random 3×4 sites in our lattice with new orbital and inter-orbital parameters for the Mo and S atoms altered with relation to the pristine case. In this case, the energies and connections relative to the impurity atoms replace the orbitals previously held as empty, disconnected mock orbitals in our modelling. The unit cells are repeated to form $54\,067\text{ nm}^2$ systems with randomly placed impurities in different sites with predetermined concentrations. Our built systems comprise approximately 7.1×10^6 orbitals on a 64×64 unit cell grid containing various disorder concentrations. In each DC conductivity calculation, we used an expansion of 2048 moments of the Chebyshev polynomials for 3 different random trace vector iterations [55].

Figure 2 shows the effect of AuCl₃ complexes and vacancies on the band gap. The vacancies are simulated by imposing a high onsite potential in the target orbitals—here the **s** and **p** orbitals of various S atoms in the 3×4 reproduction cell. The very occurrence of the unique peaks produced by the presence of disorder illustrates how the CAs with different disorder concentrations provide unique curves that populate the space of possible input signals. Here, the concentration of AuCl₃ is exaggerated to make the peaks visible on the same scale. The positions of the peaks indicate an independence between the types of disorder in the transmission gap region. Such effective orthogonality within the disorder concentration range of interest guarantees that AuCl₃ inversions and vacancy inversions can be performed separately, minimizing the overall cost of the inversion procedure from scaling exponentially with the number of disorder types to linearly in this case.

Since we aim to carry out inversion tasks not only from synthetically generated signals but also from experimentally measured ones, in what follows we include a brief

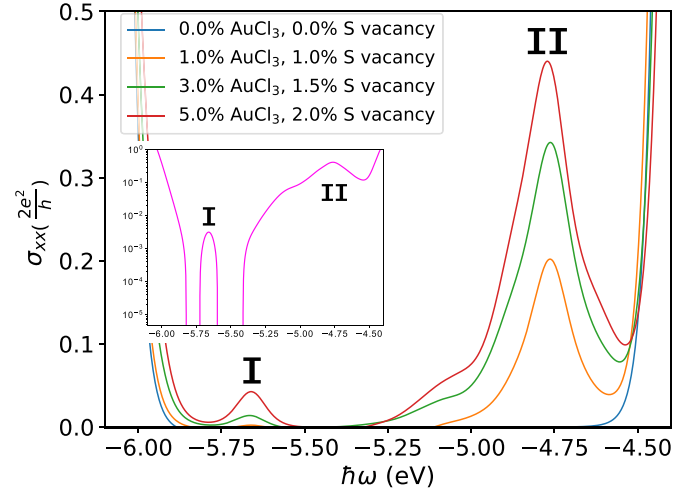


Figure 2. Intra-gap effect of the disorders shown in averages over 40 configurations. The heights of the peaks **I** are proportional to the increased presence of AuCl₃. The concentrations of AuCl₃ are exaggerated to illustrate both peaks in the same scale. The heights of the peaks **II** are proportional to the increased presence of S vacancies. In the inset, we can see the case for typical concentrations of 1% AuCl₃ and 2% S vacancies in the purple log-linear plot.

account of the experimental details of the material fabrication. Subsequent information about the conductivity measurements will also be provided.

2.2. Material fabrication

MoS₂ flakes were synthesized on 300 nm thick SiO₂/Si substrates using the chemical vapor deposition (CVD) method as in the previous work [16]. Here, MoO₃ (6mg) and NaCl (24 mg) powders were mixed and placed in an alumina boat in the high-temperature zone of the CVD furnace. The substrate was placed facedown above the alumina boat containing the NaCl/MoO₃ mixture. Meanwhile, 140 mg of pure sulfur was placed in another alumina boat in a low-temperature zone. Both alumina boats were placed at a distance of 22 cm along the quartz tube furnace. First, a flow rate of 600 sccm of high-purity argon (Ar) gas was passed through the experimental setup for 15 min to purge the quartz tube. Subsequently, a flow rate of 200 sccm (Ar) was maintained during the growth process. The Sulphur temperature was 250 °C. The growth temperature to obtain MoS₂ monolayers was $T = 850$ °C (high-temperature zone) for 10 min and after that, the system was cooled down.

