



Standardization of a NaI(Tl) scintillation detector and its use on environments for Geology teaching in a Brazilian university

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Abstract

This paper describes analytical protocols adopted for calibrating a portable gamma rays spectrometer using a 2" × 2" NaI(Tl) scintillation detector for performing exploration and environmental surveys. The steps involved the determination of counting times with good accuracy, calibration in energy and concentration, and data conversion to effective dose (Ed) for monitoring the radioactivity levels in the air, for indoor and outdoor environments. The χ^2 test was applied to the counting rate data acquired for ten different readings, allowing determine the greater accuracy among them. Some standards yielded values above of the detection limit for calibrating the K, eU, and eTh windows, whereas empirical equations provided satisfactory results. The count rates in these windows were obtained in selected environments related to Geology teaching activities in a Brazilian university. Then, they were converted to Ed, showing the applicability of the adopted procedures for environmental surveys. The radiological monitoring showed two sites exhibiting Ed values above the WHO guideline reference value of 1 mSv/year. The first was in a minerals and rocks museum and the second in a deposit (*litoteca*) dedicated to the rocks storage. Thus, the members of the public are only occasionally exposed to the natural gamma radiation there. Additionally, the Ed values are much lower than the recommended limit of 20 mSv/year for occupationally exposed individuals.

Keywords Portable gamma rays spectrometer · NaI(Tl) scintillation detector · Calibration · K, eU, and eTh windows · Effective dose

Introduction

Portable gamma rays spectrometry has been used since the 1960s for uranium exploration, geologic mapping, and environmental studies (Duval et al. 2005). The mapping of the natural radioelements K, U, and Th has been done based on windows centered over the energy photopeaks of 1461 keV (^{40}K), 1765 keV (^{214}Bi), and 2615 keV (^{208}Tl), respectively (IAEA 1989). Additionally, several methods to estimate the radionuclides concentration from the nuclear fallout have been developed. For instance, the ^{137}Cs and ^{134}Cs precipitation may be detected by monitoring two ~ 100 keV windows centered in the energies of 662 keV and 796 keV, respectively (Grasty and Cox 1997). Thus, such methods typically use windows of different γ -rays energies emitted by each

radionuclide, which are discriminated in equipment like multichannel analyzer (MCA) (IAEA 2003).

The field procedures for portable spectrometry depend on the research purpose, generally prospecting or environmental. The spectrometer type, detector volume, counting time, and measurement mode depend on the environmental radiation, which is linked to the type, size, and radioactive sources distribution. For such purpose, scintillation NaI(Tl) crystals of up to 350 cm³ have been extensively used as γ -rays detectors over the last 45 years (Adams and Freyer 1964; Adams and Gasparini 1970; Sudarshan et al. 1992; Leo 1994; Perez-Andujar and Pibida 2004; Knoll 2010; L'Annunziata 2012; Acton 2013).

Differential spectrometers are among the recent γ -rays equipment, for instance, 350 cm³ NaI(Tl) detectors that record data between 1024 and 2048 channels (energy range 0–3 MeV) (Acton 2013). The automatic spectrum stabilization occurs by the use of a reference radionuclide like ^{137}Cs , ^{40}K , or ^{208}Tl whose respective γ -rays energy photopeak corresponds to 662, 1461, and 2615 keV. The instruments can save the whole spectrum and select channels over a broad

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energy window in order to estimate the in situ concentration of the natural radioelements K, U, and Th (L'Annunziata 2012; Acton 2013). The calibration constants are stored in the equipment's memory that is able to record thousands of field measurements or hundreds of total spectra. Acceptable precision for quantitative analysis may be acquired from the use of large volume scintillation crystals or by increasing the counting time (L'Annunziata 2012; Acton 2013).

NaI(Tl) detectors (density = 3.66 g/cm³) have been largely applied to determine the formations density, chemical composition, soil salinity, and mineralogy as exhibit efficient detection, great robustness, reduced humidity sensitivity, and enhanced capacity to deal with higher counting rates without pulse piling. Additionally, scintillator crystals of higher density like CsI (4.51 g/cm³), LaBr₃ (5.29 g/cm³), and Bi₄Ge₃O₁₂ (BGO-bismuth germanate; 7.13 g/cm³) have been also used in order to improve the detection efficiency for the higher γ -rays energies (L'Annunziata 2012; Acton 2013).

It is important to highlight that smaller detectors permit a weight reduction to about 2 kg (Stoller et al. 1994). Also, BGO crystals (103 cm³) exhibit ~ 80% sensitivity of a NaI(Tl) detector (350 cm³) (Saint-Gobain 2016). NaI(Tl) crystals should be clean for applications involving low background counts and not contain K levels above 0.5 ppm in the whole mass. In contrast, the BGO detector may have a primary ²⁰⁷Bi contamination due to four γ -rays emissions from 470 to 2400 keV (Saint-Gobain 2016).

The response of a portable γ -rays spectrometer is dependent of the size, location, and geometry of the radioactive sources. Significant measures along one crossing can only be obtained if the source–detector geometry is constant to all observations. The detector must be also placed directly onto the ground surface, or it should be kept in a low but constant height for minimizing the effects of local reliefs' variation and radioelements distribution (Kogan et al. 1971; IAEA 1989). In order to obtain reliable measurements, the same source–detector geometry used to calibrate the spectrometer must be adopted in the field.

The counting time of each field measurement depends on the source radioactivity and requested precision. The K, U, and Th readings may be complicated due to the photopeaks overlap in the gamma spectra (Løfvborg and Mose 1987). In general, there is a commitment between the count rate precision and the minimum counting time required during the field reading. For example, counting times of 2 and 6 min are reasonable for highly and low radioactive rocks, respectively, when using a 350 cm³ NaI(Tl) detector (Løfvborg and Mose 1987).

Concrete pads of variable sizes possessing different concentration of natural radioelements are sometimes used for calibrating the γ -rays spectrometers. However, their access difficulty or high cost for most users justifies the adoption of more simple customized procedures as described in this

paper. It is considered here a portable γ -rays spectrometer based on a traditional NaI(Tl) scintillation detector and several standards of easy acquisition or preparation in the laboratory. The developed analytical protocols allow surveys interesting to Economic Geology (geochemical/geo-physical prospecting of radioactive ores like U-deposits) and Environmental/Medical Geology (evaluation of health risk due to the radiation exposure). A study case has been also illustrated here considering typical environments available in schools and universities dedicated to producing geologists and/or other Earth scientists.

Experimental steps

A digiDART high-performance portable digital MCA γ -rays spectrometer using a 2" × 2" (103 cm³) NaI(Tl) scintillation detector model 2BY2-DD from ORTEC-AMETEK has been used for γ -rays spectrometry in this study. The detector probe weight is ~ 4.6 kg, whose power consumption is 240 mW. It includes a preamplifier, a high-voltage power supply, and a cable for coupling to digiDART-LF. Such unit contains an amplifier, a 2048 MCA controlled by MAESTRO software and a LCD monitor, allowing visualize and store the acquired spectra.