Monolayer MoS₂ flakes morphological characterization was performed by optical and atomic force microscopies. AFM measurements were performed using an NT-MDT Solver Next microscope equipped with feedback sensors, ensuring a Noise XY of less than 0.3 nm and a Noise level Z (RMS in the band of 10–1000 Hz) typically at 0.03 nm. Non-contact mode imaging was employed, utilizing a Force Modulation AFM Probe featuring a Platinum Overall Coating, a force constant of 3 N m^{-1} , resonance frequency of 75 kHz, and dimensions of $225\text{ }\mu\text{m}$ length and $28\text{ }\mu\text{m}$ width. The

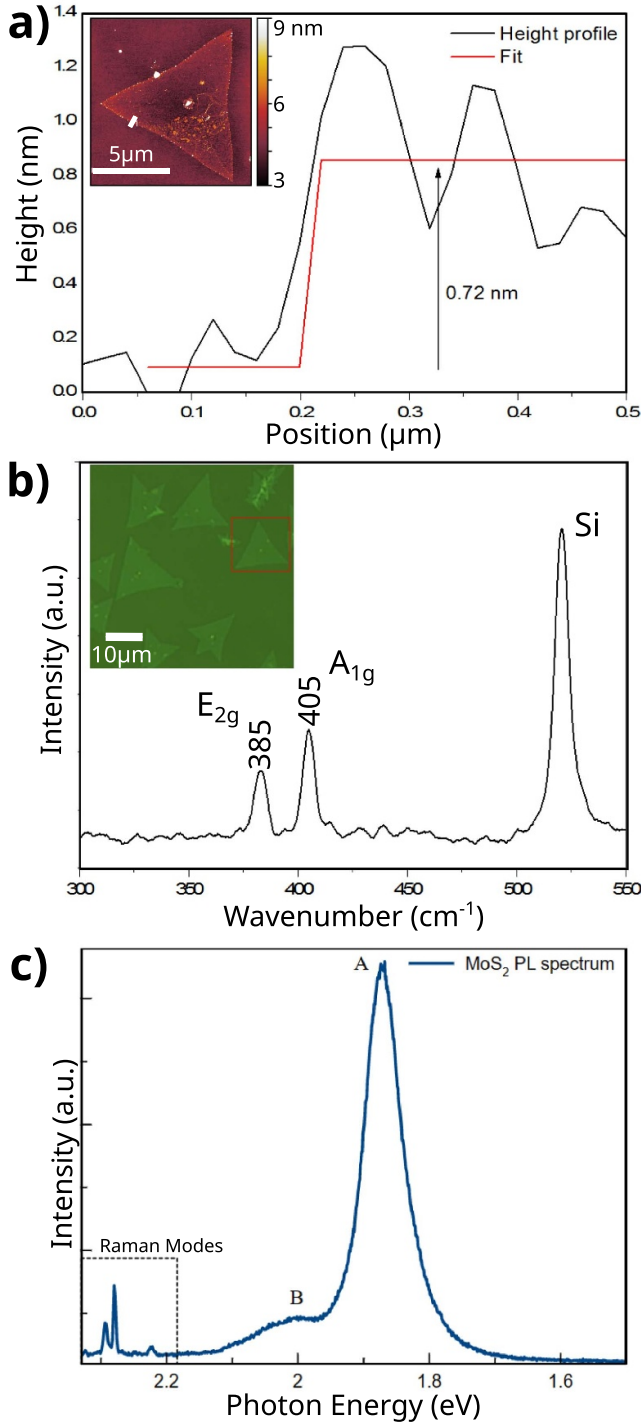


Figure 3. Characterization of monolayer MoS₂. (a) AFM height profile and topographic image (inset). (b) Optical image (inset) and Raman spectrum. (c) Photoluminescence spectrum.