This γ -rays spectrometer was initially calibrated for energy readings on using artificial and natural radioactive sources: ¹³⁷Cs, ⁶⁰Co, a solution containing ¹³³Ba, pure powdered KCl (52 wt% in K) as a ⁴⁰K source, pitchblende standard (1% U) as a ²¹⁴Bi source, and monazite sand standard (1% Th) as a ²⁰⁸Tl source. Then, for applicability in mineral deposits exploration and environmental surveys, it was calibrated for concentration readings on utilizing standards possessing variable contents of the natural radioelements K, U, and Th. Altogether 19 standards were inserted in 2.3-cm-high and 6.3-cm-diameter aluminum cans, whose radioactivity levels were taken inside a lead shielding for avoiding contribution of lateral radiation reaching the detector. The shielding consisted of two 4-cm-high, 14.5-cm-external diameter and 8.5-cm-internal diameter rings that yielded a 6-cm-thick protective lead wall.

Pitchblende and monazite sand standards having different uranium and thorium concentrations (NBL-101-A, NBL-102-A, NBL-103-A, NBL-104-A, NBL-105-A, NBL-106-A, NBL-107-A, NBL-108-A, NBL-109-A, and NBL-110-A) and providing from New Brunswick Laboratory, US Department of Energy, Argonne, Illinois, USA, were submitted to γ -rays readings after waiting ²²²Rn to reach secular radioactive equilibrium with ²²⁶Ra (at least 25 days). Such condition occurs when the number of parent atoms disintegrating per unit time equals to that of the daughter atoms formed at the same time. It happens when the parent half-life is much higher (at least 7 times) than that of the daughter

radionuclide (Ivanovich and Harmon 1992). In the ²³⁸U decay series, such time is ~ 4 weeks for the ²²⁶Ra–²²²Rn pair.

Three well-homogenized samples of stream sediments (LII-Th-1, LII-Th-2, and LII-Th-3) from Morro do Ferro, Poços de Caldas plateau were also used for yielding lower Th concentrations. Pure KCl (52 wt% in K, standard LII-KCl-1) and different mixtures prepared from this matrix and additions of pure SiO₂ were utilized to obtain variable potassium concentration. The standard K6 (LII-KCl-6) consisted of 99.99% purity SiO₂ for providing the background radioactivity level. More details of the standards composition and concentration are given in Table 1.

Saving time is very important and practical mainly on field surveys like those using the gamma rays analysis. Thus, the “optimum counting time” would correspond to the minimum counting time required for assuring the precision of the radiometric measurement within an acceptable analytical uncertainty. Such parameter may be obtained by evaluating the changes in the number of counts, counting time, or counting rate. In this paper, it was done taking into account 380 readings for the Th (160), K (120), and U (100) standards. Ten counting times were adopted for the standards and background readings: 30, 60, 90, 120, 180, 240, 300, 600, 900, and 1800 s.

The count rate sometimes is low during the gamma spectrometric readings. To determine whether the observed signal is “true” or “false,” the critical level of detection (*L_c*) (Currie 1968) may be adopted. It is expressed in Eq. 1, where *N_B* is the number of background counts and *t* is the counting time. If *t* is in min, then *L_c* is given in cpm (count

rate). Currie (1968) also defined the detection limit (*L_D*, also in cpm, if *t* is in min) according to Eq. 2, whose evaluation must be done when the net counts in a spectral region are lower than *L_c*. In this paper, a large database of background readings performed with standard K6 (pure SiO₂) for K, eU, and eTh windows allowed determine the mean *L_c* and *L_D* values.

$$L_c = \frac{N_B + (2.33 \times \sqrt{N_B})}{t} \tag{1}$$

$$L_D = \frac{N_B + (2.71 + 4.65 \times \sqrt{N_B})}{t} \tag{2}$$

Results and discussion

Calibration in energy

Table 2 reports the channels identified in the gamma spectra recorded in the MCA and the corresponding energy peaks. Such data allowed generate the energy calibration curve of the portable gamma spectrometer (Fig. 1), expressed by the following relationship:

$$E = 1.647Ch - 8.3605 \tag{3}$$

where *E* is the energy (in keV) and *Ch* is the channel number in the MCA. This curve is useful, among other purposes, for avoiding false photopeaks identification in the

Table 1 Description of the standards used for calibrating the portable γ-rays spectrometer in this study

Standard	Code	Composition	Weight (g)	Concentration
K1	LII-KCL-1	61.5 g KCl	61.50	52%
K2	LII-KCL-2	54.16 g SiO ₂ +28.9 g KCl	83.06	25%
K3	LII-KCL-3	80.12 g SiO ₂ +5.80 g KCl	85.92	5%
K4	LII-KCL-4	82.11 g SiO ₂ +2.90 g KCl	85.01	2.5%
K5	LII-KCL-5	84.72 g SiO ₂ +0.5 g KCl	85.22	0.5%
K6	LII-KCL-6	50 g SiO ₂	50	0%
U1	NBL-101-A	Pitchblende + SiO ₂	50	10,070 ppm
U2	NBL-102-A	Pitchblende + SiO ₂	50	1025 ppm
U3	NBL-103-A	Pitchblende + SiO ₂	50	499 ppm
U4	NBL-104-A	Pitchblende + SiO ₂	48.96	98.8 ppm
U5	NBL-105-A	Pitchblende + SiO ₂	50	10.2 ppm
Th1	NBL-106-A	Monazite sand + SiO ₂	50	10,290 ppm Th + 400 ppm U
Th2	NBL-107-A	Monazite sand + SiO ₂	50	1028 ppm Th + 40 ppm U
Th3	NBL-108-A	Monazite sand + SiO ₂	50	515 ppm Th + 20 ppm U
Th4	NBL-109-A	Monazite sand + SiO ₂	48.96	105.2 ppm Th + 4 ppm U
Th5	NBL-110-A	Monazite sand + SiO ₂	50	10.4 ppm Th + 0.4 ppm U
Th6	LII-Th-1	Stream sediments	50	5 ppm Th
Th7	LII-Th-2	Stream sediments	50	2.5 ppm Th
Th8	LII-Th-3	Stream sediments	50	1.25 ppm Th

Table 2 Calibration of the γ -rays spectrometric system

Radionuclide	Channel	Energy (MeV)	Half-life
^{133}Ba	225	0.36	10.51 years
^{214}Bi	1060	1.76	19.9 min
^{40}K	886	1.46	1.28×10^9 years
^{60}Co	717	1.17	5.271 years
^{60}Co	812	1.33	5.271 years
$^{60}\text{Co}^1$	1562	2.51	5.271 years
^{137}Cs	411	0.66	30.07 years
^{208}Tl	1571	2.61	183 s

Amplifier gains: coarse-1; fine-0.45. High voltage: 760–866 V. Energy and half-life values according to Chu et al. (1999)

¹Summation photopeak

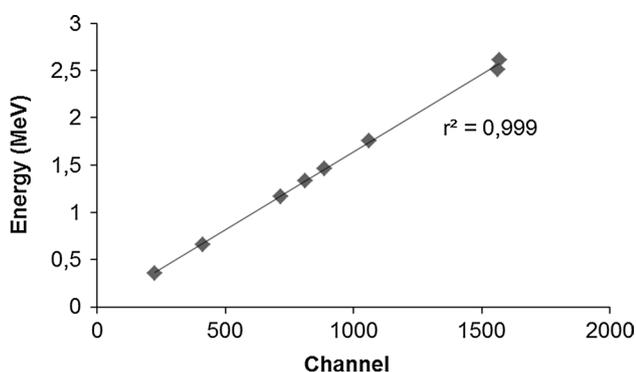


Fig. 1 Calibration of the portable gamma spectrometer for γ -rays energy readings

gamma spectra as sometimes verified in NaI(Tl) readings of the γ -rays associated with ^{40}K (1.46 MeV) and ^{214}Bi (1.76 MeV).

The detector resolution was 6.3% in the 0.66 MeV ^{137}Cs photopeak that is within the range of 6.0–7.5% as reported elsewhere (Adams and Gasparini 1970; Lønborg et al. 1976; Dickson and Scott 1997). However, the resolution improved according to the increased γ -rays energy, reaching 3.95% in the 2.61 MeV ^{208}Tl photopeak.

Table 3 shows the γ -rays energy windows adopted in this paper for radiometric monitoring of uranium (eU, equivalent uranium), thorium (eTh, equivalent thorium), and potassium

Table 3 Spectral windows adopted in this paper for evaluating uranium (eU), thorium (eTh), and potassium

Window	Energy range (MeV)	Radionuclide and monitored gamma emission (MeV)	Channel interval
Total counts	0.40–2.81	–	240–1594
Potassium (K)	1.37–1.57	^{40}K (1.46)	796–886
Uranium (eU)	1.66–1.86	^{214}Bi (1.76)	960–1070
Thorium (eTh)	2.41–2.81	^{208}Tl (2.62)	1464–1594

Energy range and monitored radionuclides according to IAEA (1991)

aimed for geological prospecting and environmental surveys. They include a significant fraction of the total absorption photopeak area, thus, avoiding the interference of other neighbor photopeaks (Lønborg et al. 1976). The total counts window comprised a large energy interval of the gamma rays emitted by the natural radionuclides.

Optimum counting time and detection limit

The acquired number of counts data in the windows reported in Table 3 by MAESTRO software was inserted in Excel spreadsheet and used for calculating the net counting rate (in cpm) after discounting the respective background values.

The average and variance values of the 10 readings were determined and used for application of the χ^2 statistical test in order to verify if the counting rates obey the Poisson distribution (Young 1962; Bevington 1969; Davis 1986; Vuolo 1996). The estimated χ^2 values were rejected when the probability was equal or lower than the pre-established limit (α significance level = 0.025) for belonging to the χ^2 distribution with $N - 1$ ($N = 10$) freedom degrees (FD).

The χ^2 reference values varied between $2.70 > \chi^2 > 19.02$ ($t \geq 30$ s, FD = 9) and $0.00 > \chi^2 > 5.02$ ($t \geq 900$ s, FD = 1). The calculated χ^2 values within the range of the expected ones indicated data obeying the Poisson distribution. Table 4 reports the optimum counting times found for the standards used in this paper.

Some unsatisfactory χ^2 values were possibly related to background fluctuations in the laboratory as a consequence of γ -rays emissions associated with the indoor ^{222}Rn activity concentration. Additional equivalent field tests should be also realized as the described experiments were held under controlled lab conditions, not considering the larger surface area and volume of the exposed rocks, which certainly would imply on enhanced radioactivity levels due to the γ -rays emissions.

The mean L_c and L_D values are reported in Table 5. Their plots for K, eU, and eTh against the counting time in Fig. 2 indicate systematic decreasing trends according to the following equations:

$$L_c \text{ and } L_D(\text{K}) = 56.071 \times t^{-0.075} (r^2 = 0.847) \tag{4}$$

$$L_c \text{ and } L_D(\text{eU}) = 24.471 \times t^{-0.109} (r^2 = 0.888) \tag{5}$$

Table 4 Minimum counting time (optimum counting time) in the standards used in this paper required for assuring the precision of the γ -rays readings within the significance level of 0.025 an 9 freedom degrees

Standard	Weight (g)		Optimum counting time (s)		
			K-40	Bi-214	Tl-208
	KCl	⁴⁰ K			
K6	0	0	–	–	–
K5	0.26	1.02×10^{-4}	–	–	–
K4	1.51	3.06×10^{-4}	≥ 600	–	–
K3	3.02	5.16×10^{-4}	≥ 300	–	–
K2	15.03	2.49×10^{-3}	≥ 90	–	–
K1	31.98	3.84×10^{-3}	≥ 30	–	–
	Pitchblende	²³⁸ U			
U5	50	5.06×10^{-4}	–	–	–
U4	50	4.90×10^{-3}	≥ 300	≥ 600	–
U3	50	2.48×10^{-2}	≥ 30	≥ 600	≥ 600
U2	50	5.09×10^{-2}	≥ 30	≥ 60	≥ 600
U1	50	5.00×10^{-1}	≥ 30	≥ 60	≥ 60
	Monazite sand	²³² Th			
Th8	50	6.25×10^{-5}	–	–	–
Th7	50	1.25×10^{-4}	–	–	–
Th6	50	2.50×10^{-4}	–	–	–
	Monazite sand	²³² Th– ²³⁸ U			
Th5	50	5.20×10^{-4} – 1.99×10^{-5}	–	–	–
Th4	50	5.26×10^{-3} – 1.99×10^{-4}	–	≥ 240	–
Th3	50	2.58×10^{-2} – 9.93×10^{-4}	≥ 240	≥ 120	≥ 30
Th2	50	5.14×10^{-2} – 1.99×10^{-3}	≥ 90	≥ 30	≥ 30
Th1	50	5.15×10^{-1} – 1.99×10^{-2}	≥ 30	≥ 30	≥ 30

Table 5 Mean L_c and L_D values for K, eU, and eTh windows determined from Eqs. 1 and 2 according to different counting times

t (min)	L_c (cpm)			L_D (cpm)		
	K	eU	eTh	K	eU	eTh
30	45.05	17.97	9.87	47.84	19.74	11.18
15	46.18	18.15	10.24	50.18	20.66	12.13
10	47.49	19.11	10.41	52.47	22.27	12.76
5	48.58	19.92	11.50	55.71	24.50	15.01
4	50.35	20.57	12.73	58.46	25.78	16.85
3	51.69	21.29	12.72	61.20	27.43	17.51
2	50.85	21.16	13.41	62.41	28.69	19.46
1.5	51.24	23.78	14.17	64.68	33.04	21.40
1	56.04	23.64	15.63	73.20	34.94	24.98
0.5	64.18	29.12	17.70	90.30	47.17	32.15

$$L_c \text{ and } L_D(\text{eTh}) = 15.284 \times t^{-0.148} (r^2 = 0.961) \quad (6)$$

Calibration in concentration

Grasty and Darnley (1971) suggested the stripping correction technique for converting the counting rates obtained in the K, eU, and eTh windows into concentrations, without

interference among them. The stripping factors α , β , and γ take into account the Compton effects in the γ -rays spectra, considering the counting rate in a lower energy window relatively to the higher one: $\alpha = \alpha_{U,Th}/\alpha_{Th,Th}$; $\beta = \alpha_{K,Th}/\alpha_{Th,Th}$; $\gamma = \alpha_{K,U}/\alpha_{U,U}$. Contrarily, the reverse stripping factors a , b , and g take into account the counting rate in a higher energy window relatively to the lower one: $a = \alpha_{Th,U}/\alpha_{U,U}$; $b = \alpha_{Th,K}/\alpha_{K,K}$; $g = \alpha_{U,K}/\alpha_{K,K}$. In addition to Compton effects, sensitivity (S) constants must be also taken into account, which are dependent of the detector size/efficiency/