formation of triangle-shaped nanocrystals with a mean lateral dimension of approximately 20 μm is observed (inset in figures 3(a) and (b)). The thickness of these triangular crystals determined by the height profile of AFM measurements (figure 3(a)) is around 0.7 nm, confirming the formation of monolayers. The Raman spectra (figure 3(b)) also confirm the flake's monolayer nature by the presence of two Raman-active

modes at the Γ point: E_{2g}^1 at around 385 cm⁻¹ and A_{1g} at 405 cm⁻¹, with a frequency difference of 20 cm⁻¹, typical of monolayer. A strong signal between 665 and 620 nm can be observed in the Photoluminescence spectra (figure 3(c)), which corresponds to *A* and *B* direct excitonic transitions. For Raman and Photoluminescence measurements, a confocal Raman spectrometer (WITec Alpha 300R) was utilized, incorporating a 100X objective lens and a 532 nm laser line with a laser power of 0.5 mW.

Back-gated field effect transistors were fabricated according to our previous work [16], using a 50 nm alumina (Al₂O₃) film as the gate oxide and a stack of Pt/TiN/p++Si as the back gate electrode. First, CVD-grown MoS₂ was transferred onto the alumina sample. The sample was spin-coated with A6 PMMA, followed by electron-beam (e-beam) lithography to specify the 2 μm channels, and then separated out by sulfur hexafluoride (SF₆) etch under 5 °C for 30 s. After each step, the sample was rinsed in acetone for 30 min, followed by a rinse in IPA. To define the source and drain contacts, the sample was spin-coated with methyl methacrylate (MMA) followed by A3 PMMA. The electron-beam lithography source and drain contacts were then patterned and further developed using a 1:1 mixture of 4-methyl-2-pentanone (MIBK) and IPA for 60 s. 40 nm of Ni and 30 nm of Au were deposited onto the patterns using electron beam (E-beam) evaporation. The evaporated materials were lifted off by immersing the sample in acetone for 30 min, followed by an IPA final rinse.

3. Results and discussion

Figure 4(a) displays in green the integrand in the right-hand side of equation (1) calculated over the whole spectrum, depicting the CA in red and the input signal in blue, both for the case of theoretically calculated DC conductivity for MoS₂. To isolate the effect of the presence of AuCl₃ coordinated complexes, every system in these calculations was subject to the same S vacancy concentration of 1%. Each CA was calculated over configurations containing a certain concentration of AuCl₃ from 0% to 1%. The red line in figure 4(a) represents one of these CAs, which was calculated over configurations containing 1% concentration of AuCl₃ impurity. The input function in blue has an unknown AuCl₃ concentration, to be determined via the inversion from the misfit depicted in green. Figure 4(b) shows the resulting misfit function, which has a very distinctive minimum indicating the impurity concentration that is likely to produce the smallest discrepancy when compared to the synthetic input signal. It is worth noting that the CA defined in equation (2) was calculated using only $N_c = 20$ configurations for seven different disorder concentration values n , seen on the horizontal axis. Given the large integration range of this particular inversion, a relatively low N_c number is enough to compute a high accuracy inversion considering the ergodic properties of the problem [26]. We, then, selected an arbitrary disorder concentration (in this case 0.5%, indicated by the vertical red dashed line) to characterize our synthetically generated input function $\Gamma(E)$ and compared it with the previously obtained CAs calculated via