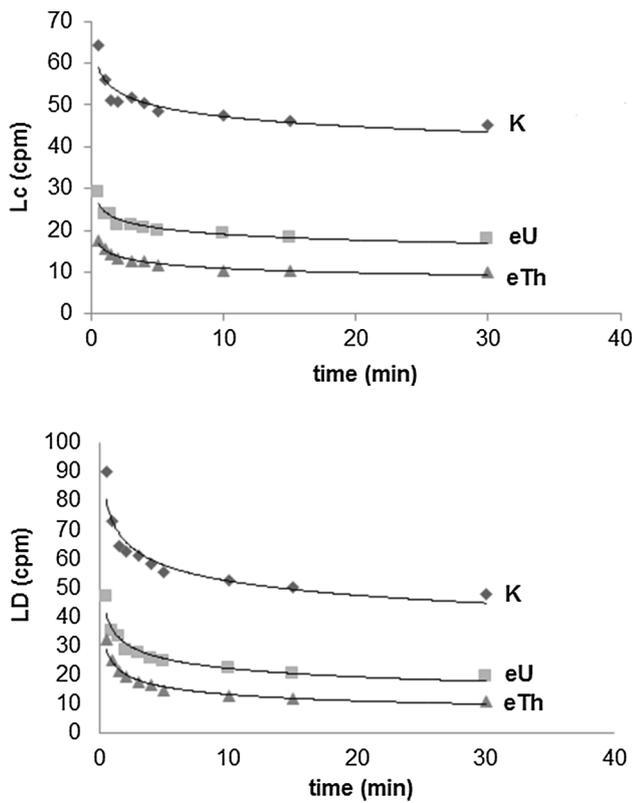


Fig. 2 L_c and L_D values for K, eU, and eTh plotted against the counting time (in min)

geometry and discrimination limit. They can be expressed in each window by $S_K = \alpha_{K,K}$, $S_U = \alpha_{U,U}$, and $S_{Th} = \alpha_{Th,Th}$, allowing determine the concentrations of potassium (C_K), uranium (C_U), and thorium (C_{Th}) by $C_K = (nK)_{net}/S_K$, $C_U = (nU)_{net}/S_U$, and $C_{Th} = (nTh)_{net}/S_{Th}$, where $(nK)_{net}$, $(nU)_{net}$, and $(nTh)_{net}$ correspond to the net count rates in the K, eU, and eTh windows, respectively. Equations 7–9 express the $(nK)_{net}$, $(nU)_{net}$, and $(nTh)_{net}$ values taking into account the interference factors:

$$(nTh)_{net} = (nTh) - a(nU)_{net} - b(nK)_{net} \tag{7}$$

$$(nU)_{net} = (nU) - \alpha(nTh)_{net} - g(nK)_{net} \tag{8}$$

$$(nK)_{net} = (nK) - \gamma(nU)_{net} - \beta(nTh)_{net} \tag{9}$$

The matricial Eqs. 10 and 11 allow determine the K, U, and Th concentrations from the net count rates in the K, eU, and eTh windows.

$$\begin{bmatrix} C_K \\ C_U \\ C_{Th} \end{bmatrix} = \begin{bmatrix} \beta_{K,K} & \beta_{K,U} & \beta_{K,Th} \\ \beta_{U,K} & \beta_{U,U} & \beta_{U,Th} \\ \beta_{Th,K} & \beta_{Th,U} & \beta_{Th,Th} \end{bmatrix} \begin{bmatrix} (nK)_{net} \\ (nU)_{net} \\ (nTh)_{net} \end{bmatrix} \tag{10}$$

$$\begin{bmatrix} \beta_{K,K} & \beta_{K,U} & \beta_{K,Th} \\ \beta_{U,K} & \beta_{U,U} & \beta_{U,Th} \\ \beta_{Th,K} & \beta_{Th,U} & \beta_{Th,Th} \end{bmatrix} = \begin{bmatrix} \alpha_{K,K} & \alpha_{K,U} & \alpha_{K,Th} \\ \alpha_{U,K} & \alpha_{U,U} & \alpha_{U,Th} \\ \alpha_{Th,K} & \alpha_{Th,U} & \alpha_{Th,Th} \end{bmatrix}^{-1} \tag{11}$$

In general, $\alpha_{Th,K}$, $\alpha_{U,K}$, $\beta_{U,K}$, and $\beta_{Th,K}$ are null and other stripping factors are sometimes low, like a . In this paper, the large number of readings (380) for the standards permitted uses the net count rates for estimating the following stripping factors: $\alpha = 0.8848 \pm 0.0273$, $\beta = 0.8558 \pm 0.0291$, $\gamma = 1.1314 \pm 0.0009$, $a = 0.0338 \pm 0.0084$, $S_K = 1.5320 \pm 0.1146$, $S_U = 0.1021 \pm 0.0006$, and $S_{Th} = 0.0261 \pm 0.0001$.

The same database, stripping factors, and Eqs. 7–11 allowed generate Eqs. 12–14 for estimating concentration values from the count rate readings, which are useful in geochemical prospecting of mineral deposits (for instance, eU values above 100 ppm may indicate areas enriched in uranium, perhaps of economic interest).

$$C_K = (0.6527 \pm 9.65 \times 10^{-16}) (nK)_{net} - (0.7418 \pm 1.10 \times 10^{-15}) (nU)_{net} + (0.0977 \pm 3.68 \times 10^{-16}) (nTh)_{net} \tag{12}$$

$$C_U = (1.0092 \pm 2.28 \times 10^{-16}) (nU)_{net} - (8.9293 \pm 8.92 \times 10^{-16}) (nTh)_{net} \tag{13}$$

$$C_{Th} = -(1.3329 \pm 5.07 \times 10^{-16}) (nU)_{net} + (39.4867 \pm 1.99 \times 10^{-15}) (nTh)_{net} \tag{14}$$

Table 6 shows the counting rates in the K, eU, and eTh windows for each standard, as well the estimated concentration values from the developed equations with their respective statistical residues (unexplicable variation).

Exposure rate and dose rate calibrations

The radioelements concentration allows evaluating their activity concentration from the corresponding specific activities, i.e., 2.589×10^5 Bq/g for ^{40}K , 1.246×10^4 Bq/g for ^{238}U , and 4.066×10^3 Bq/g for ^{232}Th (IAEA 2008). The air exposure rate (X_{air}) associated with the gamma activity from a radioactive source may be estimated by $X_{air} = \Gamma A^2/d^2$, where A is the activity (in Ci), d is the source-measuring point distance (in m), and Γ is the exposure rate constant (in $\text{R m}^2/\text{h Ci}$) (Tauhata et al. 2013). Typical Γ values are 0.779, 7.48, and 15.2 $\text{R cm}^2/\text{h mCi}$ for ^{40}K , ^{214}Bi (^{238}U), and ^{208}Tl (^{232}Th), respectively (Tauhata et al. 2013). The parameter

Table 6 Net count rate for each standard in the K, eU, and eTh windows and respective estimated values of the K, U, and Th concentration

Standard	$(nK)_{\text{net}}$ (cpm)	K (%)	K (%) estimated	Residue
K6	0.9371 ± 1.5741	0	1.2922 ± 0.3456	1.2922
K5	0.3672 ± 1.6545	0.5	0.8352 ± 0.2906	- 0.1648
K4	1.1816 ± 2.0438	2.5	0.7297 ± 0.5576	- 2.2703
K3	7.7089 ± 2.1867	5	8.2579 ± 0.7446	3.2579
K2	36.7834 ± 2.0023	25	35.5733 ± 0.7061	10.5733
K1	80.6069 ± 2.6660	52	81.2906 ± 1.0191	29.2906
U5	- 0.0727 ± 1.7374	0	- 2.0032 ± 0.3639	- 2.0032
U4	14.1559 ± 1.8138	0	3.9979 ± 0.3004	3.9979
U3	73.3601 ± 2.5186	0	3.5824 ± 0.5313	3.5824
U2	138.9254 ± 3.2160	0	- 6.0207 ± 0.0670	- 6.0207
U1	1160.7233 ± 8.0024	0	5.5029 ± - 0.1362	5.5029
Th8	0.1240 ± 2.1192	0	- 1.8190 ± 0.5027	- 1.8190
Th7	0.4156 ± 1.8945	0	- 0.8935 ± 0.4487	- 0.8935
Th6	2.5310 ± 1.8473	0	3.3367 ± 0.4781	3.3367
Th5	- 0.1059 ± 1.7812	0	- 1.5934 ± 0.4045	- 1.5934
Th4	2.7312 ± 2.0482	0	- 0.9387 ± 0.4672	- 0.9387
Th3	13.6939 ± 2.0148	0	- 0.5541 ± 0.2473	- 0.5541
Th2	27.2897 ± 1.9172	0	- 1.1007 ± 0.1189	- 1.1007
Th1	276.1431 ± 4.1547	0	0.3504 ± 0.4781	0.3504
Standard	$(nU)_{\text{net}}$ (cpm)	U (ppm)	eU (ppm) estimated	Residue
K6	- 0.2517 ± 1.0290	0	- 6.8653 ± - 6.8653	- 6.8653
K5	- 0.2294 ± 1.1756	0	- 15.1309 ± - 15.1309	- 15.1309
K4	0.5071 ± 1.1409	0	- 2.3735 ± - 2.3735	- 2.3735
K3	- 0.5193 ± 1.0488	0	- 2.8738 ± - 2.8738	- 2.8738
K2	0.8943 ± 0.9184	0	21.2216 ± 21.2216	21.2216
K1	- 0.4444 ± 1.0885	0	- 15.6184 ± - 15.6184	- 15.6184
U5	1.6967 ± 1.1279	10.2	18.3911 ± 18.3911	8.1911
U4	8.9195 ± 1.3227	98.8	94.1238 ± 94.1238	- 4.6762
U3	61.7096 ± 1.6204	499	620.4086 ± 620.4086	121.4086
U2	128.5043 ± 2.8712	1025	1269.8436 ± 1269.8436	244.8436
U1	1025.5498 ± 7.4522	10070	10039.0994 ± 10039.0994	- 30.9006
Th8	1.7953 ± 1.2765	0	12.6991 ± 12.6991	12.6991
Th7	1.2506 ± 1.1714	0	6.1158 ± 6.1158	6.1158
Th6	- 0.6631 ± 1.0804	0	- 10.1026 ± - 10.1026	- 10.1026
Th5	1.3627 ± 1.1374	0.4	10.4180 ± 10.4180	10.0180
Th4	3.7431 ± 1.2686	4	3.0212 ± 3.0212	- 0.9788
Th3	14.3912 ± 1.5825	20	20.1411 ± 20.1411	0.1411
Th2	28.9712 ± 1.6965	40	22.3727 ± 22.3727	- 17.6273
Th1	278.3644 ± 3.4990	400	401.7557 ± 401.7557	1.7557
Standard	$(nTh)_{\text{net}}$ (cpm)	Th (ppm)	eTh (ppm) estimated	Residue
K6	0.4844 ± 0.8331	0	19.4623 ± 31.5241	19.4623
K5	1.4352 ± 0.8470	0	56.9774 ± 31.8772	56.9774
K4	0.8389 ± 0.7149	0	32.4501 ± 26.7065	32.4501
K3	- 0.2651 ± 0.9756	0	- 9.7754 ± 37.1270	- 9.7754
K2	- 1.3658 ± 0.8223	0	- 55.1227 ± 31.2475	- 55.1227
K1	1.2469 ± 0.8831	0	49.8270 ± 33.4214	49.8270
U5	- 0.1532 ± 0.6812	0	- 8.2988 ± 25.3948	- 8.2988
U4	- 0.4597 ± 0.9993	0	- 30.0387 ± 37.6970	- 30.0387

Table 6 (continued)

Standard	(<i>nTh</i>) _{net} (cpm)	Th (ppm)	eTh (ppm) estimated	Residue
U3	0.2677 ± 0.9145	0	- 71.6812 ± 33.9496	- 71.6812
U2	3.0320 ± 0.9998	0	- 51.5535 ± 35.6501	- 51.5535
U1	34.8476 ± 1.7231	0	9.1031 ± 58.1079	9.1031
Th8	0.6069 ± 0.6786	1.25	21.5723 ± 25.0927	20.3223
Th7	0.7286 ± 0.8290	2.5	27.1018 ± 31.1729	24.6018
Th6	0.3819 ± 0.7547	5	15.9656 ± 28.3611	10.9611
Th5	0.3735 ± 0.8756	10.4	12.9304 ± 33.0579	2.5304
Th4	3.8924 ± 0.7297	105.2	148.7072 ± 27.1233	43.5072
Th3	14.0102 ± 1.0857	515	534.0347 ± 40.7620	19.0347
Th2	30.2429 ± 1.2891	1028	1155.5733 ± 48.6397	127.5733
Th1	269.6317 ± 3.7020	10290	10275.8413 ± 141.5149	- 14.1587

X_{air} allows calculating the air absorbed dose (D_{air}) according to the relationship $D_{\text{air}} = 0.869 \times X_{\text{air}}$, where the constant 0.869 corresponds to the mean energy necessary for producing an ions pair in air per electron charge (Tauhata et al. 2013). In this paper, X_{air} and D_{air} have been estimated from the net count rates in the K, eU, and eTh windows according to the equations:

$$X_{\text{air}} = (0.0248 \pm 0.00722)(nK)_{\text{net}} + (0.0738 \pm 0.00823)(nU)_{\text{net}} + (0.0097 \pm 0.00277)(nTh)_{\text{net}} \quad (15)$$

$$D_{\text{air}} = (0.2395 \pm 0.0697)(nK)_{\text{net}} + (0.7120 \pm 0.0795)(nU)_{\text{net}} + (0.0925 \pm 0.0268)(nTh)_{\text{net}} \quad (16)$$

The air exposure rate may be used to provide the exposure rate in another medium, for instance, water or tissue by adopting an appropriate conversion factor (f). For water, some typical f values vary from 0.881 rad/R (20 keV) to 0.964 rad/R (150 keV) (Tauhata et al. 2013). In this paper, X_{tissue} has been calculated from the net count rate in the K, eU, and eTh windows, according to the equation:

$$X_{\text{tissue}} = (0.0276 \pm 0.00802)(nK)_{\text{net}} + (0.0819 \pm 0.00914)(nU)_{\text{net}} + (0.0106 \pm 0.00308)(nTh)_{\text{net}} \quad (17)$$

Finally, the effective dose (Ed, in mSv/year) may be estimated from the net count rate in the K, eU, and eTh windows by $Ed = \Gamma_K (nK)_{\text{net}} + \Gamma_U (nU)_{\text{net}} + \Gamma_{Th} (nTh)_{\text{net}}$ (Tauhata et al. 2013), whose empirical equation adopted in this paper is:

$$Ed = (0.0016 \pm 0.000452)(nK)_{\text{net}} + (0.0046 \pm 0.000515)(nU)_{\text{net}} + (0.0006 \pm 0.000174)(nTh)_{\text{net}} \quad (18)$$

Table 7 reports the values calculated for X_{air} , D_{air} , X_{tissue} , and Ed from the counting rates given in Table 6, illustrating the applicability of the database for environmental approaches.