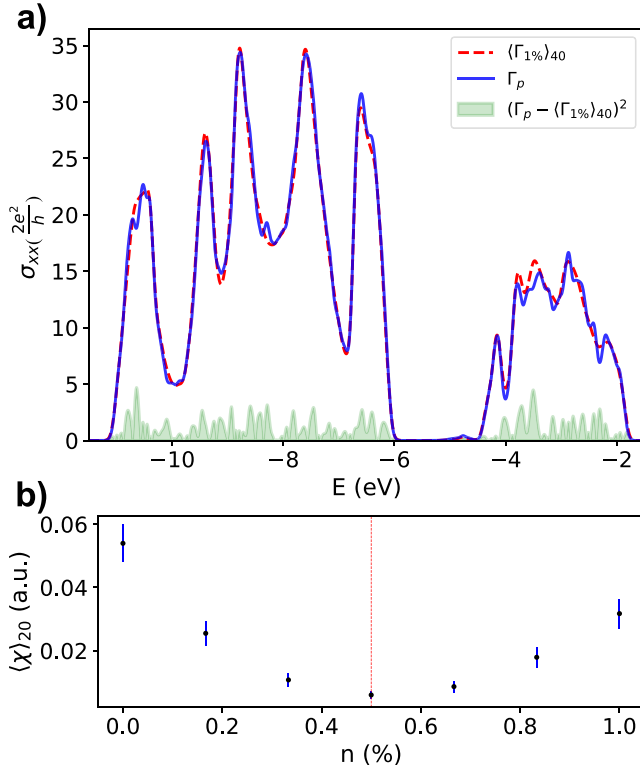


Figure 4. (a) Longitudinal DC conductivity at temperature $T = 300$ K comparison (filled green region) between 1% AuCl_3 concentration CA (dashed red line) and input configuration signal (solid blue line) containing unknown AuCl_3 concentration. The CA is composed of 40 configurations and all systems have a 1% concentration of S vacancies. (b) Misfit function $\chi(n)$ in arbitrary units averaged over 20 distinct inversion instances with minima at 0.5% AuCl_3 concentration. The vertical (red) line indicates the real concentration of the input configurations.

equation (1). Repeating this procedure for multiple input functions with the same disorder concentration, we generate the average misfit values and respective standard deviations indicated by the error bars around the data points. The presence of such a clear minimum that coincides with the red line indicates that the inversion was successful. Furthermore, from the minimum relative depth and the respective standard deviations, we can conclude that the inversion was not only successful but also very accurate.

Remarkably, this result reveals that this inversion methodology is indeed capable of resolving the contribution of a minute concentration of adsorbed atoms and molecules to the DC conductivity of a disordered structure, even when such complexes weakly affect the charge mobility signature of the system [16]. This configures an expansion of the method's applicability now encompassing a new type of complex disorder, also testifying to its general accuracy and robustness.

To demonstrate that this method has a practical use beyond the case of synthetically generated Hamiltonians, it remains to prove that the inversion can be carried out from realistic experimental data. To answer this question resourcefully, we can approach the same inversion problem with a different lens—instead of determining the concentration of disorder for

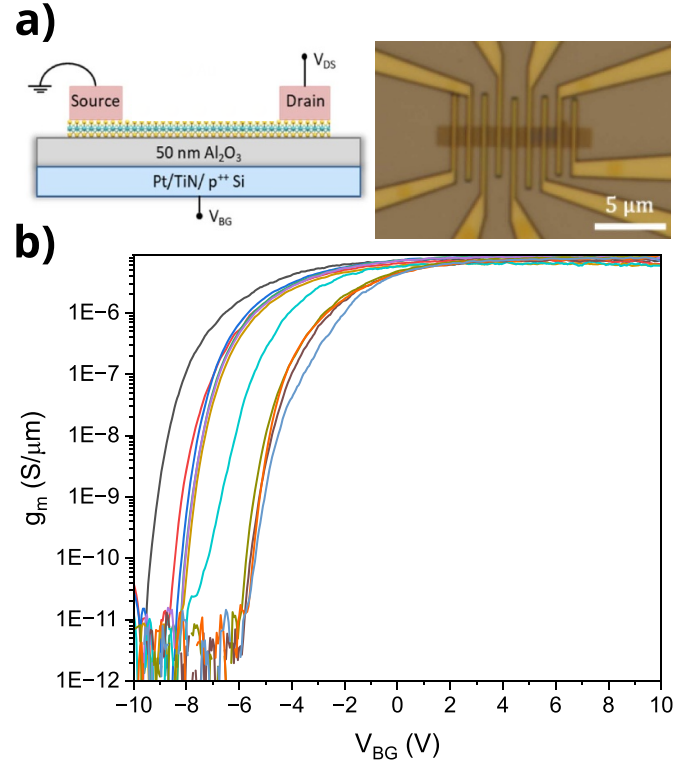


Figure 5. MoS₂ BGFET devices (a) Schematic of a back-gated field effect transistor on the MoS₂ channel and optical image of the fabricated devices showing the lateral structure. (b) Transconductance ($g_m = dI_{DS}/dV_{BG}$) vs back-gate voltage (V_{BG}) with a drain bias of 1 V for 12 different pristine MoS₂ BGFETs.

designing a material with certain properties, we can attempt to decode the impurity concentration in a sample by using its experimental signal as the input function $\Gamma(E)$. It is then useful to simplify the system to mono-layer MoS₂ containing vacancies as the only source of disorder due to the current data availability. Verily, chalcogen vacancies in the MoS₂ are one of the most common types of defects that occur during fabrication, along with antisite or atomic defects depending on the sample preparation method [40]. This implies that for any attempt to design TMDs by engineering a specific defect concentration, it is very likely that the effect of vacancies will also be present. The presence of vacancies is also effective in unlocking properties on MoS₂ such as the introduction of a local magnetic moment, as well as the conversion of the compound into p- or n-type semiconductors [56, 57]. Therefore, the ability to identify vacancy concentrations in MoS₂ samples becomes exceptionally valuable. In previous work, we successfully performed an inversion of MoS₂ containing chalcogen vacancies from theoretically calculated input signals [26]. In the following result, we present an experimental counterpart of that inversion.

Twelve back-gated field effect transistors (BGFETs) were designed and fabricated, according to our previous work [16] (figure 5), to extract the drain current (I_{DS}) versus the back gate voltage (V_{BG}). The transconductance (g_m) versus V_{BG} , defined as $g_m = dI_{DS}/dV_{BG}$, was used as input function $\Gamma(E)$ to calculate the vacancy concentration on the MoS₂ flakes.

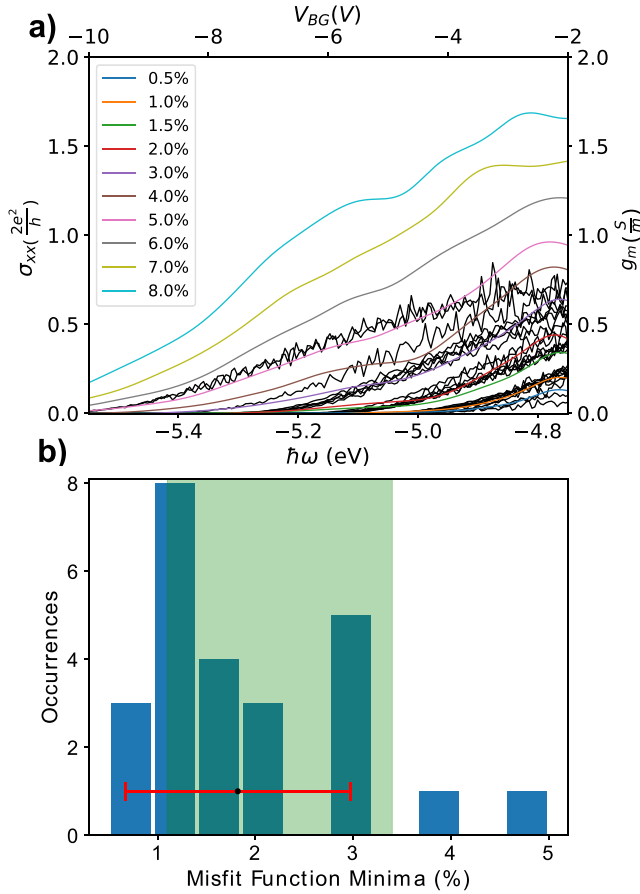


Figure 6. (a) Comparison between simulated MoS₂ DC conductivity at $T = 300$ K containing different vacancy concentrations (colored) and experimental conductivity data— g_m values normalized by the ratio between the gate width and length—for various MoS₂ systems (black), each of which contains unknown vacancy concentration. (b) Histogram of the minima of the different misfit functions for each black curve of (a). The black dot and red bar indicate the Gaussian mean and standard deviation of the minima distribution at $1.8 \pm 1.1\%$. The area in green represents the expected range of vacancy concentration previously reported for this method of fabrication.