Effective dose in environments for Geology teaching

The described procedures allowed perform γ -rays measurements in different environments related to activities developed for undergraduate and graduate Geology teaching at UNESP-São Paulo State University, IGCE-Geosciences and Exacts Sciences Institute, Rio Claro, São Paulo State, Brazil. The investigated sites were: LIT1-rocks deposit (*litoteca*) from DPM-Petrology and Metallogeny Department; MUS-“Heinz Ebert” Minerals and Rocks Museum; TEA-office of one professor responsible by the Mineralogy discipline located at DPM; LIT2-*litoteca* from UNESPetro (Geosciences Center Applied to Petroleum); CTD-courtyard of UNESPetro.

The net count rates (in cpm) in K, eU, and eTh windows obtained in those sites are reported in Table 8, which were integrated and converted to effective dose (Ed, mSv/year) on using Eq. 18. Then, the values were modeled in Surfer software (Golden Software) by the kriging method through the spherical semivariogram model. Figures 3, 4, 5, 6, and 7 show the isolines geostatistical models generated for the Ed values calculated in the selected points. Some crystalline rocks (possibly enriched in natural radioelements) are stored in points P3, P4, and P5 at TEA site, justifying the enhanced Ed levels shown in Fig. 3. The elevated Ed value in P1 measuring point at LIT1 site (Fig. 4) is due to a highly radioactive rock sample stored in a rectangular lead shield in order to minimize the γ -rays emissions. The inferred mineralogical composition of this sample is thorianite (ThO_2), thorite (ThSiO_2), uranothorite [(Th,U)SiO₄], or uranothorianite [(Th,U)O₂] (Bonotto 1996). This is reinforced by the high counting rate in the eTh window (^{208}Tl),

Table 7 Estimated values of the air exposure rate (X_{air}), air absorbed dose (D_{air}), tissue exposure rate (X_{tissue}), and effective dose (Ed) from the net count rate in the K, eU, and eTh windows

Standard	X_{air} ($\mu\text{R/h}$)	D_{air} (nSv/h)	X_{tissue} ($\mu\text{R/h}$)	Ed (mSv/year)
K6	0.0950 ± 0.2635	0.9157 ± 2.5402	0.1054 ± 0.2923	0.0080 ± 0.0223
K5	0.2600 ± 0.2781	2.5046 ± 2.6812	0.2882 ± 0.3085	0.0219 ± 0.0235
K4	0.2192 ± 0.2603	2.1129 ± 2.5105	0.2431 ± 0.2889	0.0185 ± 0.0220
K3	0.1089 ± 0.3394	0.0980 ± 0.3059	0.1089 ± 0.3394	0.0083 ± 0.0258
K2	0.6817 ± 0.2640	6.5830 ± 2.5454	0.7575 ± 0.2929	0.0577 ± 0.0223
K1	2.1169 ± 0.3028	20.4259 ± 2.9200	2.3505 ± 0.3360	0.1789 ± 0.0256
U5	0.0862 ± 0.2459	0.8323 ± 2.3712	0.0958 ± 0.2729	0.0073 ± 0.0208
U4	0.8666 ± 0.3204	8.3645 ± 3.0889	0.9625 ± 0.3555	0.0733 ± 0.0271
U3	6.0528 ± 0.3419	58.4149 ± 3.2967	6.7221 ± 0.3794	0.5117 ± 0.0289
U2	12.7360 ± 0.4607	122.9061 ± 4.4428	14.1434 ± 0.5113	1.0767 ± 0.0389
U1	104.8596 ± 1.0252	1011.8939 ± 9.8893	116.4435 ± 1.1380	8.8642 ± 0.0866
Th8	0.2398 ± 0.2647	2.3126 ± 2.5530	0.2661 ± 0.2938	0.0203 ± 0.0224
Th7	0.2318 ± 0.2801	2.2347 ± 2.7012	0.2572 ± 0.3108	0.0196 ± 0.0237
Th6	0.0855 ± 0.2589	0.8235 ± 2.4966	0.0948 ± 0.2873	0.0072 ± 0.0219
Th5	0.1610 ± 0.2838	1.5531 ± 2.7361	0.1787 ± 0.3149	0.0136 ± 0.0240
Th4	1.0476 ± 0.2720	10.0987 ± 2.6232	1.1621 ± 0.3019	0.0885 ± 0.0230
Th3	3.9258 ± 0.3591	37.8471 ± 3.4629	4.3552 ± 0.3985	0.3315 ± 0.0303
Th2	8.2762 ± 0.4025	79.7842 ± 3.8811	9.1812 ± 0.4466	0.6989 ± 0.0340
Th1	75.9504 ± 1.0291	732.2042 ± 9.9212	84.2582 ± 1.1417	6.4141 ± 0.0869

suggesting significant presence of gamma emissions from ^{232}Th -descendants. The enhanced Ed value in P7 measuring point at MUS site (Fig. 5) is linked to the presence of an enriched U-ore. Some carbonate rocks slightly enriched in uranium and thorium in point P2 at LIT2 site justify the higher Ed values shown in Fig. 6, whereas the more elevated Ed levels in P1 at CTD site (Fig. 7) are possibly related to its proximity to the shelf for storing the radioactive standards used in our experiments.

All Ed values are below the guideline reference value of 20 mSv/year as recommended by International Commission on Radiological Protection (ICRP 2005) for the average of 5 years directed to occupationally exposed individuals. However, two values are higher than the guidance level of 1 mSv/year for members of the public as proposed by ICRP (2005) and adopted by CNEN-National Commission on Nuclear Energy, Brazil (CNEN 2006). They were found in monitoring points located at sites LIT1 and MUS that are only occasionally utilized by professors and undergraduate and graduate students, thus, not implying on significant health risk for people. LIT1 is primarily used to store and catch samples of rocks, ores, and minerals, whereas MUS is only sometimes accessed for short duration visitation. In LIT1 site, the highest Ed value was found at P1 monitoring point after positioning the NaI(Tl) detector frontally over the lead box lid. In other monitoring points, the γ -rays readings were realized in air at 0.75–1.8 m above ground surface. In MUS site, the high Ed value was measured at P7 monitoring point.

Another independent radiological monitoring was done in P1 at LIT1 site (Fig. 4) using a Ludlum Model 193-6 Survey

Wand employing a 6" \times 1" EJ212 Plastic Gamma Scintillator, whose calibration certificate 12-1510 was issued by METROBRAS (Center for Trials and Researches in Metrology) in June 21, 2012 (Borges et al. 2012). This instrument recorded the following air exposure rate data (X_{air} , in $\mu\text{R/h}$): (a) in the sides and top (closed box)—from 30 to 40 $\mu\text{R/h}$; (b) in the top, close to the edge (open box)—630 $\mu\text{R/h}$. The highest Ed value recorded in this paper at P1 corresponded to ~ 6 mSv/year (Fig. 4) that is equivalent to 68 $\mu\text{R/h}$. It is a value compatible with those reported by Borges et al. (2012) as the detectors efficiencies are different in both cases (the NaI(Tl) detector used in this paper exhibits better detection efficiency than the plastic gamma scintillator as reported by Saint-Gobain 2016).

Finally, an additional merit of the analytical protocol described in this paper consists on the possibility of converting activity concentration values in doses, which is, in general, a very difficult task due to the several parameters involved (see, for instance, the comments of PSE-Physics Stack Exchange; PSE 2017).

Conclusions

The χ^2 test for the counting rates data of the γ -rays NaI(Tl) scintillation detector for different times (1800, 900, 600, 300, 240, 180, 120, 90, 60, and 30 s) permitted to determine the optimum counting time for gamma rays readings of the standards used. The ideal counting time for in situ γ -rays determination should be done considering the contribution of the

Table 8 Net count rate in the K, eU, and eTh windows for γ -rays readings realized in air at 0.75–1.8 m above ground surface at different environments from IGCE-UNESP-Rio Claro campus

ID	X (m)	Y (m)	t (min)	K (cpm)	eU (cpm)	eTh (cpm)
MUS						
P1	0.25	3.07	30.29	136.96	28.92	23.05
P2	0.25	6.69	30.43	144.54	34.57	26.12
P3	2.84	7.02	30.90	133.06	34.08	28.61
P4	6.87	7.02	30.58	198.33	85.38	29.79
P5	10.85	7.02	30.46	176.58	60.99	69.50
P6	10.85	6.07	48.98	169.25	54.29	42.91
P7	6.87	6.07	36.82	258.58	150.68	38.57
P8	2.84	6.07	30.41	135.57	33.83	28.44
P9	2.84	3.72	54.94	137.19	28.21	22.92
P10	6.87	3.72	30.53	133.70	35.54	22.60
P11	10.85	3.72	30.68	130.55	30.71	25.49
P12	10.85	2.76	30.52	126.01	29.91	23.10
P13	6.87	2.76	30.76	124.99	27.70	21.88
P14	2.84	2.76	30.30	127.48	30.37	22.38
P15	2.01	6.55	30.75	136.66	32.52	24.94
P16	3.69	6.55	30.50	144.32	35.81	28.66
P17	6.05	6.55	30.61	198.77	88.01	27.67
P18	7.72	6.55	30.69	163.06	60.48	28.35
P19	10.01	6.55	30.37	183.27	67.61	71.10
P20	12.86	8.13	30.84	166.47	41.73	38.85
P21	11.70	3.25	69.56	182.35	36.79	39.74
P22	10.01	3.25	30.51	128.55	30.81	22.03
P23	7.72	3.25	30.60	130.07	29.58	22.71
P24	6.05	3.25	30.19	130.96	29.81	19.18
P25	3.69	3.25	30.31	135.05	31.02	22.77
P26	2.84	0.42	30.17	121.10	27.61	20.75
P27	2.84	9.43	30.41	115.60	31.67	22.86
P28	10.85	9.43	29.95	127.05	33.39	27.75
P29	10.47	0.42	31.02	116.23	28.14	24.75
P30	6.87	0.42	33.99	118.55	27.06	21.74
P31	6.87	9.43	30.68	129.29	43.35	24.12
CTD						
P1	38.99	1.72	28.37	93.90	23.97	17.48
P2	4.82	4.22	30.15	61.20	16.48	12.01
P3	10.42	19.56	30.44	58.14	18.03	15.01
P4	38.99	18.76	30.30	65.41	20.69	14.95
TEA						
P1	0.32	1.47	36.23	120.41	27.65	21.33
P2	3.08	3.79	30.10	114.86	27.18	21.43
P3	1.02	4.46	30.23	145.30	39.03	24.87
P4	2.47	0.93	30.99	124.55	36.53	28.98
P5	1.65	4.46	30.96	137.55	30.23	24.09
LIT2						
P1	12.49	1.67	30.44	97.81	22.64	16.39
P2	6.52	4.80	30.38	99.29	27.16	19.85
P3	23.56	2.18	30.60	66.86	19.90	11.60
P4	22.27	4.56	30.47	77.27	20.84	13.13
P5	25.89	3.71	31.12	80.01	20.15	14.91
LIT1						
P1	9.19	3.05	30.16	1358.58	905.99	850.07

Table 8 (continued)

ID	X (m)	Y (m)	t (min)	K (cpm)	eU (cpm)	eTh (cpm)
P2	16.23	3.05	30.66	134.33	27.07	19.96
P3	16.23	1.08	30.64	133.44	32.35	21.81
P4	16.23	4.96	30.74	114.04	24.33	20.88
P5	- 3.07	5.57	69.69	112.92	27.08	18.84
P6	2.05	3.05	30.74	164.60	37.64	33.67
P7	2.05	1.08	30.44	135.61	46.88	35.71
P8	2.05	4.96	31.36	125.35	29.85	23.82

“X” and “Y” correspond, respectively, to horizontal and vertical distances in the sketch diagrams shown in Figs. 3, 4, 5, 6, and 7 for the surveyed areas