Figure 5(b) shows $g_m \times V_{BG}$ curves of representatives of the twelve pristine MoS₂ BGfET. The observed threshold voltage (V_T) variation, between -10 and -5 V, can be attributed to different defect densities on the MoS₂ flake affecting the Fermi energy and electron mobility. The calculated average field effect mobility (μ_{FE}) was 15 ± 10 cm²/V^s, which is comparable to the mobility values reported for exfoliated single crystal material [58]. The ON state current was found to be approximately $120 \mu A \mu m^{-1}$ at $V_{DS} = 8$ V and $V_{BG} = 10$ V. The calculated subthreshold slope (SS) was 350 mV/decade.

Using the parametrization adopted in the earlier DFT calculations for the pristine mono-layer MoS₂, we can introduce chalcogen vacancies through KITE's environment. Performing calculations with similar parameters as the ones discussed previously, we arrive at CAs for MoS₂ at various vacancy concentrations. Figure 6(a) presents the experimental transconductivity data (black) and

the calculated CAs conductivity in colored curves, establishing a visual comparison that highlights the effect of vacancies on the conductivity curves and the possible concentration of vacancies present in the experimental data. Assuming the system is composed of pure MoS₂ and S vacancies, we can invert the experimental data to obtain the misfit functions minima as presented in figure 6(b). Through the inversions, we find vacancy concentrations falling within and around the expected range of 1.2%–3.4% of vacancy concentrations for CVD-grown MoS₂ [40, 59, 60].

The direct application of the presented method on an input signal generated from experimental data closes the gap between inversion tools being potentially useful and being of practical utility. Furthermore, it establishes that tools of that type can be employed as a reliable mechanism for material design and characterization.

4. Conclusion

In conclusion, we tackled two fronts of the presented inversion method application. We established that the method can be combined with first-principles calculations to perform complex, more realistic simulations and that its robustness and accuracy are preserved even when dealing with scatterers which possess extremely weak and feeble signatures.

This study also pioneered the method's implementation along with experimental data, where it displayed clear compatibility, paving its way for realistic applications in characterization and material design. To complement the results presented in the experimental-signal inversion part of this work, a systematic characterization of the material samples quantifying the different types of disorder present in the system will allow for a quantitative evaluation of the method's accuracy. It is also important to emphasize that, in cases where the inverted function is subject to impurities with extremely weak signatures, the accuracy of the method relies on the precision and resolution of the input function. If one would attempt an inversion of an experimental signal containing AuCl₃ complexes, for example, the conductivity experimental measurements must be correspondingly sensitive to the characteristic signatures of this type of disorder. At this higher level of precision not only the characterization of disorder concentration of AuCl₃ is needed to evaluate the method's accuracy, but also characterizations of the other possible impurities that may have stronger signatures, such as vacancies.

Finally, given the growing interest in material design and the demonstration that inversion methodologies do play a significant role in our capacity to engineer the physical properties of materials, it is encouraging to see that such tools are becoming more reliable, even in the presence of disorder. This is an important step towards the goal of reducing the computational complexity that is involved in the search for new materials and, in a period when contemporary material design research is greatly driven by machine learning algorithms, this work offers a transparency that is essential for the overall understanding of the research area, in contrast to full black-box approaches.

Data availability statement

All data that support the findings of this study are included within the article (and any supplementary files).

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Author contributions

FRD co-wrote the paper and conducted all the spectral simulations and data analysis. FM conducted the TB parametrization from DFT data and co-wrote the paper. DG conducted the experimental part and co-wrote the paper. LS conducted the DFT calculations. ARR led the TB parametrization. SM aided the theoretical implementations. CJSdM co-led the experimental part of the research. MSF led the research. All authors discussed and commented on the manuscript and on the results.

Conflict of interest

The authors declare no Competing Financial or Non-Financial Interests.

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