Fig. 3 Isolines map of the effective dose at TEA site

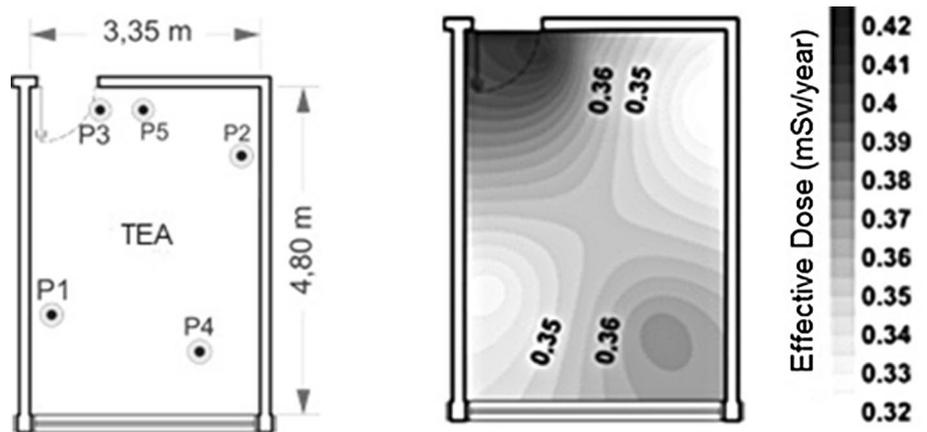
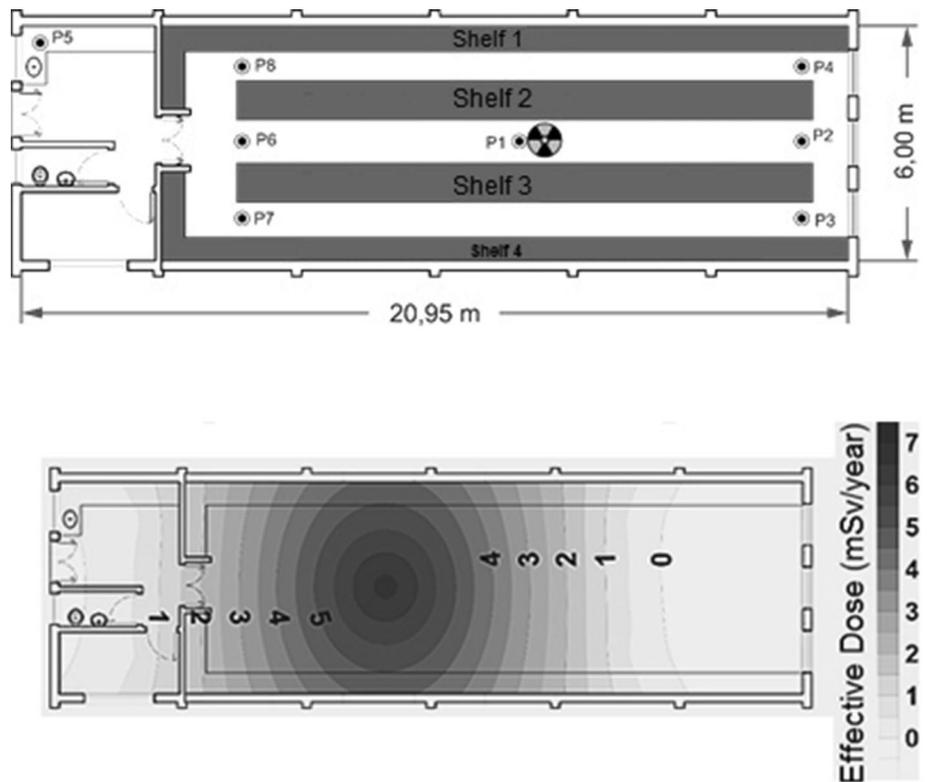


Fig. 4 Isolines map of the effective dose at LIT1 site



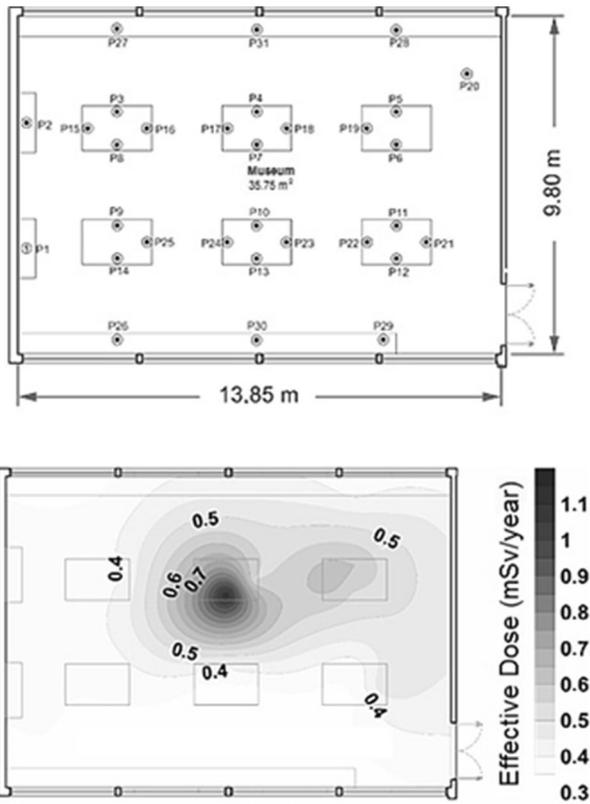


Fig. 5 Isolines map of the effective dose at MUS site

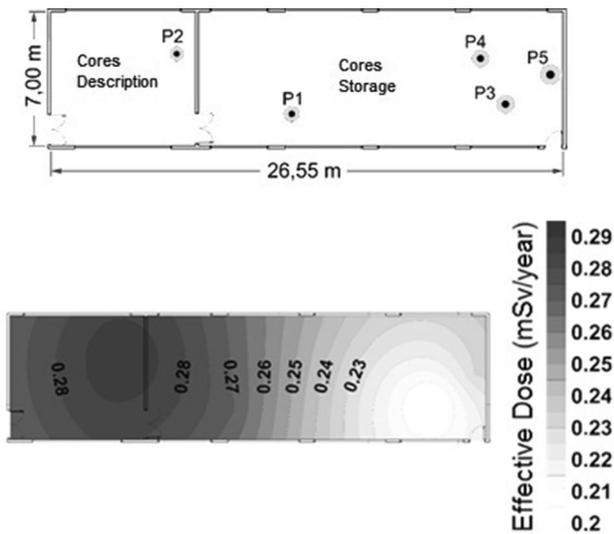


Fig. 6 Isolines map of the effective dose at LIT2 site

source (soil or rock) volume, mean density, and average radionuclide concentration in the materials. K1, K2, U1, U2, U3, Th1, Th2, and Th3 standards used during the calibration steps provided values above the detection limit for each counting

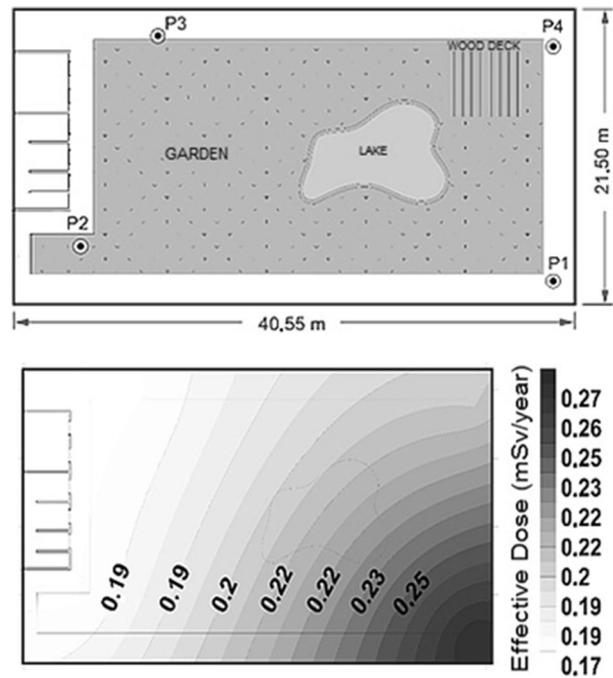


Fig. 7 Isolines map of the effective dose at CTD site

time adopted. In other cases, the count rates didn't exceed the background values, within the uncertainty levels adopted in this paper. Possible reasons could be the lack of a lead shield wall at the detector's frontal window and/or detector directed to ~ 1 m distant ground surface from its positioning over a bench. Despite these limitations, the empirical calibration equations yielded satisfactory results as described along this paper. Geostatistical modeling allowed the data integration for calculating the effective dose (Ed) associated with sites related to activities developed for Geology teaching. Although some levels were above the guideline reference value of 1 mSv/year, none intervention is required as the sites are primarily used to store/display geological materials and only occasionally accessible for members of the public during short duration visits. The highest calculated Ed levels were much lower than the limiting value of 20 mSv/year for occupationally exposed individuals.

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